

Title: PSI 2016/2017 Condensed Matter - Lecture 1

Date: Nov 07, 2016 10:45 AM

URL: <http://pirsa.org/16110052>

Abstract:

Perimeter: Condensed Matter Fall 2016



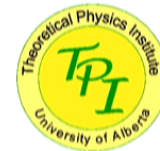
# ...Metals, Insulators, Magnets, and Superconductors...

F. Marsiglio

[fm3@ualberta.ca](mailto:fm3@ualberta.ca)



UNIVERSITY OF  
ALBERTA



# PACS (Physics and Astronomy Classification Scheme)

**00**—General

**10**—The Physics of Elementary Particles and Fields

**20**—Nuclear Physics

**30**—Atomic and Molecular Physics

**40**—Electromagnetism, Optics, Acoustics, Heat Transfer, Classical Mechanics, and Fluid Dynamics

**50**—Physics of Gases, Plasmas, and Electric Discharges

**60**—Condensed Matter: Structural, Mechanical and Thermal Properties

**70**—Condensed Matter: Electronic Structure, Electrical, Magnetic, and Optical Properties

**80**—Interdisciplinary Physics and Related Areas of Science and Technology

**90**—Geophysics, Astronomy, and Astrophysics

**60—Condensed Matter: Structural, Mechanical and Thermal Properties**

- 61. Structure of solids and liquids; crystallography
- 62. Mechanical and acoustical properties of condensed matter
- 63. Lattice dynamics
- 64. Equations of state, phase equilibria, and phase transitions
- 65. Thermal properties of condensed matter
- 66. Nonelectronic transport properties of condensed matter
- 67. Quantum fluids and solids
- 68. Surfaces and interfaces; thin films and nanosystems (structure and nonelectronic properties)

**70—Condensed Matter: Electronic Structure, Electrical, Magnetic, and Optical Properties**

- 71. Electronic structure of bulk materials
- 72. Electronic transport in condensed matter
- 73. Electronic structure and electrical properties of surfaces, interfaces, thin films, and low-dimensional structures
- 74. Superconductivity
- 75. Magnetic properties and materials
- 76. Magnetic resonances and relaxations in condensed matter, Mössbauer effect
- 77. Dielectrics, piezoelectrics, and ferroelectrics and their properties
- 78. Optical properties, condensed-matter spectroscopy and other interactions of radiation and particles with condensed matter
- 79. Electron and ion emission by liquids and solids; impact phenomena

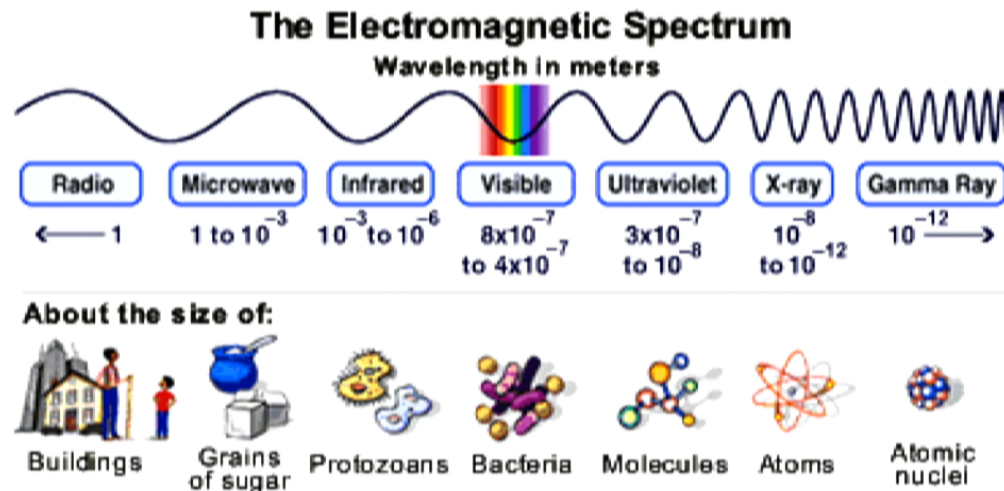


Perimeter Condensed Matter Course, Frank Marsiglio,  
November, 2016

The “plan”

- Lecture 1: A survey of Condensed Matter Physics: Introduction + some special topics
  - Lecture 2: Bravais lattices, Reciprocal Lattice, Brillouin Zones, Bloch states, Kronig-Penney model
  - Lecture 3: Numerical Band Structure Calculations, Weak Potentials
  - Lecture 4: Tight-binding models, Density of States, Density Functional Theory
  - Lecture 5: Ions, Ionic vibrations, phonons
  - Lecture 6 : More phonons, Einstein and Debye models, realistic calculations
  - Lecture 7: Detecting phonons: neutron scattering
  - Lecture 8: Interacting electrons: Hartree Fock + Screening; Optical Conductivity
  - Lecture 9: Case study: the polaron
  - Lecture 10: Magnetism: exchange, paramagnetic insulators, Heisenberg
  - Lecture 11: Spin waves, Holstein-Primakov transformation
  - Lecture 12: Hubbard Model; Hubbard to t-J model, other electron correlations
  - Lecture 13: Superconductivity: phenomenology + the electron-phonon interaction
  - Lecture 14: BCS microscopic theory
  - Lecture 15: Observables/Eliashberg Theory
- or
- Lecture 15: Inhomogeneous superconductivity: Anderson vs. Bogoliubov-de Gennes

# Units



Units:      $1 \text{ meV} = 0.24 \text{ THz}$   
                   $8.066 \text{ cm}^{-1}$   
                   $11.605 \text{ K}$   
                   $1.73 [\Omega\text{-cm}]^{-1}$

# PERIODIC TABLE OF THE ELEMENTS

<http://www.kf-split.hr/periodni/en/>

PERIOD	GROUP I IA		GROUP IIA		GROUP IUPAC										GROUP IIIA		GROUP IVA		GROUP VA		GROUP VIA		GROUP VIIA		GROUP VIIIA											
1	1 1.0079 <b>H</b> HYDROGEN		2 9.0122 <b>He</b> HELIUM												10 20.180 <b>Ne</b> NEON		18 39.948 <b>Ar</b> ARGON		36 83.80 <b>Kr</b> KRYPTON		54 131.29 <b>Xe</b> XENON		86 (222) <b>Rn</b> RADON													
2	3 6.941 <b>Li</b> LITHIUM		4 9.0122 <b>Be</b> BERYLLIUM												5 10.811 <b>B</b> BORON		6 12.011 <b>C</b> CARBON		7 14.007 <b>N</b> NITROGEN		8 15.999 <b>O</b> OXYGEN		9 18.998 <b>F</b> FLUORINE		10 20.180 <b>Ne</b> NEON											
3	11 22.990 <b>Na</b> SODIUM		12 24.305 <b>Mg</b> MAGNESIUM												13 26.982 <b>Al</b> ALUMINIUM		14 28.086 <b>Si</b> SILICON		15 30.974 <b>P</b> PHOSPHORUS		16 32.065 <b>S</b> SULPHUR		17 35.453 <b>Cl</b> CHLORINE		18 39.948 <b>Ar</b> ARGON											
4	19 39.098 <b>K</b> POTASSIUM		20 40.078 <b>Ca</b> CALCIUM		21 44.956 <b>Sc</b> SCANDIUM		22 47.867 <b>Ti</b> TITANIUM		23 50.942 <b>V</b> VANADIUM		24 51.996 <b>Cr</b> CHROMIUM		25 54.938 <b>Mn</b> MANGANESE		26 55.845 <b>Fe</b> IRON		27 58.933 <b>Co</b> COBALT		28 58.693 <b>Ni</b> NICKEL		29 63.546 <b>Cu</b> COPPER		30 65.39 <b>Zn</b> ZINC		31 69.723 <b>Ga</b> GALLIUM		32 72.64 <b>Ge</b> GERMANIUM		33 74.922 <b>As</b> ARSENIC		34 78.96 <b>Se</b> SELENIUM		35 79.904 <b>Br</b> BROMINE		36 83.80 <b>Kr</b> KRYPTON	
5	37 85.468 <b>Rb</b> RUBIDIUM		38 87.62 <b>Sr</b> STRONTIUM		39 88.906 <b>Y</b> YTRIUM		40 91.224 <b>Zr</b> ZIRCONIUM		41 92.906 <b>Nb</b> NIOBIUM		42 95.94 <b>Mo</b> MOLYBDENUM		43 (98) <b>Tc</b> TECHNETIUM		44 101.07 <b>Ru</b> RUTHENIUM		45 102.91 <b>Rh</b> RHODIUM		46 106.42 <b>Pd</b> PALLADIUM		47 107.87 <b>Ag</b> SILVER		48 112.41 <b>Cd</b> CADMIUM		49 114.82 <b>In</b> INDIUM		50 118.71 <b>Sn</b> TIN		51 121.76 <b>Sb</b> ANTIMONY		52 127.60 <b>Te</b> TELLURIUM		53 126.90 <b>I</b> IODINE		54 131.29 <b>Xe</b> XENON	
6	55 132.91 <b>Cs</b> CAESIUM		56 137.33 <b>Ba</b> BARIUM		57-71 <b>La-Lu</b> Lanthanide		72 178.49 <b>Hf</b> HAFNIUM		73 180.95 <b>Ta</b> TANTALUM		74 183.84 <b>W</b> TUNGSTEN		75 186.21 <b>Re</b> RHENIUM		76 190.23 <b>Os</b> OSMIUM		77 192.22 <b>Ir</b> IRIDIUM		78 195.08 <b>Pt</b> PLATINUM		79 196.97 <b>Au</b> GOLD		80 200.59 <b>Hg</b> MERCURY		81 204.38 <b>Tl</b> THALLIUM		82 207.2 <b>Pb</b> LEAD		83 208.98 <b>Bi</b> BISMUTH		84 (209) <b>Po</b> POLONIUM		85 (210) <b>At</b> ASTATINE		86 (222) <b>Rn</b> RADON	
7	87 (223) <b>Fr</b> FRANCIUM		88 (226) <b>Ra</b> RADIUM		89-103 <b>Ac-Lr</b> Actinide		104 (261) <b>Rf</b> RUTHERFORDIUM		105 (262) <b>Db</b> DUBNIUM		106 (266) <b>Sg</b> SEABORGIUM		107 (264) <b>Bh</b> BOHRNIUM		108 (277) <b>Hs</b> HASSIUM		109 (268) <b>Mt</b> MEITNERIUM		110 (281) <b>Uu</b> UNUNNIUM		111 (272) <b>Uuu</b> UNUNUNIUM		112 (285) <b>Uub</b> UNUNBIUM		114 (289) <b>Uuq</b> UNUNQUADIUM											

RELATIVE ATOMIC MASS (1)

GROUP IUPAC

GROUP CAS

ATOMIC NUMBER

SYMBOL

ELEMENT NAME

Legend:

- Metal
- Semimetal
- Nonmetal
- 1 Alkali metal
- 2 Alkaline earth metal
- Transition metals
- Lanthanide
- Actinide
- 16 Chalcogens element
- 17 Halogens element
- 18 Noble gas

STANDARD STATE (25 °C; 101 kPa)

Ne - gas Fe - solid Ga - liquid Yc - synthetic

## LANTHANIDE

57 138.91 <b>La</b> LANTHANUM	58 140.12 <b>Ce</b> CERIUM	59 140.91 <b>Pr</b> PRASEODYMIUM	60 144.24 <b>Nd</b> NEODYMIUM	61 (145) <b>Pm</b> PROMETHIUM	62 150.36 <b>Sm</b> SAMARIUM	63 151.96 <b>Eu</b> EUROPIUM	64 157.25 <b>Gd</b> GADOLINIUM	65 158.93 <b>Tb</b> TERBIUM	66 162.50 <b>Dy</b> DYSPROSIUM	67 164.93 <b>Ho</b> HOLMIUM	68 167.26 <b>Er</b> ERBIUM	69 168.93 <b>Tm</b> THULIUM	70 173.04 <b>Yb</b> YTTERIUM	71 174.97 <b>Lu</b> LUTETIUM
-------------------------------------	----------------------------------	--	-------------------------------------	-------------------------------------	------------------------------------	------------------------------------	--------------------------------------	-----------------------------------	--------------------------------------	-----------------------------------	----------------------------------	-----------------------------------	------------------------------------	------------------------------------

## ACTINIDE

89 (227) <b>Ac</b> ACTINIUM	90 232.04 <b>Th</b> THORIUM	91 231.04 <b>Pa</b> PROTACTINIUM	92 238.03 <b>U</b> URANIUM	93 (237) <b>Np</b> NEPTUNIUM	94 (244) <b>Pu</b> PLUTONIUM	95 (243) <b>Am</b> AMERICIUM	96 (247) <b>Cm</b> CURIUM	97 (247) <b>Bk</b> BERKELIUM	98 (251) <b>Cf</b> CALIFORNIUM	99 (252) <b>Es</b> EINSTEINIUM	100 (257) <b>Fm</b> FERMIUM	101 (258) <b>Md</b> MENDELEVIUM	102 (259) <b>No</b> NOBELIUM	103 (262) <b>Lr</b> LAWRENCIUM
-----------------------------------	-----------------------------------	--	----------------------------------	------------------------------------	------------------------------------	------------------------------------	---------------------------------	------------------------------------	--------------------------------------	--------------------------------------	-----------------------------------	---------------------------------------	------------------------------------	--------------------------------------

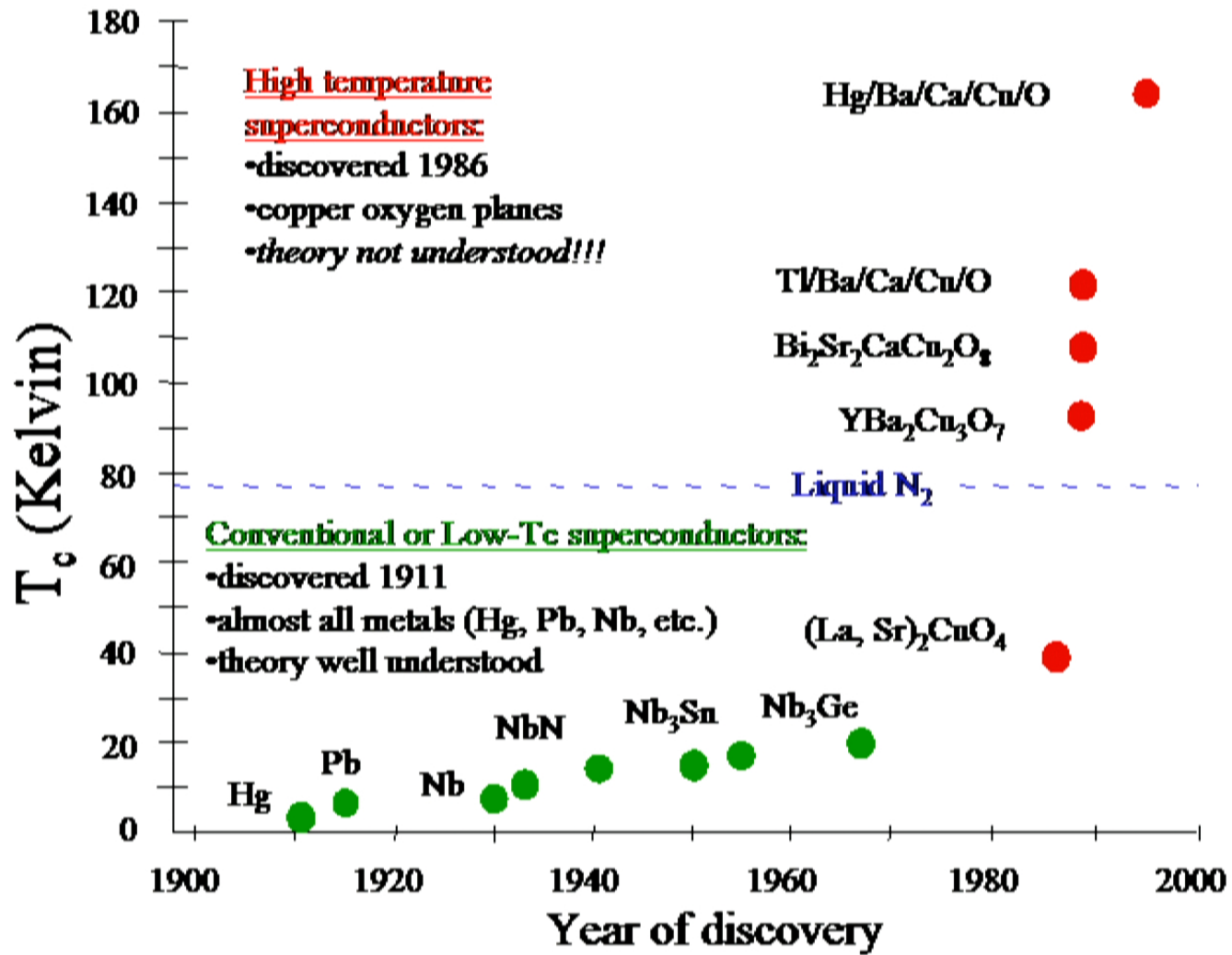
(1) Pure Appl. Chem., 73, No. 4, 667-683 (2001)  
Relative atomic mass is shown with five significant figures. For elements having no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element.  
However three such elements (Th, Pa, and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.

Editor: Aditya Vardhan (adivar@netlinx.com)

N.W. Ashcroft, Nature  
News and Views, 2002

H 1																	He 2
Li 3	Be 4											B 5	C 6	N 7	O 8	F 9	Ne 10
Na 11	Mg 12											Al 13	Si 14	P 15	S 16	Cl 17	Ar 18
K 19	Ca 20	Sc 21	Ti 22	V 23	Cr 24	Mn 25	Fe 26	Co 27	Ni 28	Cu 29	Zn 30	Ga 31	Ge 32	As 33	Se 34	Br 35	Kr 36
Rb 37	Sr 38	Y 39	Zr 40	Nb 41	Mo 42	Tc 43	Ru 44	Rh 45	Pd 46	Ag 47	Cd 48	In 49	Sn 50	Sb 51	Te 52	I 53	Xe 54
Cs 55	Ba 56	La 57	Hf 72	Ta 73	W 74	Re 75	Os 76	Ir 77	Pt 78	Au 79	Hg 80	Tl 81	Pb 82	Bi 83	Po 84	At 85	Rn 86
Fr 87	Ra 88	Ac 89	Ru 104	Ha 105	Unh 106	Uns 107	Uno 108	Une 109									
		Ce 58	Pr 59	Nd 60	Pm 61	Sm 62	Eu 63	Gd 64	Tb 65	Dy 66	Ho 67	Er 68	Tm 68	Yb 70	Lu 71		
		Th 90	Pa 91	U 92	Np 93	Pu 94	Am 95	Cm 96	Bk 97	Cf 98	Es 99	Fm 100	Md 101	No 102	Lr 103		



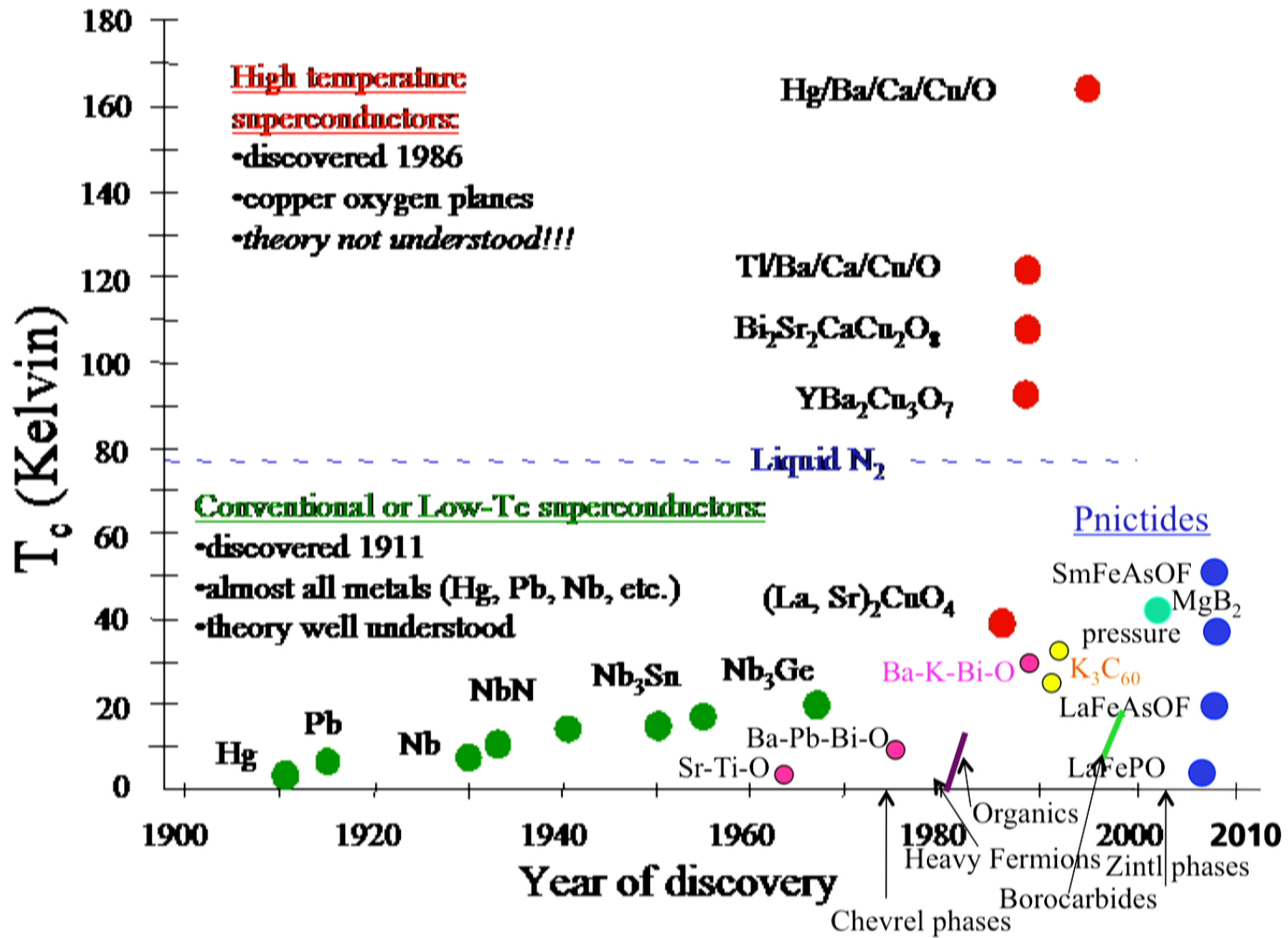


<http://spot.colorado.edu/~dessau/HighTc.shtml>

**Table 1**

Classes of superconducting materials. C (conventional), P (possibly unconventional) and U (unconventional). The 'Year' indicates which year the first material in the class was discovered. The 'Max  $T_c$ ' refers to ambient pressure except for C4 and C6. For 'mag?', y/n indicates whether or not there are magnetic phases nearby in the phase diagram. 'dim'=dimensionality of the structural part of the material believed to drive superconductivity. 'symm'=symmetry of the order parameter. Typical values of coherence length  $\xi$ , penetration depth  $\lambda_L$  and gap ratio are given.  $dT_c/dP$  indicates the sign of the change of  $T_c$  with pressure for most materials in the class.

	Material class	Year	Max $T_c$ material	$T_c^{max}$ (K)	$\xi$ (Å)	$\lambda_L$ (Å)	$2\Delta/k_B T_c$	$dT_c/dP$	mag?	dim	symm	Category
C1	Elements,	1911	Nb	9.5	380	390	3.80	+/-	n	3	s	conv
	alloys and simple compounds	1912	NbN	17	50	2000	4.1	+/-	n	3	s	conv
C2	A15's	1954	Nb <sub>3</sub> Ge	23.2	55	1000	4.2	+	n	3	s	conv
C3	Doped semiconductors	1964	CB <sub>x</sub>	10	950	720	3.5	-	n	3	s	conv
C4	Insul. elements under pressure	1964	S	17				+	n	3	s	conv
C5	Intercalated graphite	1965	C <sub>6</sub> Ca	11.5	380	720	3.6	+	n	2	s	conv
C6	Metallic elements under pressure	1968	Ca	25				+/-	n	3	s	conv
C7	Hydrogen-rich materials	1970	PdD	10.7	400		3.8	+/-	n	3	s	conv
C8	Layered t. m. dichalcogenides	1970	NbS <sub>2</sub>	7.2	100	1250	3.7	-	n	2	s	conv.
C9	Chevreil phases	1971	PbMo <sub>6</sub> S <sub>8</sub>	15	30	3000	4.7	+/-	y	3	s	conv
C10	Magnetic superconductors	1972	ErRh <sub>4</sub> B <sub>4</sub>	8.7	180	830	4	+/-	y	3	s	conv
C11	Thin films	1978							n	2	s	conv
C12	Magnesium diboride	2001	MgB <sub>2</sub>	39	52	1400	4.5	-	n	2	s	conv
P1	Bismuthates	1975	Ba <sub>1-x</sub> K <sub>x</sub> BiO <sub>3</sub>	34	50	5500	4	-	n	3	s	poss unc
P2	Fullerenes	1991	RbC <sub>52</sub> C <sub>60</sub>	33	30	4500	3.5-5.0	-	n	0	s	poss unc
P3	Borocarbides	1993	YPd <sub>3</sub> B <sub>3</sub> C <sub>0.3</sub>	23	100	1000	4	+/-	y,n	2	s + g?	poss unc
P4	Plutonium compounds	2002	PuCoGa <sub>5</sub>	18.5	16	2400	5-8	+/-	y	2	d	poss unc
P5	Interface superconductivity	2007	LaAlO <sub>3</sub> /SrTiO <sub>3</sub>	.35	600				y	2		poss unc
P6	Aromatic hydrocarbons	2010	K-doped DBP	33	180	770		+/-	n	3		poss unc
P7	Doped top. ins.	2010	Cu <sub>x</sub> (PbSe) <sub>5</sub> (Bi <sub>2</sub> Se <sub>3</sub> ) <sub>6</sub>	3	110	13000			n	2		poss unc
P8	BiS <sub>2</sub> -based materials	2012	YbO <sub>0.5</sub> F <sub>0.5</sub> BiS <sub>2</sub>	5.4	53	5000	7.2	+/-	n	2	s	poss unc
P9	Unstable/elusive sc	1946	C-S	300?					n	2		poss unc
U1	Heavy fermions	1979	UPd <sub>2</sub> Al <sub>3</sub>	2	50	4000		+/-	y	3	d, p	unconv
U2	Organic charge-transfer	1980	(BEDT-TTF) <sub>2</sub> X	13.4	100	5000	4.4	-	y	1, 2	d	unconv
U3	Cuprates hole-doped	1986	HgBa <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>9</sub>	134	20	1200	4.3	+	y	2	d	unconv
U4	Cuprates e-doped	1989	Sr <sub>0.9</sub> La <sub>x</sub> CuO <sub>2</sub>	40	50	2500	3.5	-	y	2	d	unconv
U5	Strontium ruthenate	1994	Sr <sub>2</sub> RuO <sub>4</sub>	1.5	660	1500		-	y	2	p	unconv
U6	Layered nitrides	1996	Ca(THF)HfNCl	26	60	4700	2.9-10	-	n	2	d + id	unconv
U7	Ferromagnetic sc	2000	UGe <sub>2</sub>	0.8	100	~ 10 <sup>4</sup>		+/-	y	3	p	unconv
U8	Cobalt oxyde hydrate	2003	Na <sub>x</sub> (H <sub>3</sub> O) <sub>z</sub> CoO <sub>2</sub> ·yH <sub>2</sub> O	4.7	100	7000	4.3-4.6	-	y	2	?	unconv
U9	Non-centro-symmetric	2004	SrPtSi <sub>3</sub>	2	60	8000			y	3	s/p	unconv
U10	Iron pnictides	2008	SmFeAsO <sub>0.85</sub>	55	10-50	2000	7.5	+/-	y	2	s±	unconv
U11	Iron chalcogenides	2008	Na <sub>x</sub> Fe <sub>2</sub> Se <sub>2</sub>	46	20	2000	3.8	+	y	2	s	unconv

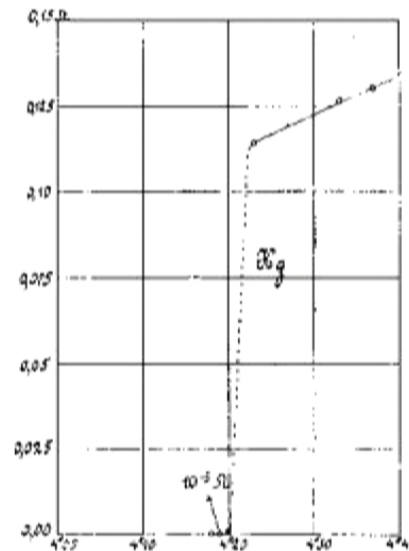


# Superconductivity

Kamerlingh Onnes, H.,  
"The Superconductivity of Mercury."  
*Comm. Phys. Lab. Univ. Leiden; Nos.*  
*122 and 124, 1911.*



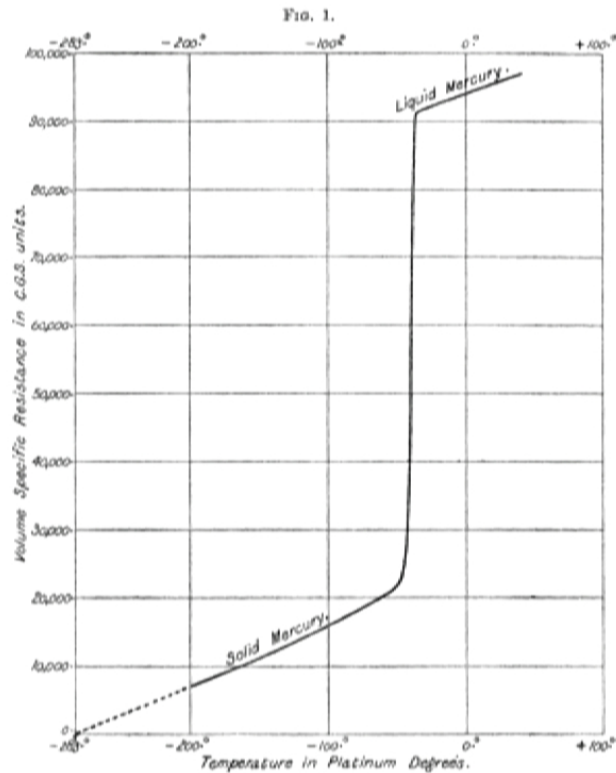
Heike Kamerlingh Onnes





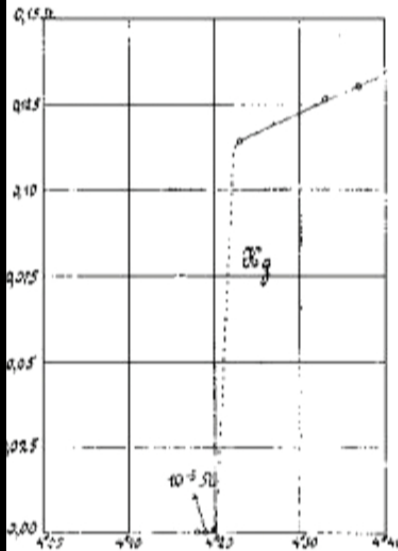
# Superconductivity

## On the Electrical Resistivity of Pure Mercury at the Temperature of Liquid Air James Dewar and J. A. Fleming



*Proceedings of the Royal Society of London,*  
Vol. 60 (1896 – 1897), pp. 76-81

Aside: **Gilles Holst**, the student



- Really the discoverer of superconductivity
- Founding director of Philips labs in Netherlands
  - Holst Memorial Lectures, started 1977
- Acknowledged by Kamerlingh-Onnes in papers, and in a letter for membership in Royal Dutch Academy

***“As a research director at Philips, according to Casimir, Holst would never insist on being coauthor, let alone sole author of papers based essentially on the work of others.”***

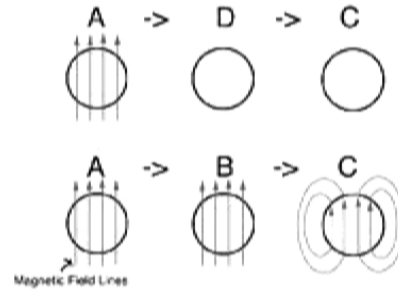
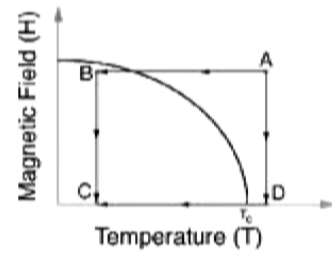
As quoted from Casimir, in Dahl: *Superconductivity*



Walther Meißner  
1882 - 1974

## Meissner-Ochsenfeld Effect

### Perfect Conductor



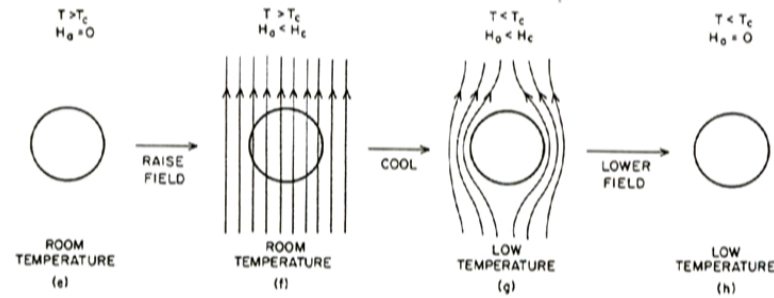
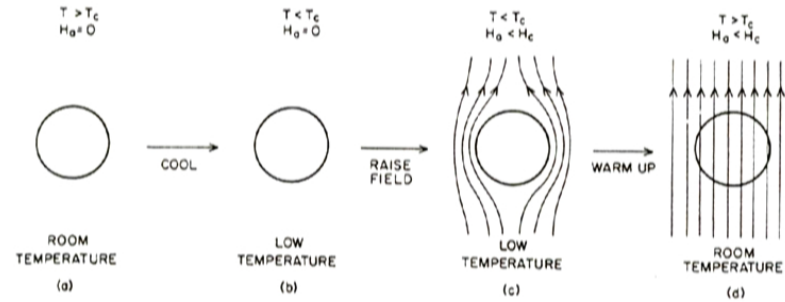
Robert Ochsenfeld  
1901 - 1993



Walther Meißner  
1882 - 1974

# Meissner-Ochsenfeld Effect

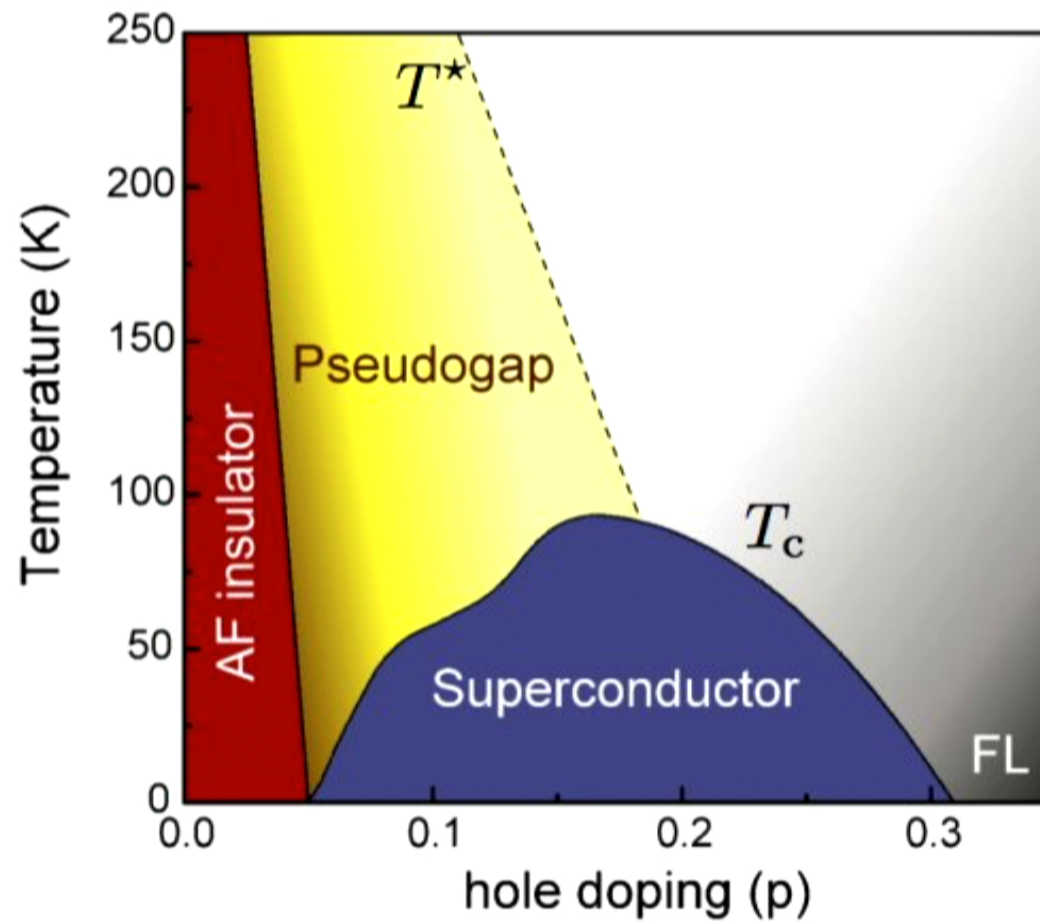
SUPERCONDUCTOR  
(ZERO INDUCTION)



Robert Ochsenfeld  
1901 - 1993

Dahl, p.180

“high  $T_c$ ” cuprates



<http://www.toulouse.lncmi.cnrs.fr/spip.php?rubrique149&lang=en>



Frank Marsiglio (Univ. of Alberta)

$\text{La}_2\text{CuO}_4$  Rm. 272

fm3@ua'

$\text{La}_{2-x}\text{Sr}_x\text{CuO}_4$

**How we view  
electronic properties  
in  
Condensed Matter**

# The Theory of Everything

R. B. Laughlin\* and David Pines<sup>†‡§</sup>



Bob Laughlin



David Pines

28–31 PNAS January 4, 2000 vol. 97 no. 1

$$i\hbar \frac{\partial}{\partial t} |\Psi\rangle = \mathcal{H}|\Psi\rangle \quad [1]$$

where

$$\begin{aligned} \mathcal{H} = & - \sum_j^{N_e} \frac{\hbar^2}{2m} \nabla_j^2 - \sum_\alpha^{N_i} \frac{\hbar^2}{2M_\alpha} \nabla_\alpha^2 \\ & - \sum_j^{N_e} \sum_\alpha^{N_i} \frac{Z_\alpha e^2}{|\vec{r}_j - \vec{R}_\alpha|} + \sum_{j \ll k}^{N_e} \frac{e^2}{|\vec{r}_j - \vec{r}_k|} + \sum_{\alpha \ll \beta}^{N_i} \frac{Z_\alpha Z_\beta e^2}{|\vec{R}_\alpha - \vec{R}_\beta|}. \end{aligned} \quad [2]$$

‘ We have succeeded in reducing all of ordinary physical behavior to a simple, correct Theory of Everything only to discover that it has revealed exactly nothing about many things of great importance.’



# The Theory of Everything

R. B. Laughlin\* and David Pines<sup>†‡§</sup>



Bob Laughlin



David Pines

28–31 PNAS January 4, 2000 vol. 97 no. 1

$$i\hbar \frac{\partial}{\partial t} |\Psi\rangle = \mathcal{H}|\Psi\rangle \quad [1]$$

See also, P.W. Anderson's 'More is Different' in Science, 1972

$$\mathcal{H} = - \sum_j^{N_e} \frac{\hbar^2}{2m} \nabla_j^2 - \sum_\alpha^{N_i} \frac{\hbar^2}{2M_\alpha} \nabla_\alpha^2 - \sum_j^{N_e} \sum_\alpha^{N_i} \frac{Z_\alpha e^2}{|\vec{r}_j - \vec{R}_\alpha|} + \sum_{j \ll k}^{N_e} \frac{e^2}{|\vec{r}_j - \vec{r}_k|} + \sum_{\alpha \ll \beta}^{N_i} \frac{Z_\alpha Z_\beta e^2}{|\vec{R}_\alpha - \vec{R}_\beta|} \quad [2]$$



'We have succeeded in reducing all of ordinary physical behavior to a simple, correct Theory of Everything only to discover that it has revealed exactly nothing about many things of great importance.'

## A big “rug”: Fermi Liquid Theory

nothing very exceptional about the  
normal state of electrons in a metal  
(pretend they don't interact)

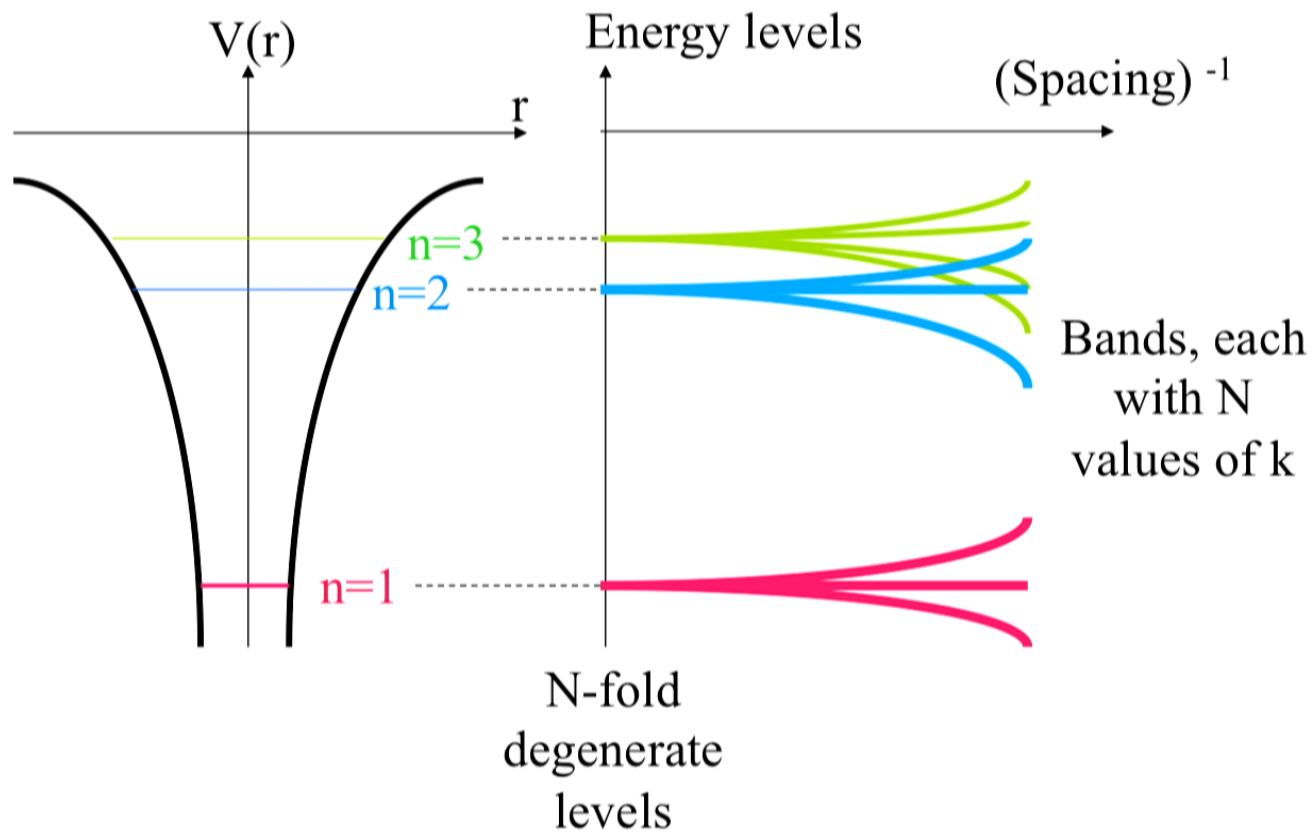
## A big “rug”: Fermi Liquid Theory

nothing very exceptional about the  
normal state of electrons in a metal  
(pretend they don't interact)

... a premise for ‘conventional’ superconductivity



## Band theory of metals

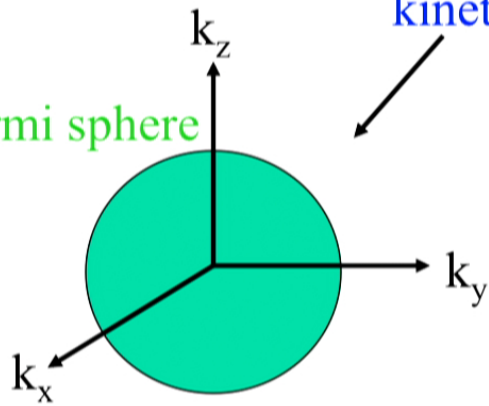


# Electrons in solids

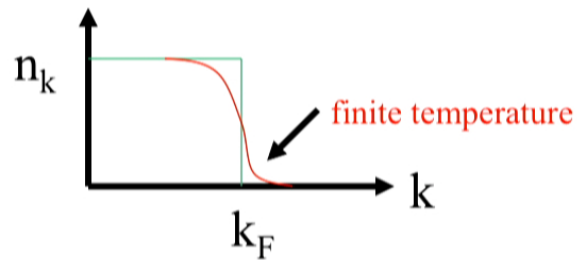
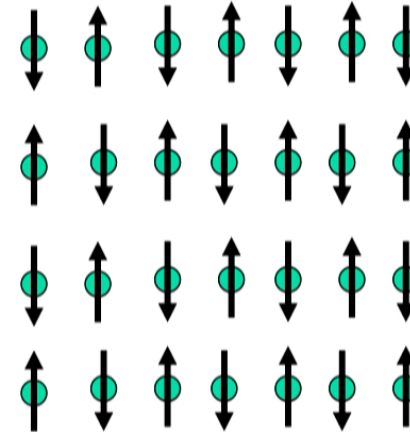


kinetic vs. potential energy

Fermi sphere



Mott insulator



$$E_{\text{kin}} = 2 \sum \epsilon_k n_k$$

why? Coulomb:  
keep electrons away from one another!

Rm. 272

fm3@ualberta.ca

$$E_k = \frac{\hbar^2 |\vec{k}|^2}{2m}$$

u04



# Ultrathin films

VOLUME 62, NUMBER 18

PHYSICAL REVIEW LETTERS

1 MAY 1989

## Onset of Superconductivity in the Two-Dimensional Limit

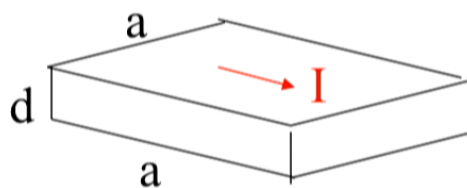
D. B. Haviland, Y. Liu, and A. M. Goldman

*School of Physics and Astronomy, University of Minnesota, Minneapolis, Minnesota 55455*

(Received 2 February 1989)

$$R = \rho L / A$$

$$R_s = \rho a / (ad) = \rho / d$$



sheet resistance  
resistance/sheet

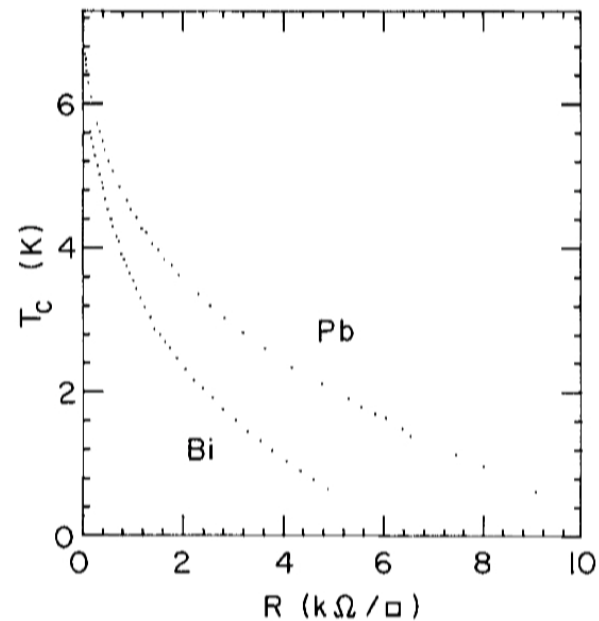


FIG. 3. Dependence of the mean-field transition temperature of Bi and Pb films on sheet resistance.

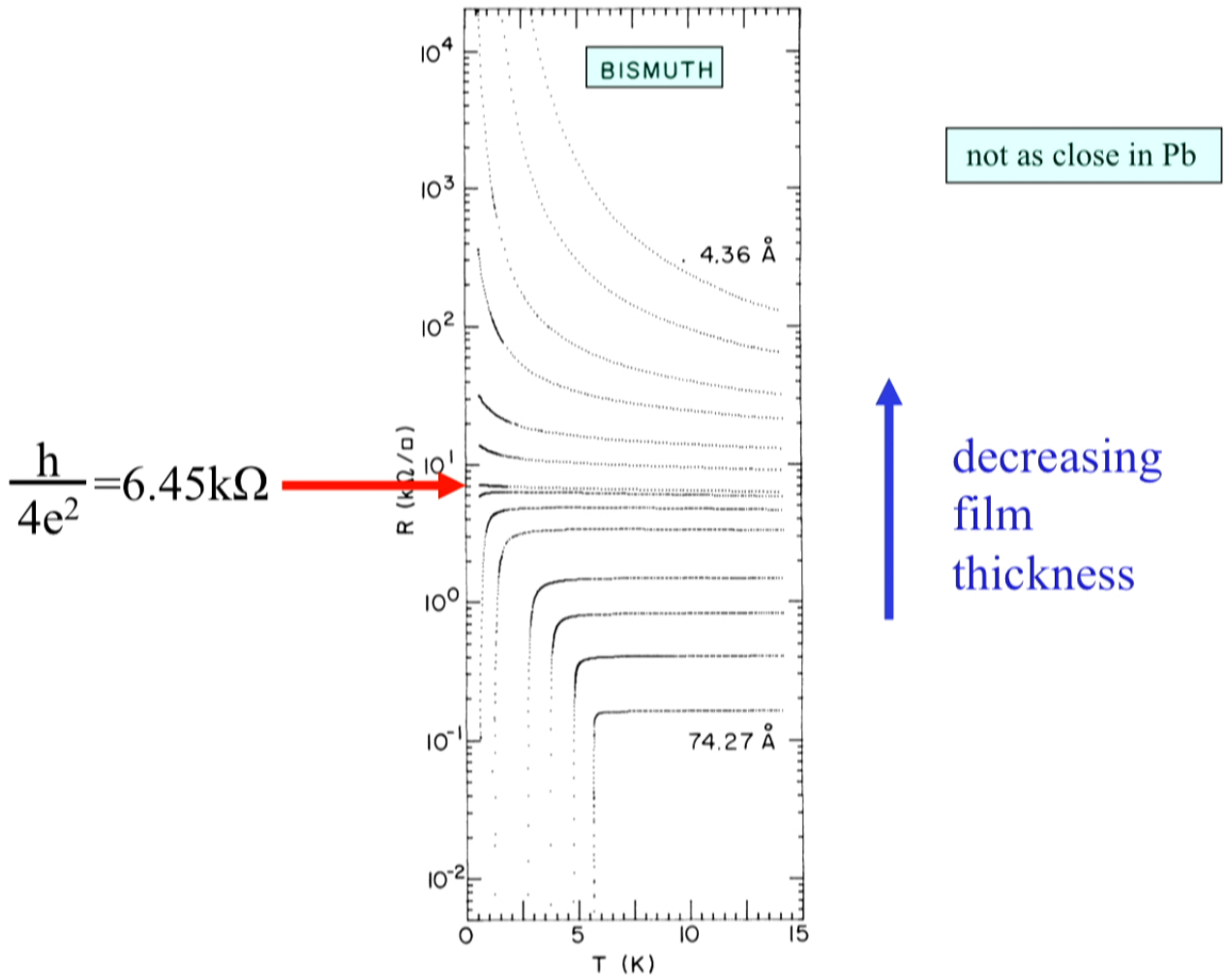


FIG. 1. Evolution of the temperature dependence of the sheet resistance  $R(T)$  with thickness for a Bi film deposited onto Ge. Fewer than half of the traces actually acquired are shown. Film thicknesses shown range from 4.36 to 74.27 Å.



[http://www.jst.go.jp/sicp/ws2009\\_ge3rd/presentation/14.pdf](http://www.jst.go.jp/sicp/ws2009_ge3rd/presentation/14.pdf)



## Two Dimensional Electron Gases at Oxide Interfaces

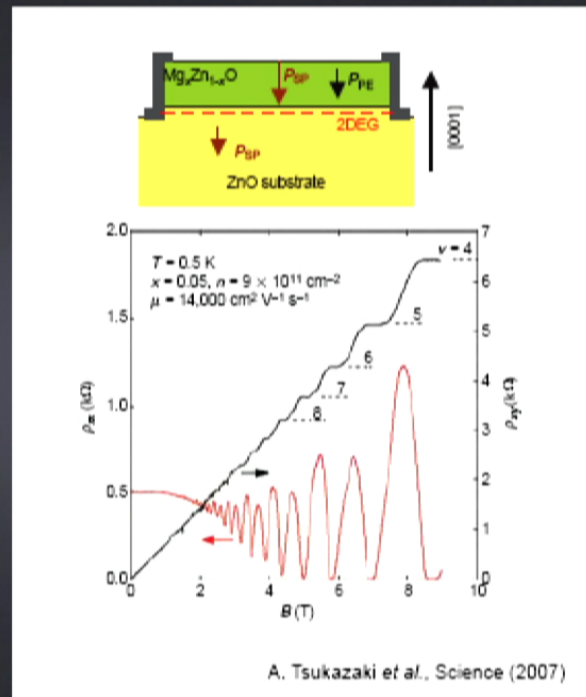
Jochen Mannhart

Center for Electronic Correlations and Magnetism  
University of Augsburg

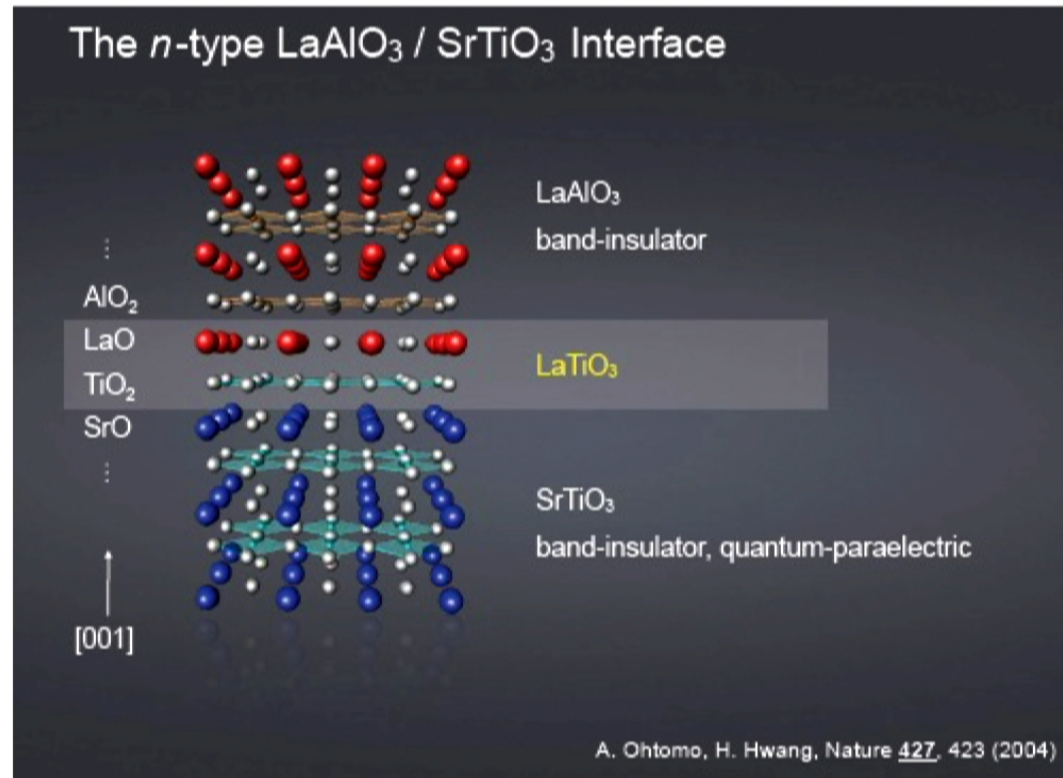
JST-DFG Workshop on Nanoelectronics, Kyoto, Jan. 21, 2009

Slide from Mannhart Kyoto talk

## 2-DEGs Can Be Realized in Oxides

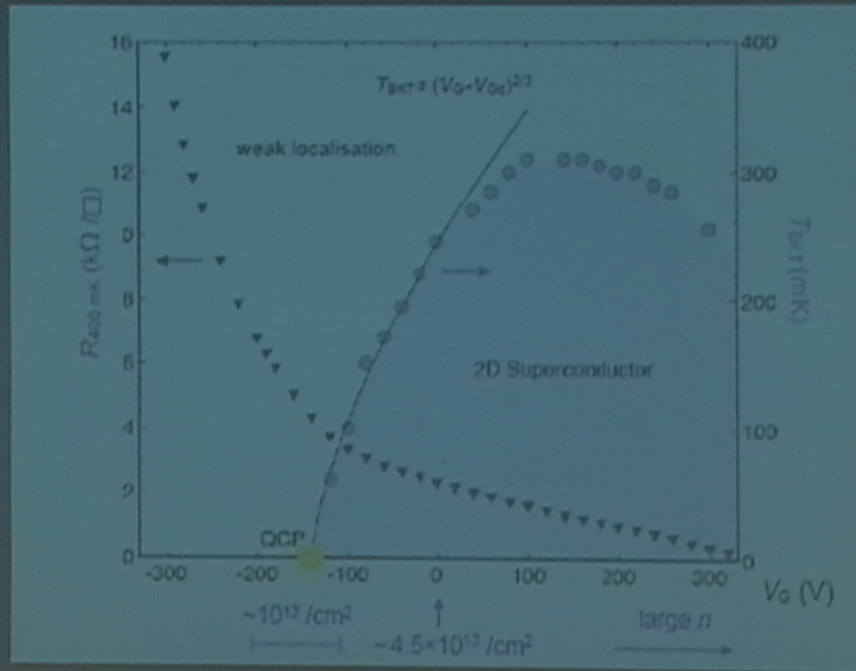


Slide from Mannhart Kyoto talk



Slide from Mannhart Kyoto talk

### Measured Phase Diagram of the LaAlO<sub>3</sub>/SrTiO<sub>3</sub> Interface

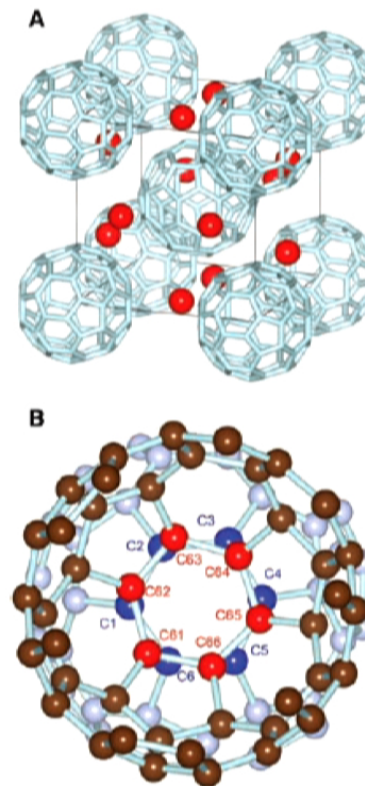
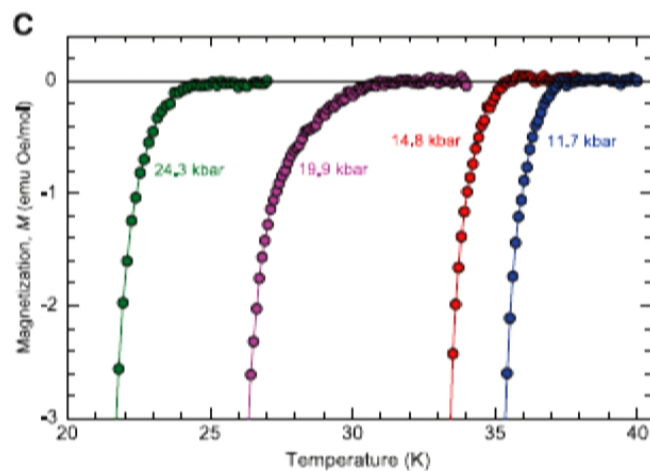


A.D. Caviglia et al., nature 2008

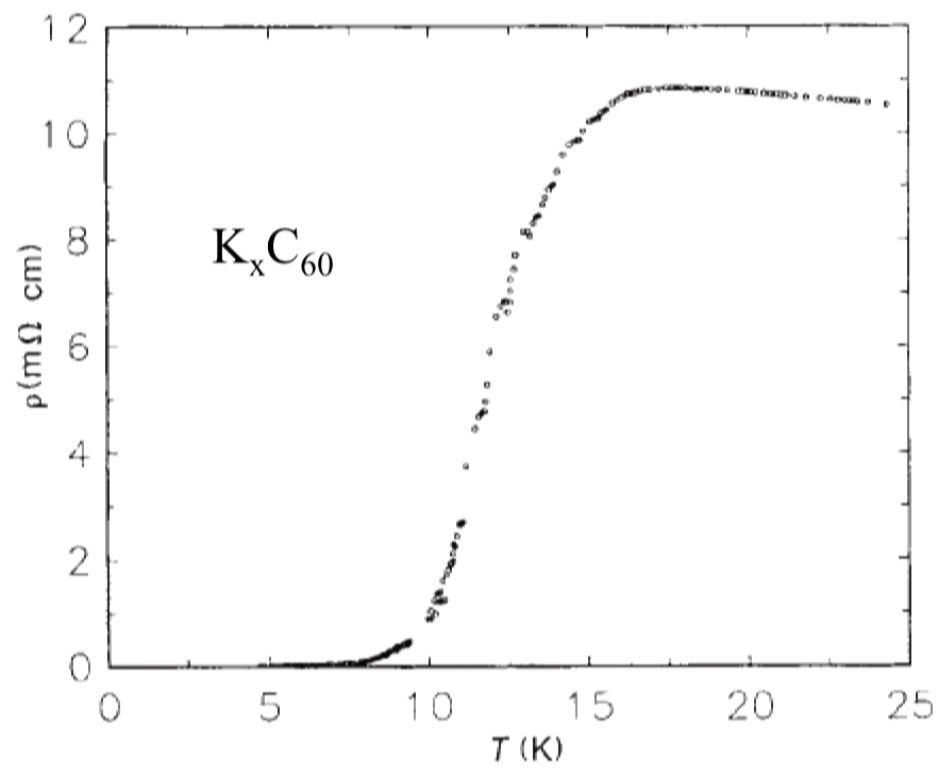


# The Disorder-Free Non-BCS Superconductor $\text{Cs}_3\text{C}_{60}$ Emerges from an Antiferromagnetic Insulator Parent State

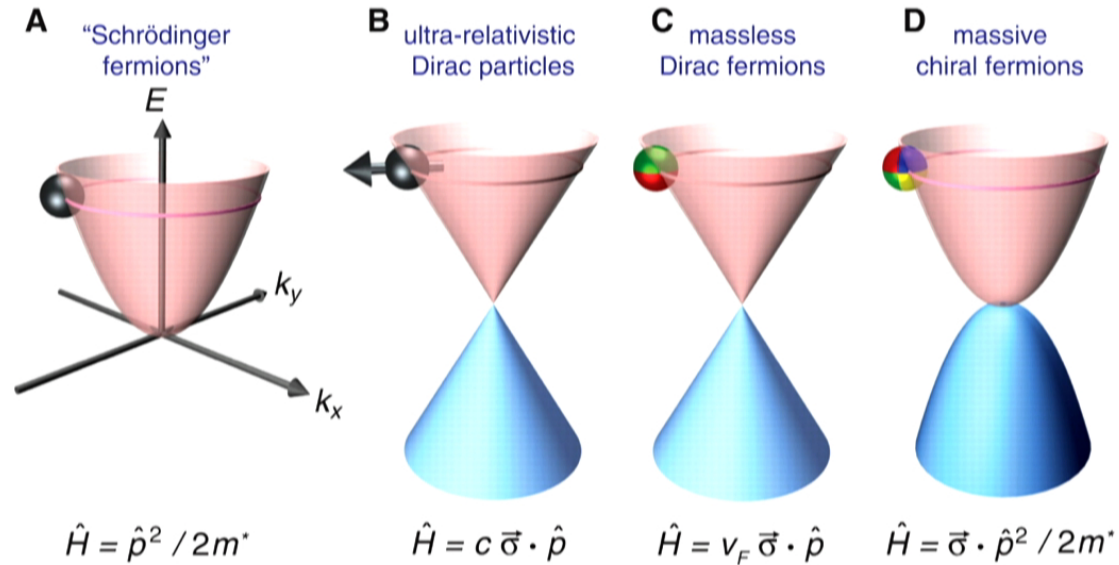
Yasuhiro Takabayashi,<sup>1\*</sup> Alexey Y. Ganin,<sup>2\*</sup> Peter Jeglič,<sup>3</sup> Denis Arčon,<sup>3,4</sup> Takumi Takano,<sup>5</sup> Yoshihiro Iwasa,<sup>5</sup> Yasuo Ohishi,<sup>6</sup> Masaki Takata,<sup>6,7</sup> Nao Takeshita,<sup>8</sup> Kosmas Prassides,<sup>1†</sup> Matthew J. Rosseinsky<sup>2†</sup>



A.F. Hebard et al. Nature 350, 600 (1991)



**Fig. 2 Quasi-particle zoo**



**A. K. Geim Science 324, 1530 -1534 (2009)**

Fig. 2 Quasi-particle zoo. (A) Charge carriers in condensed matter physics are normally described by the Schrödinger equation with an effective mass  $m^*$  different from the free electron mass ( $\hat{p}$  is the momentum operator). (B) Relativistic particles in the limit of zero rest mass follow the Dirac equation, where  $c$  is the speed of light and  $\vec{\sigma}$  is the Pauli matrix. (C) Charge carriers in graphene are called massless Dirac fermions and are described by a 2D analog of the Dirac equation, with the Fermi velocity  $v_F \approx 1 \times 10^6$  m/s playing the role of the speed of light and a 2D pseudospin matrix  $\vec{\sigma}$  describing two sublattices of the honeycomb lattice (3). Similar to the real spin that can change its direction between, say, left and right, the pseudospin is an index that indicates on

Published by AAAS



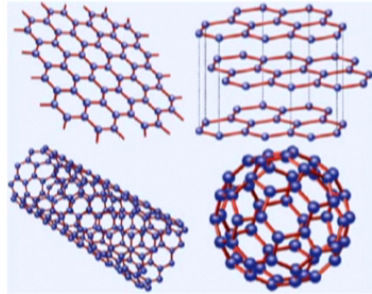


FIG. 1. (Color online) Graphene (top left) is a honeycomb lattice of carbon atoms. Graphite (top right) can be viewed as a stack of graphene layers. Carbon nanotubes are rolled-up cylinders of graphene (bottom left). Fullerenes ( $C_{60}$ ) are molecules consisting of wrapped graphene by the introduction of pentagons on the hexagonal lattice. From [Castro Neto \*et al.\*, 2006a](#).

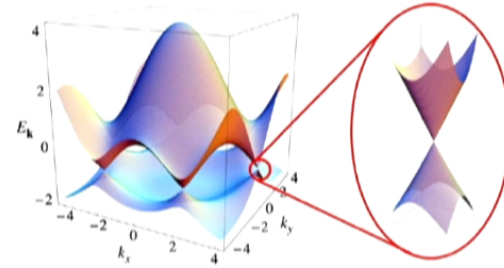
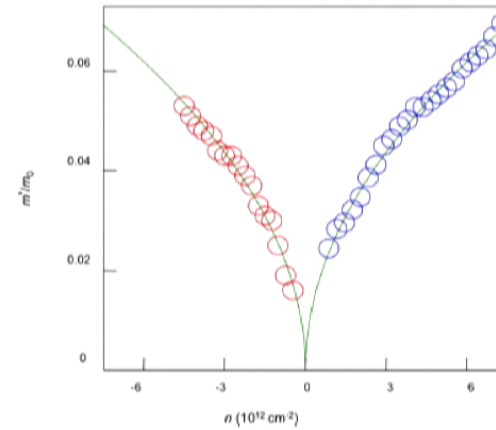


FIG. 3. (Color online) Electronic dispersion in the honeycomb lattice. Left: energy spectrum (in units of  $t$ ) for finite values of  $t$  and  $t'$ , with  $t=2.7$  eV and  $t'=-0.2t$ . Right: zoom in of the energy bands close to one of the Dirac points.



$$m^* = \frac{\sqrt{\pi}}{v_F} \sqrt{n}. \quad (13)$$

perature dependence of the SdH oscillations; solid curves are the best fit by Eq. (13).  $m_0$  is the free-electron mass. Adapted from [Novoselov, Geim, Morozov, \*et al.\*, 2005](#).



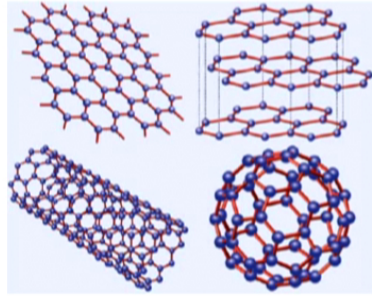


FIG. 1. (Color online) Graphene (top left) is a honeycomb lattice of carbon atoms. Graphite (top right) can be viewed as a stack of graphene layers. Carbon nanotubes are rolled-up cylinders of graphene (bottom left). Fullerenes ( $C_{60}$ ) are molecules consisting of wrapped graphene by the introduction of pentagons on the hexagonal lattice. From Castro Neto *et al.*, 2006a.

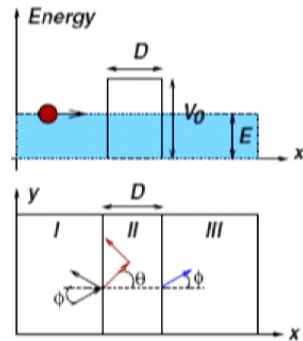


FIG. 6. (Color online) Klein tunneling in graphene. Top: schematic of the scattering of Dirac electrons by a square potential. Bottom: definition of the angles  $\phi$  and  $\theta$  used in the scattering formalism in regions I, II, and III.

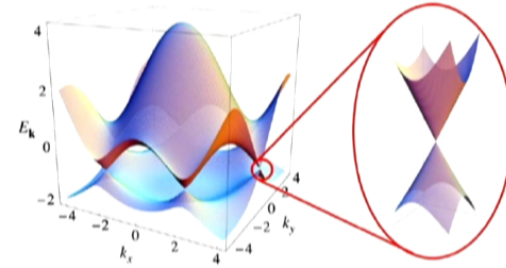
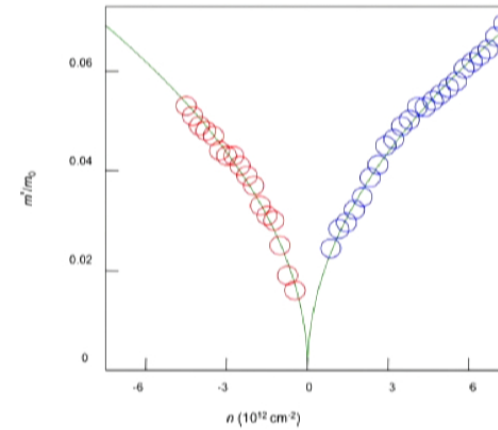


FIG. 3. (Color online) Electronic dispersion in the honeycomb lattice. Left: energy spectrum (in units of  $t$ ) for finite values of  $t$  and  $t'$ , with  $t=2.7$  eV and  $t'=-0.2t$ . Right: zoom in of the energy bands close to one of the Dirac points.

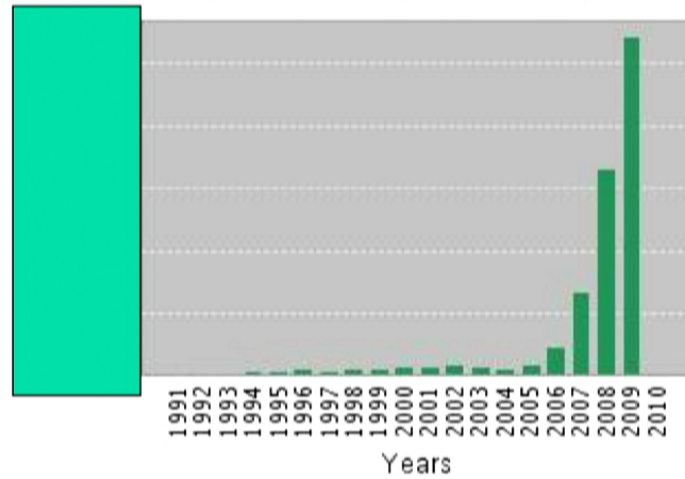


$$m^* = \frac{\sqrt{\pi}}{v_F} \sqrt{n}. \quad (13)$$

perature dependence of the SdH oscillations; solid curves are the best fit by Eq. (13).  $m_0$  is the free-electron mass. Adapted from Novoselov, Geim, Morozov, *et al.*, 2005.

A.K. Geim

**Citations in Each Year**



citations

2004 Science paper:  
2005 Nature paper:  
2007 Nature Materials:  
2009 Rev. Mod. Phys:



Use the checkboxes to remove individual items from this Citation Report

or restrict to items published between  and

1. **Generalized gradient approximation made simple**  
By: Perdew, JP; Burke, K; Ernzerhof, M  
[PHYSICAL REVIEW LETTERS](#) Volume: 77 Issue: 18 Pages: 3865-3868  
Published: OCT 28 1996

2013	2014	2015	2016	2017	Total	Average Citations per Year
12517	13479	14545	8941	0	143820	3344.65
5780	6465	7459	4835	0	50556	2407.43

## Possible High $T_c$ Superconductivity in the Ba-La-Cu-O System

J.G. Bednorz and K.A. Müller

IBM Zürich Research Laboratory, Rüschlikon, Switzerland

Received April 17, 1986

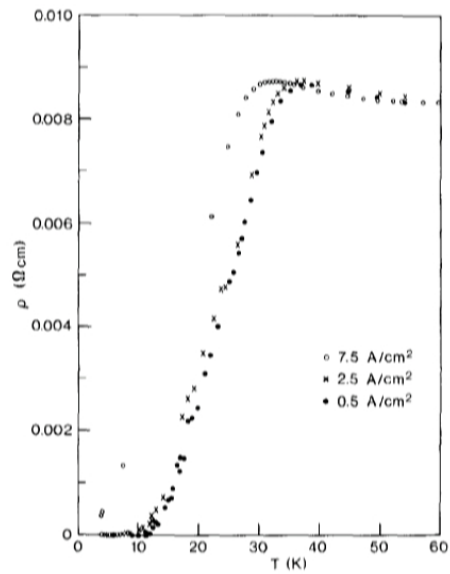
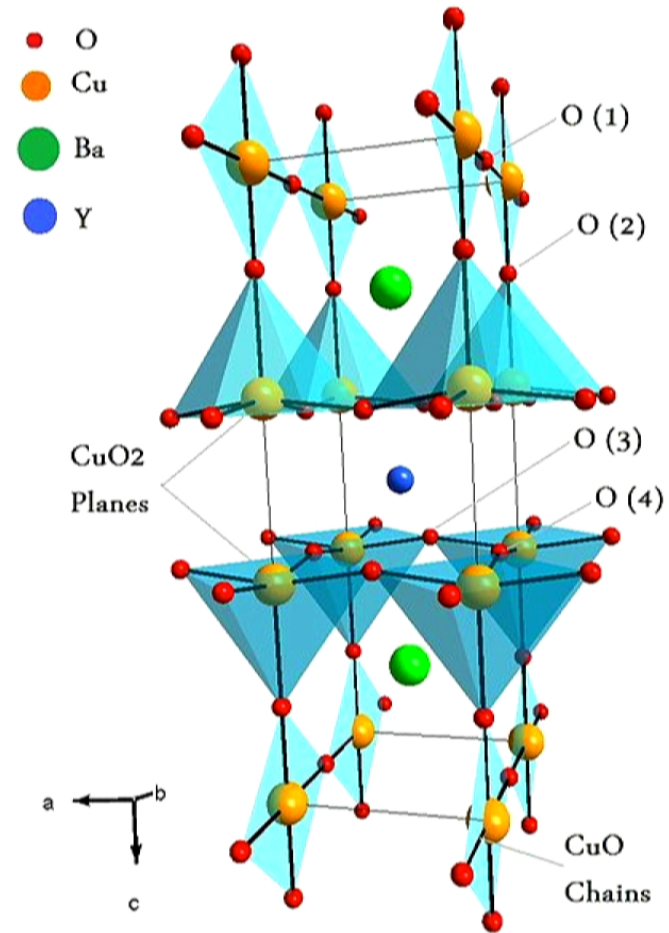
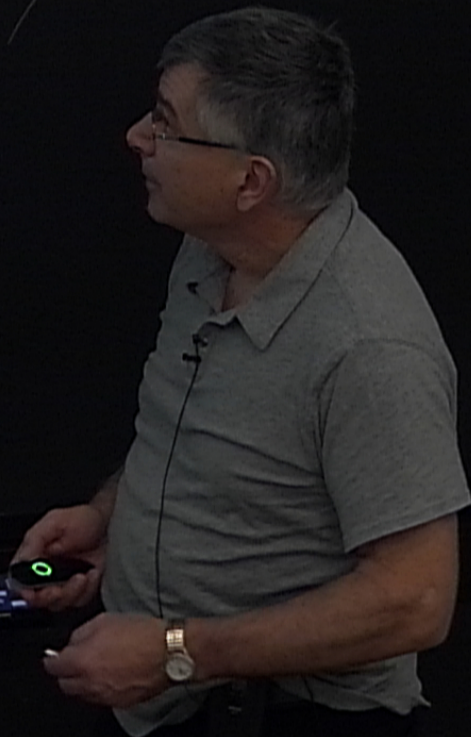
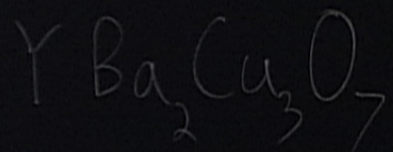


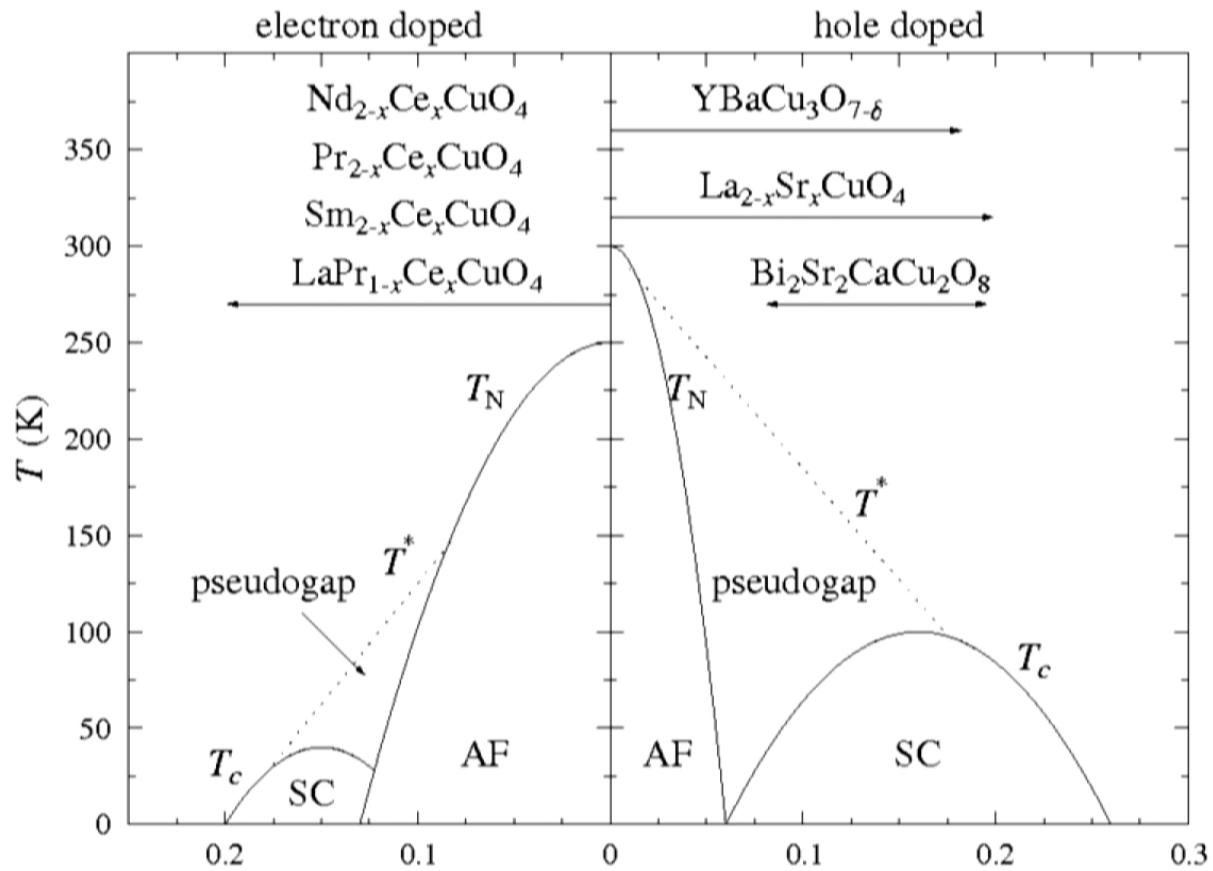
Fig. 3. Low-temperature resistivity of a sample with  $x(\text{Ba})=0.75$ , recorded for different current densities







# Phase diagram for cuprate materials



<http://www.unine.ch/phys/theocond/Research/supra/PhaseDiagramCuprates.gif>

<http://en.wikipedia.org/wiki/File:Cuphasediag.png>



# BCS formalism vs. Pairing Mechanism

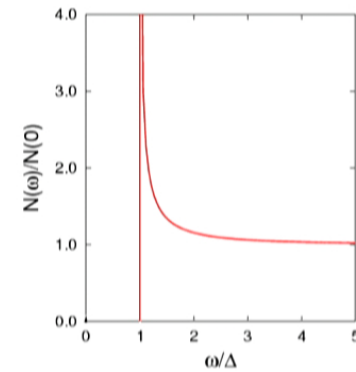
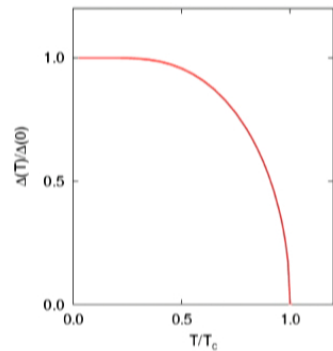
$$\Delta = |V| \frac{1}{N} \sum_k \frac{\Delta}{2E_k}$$

$T_c$  equation (useless)

$$\frac{2\Delta}{k_B T_c} = 3.53$$

Universality

$$\frac{\Delta C}{\gamma T_c} = 1.43$$



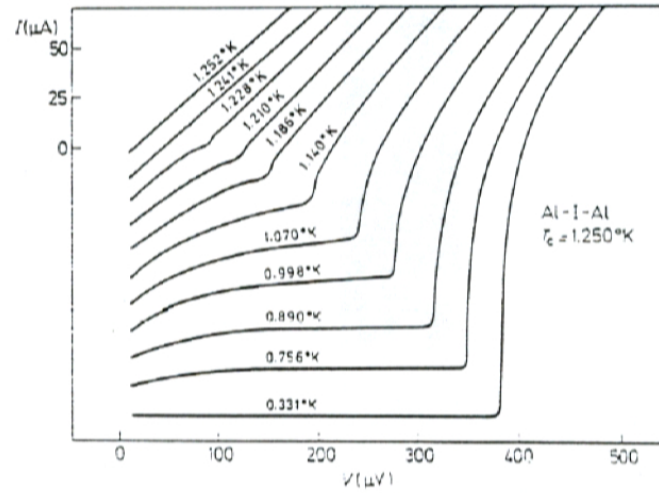
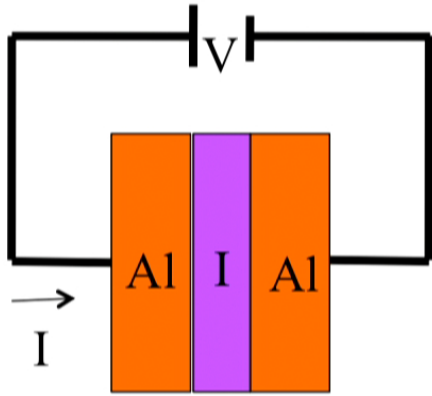
BCS

Bardeen

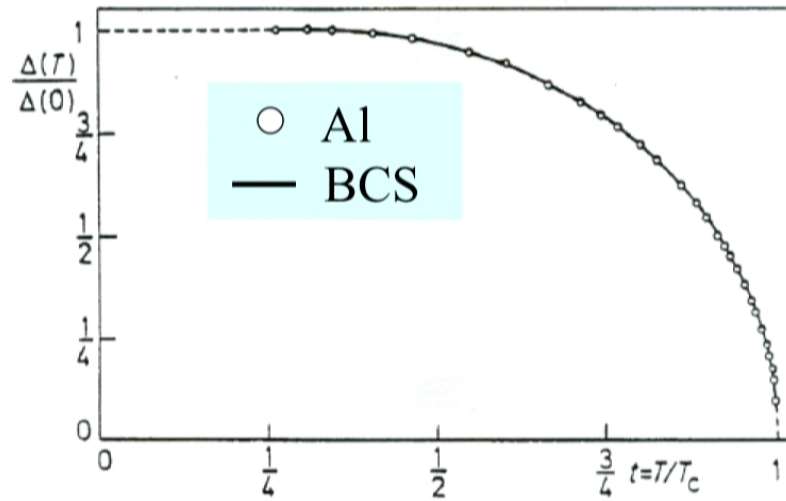
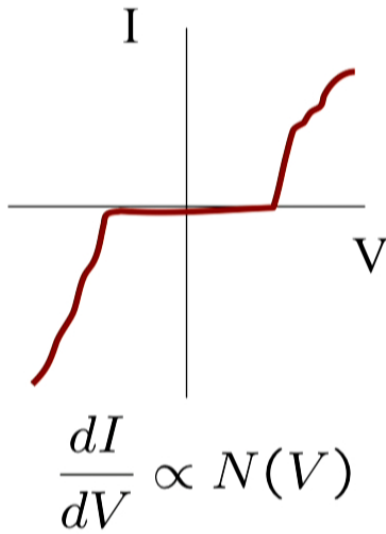
Cooper

Schrieffer

} 1957

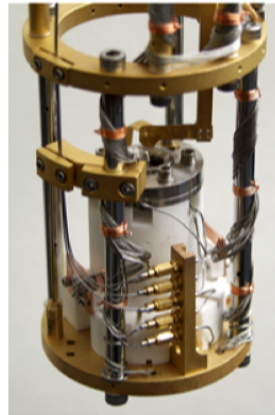
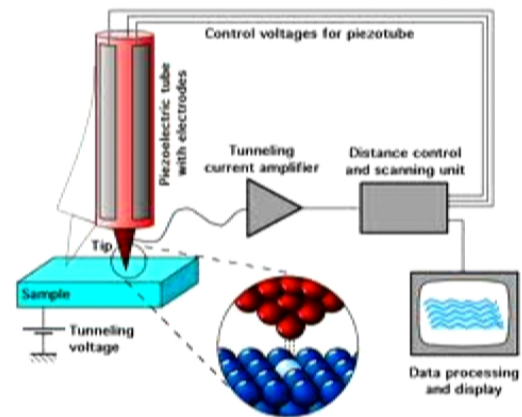


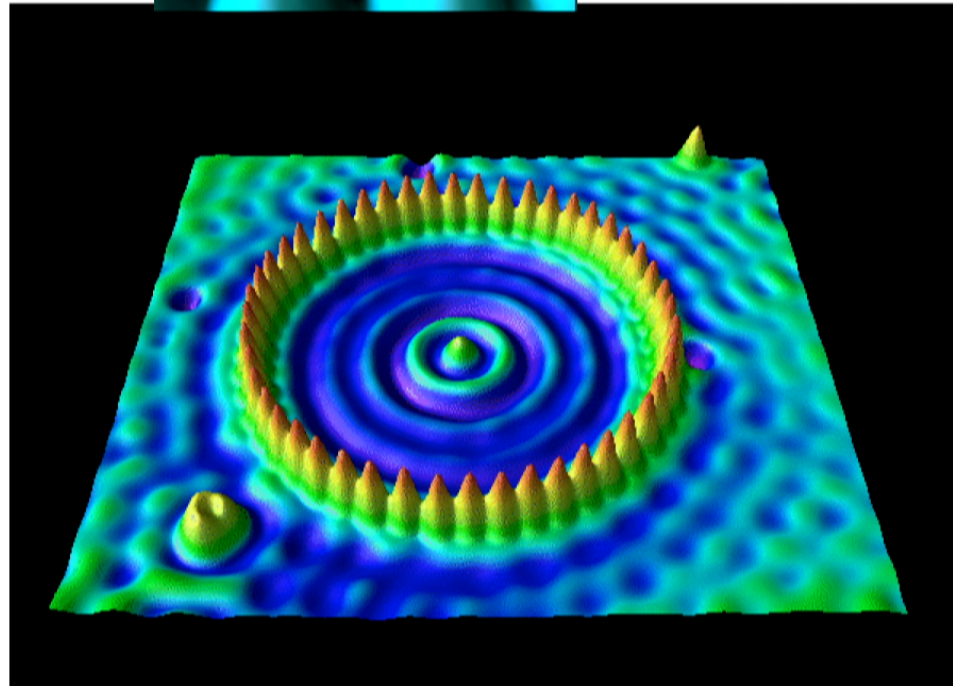
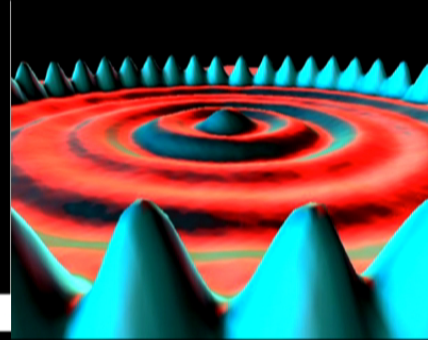
B.L. Blackford and R.H. March, *Can. J. Phys.* **46**, 141 (1968)



Nowadays...

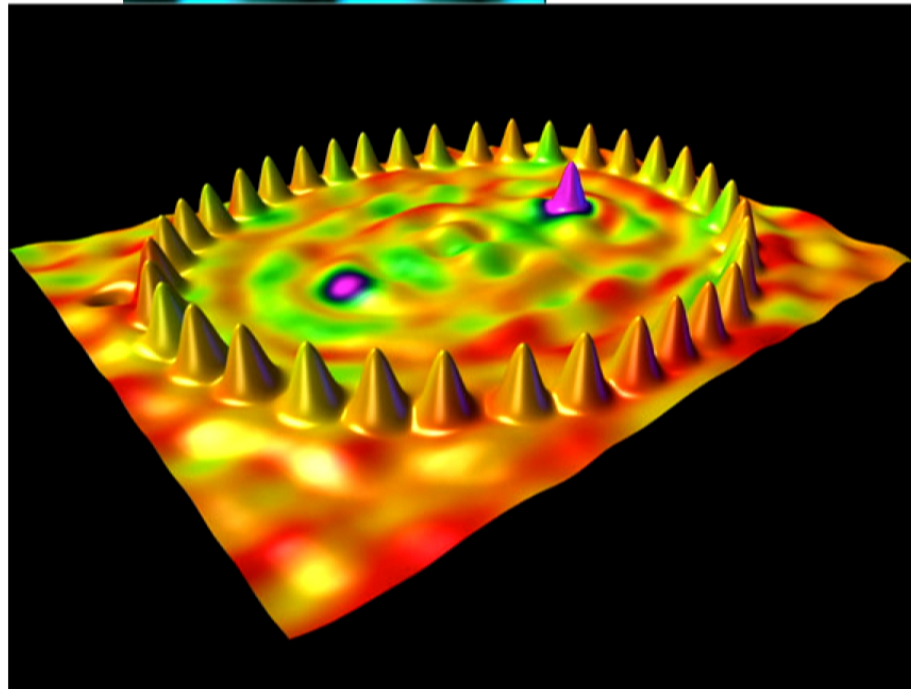
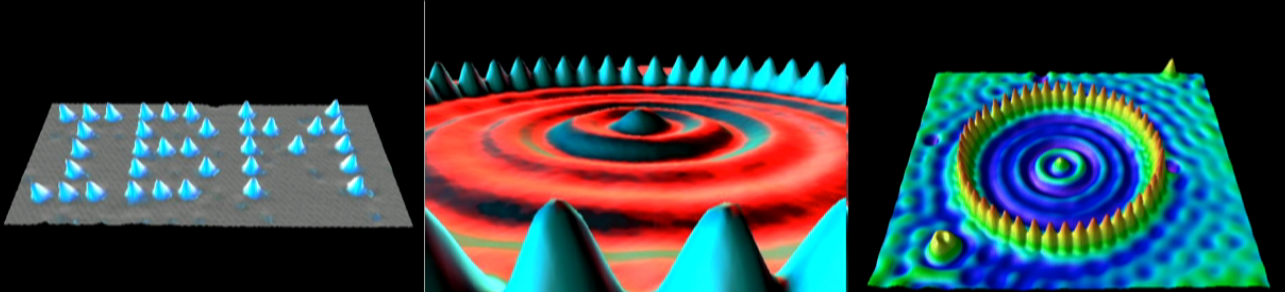
## Scanning Tunneling Microscope (STM)





Crommie, Lutz, and Eigler <http://www.almaden.ibm.com/vis/index.html>



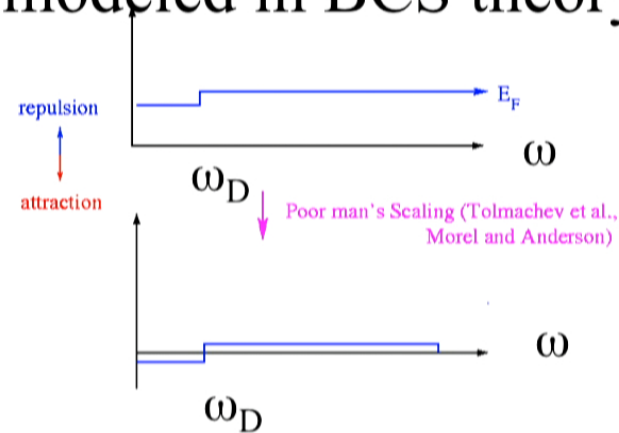


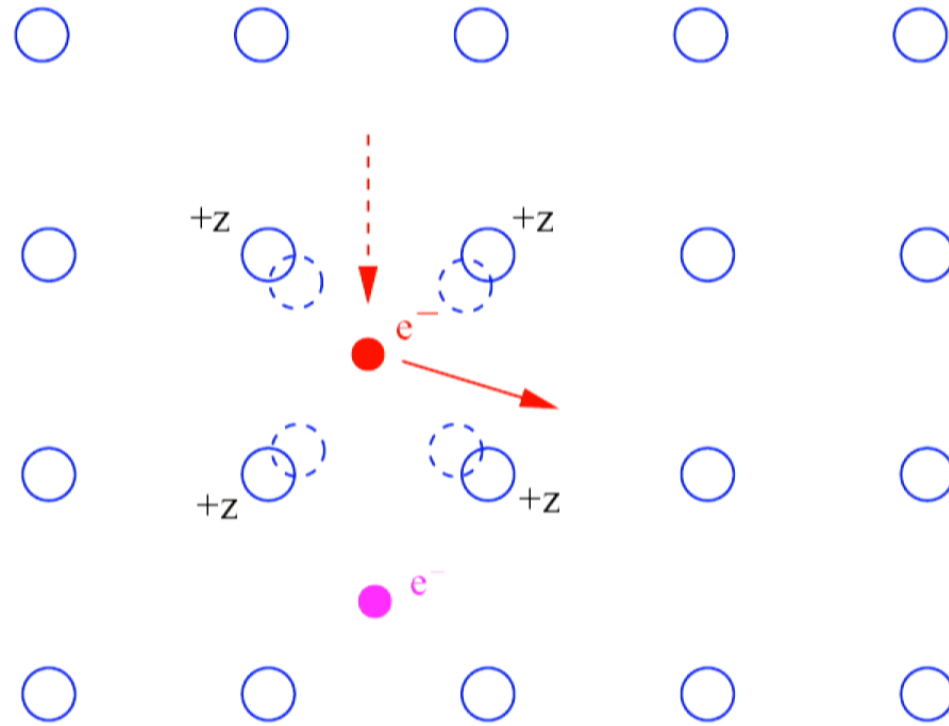
Crommie, Lutz, and Eigler <http://www.almaden.ibm.com/vis/index.html>



# Eliashberg Theory

- Extension of BCS formalism to include dynamical electron-phonon interaction
- builds on Migdal theory in the normal state
- loosely modeled in BCS theory





effective attraction

Direct measurements of the *L*-gap surface states on the (111) face of noble metals by photoelectron spectroscopy

F. Reinert,\* G. Nicolay, S. Schmidt, D. Ehm, and S. Hüfner

# Photoemission

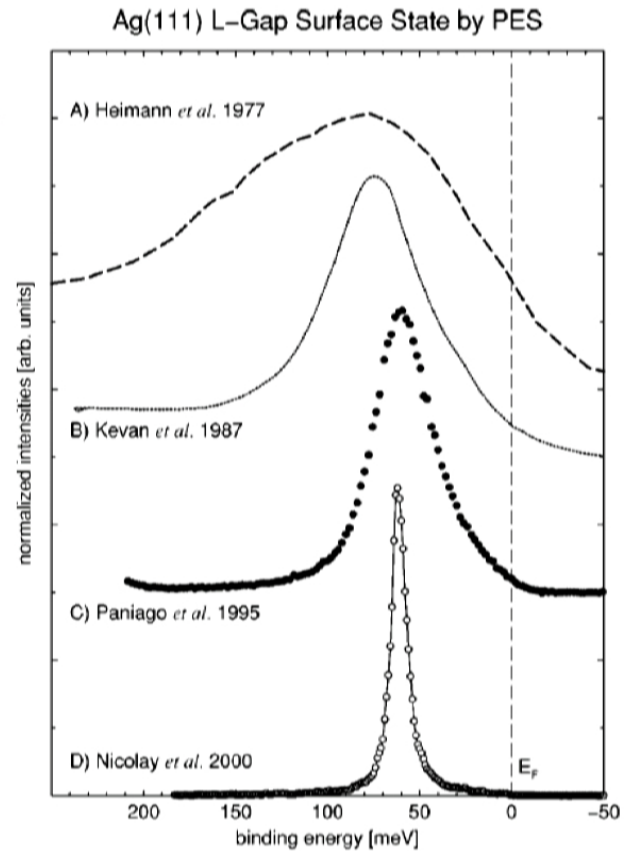
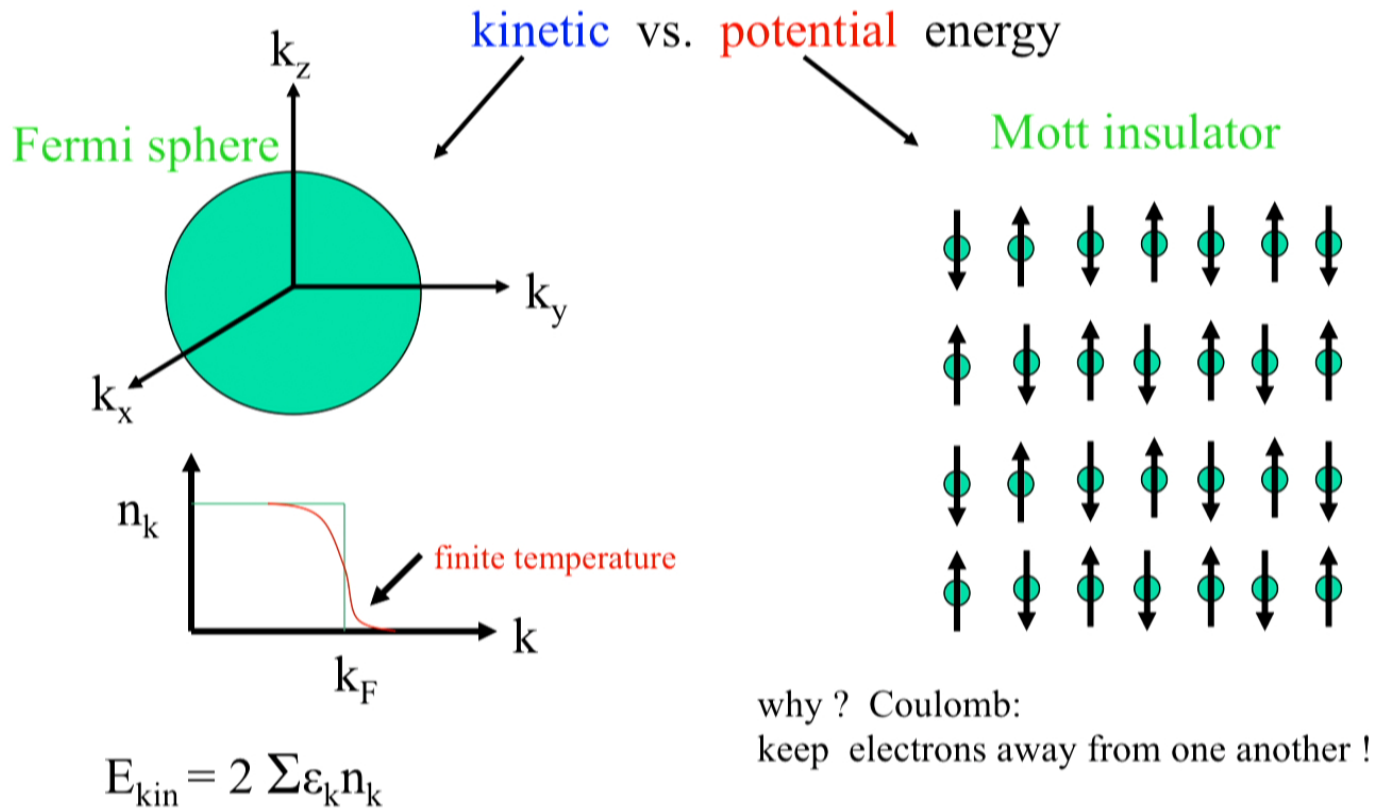


FIG. 1. Technological development in PES since the first observation of the Ag(111) surface state in photoemission spectra: (A) from Ref. 2 measured at room temperature (RT) with Ar I ( $h\nu = 11.83$  eV), angular integrated; (B) from Ref. 30 at RT with  $h\nu = 13$  eV,  $\Delta E \approx 60$  meV and  $\Delta\theta = 1^\circ$ ; (C) from Ref. 13 at  $T = 56$  K with Ar I,  $\Delta E = 21$  meV, and  $\Delta\theta = 0.9^\circ$ ; (D) present data at  $T = 30$  K with He I ( $h\nu = 21.23$  eV),  $\Delta E = 3.5$  meV and  $\Delta\theta = \pm 0.15^\circ$ .

## Electrons in solids



## The conventional scenario: BCS



J. Bardeen L.N. Cooper J.R. Schrieffer

$$\psi_{\text{BCS}} = \prod_{\mathbf{k}} (u_{\mathbf{k}} + v_{\mathbf{k}} c_{\mathbf{k}\uparrow}^{\dagger} c_{-\mathbf{k}\downarrow}^{\dagger}) |0\rangle$$

$$\psi_{2\nu} = \int_0^{2\pi} \frac{d\theta}{2\pi} e^{-i\nu\theta} \prod_{\mathbf{k}} (u_{\mathbf{k}} + e^{i\theta} v_{\mathbf{k}} c_{\mathbf{k}\uparrow}^{\dagger} c_{-\mathbf{k}\downarrow}^{\dagger}) |0\rangle$$

It's all about pairs...

In Ogg's theory it was his intent  
That the current keep flowing, once sent;  
So to save himself trouble,  
He put them in double,  
And instead of stopping, it went.

George Gamow

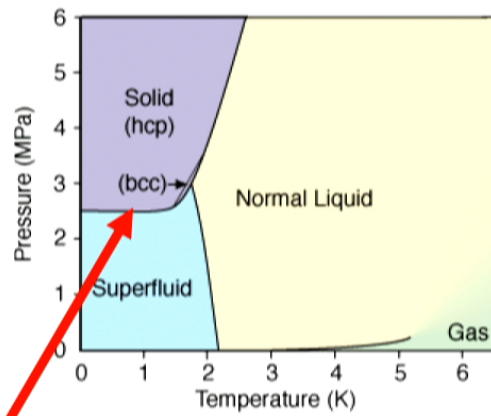
**Bose-Einstein Condensation of Trapped Electron  
Pairs. Phase Separation and Super-  
conductivity of Metal-Ammonia  
Solutions**

RICHARD A. OGG, JR.  
*Department of Chemistry, Stanford University, California*  
March 2, 1946

...Cooper pairs

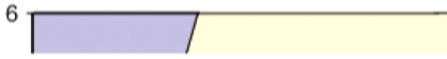


## Supersolid phase in He<sup>4</sup>?



Supersolid phase here?

# Supersolid phase in He<sup>4</sup>?



PRL 109, 155301 (2012)

PHYSICAL REVIEW LETTERS

week ending  
12 OCTOBER 2012

## Absence of Supersolidity in Solid Helium in Porous Vycor Glass

Duk Y. Kim and Moses H. W. Chan\*

*Department of Physics, Pennsylvania State University, University Park, Pennsylvania 16802, USA*

(Received 24 July 2012; published 8 October 2012)

In 2004, Kim and Chan carried out torsional oscillator measurements of solid helium confined in porous Vycor glass and found an abrupt drop in the resonant period below 200 mK. The period drop was interpreted as probable experimental evidence of nonclassical rotational inertia. This experiment sparked considerable activities in the studies of superfluidity in solid helium. More recent ultrasound and torsional oscillator studies, however, found evidence that shear modulus stiffening is responsible for at least a fraction of the period drop found in bulk solid helium samples. The experimental configuration of Kim and Chan makes it unavoidable to have a small amount of bulk solid inside the torsion cell containing the Vycor disk. We report here the results of a new helium in Vycor experiment with a design that is completely free from any bulk solid shear modulus stiffening effect. **We found no measurable period drop that can be attributed to nonclassical rotational inertia.**

# Topological Superconductivity

See [http://www.princeton.edu/~psscmp/ss2010/Lecture\\_Notes\\_files/Lecture3.pdf](http://www.princeton.edu/~psscmp/ss2010/Lecture_Notes_files/Lecture3.pdf)

(Charlie Kane)

# Cold Atoms and Optical Lattices

BCS-BEC Crossover