

Title: On the logical complexity of tiny heat engines -- and whether they can really be reversible

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Abstract: I consider systems that consist of a few hot and a few cold two-level systems and define heat engines as unitaries that extract energy. These unitaries perform logical operations whose complexity depends on both the desired efficiency and the temperature quotient. I show cases where the optimal heat engine solves a hard computational task (e.g. an NP-hard problem) [2]. Heat engines can also drive refrigerators and use the temperature difference between two systems for cooling a third one. I argue that these triples of systems define a classification of thermodynamic resources [1]. All the above assumes that unitaries are implemented by an external controller. To get a thermodynamically reversible process, the joint process on system and controller must be reversible. Then, the implementation of the joint process requires a "meta-controller", and so on. To study thermodynamic limits without such an infinite sequence of controllers, I introduce the model of "physically universal cellular automata", in which the boundary between system and controller can be shifted (in analogy to the Heisenberg-cut for the quantum measurement problem). I show that this model raises a lot of fundamental questions [3]. Literature: [1] Janzing et al: Thermodynamic cost of reliability and low temperatures: Tightening Landauer's principle and the second law, J. Stat. Phys. 2000 [2] Janzing: On the computation power of molecular heat engines, J. Stat. Phys. 2006 [3] Janzing: Is there a physically universal cellular automaton or Hamiltonian? arXiv:1009.1720

On the logical complexity of tiny heat engines – and whether they can really be reversible



Dominik Janzing

MPI for Intelligent Systems, Tübingen, Germany

31 March 2011

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- ▶ definitely about thermodynamics
- ▶ don't consider quantum superpositions
- ▶ “quantum” only in the sense of discrete energy levels

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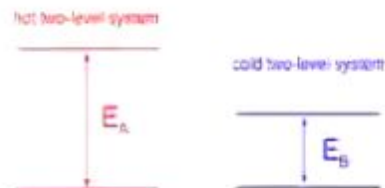
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Why I started to think about quantum heat engines in 1999...

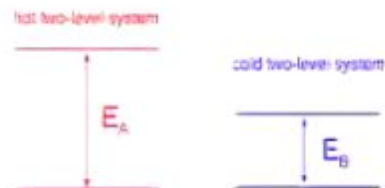
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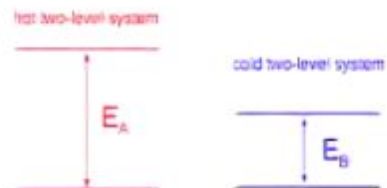
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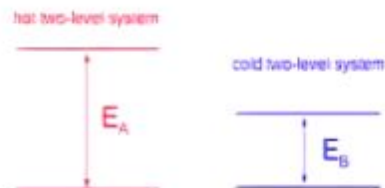
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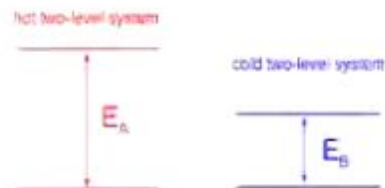
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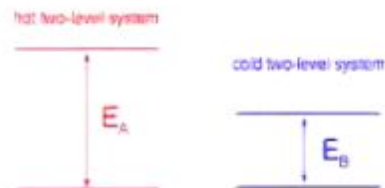
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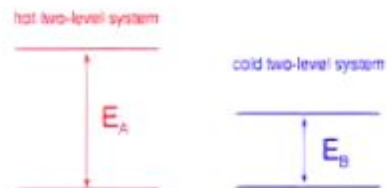
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This talk

- ▶ is far from applications
- ▶ like a game that teaches thermodynamics
- ▶ game with simple rules, but challenging questions

Unitary work extraction

(Allahverdyan, Balian, Nieuwenhuizen 2004)

Given a quantum system with state ρ and Hamiltonian H

- ▶ unitary U extracts the work

$$\Delta W := \text{tr}(\rho H) - \text{tr}(U\rho U^\dagger H).$$

- ▶ U is optimal if it maximizes ΔW

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Passive systems and Gibbs equilibrium

(Lenard 1978)

- ▶ ρ is called **passive** if no unitary can extract energy
($\Leftrightarrow \rho$ diagonal in the energy basis and eigenvalues are ordered according to the energy)
- ▶ ρ is called **completely passive** iff $\rho^{\otimes n}$ is passive for all $n \in \mathbb{N}$
($\Leftrightarrow \rho$ is a Gibbs state to some temperature $T \in [0, \infty]$)

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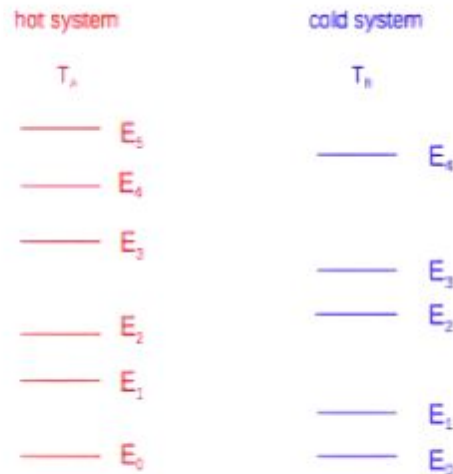
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Toy Model of Molecular Heat Engine

quantum systems as hot and cold reservoirs

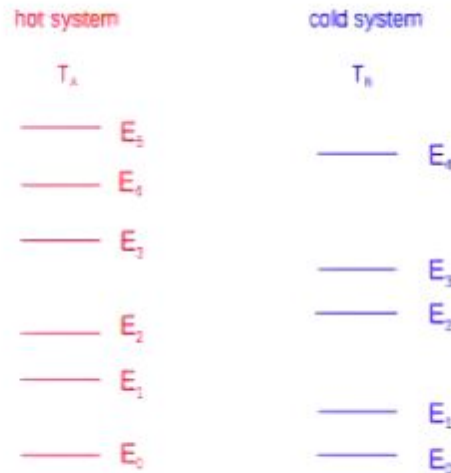


- ▶ heat engine: unitary that extracts work $\Delta W > 0$
(not always possible: joint system can be passive, but never completely passive)
- ▶ optimal heat engine: unitary that maximizes ΔW

(many quantum heat engines have been constructed meanwhile, I focus on the relation to computation)

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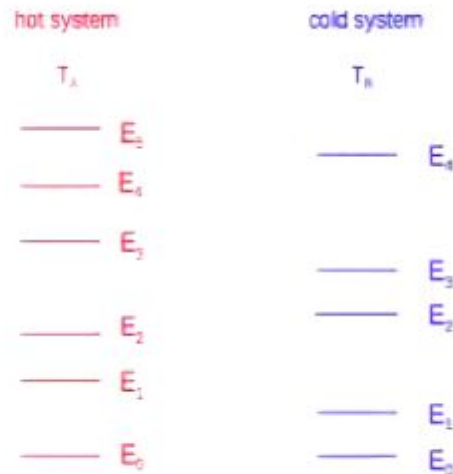


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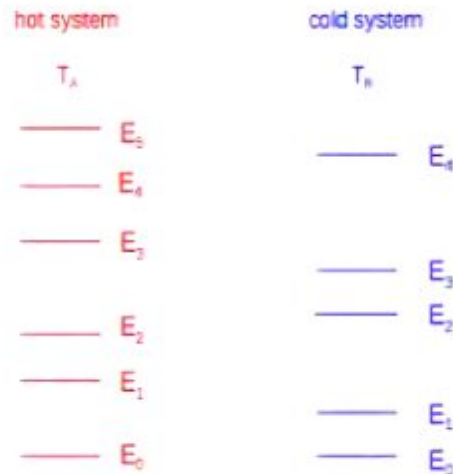


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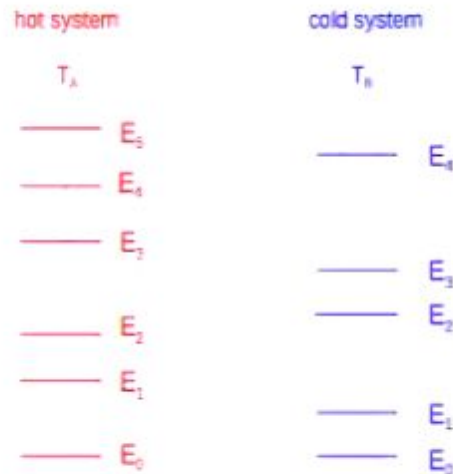


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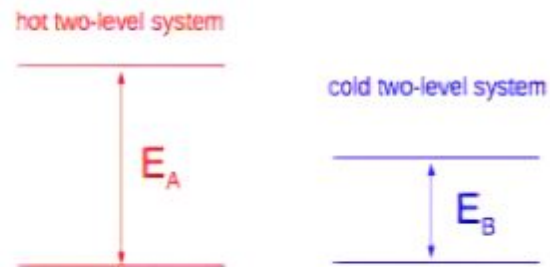


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Simplest Heat Engine: SWAP gate

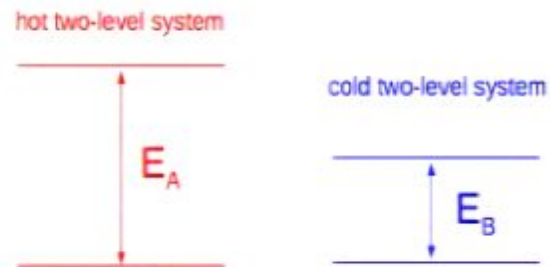
Two 2-level systems with energy gaps $E_B < E_A < E_B \frac{T_A}{T_B}$.



- ▶ $|10\rangle$ more likely than $|01\rangle$ although its energy is larger.
- ▶ $|01\rangle \leftrightarrow |10\rangle$ releases the energy $(p_{10} - p_{01})(E_A - E_B)$.
- ▶ **SWAP gate is the unique optimal heat engine.**

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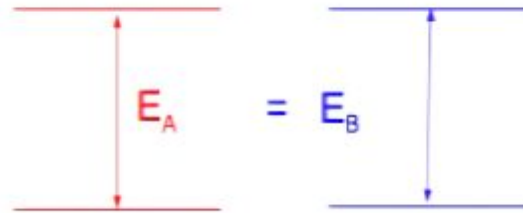
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This already shows...

whether or not a heat engine exists depends on

- ▶ the energy level structure of both systems
- ▶ their temperature ratio

Two qubits with equal energy gap



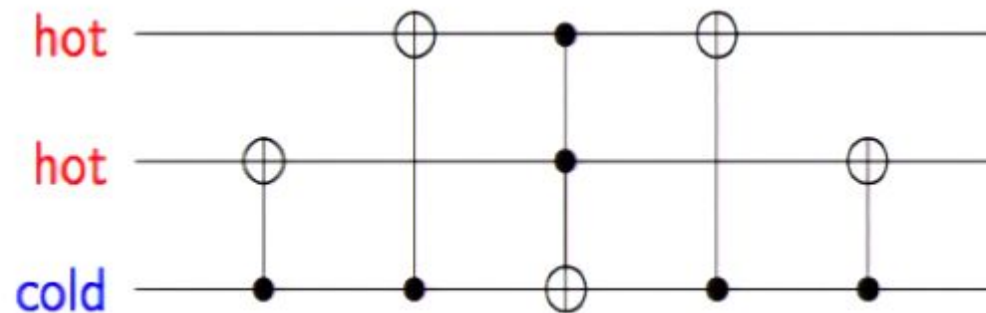
- ▶ no energy release – SWAP permutes states with equal energy
- ▶ state of the joint system is passive
- ▶ not completely passive because the joint state is not Gibbs (\Rightarrow many copies of this pair allow for a heat engine)

Qubits with equal energy gaps
require more complex gates

Simplest example:

- ▶ 2 hot and 1 cold qubit with $T_A > 2T_B$.
- ▶ $|110\rangle$ more likely than $|001\rangle$ although its energy is larger.
- ▶ Permutation $|001\rangle \leftrightarrow |110\rangle$ releases some energy.

Implemented by



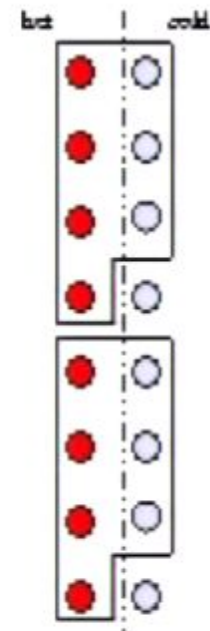
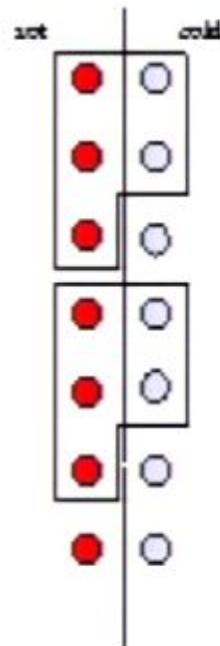
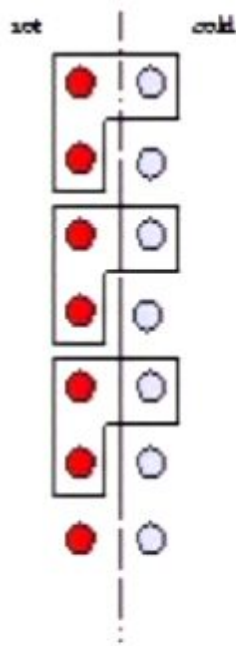
(Note: for $T_A = T_B$ the circuit is a refrigerator)

Small temperature differences require multi-qubit gates

(Janzing, 2006)

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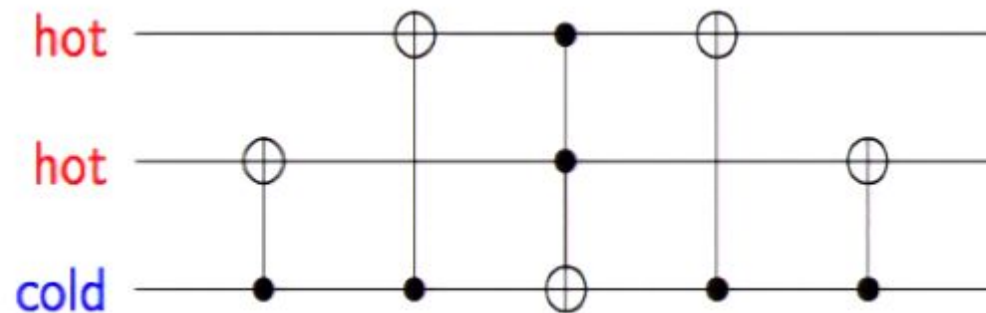
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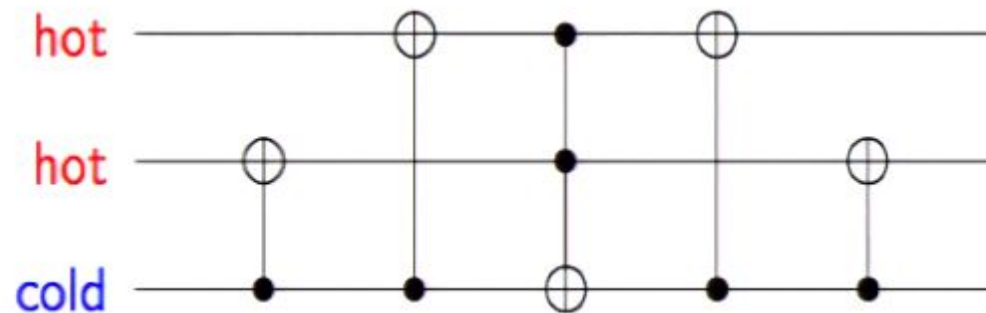


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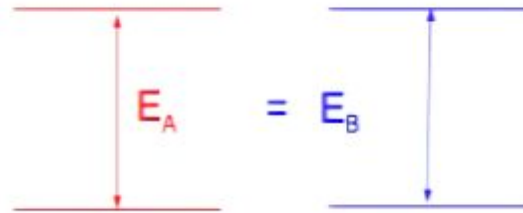
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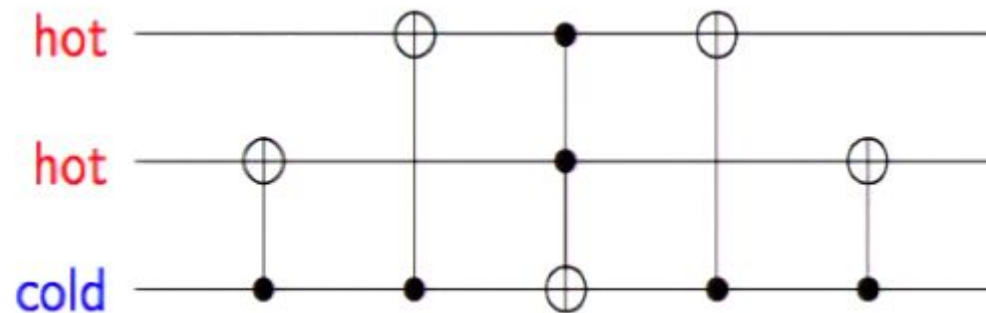


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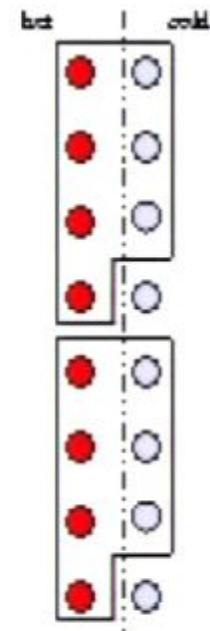
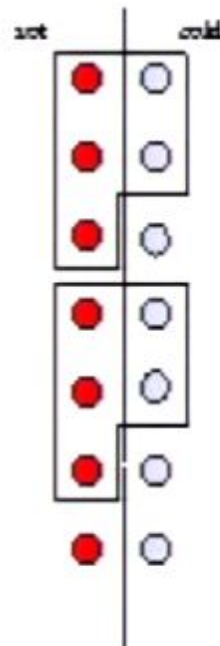
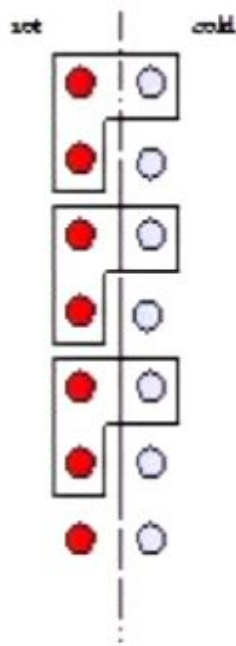
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Small temperature differences require multi-qubit gates

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Heat engine with n_A hot and n_B cold qubits with $E_A = E_B$ exists iff

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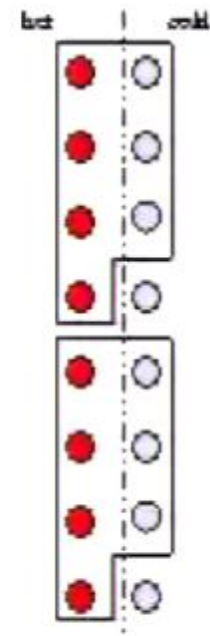
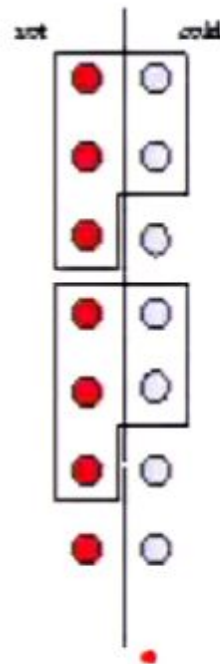
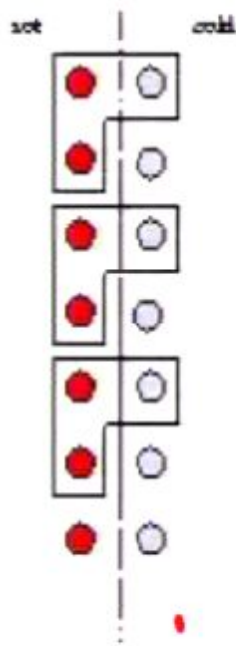
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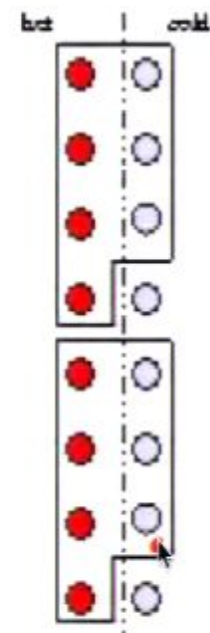
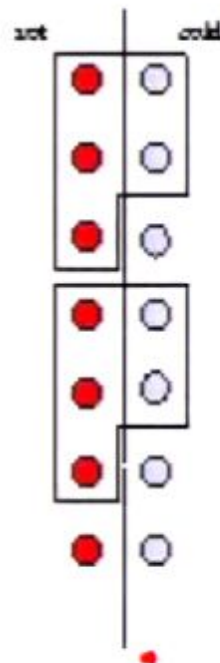
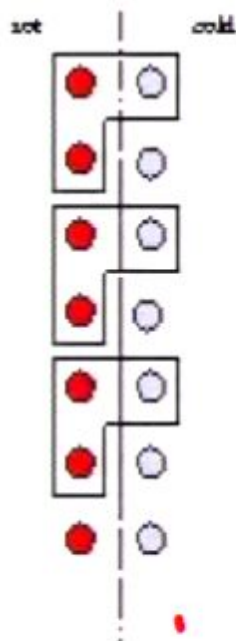


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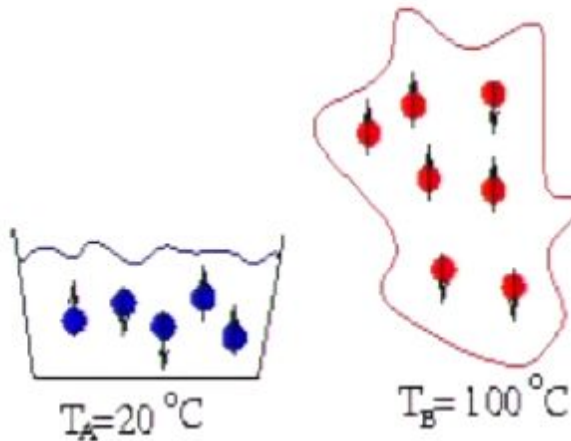
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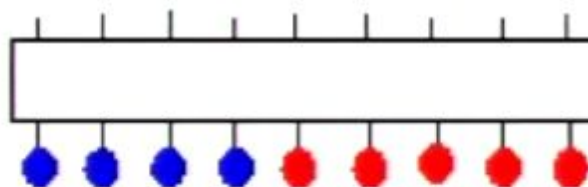
Example:

$T_A := 100^\circ C$ (vapor)

$T_B := 20^\circ C$ (room temperature)



requires 9-Qubit gate (4 cold and 5 hot qubits).



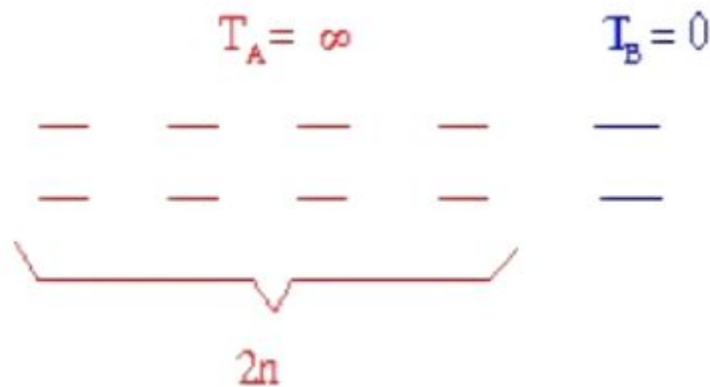
Heat engines are non-trivial multi-qubit gates –

what's their computational power?

(SWAP is not too exciting as a logical gate)

Heat Engine whose inverse computes

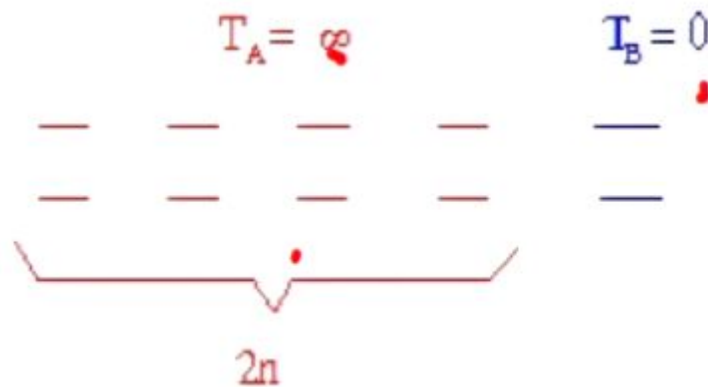
Given $2n$ hot and 1 cold qubits:



- ▶ optimal U maps all words with suffix 0 to words with Hamming weight at most n
- ▶ Hence, U^{-1} computes MAJORITY
- ▶ logical depth of U is at least $\log_2(2n + 1)$.

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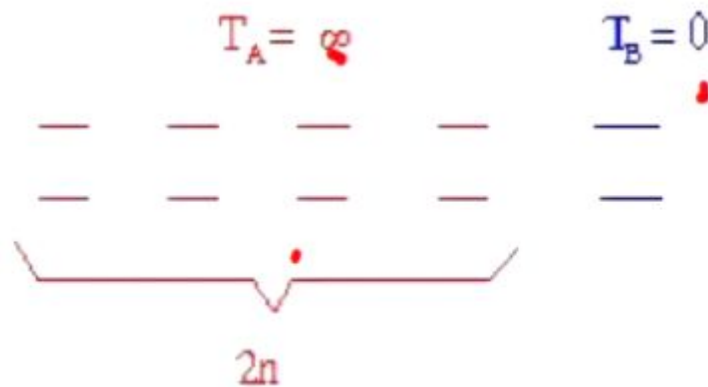
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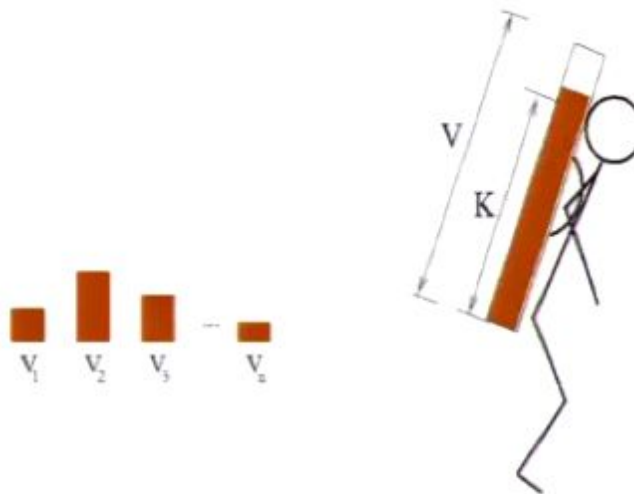
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which hard computational problems
can be solved by heat engines?

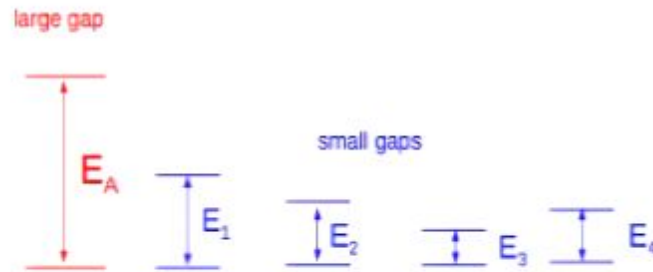
NP-complete problem KNAPSACK

Special instance (still NP-hard, see Papadimitrou: Computational Complexity):

- ▶ Given n items with volumes V_1, \dots, V_n .
- ▶ Given a knapsack with volume V
- ▶ Given some $K \leq V$, is there a subset of items whose total volume is at least K ?



Optimal heat engine solves KNAPSACK



- ▶ volume of knapsack = E_A , volumes of items = E_1, \dots, E_n
- ▶ $K = \frac{T_B}{T_A} E_A$
- ▶ heat engine exists iff there is a subset S such that

$$E_A > \sum_{j \in S} E_j > K.$$

- ▶ if the answer is “yes”, the heat engine finds the proof:

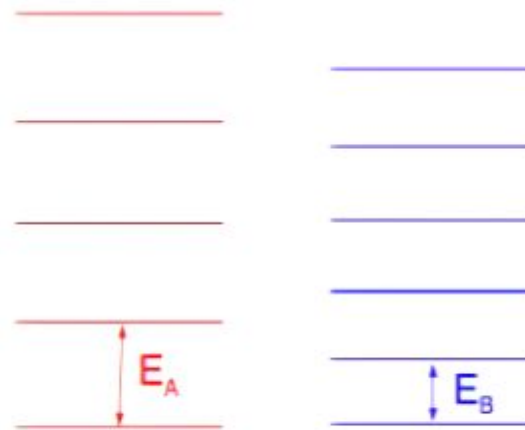
$$U(|1\rangle \otimes |0 \cdots 0\rangle) \in |0\rangle \otimes \mathcal{H}_S,$$

where \mathcal{H}_S is the space of solutions.

admittedly, this example was a bit artificial

let's go back to more physical systems...

Optimal heat engine on two harmonic oscillators



Assume

$$\frac{E_A}{E_B} \notin \mathbb{Q} \quad \text{and} \quad \frac{T_A/E_A}{T_B/E_B} \notin \mathbb{Q}$$

such that joint density operator and joint Hamiltonian are non-degenerate.

\Rightarrow Optimal heat engine is a unique permutation of basis states, i.e., permutation on $\mathbb{N}_0 \times \mathbb{N}_0$

Optimal heat engine on two harmonic oscillators



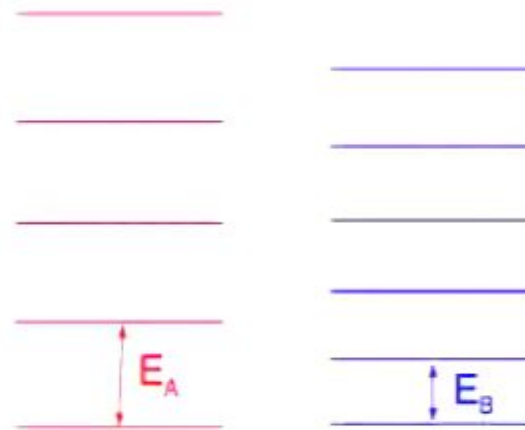
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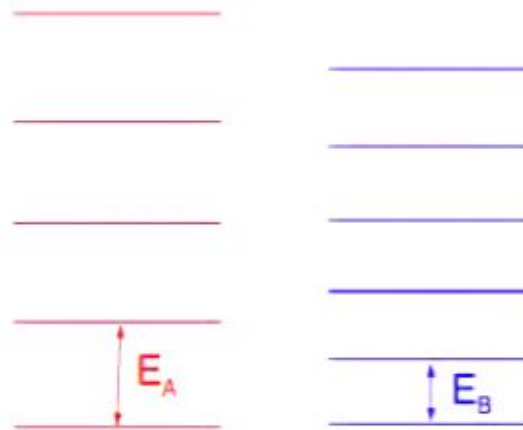
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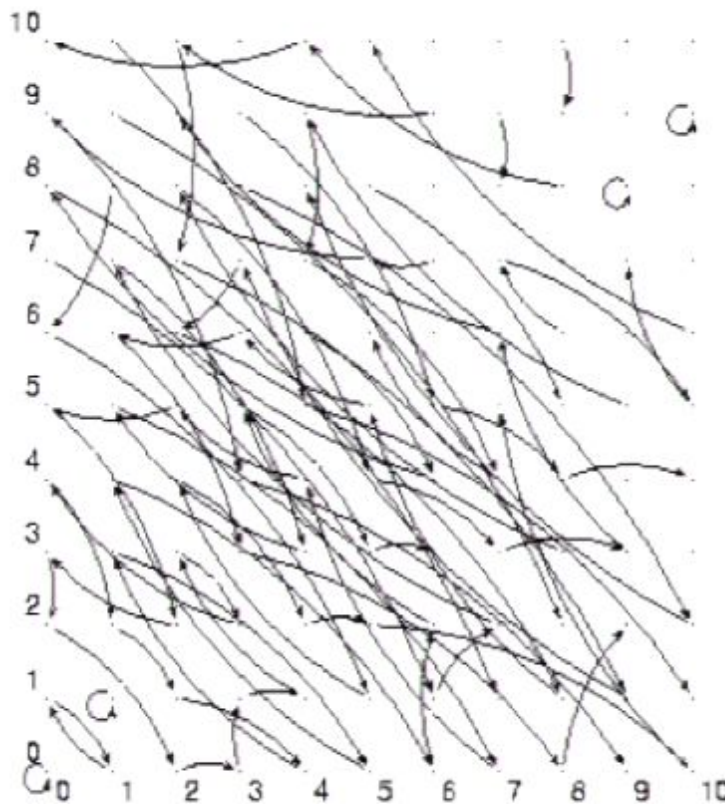
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$$\frac{E_A}{E_B} \notin \mathbb{Q} \quad \text{and} \quad \frac{T_A/E_A}{T_B/E_B} \notin \mathbb{Q}$$

such that joint density operator and joint Hamiltonian are non-degenerate.

⇒ Optimal heat engine is a unique permutation of basis states, i.e., permutation on $\mathbb{N}_0 \times \mathbb{N}_0$

Example with $E_A/E_B = \sqrt{2}$ and $T_B/T_A = E_B/(\sqrt{3}E_A)$



Example with $E_A/E_B = \sqrt{2}$ and T_B/I_B



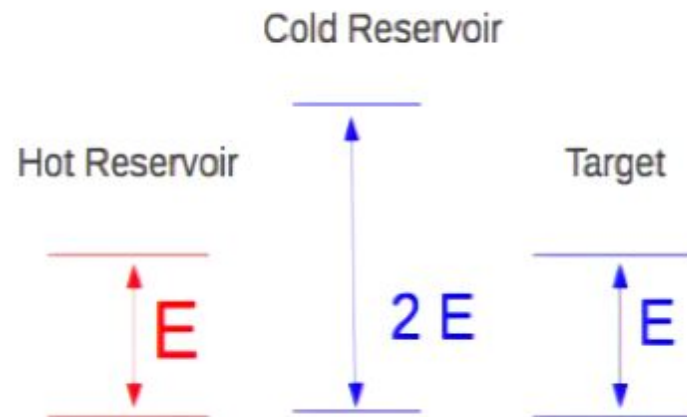
Our toy heat engines show that...

- ▶ having different temperatures is necessary for work extraction (derive and “feel” second law from first principles)
- ▶ ...but not sufficient (additional constraints for micro-systems)
- ▶ more thermodynamic laws for micro-physics yet to be discovered

What was missing...

how can we use the extracted energy?

Using heat for cooling: “quantum absorption heat pump”



- ▶ permutation $|010\rangle \leftrightarrow |101\rangle$ reduces temperature of system 3
- ▶ implementation does not require energy because it acts on a degenerate subspace
- ▶ temperature difference between hot and cold reservoir is used for cooling the target

this inspired a new framework
for thermodynamics at the nanoscale...

Idea (Janzing et al, 2000)

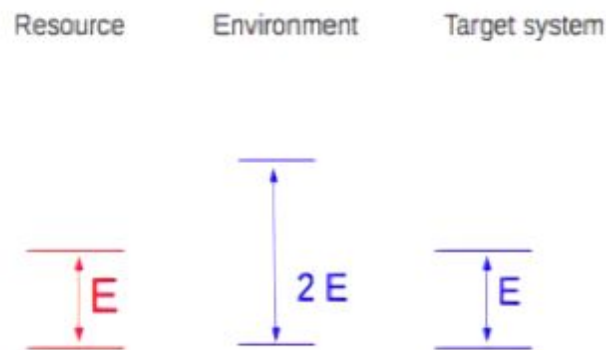
- ▶ Assume the whole world is full of quantum systems with uniform temperature T
- ▶ except for a few systems called “resources”
(resources can be hotter or colder or be in a non-Gibbs state)
- ▶ “processes” can transform resources into other resources
- ▶ processes are unitaries that commute with the uncoupled joint Hamiltonian

Idea (Janzing et al, 2000)

is full
ture
system called source

- "processes" can transform resources into other resources
- proc
- hamiltonian

Quasi-order of resources

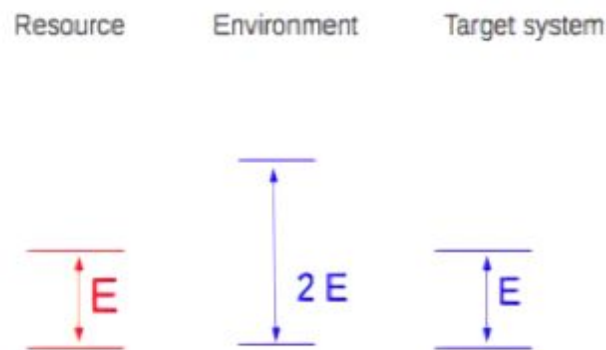


- ▶ process $|010\rangle \leftrightarrow |101\rangle$ generates a cold resource from a hot one
- ▶ perfect cooling of a single qubit requires infinitely many copies of the resource
- ▶ perfect erasure of a qubit requires infinitely large resources (“Tightening Landauer’s principle”, Janzing et al 2000)

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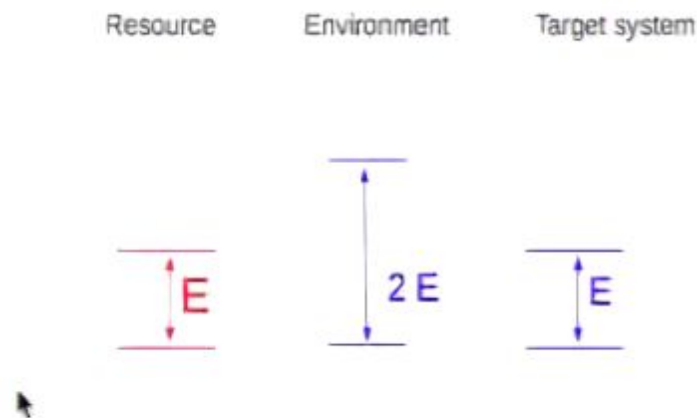
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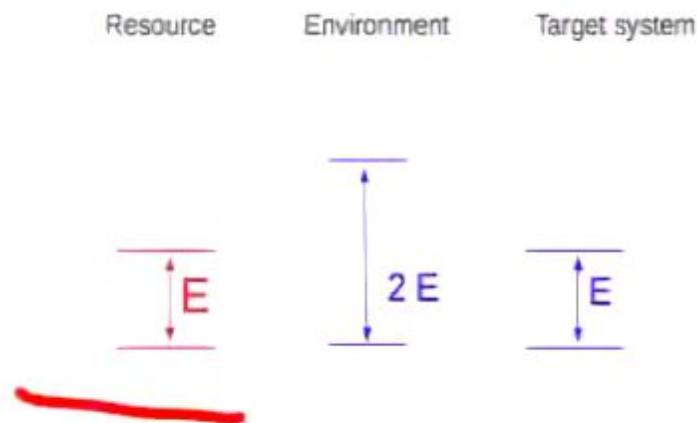
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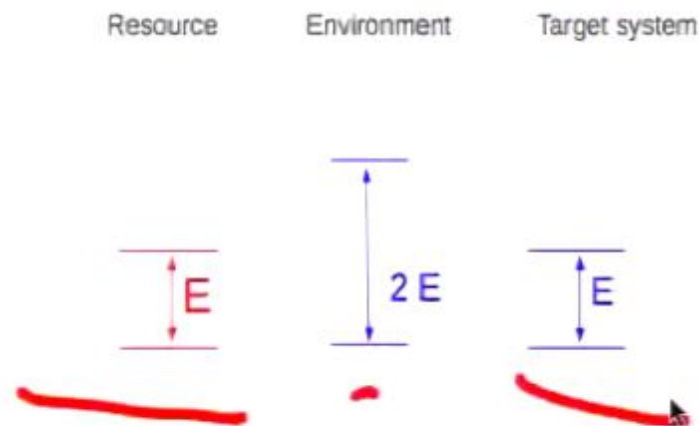
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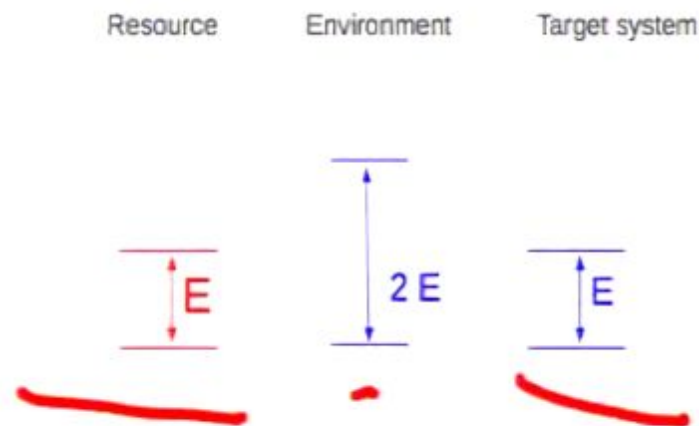
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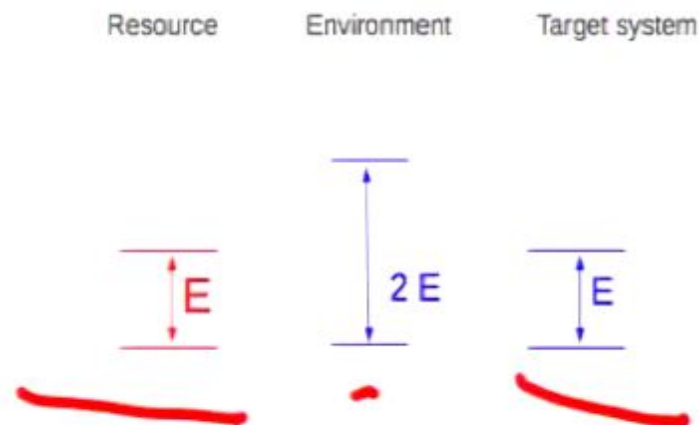
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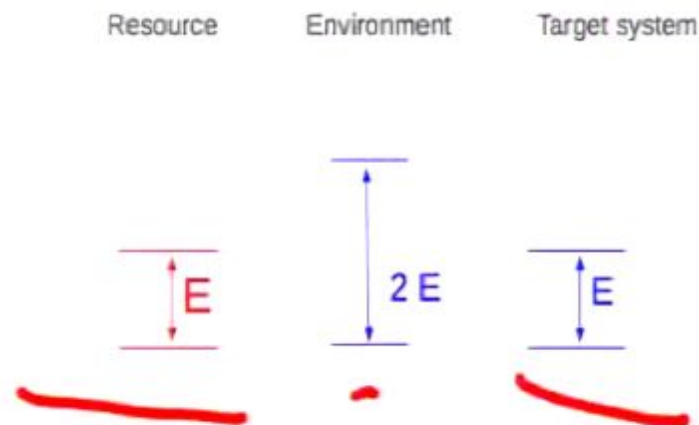
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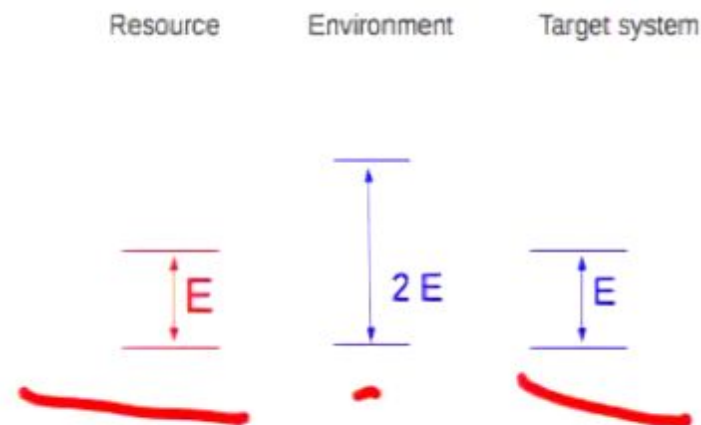
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Quasi-order formally

- ▶ Characterize every system by a pair

$$(\rho, H) = (\text{density matrix, Hamiltonian}).$$

- ▶ Define an order relation:

$$(\rho, H) \geq (\tilde{\rho}, \tilde{H})$$

if there is an environment $(\hat{\gamma}, \hat{H})$ and a unitary U with $[U, H + \hat{H} + \tilde{H}] = 0$ such that

$$\text{tr}_{12} \left(U(\rho \otimes \hat{\gamma} \otimes \tilde{\gamma}) U^\dagger \right) = \tilde{\rho}.$$

- ▶ restricted to diagonal states, quasi-order given by a majorization-like criterion

Are these processes thermodynamically reversible?

- ▶ formally, every unitary U has an inverse U^{-1}
- ▶ but the macroscopic controller implementing U or U^{-1} may consume a lot of work
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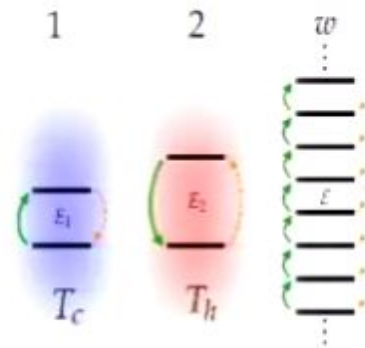
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Closed system as heat engine (Linden, Popescu, and Skrzypczyk, 2010)



- ▶ consists of 2 qubits and a multi-level system (“weight to be lifted”)
- ▶ qubits are coupled to hot and cold reservoirs
- ▶ tripartite interaction Hamiltonian H lifts the weight without external energy source or control operation

Sounds pretty much “reversible” because $-H$ implements the reverse process...

...but who turns H into $-H$?

actually, we cannot change Hamiltonians,
they are just given and we can only
change *effective* Hamiltonians
by changing the states of matter...

Limits of thermodynamic reversibility?

these questions remain:

- ▶ are there thermodynamic costs of switching from one effective Hamiltonian to another?
- ▶ all systems are coupled to the environment, are there thermodynamic costs of effectively isolating them?

I don't have an answer, but I propose the following approach to address these questions...

Hamiltonian or cellular automaton model of quantum control

(informal version, for details see arXiv:1009.1720)

- ▶ the world W consists of a lattice \mathbb{Z}^d of cells, each cell containing a quantum system
- ▶ dynamics of the world: group (α_t)
($t \in \mathbb{Z}$: cellular automaton, $t \in \mathbb{R}$: translation invariant finite range Hamiltonian)
- ▶ consider a region R consisting of finitely many cells
- ▶ depending on the state of $W \setminus R$, the dynamics implements a family of completely positive maps G_t on R

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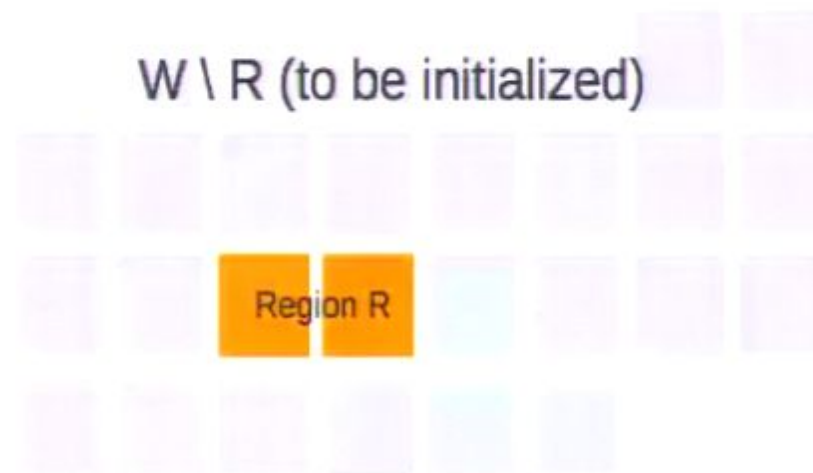
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Physically universal CAs or Hamiltonians

dynamics α_t is called physically universal if

the set of all G_t generated by all possible initializations of $W \setminus R$ is dense in the set of completely positive trace preserving maps.

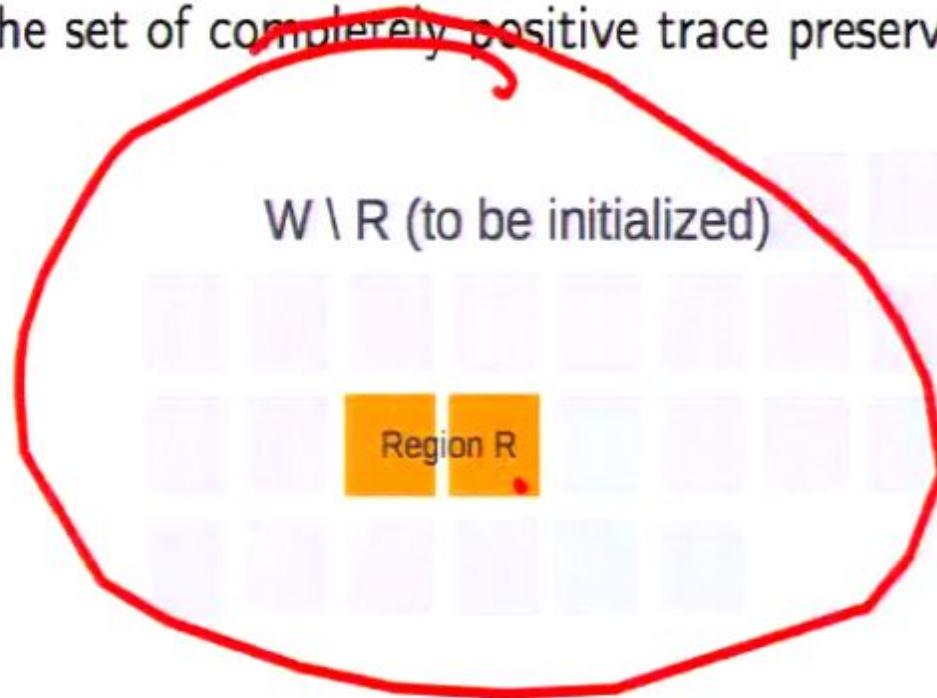


(do they exist? I don't know...)

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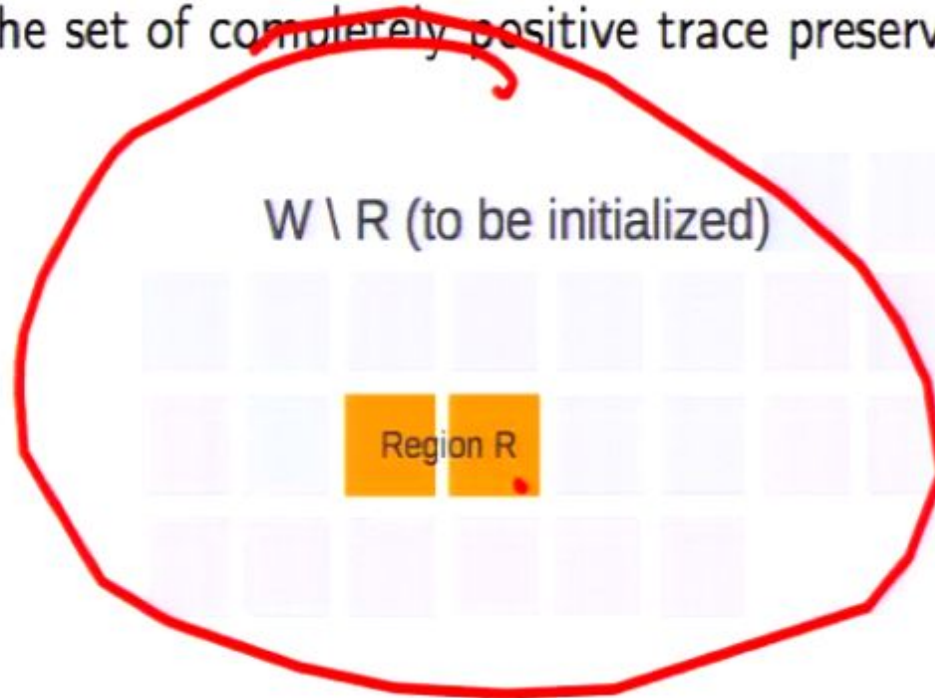
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- ▶ they consist of program and data space
- ▶ they can implement any desired transformation on the **data space**

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Physically universal CAs or Hamiltonians...

- ▶ provide a model of quantum control where the controller is not external
- ▶ boundary between system R and its controller $W \setminus R$ can be shifted (like Heisenberg cut for quantum measurement)
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I would be delighted if
someone could prove existence or non-existence
of physically universal Hamiltonians!

Both would have interesting implications
for the foundations of physics

Thanks for your attention!

Although this talk was only a game –
energy conversion at nanoscale
could become even more relevant in the future

I just read about recent progress
in storing solar energy via artificial photosynthesis
<http://www.sciencemag.org/content/321/5892/1072.abstract>