

Title: Small thermal machines

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Abstract: The second law of thermodynamics tells that physics imposes a fundamental constraint on the efficiency of all thermal machines. Here I will address the question of whether size imposes further constraints upon thermal machines, namely whether there is a minimum size below which no machine can run, and whether when they are small if they can still be efficient? I will present a simple model which shows that there is no size limitation and no limit on the efficiency of thermal machine and that this leads to a unified view of small refrigerators, pumps and engines.

Small thermal machines

Paul Skrzypczyk

Joint work with Sandu Popescu, Noah Linden, Nicolas Brunner



outline

outline

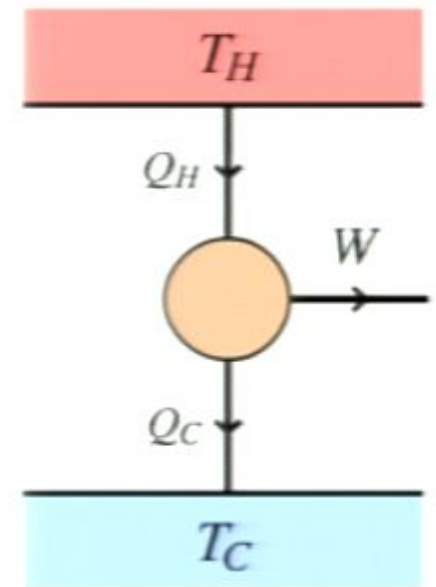
- introduction & motivation
- basic functioning principle
- refrigerator - model and efficiency



introduction

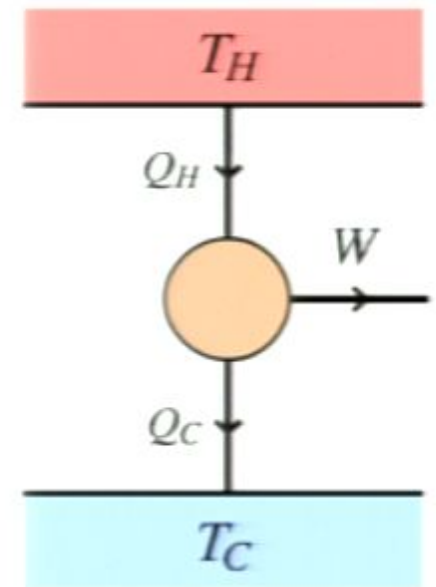
introduction

- Sadi Carnot: abstract, model independent machines



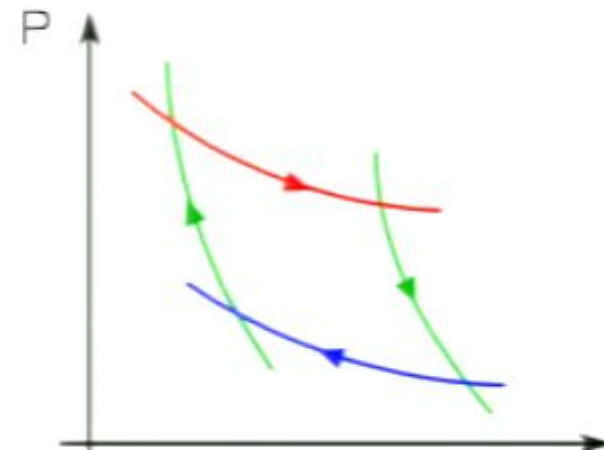
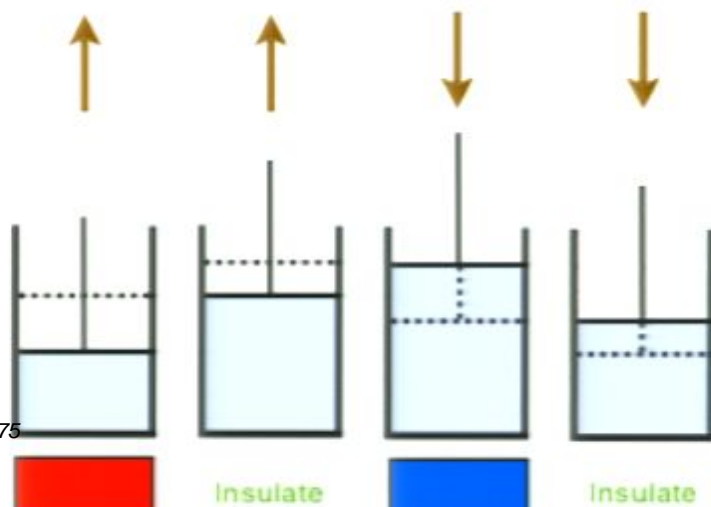
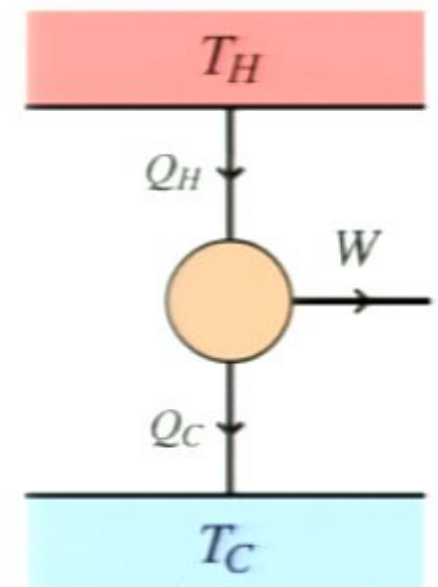
introduction

- Sadi Carnot: abstract, model independent machines
 - nature imposes a fundamental limit on efficiency of all thermal machines



introduction

- Sadi Carnot: abstract, model independent machines
 - nature imposes a fundamental limit on efficiency of all thermal machines
 - machines become maximally efficient when they become reversible





additional fundamental constraints

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- question: what happens if physical properties are specified?
specifically 'size'

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→ is there a minimum size below which no thermal machine can exist?

additional fundamental constraints

- question: what happens if physical properties are specified?
specifically 'size'
 - ➔ is there a minimum size below which no thermal machine can exist?
 - ➔ if a thermal machine is small does this constrain its performance? in particular its efficiency?

small and self-contained

- size: dimension of Hilbert space of machine

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important: '*self-contained*' - i.e. must ensure that
the counting is fair

→ no time dependent interactions: t external parameter

small and self-contained

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important: '*self-contained*' - i.e. must ensure that the counting is fair

→ no time dependent interactions: t external parameter

↳ macroscopic control

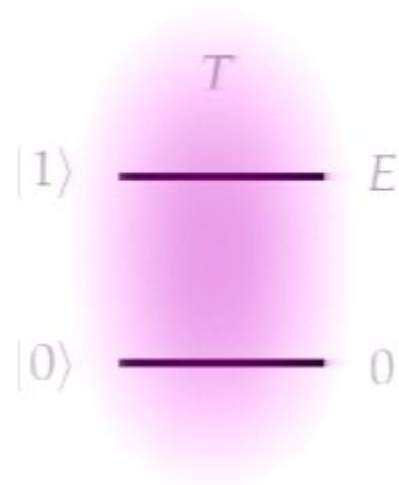
→ only external interaction: incoherent interaction with thermal baths

basic functioning principle I

how can you cool a qubit?

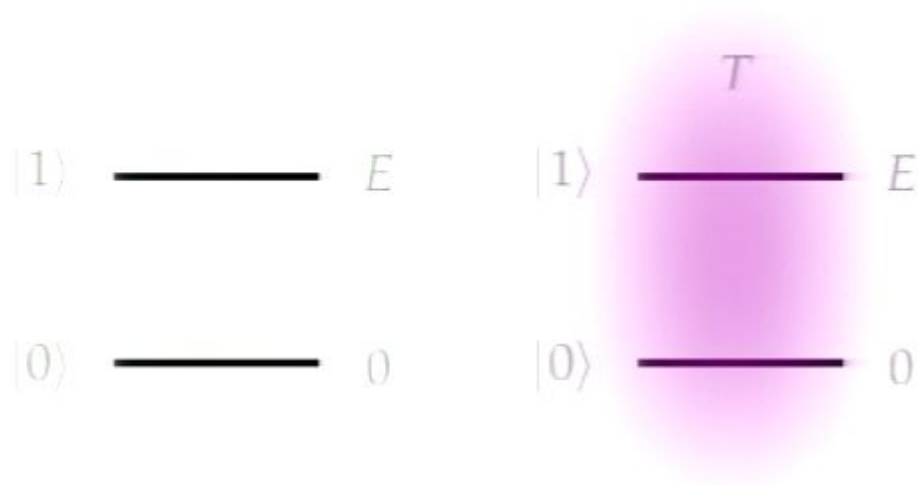
basic functioning principle I

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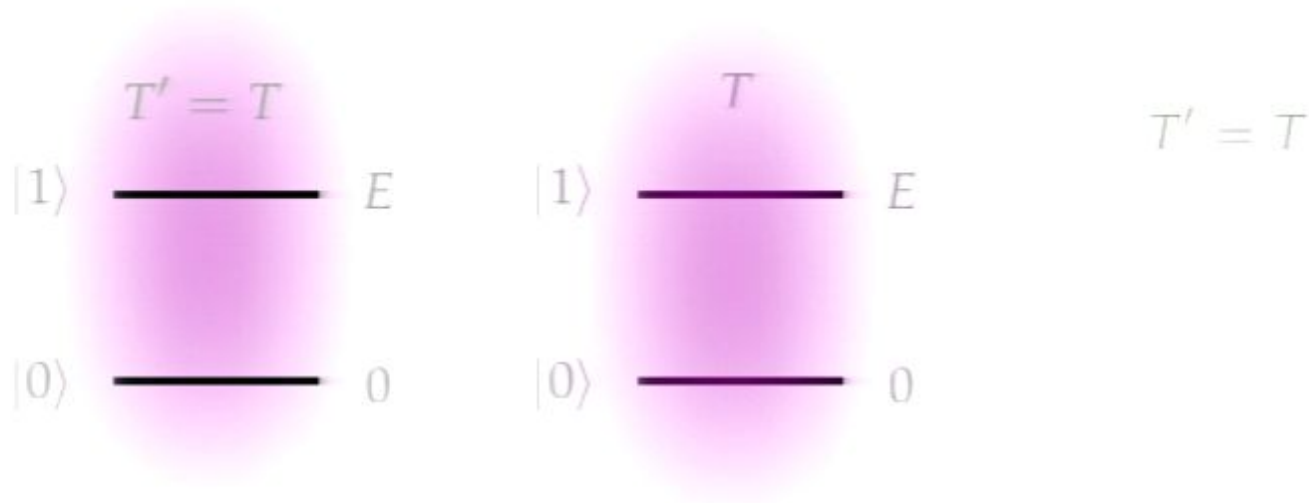


$$\rho = \frac{1}{Z} e^{-H_0/kT}$$

$$\rho = \frac{1}{1 + e^{-E/kT}} |0\rangle\langle 0| + \frac{e^{-E/kT}}{1 + e^{-E/kT}} |1\rangle\langle 1|$$

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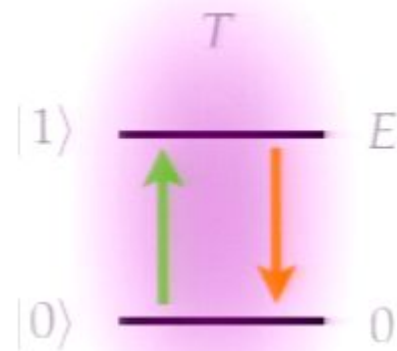
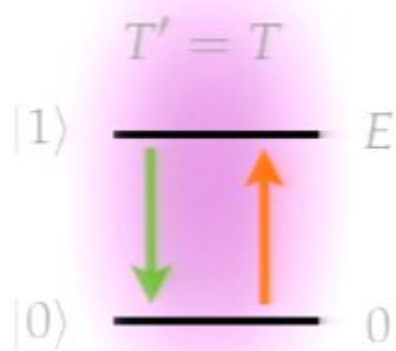


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$$T' = T$$

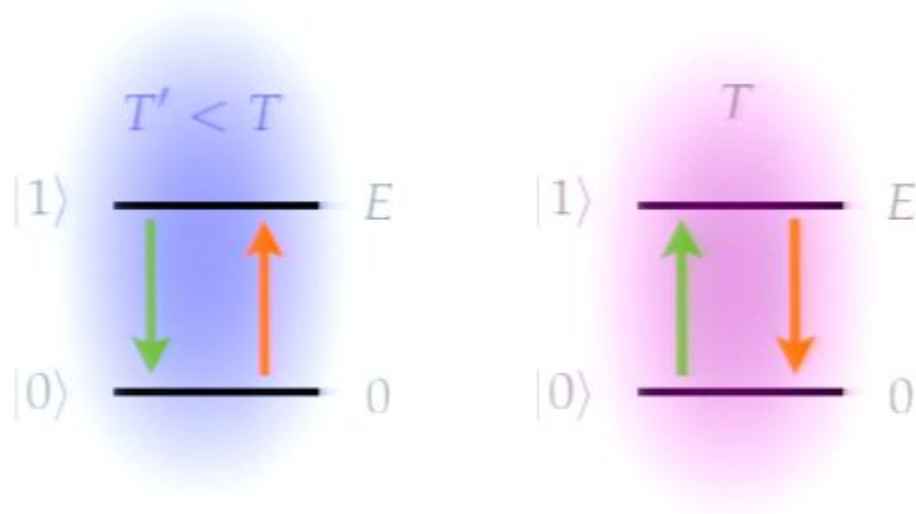
$p(10) = p(01)$
nothing happens

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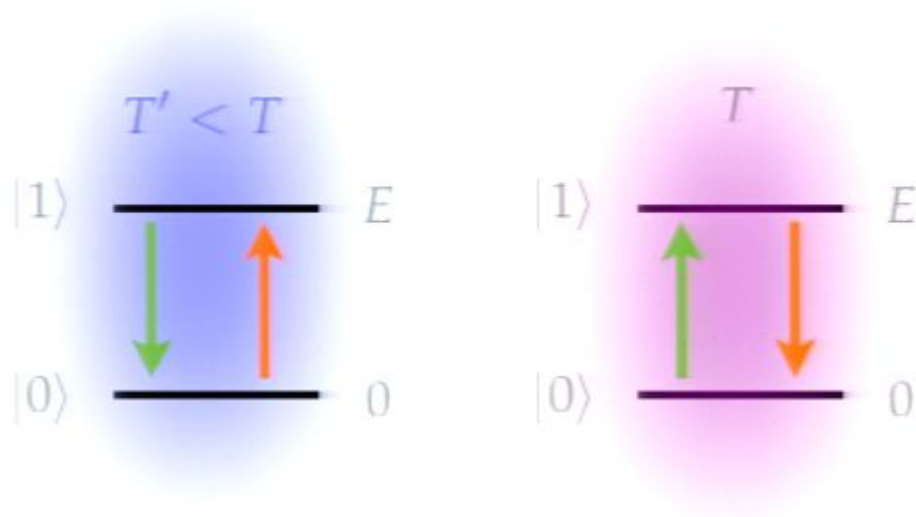
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cools down qubit

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basic functioning principle I

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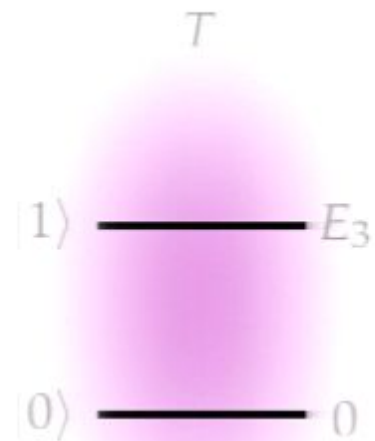
$$p = \frac{1}{Z} e^{-H_0/kT}$$

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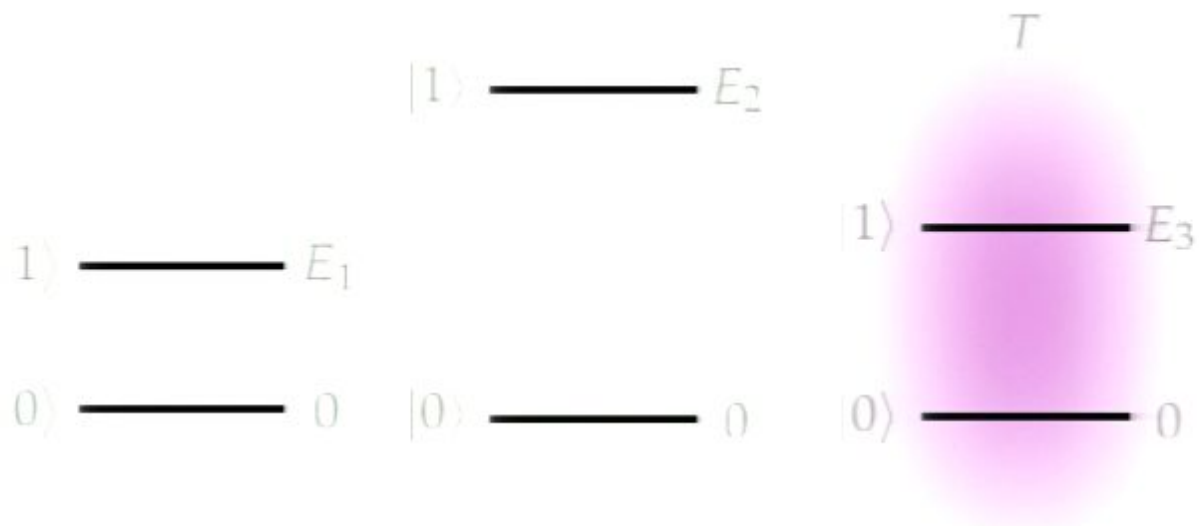
qubit 1 is like an **ice-cube**



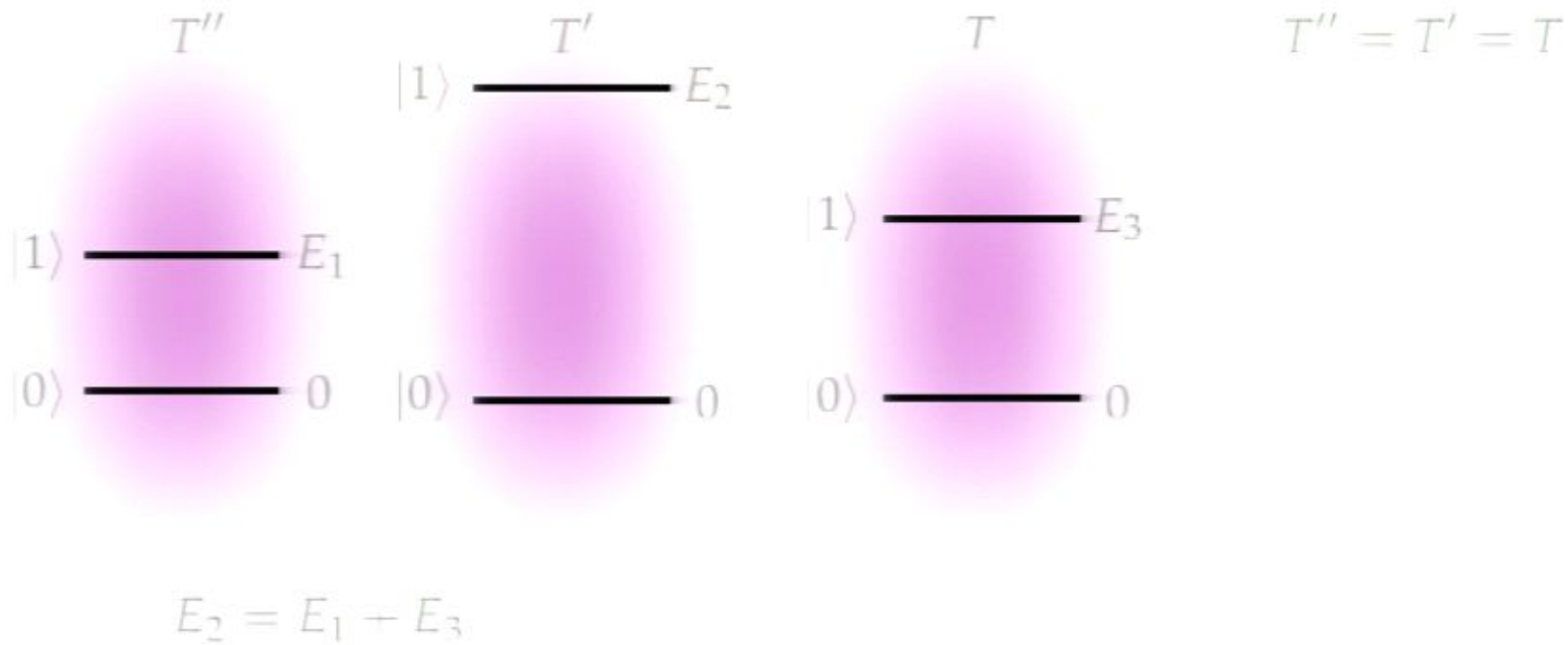
basic functioning principle II



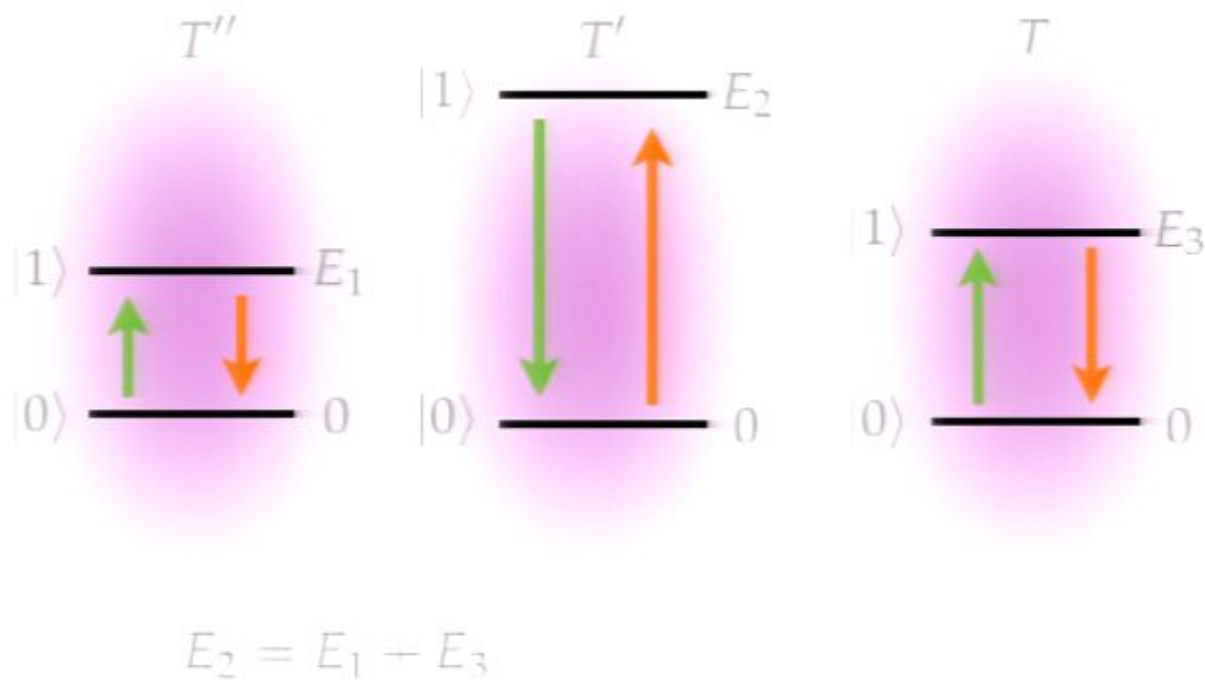
basic functioning principle II



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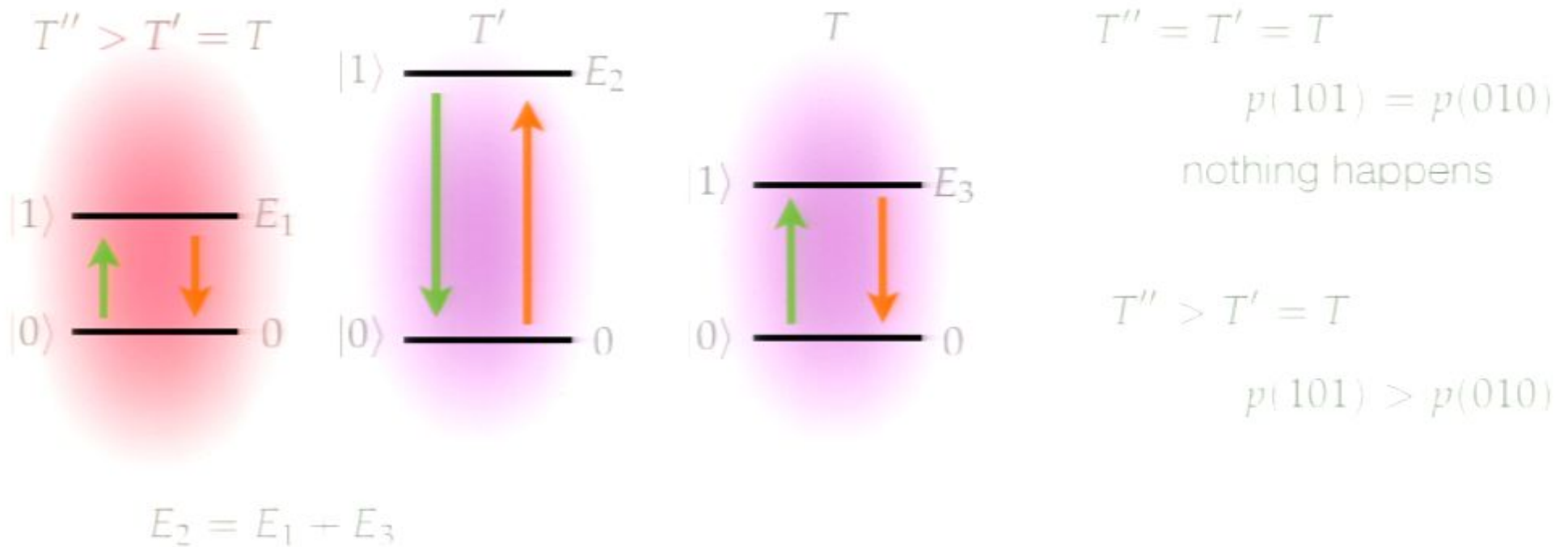


$$T'' = T' = T$$

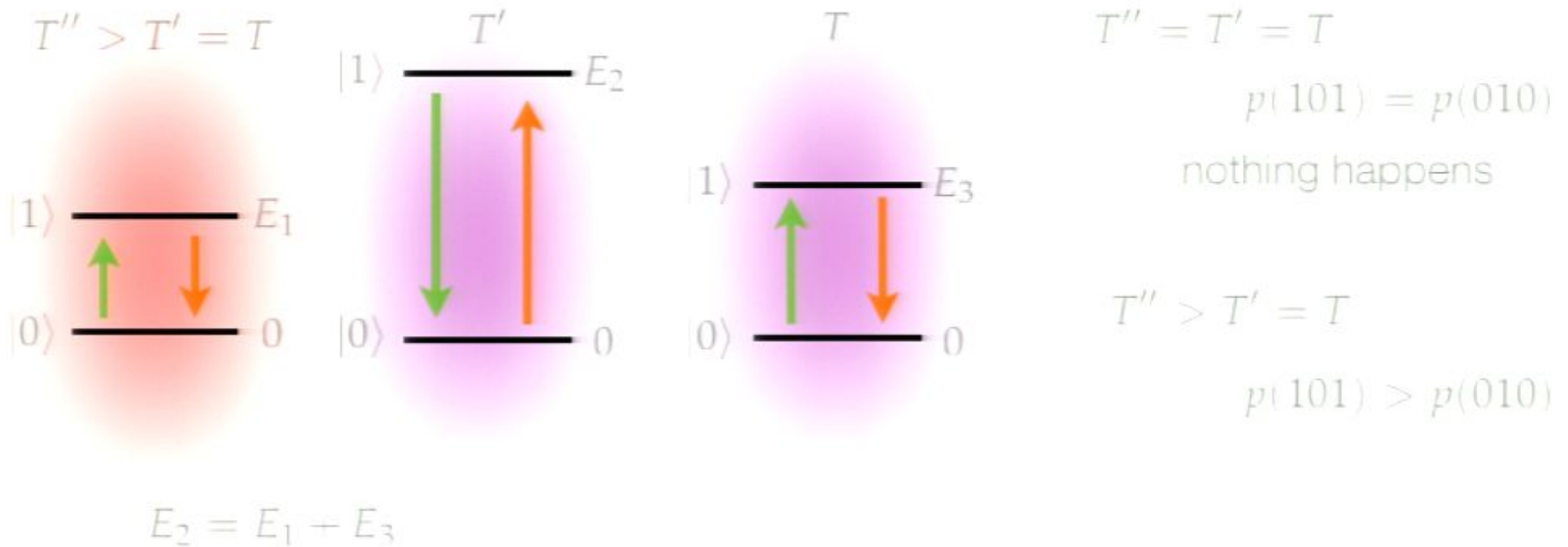
$$p(101) = p(010)$$

nothing happens

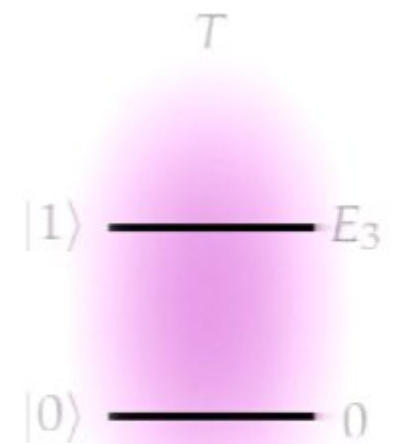
basic functioning principle II



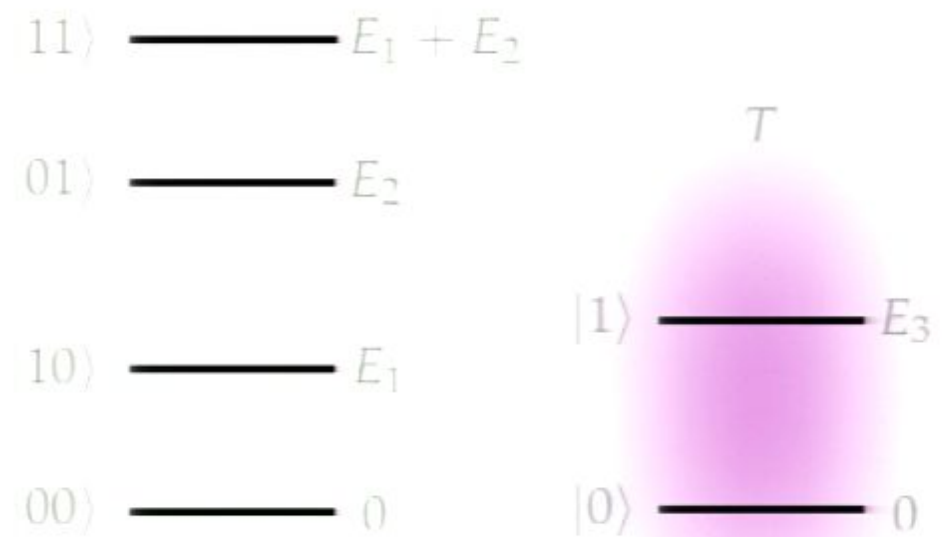
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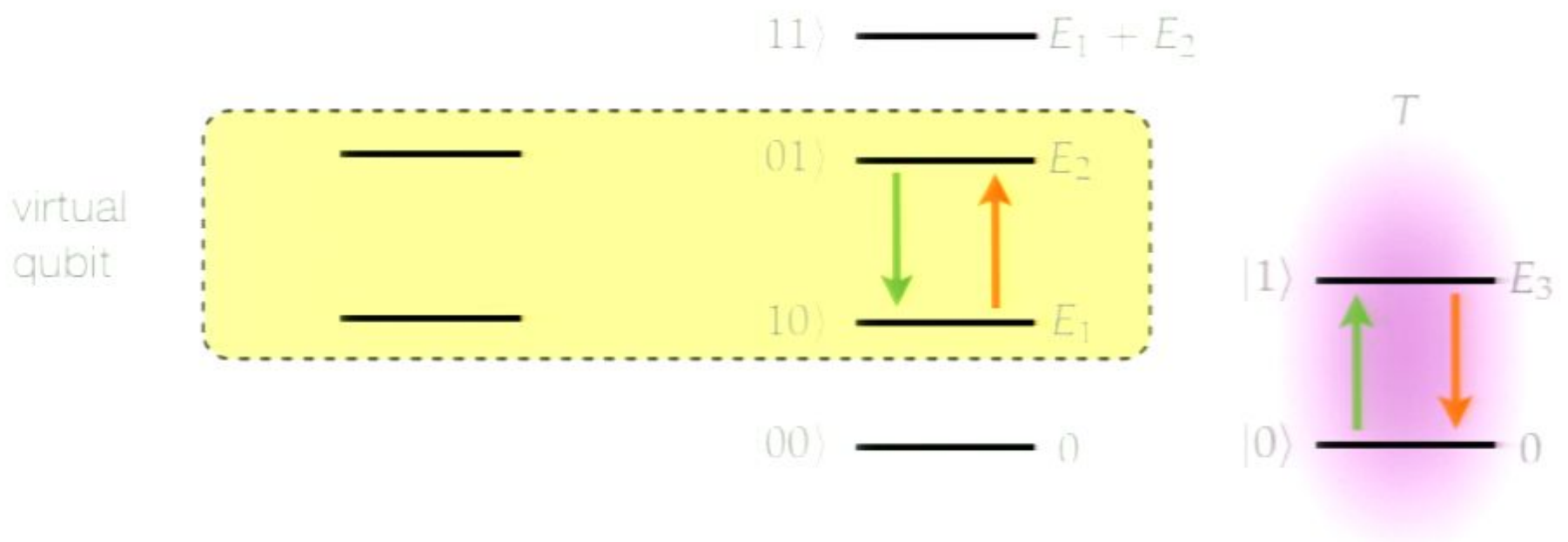
basic functioning principle III



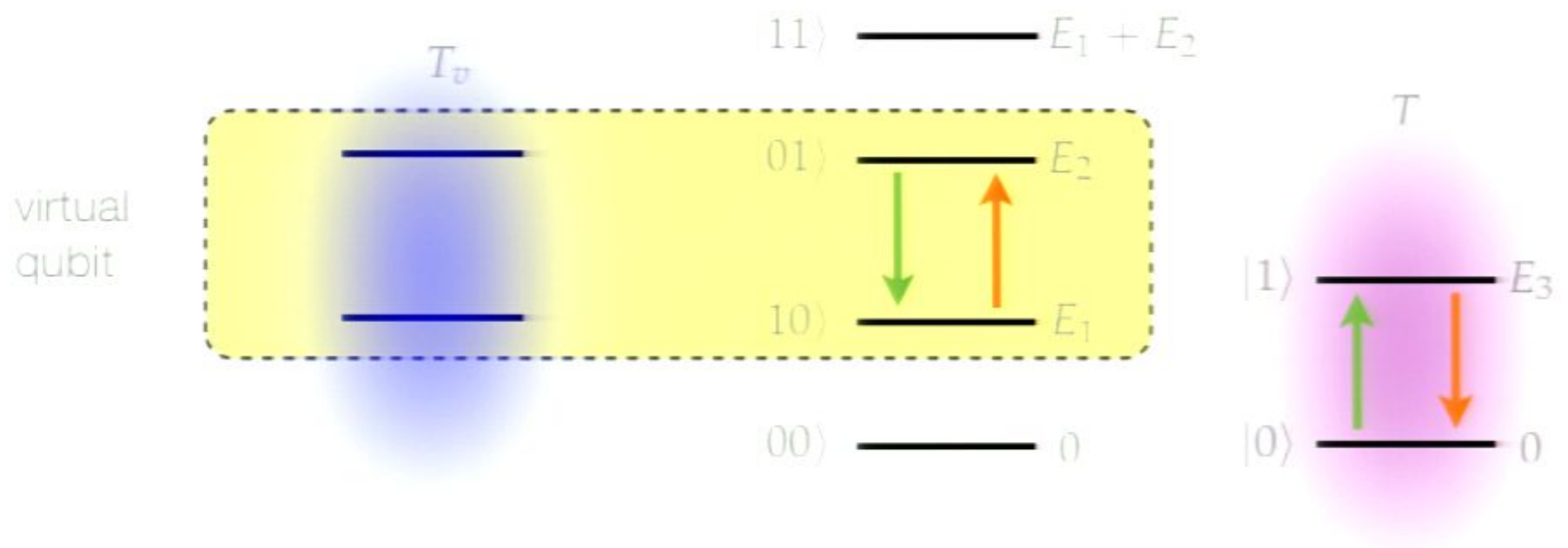
basic functioning principle III



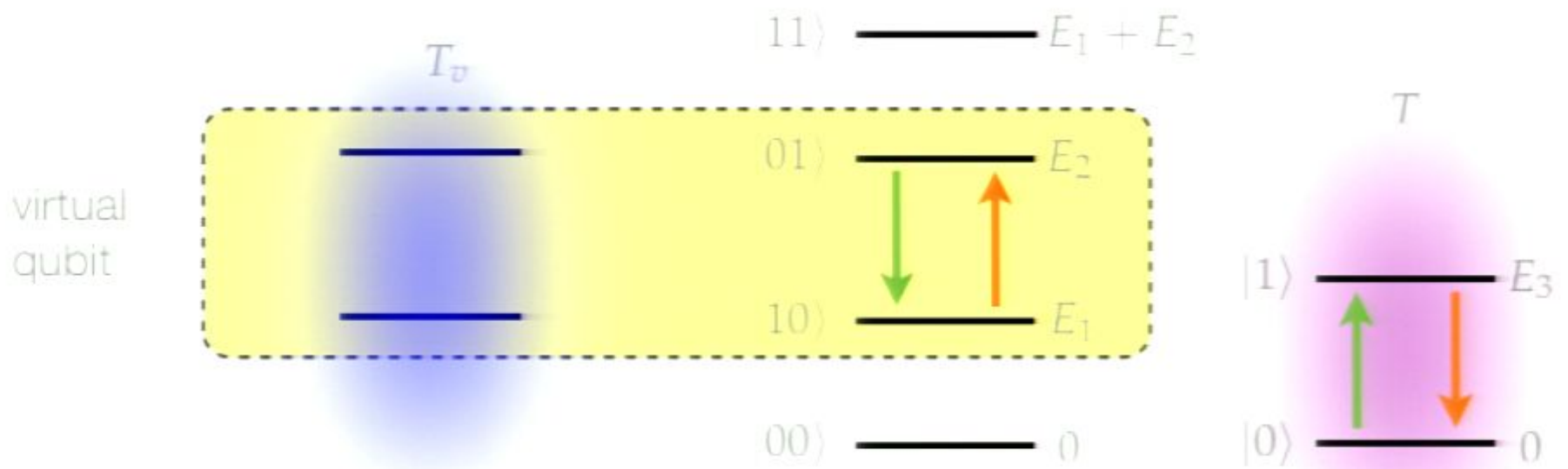
basic functioning principle III



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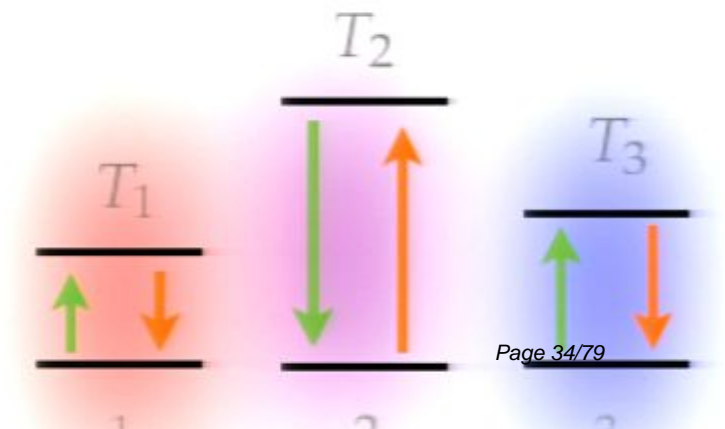
basic functioning principle III



$$\frac{p(01)}{p(10)} = \frac{e^{-E_2/T'}}{e^{-E_1/T''}} = e^{-E_3/T_v}$$

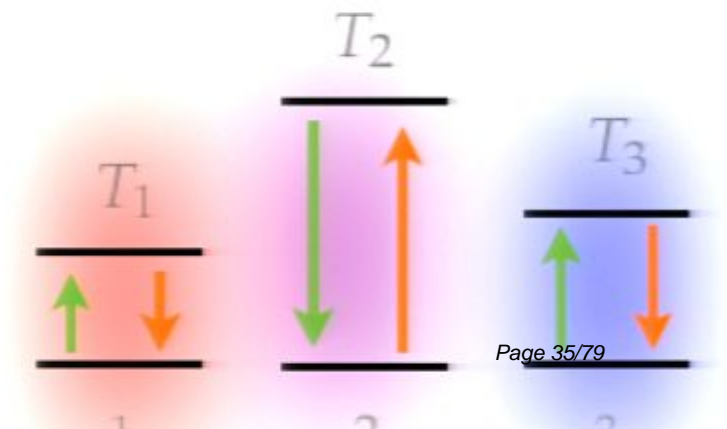
$$T_v = \frac{E_2 - E_1}{\frac{E_2}{T'} - \frac{E_1}{T''}}$$

model refrigerator



model refrigerator

- dissipation: prob. $p_i \delta t$ $\rho \mapsto \tau_i \otimes \text{tr}_i \rho$ $\tau_i = \frac{1}{Z_i} e^{-H_i/kT_i}$
 $1 - p_i \delta t$ $\rho \mapsto \rho$
- interaction Hamiltonian: $H_{int} = g (|010\rangle\langle 101| + |101\rangle\langle 010|)$

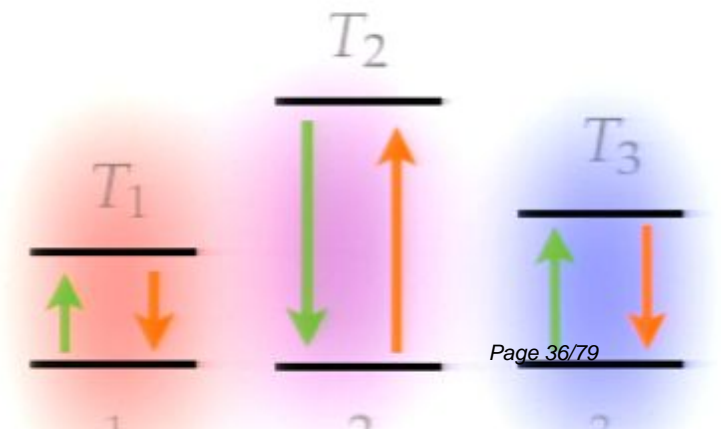


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→ Master equation:
$$\frac{\partial \rho}{\partial t} = -i[H_0 + H_{int}, \rho] + \sum_i p_i (\tau_i \otimes \text{tr}_i \rho - \rho)$$



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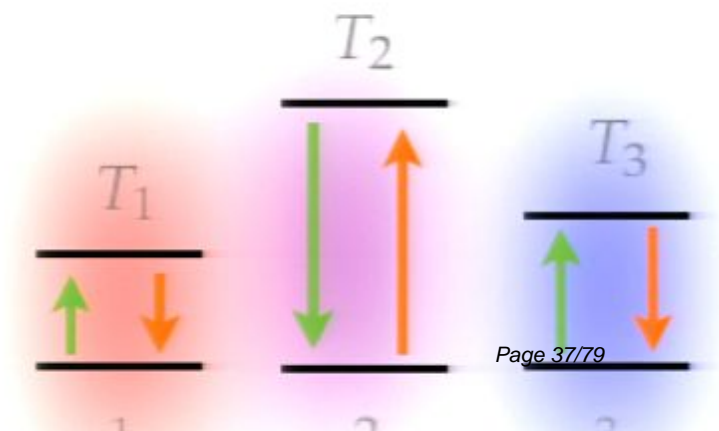
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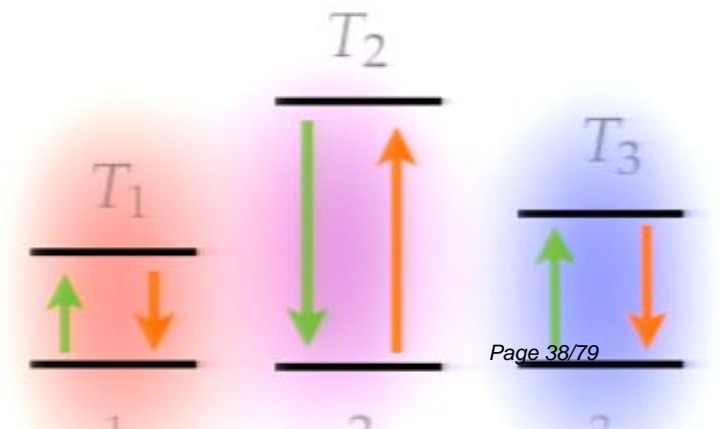
subtleties:

- T defined relative to H_0 not $H_0 + H_{int}$
 $\hookrightarrow g \ll E_i$

- no coupling between interaction and dissipation



refrigerator solution



model refrigerator

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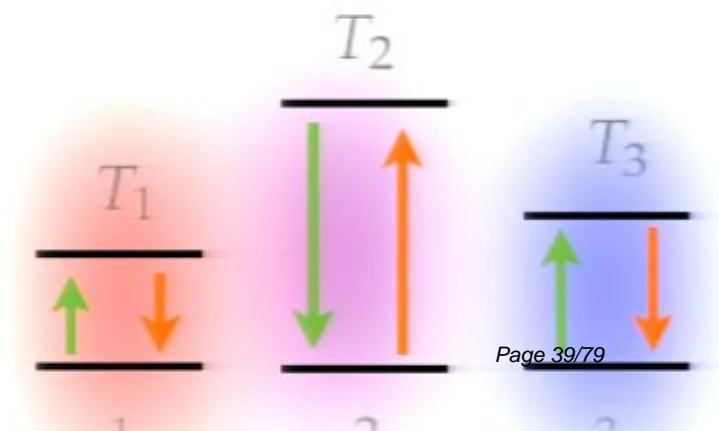
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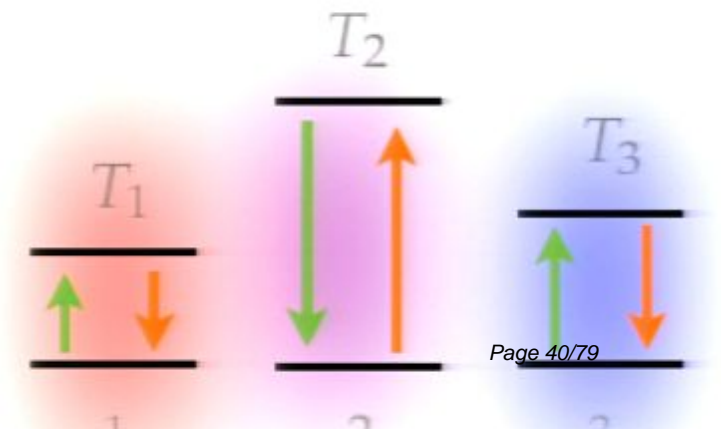
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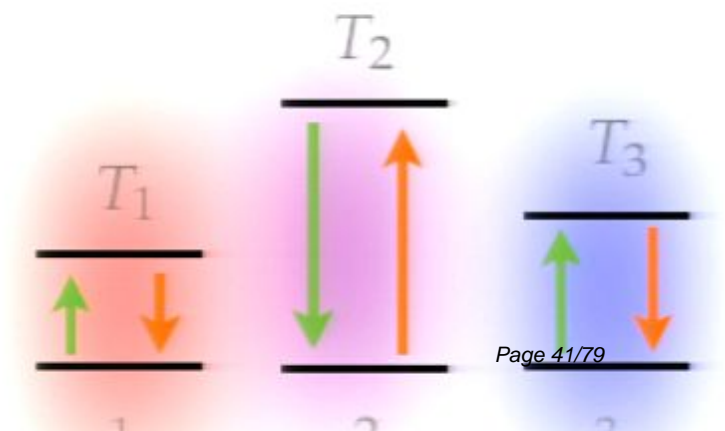
refrigerator solution



refrigerator solution

→ stationary solution to Master equation: $\frac{\partial \rho^S}{\partial t} = 0$

$$\hookrightarrow 0 = -i[H_0 + H_{int}, \rho^S] + \sum_i p_i (\tau_i \otimes \text{tr}_i \rho^S - \rho^S)$$



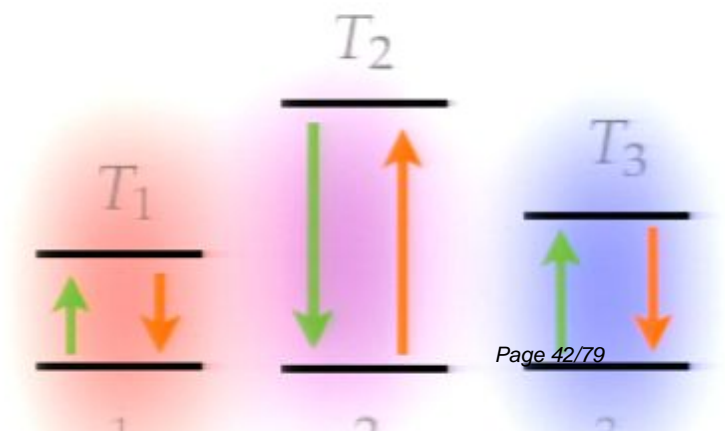
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such that $T(\rho_3^S) < T_3$

• analytic solution $\rho_3^S = \tau_3 + p_3 \gamma \sigma_z$ $\gamma(g, p_i, E_i, T_i)$



refrigerator solution

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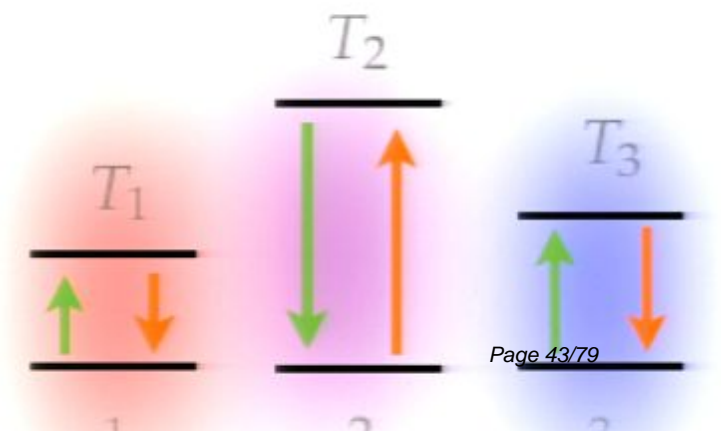
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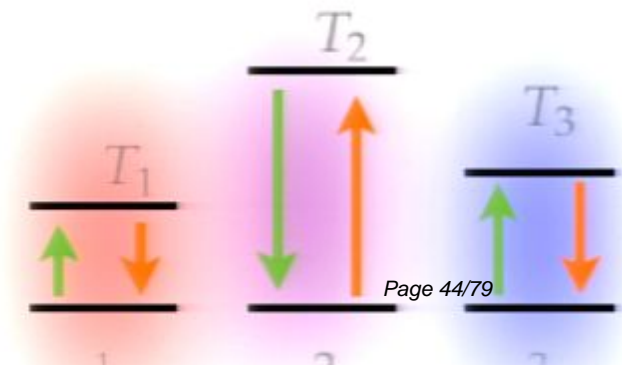
• analytic solution $\rho_3^S = \tau_3 + p_3 \gamma \sigma_z$ $\gamma(g, p_i, E_i, T_i)$

$$\gamma \propto p(101) - p(010) \propto e^{-E_3/T_v} - e^{-E_3/T_3}$$

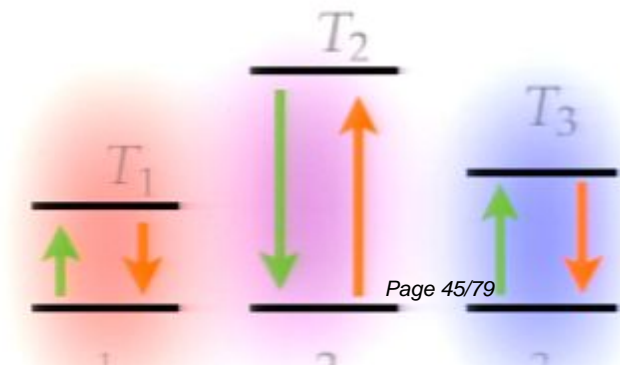
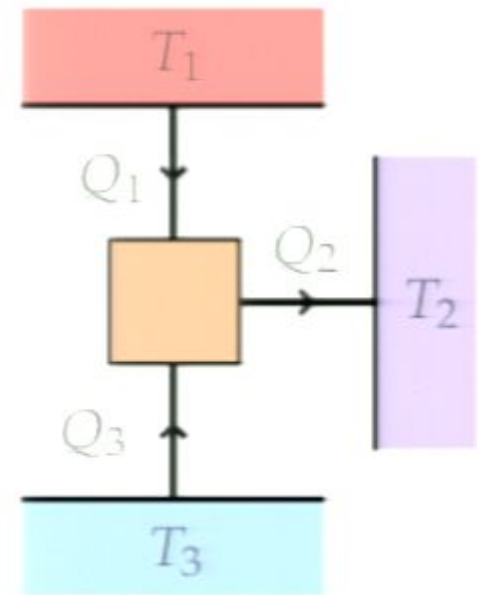
$$\hookrightarrow \boxed{T(\rho_3^S) < T_3 \quad \text{iff} \quad T_v < T_3}$$



efficiency



efficiency

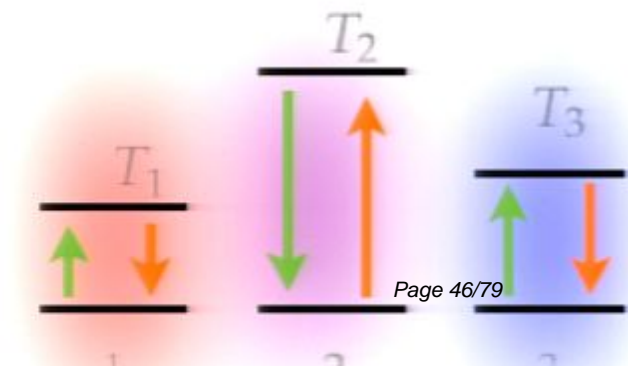
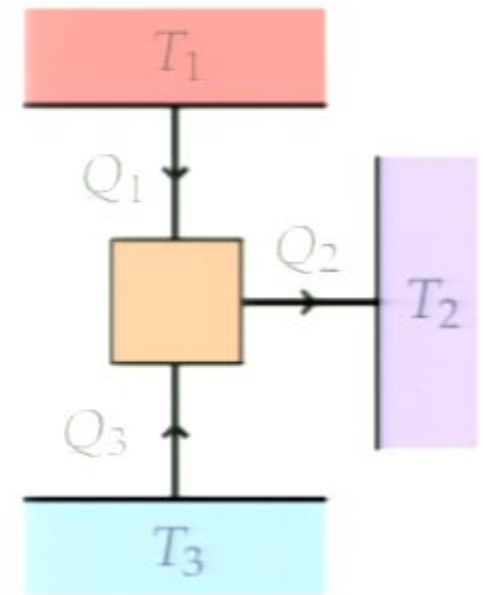


efficiency

$$\eta = \frac{Q_3}{Q_1} \quad \rightarrow \quad \eta^c = \frac{1 - \frac{T_2}{T_1}}{\frac{T_2}{T_3} - 1}$$

$$\frac{dQ_i}{dt} = \text{tr}(H_i \mathcal{D}_i(\rho)) \quad \mathcal{D}_i(\rho) = p_i(\tau_i \otimes \text{tr}_i \rho - \rho)$$

$$\frac{dQ_i}{dt} = \gamma \bar{E}_i \quad \rightarrow \quad \eta^q = \frac{E_3}{E_1}$$



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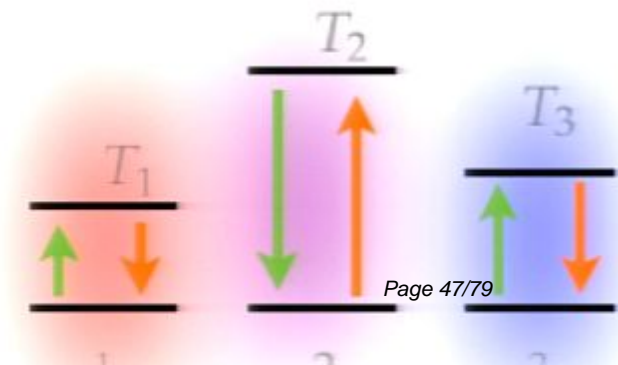
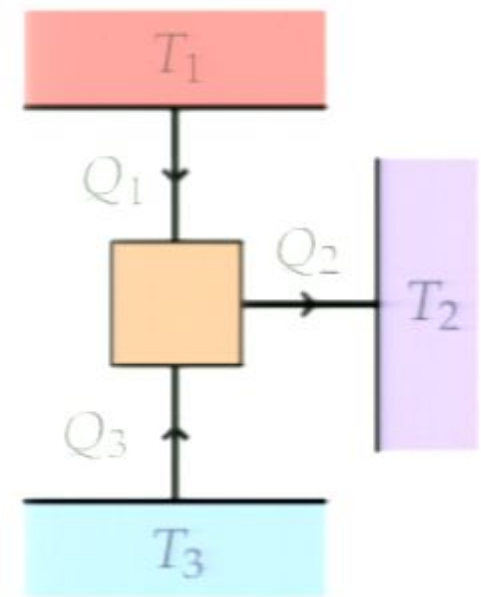
$$\frac{dQ_i}{dt} = \text{tr}(H_i D_i(\rho)) \quad D_i(\rho) = p_i(\tau_i \otimes \text{tr}_i \rho - \rho)$$

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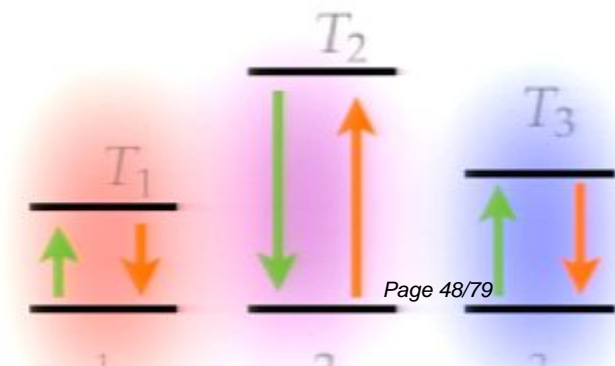
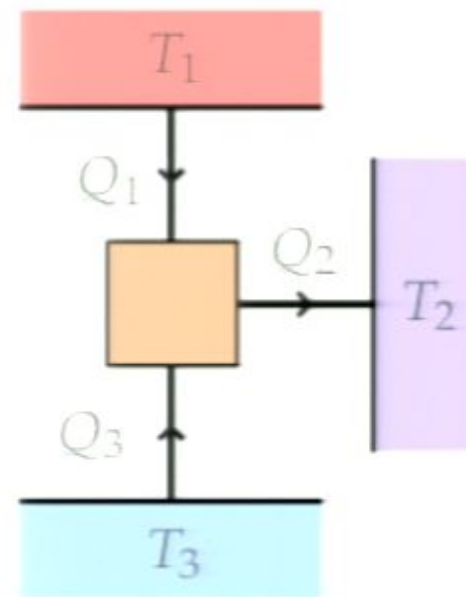
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$$\eta^q < \eta^c$$

- $T_v < T_3$

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but how to model an engine?

- question: what is *work* in this context?

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“motive power (work) is the useful effect that a motor is capable of producing. This effect can always be likened to the elevation of a weight to a certain height.”

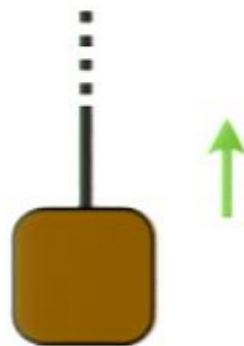
Sadi Carnot, reflexions sur la puissance motrice du feu

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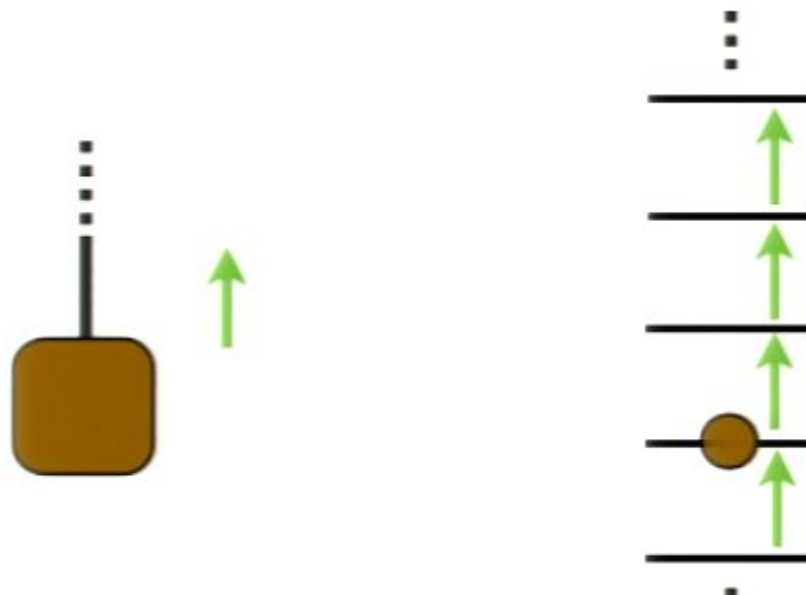


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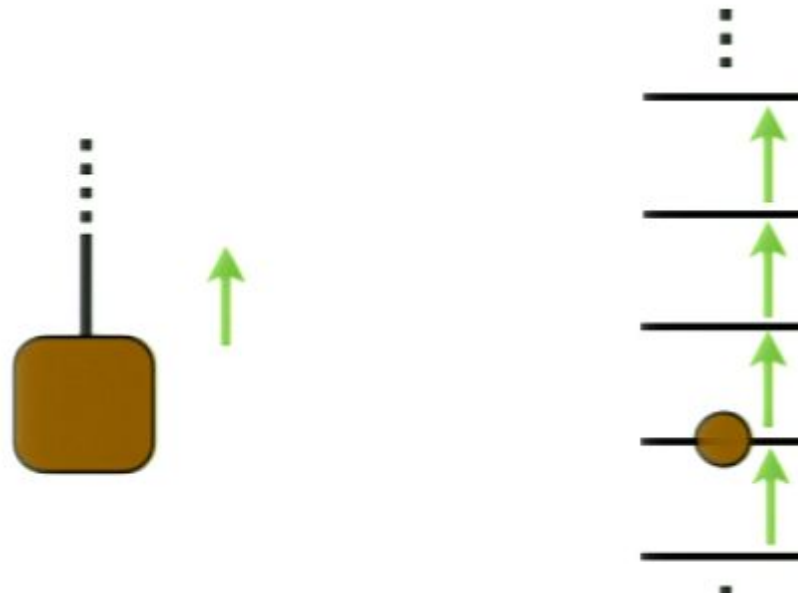


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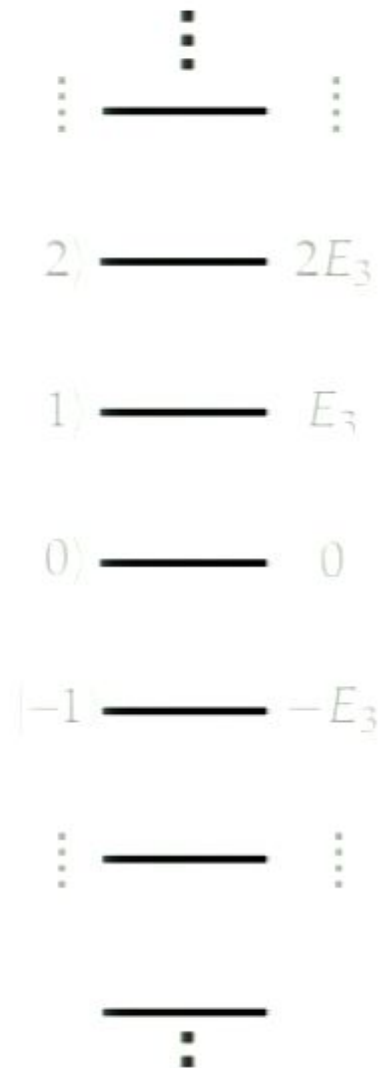
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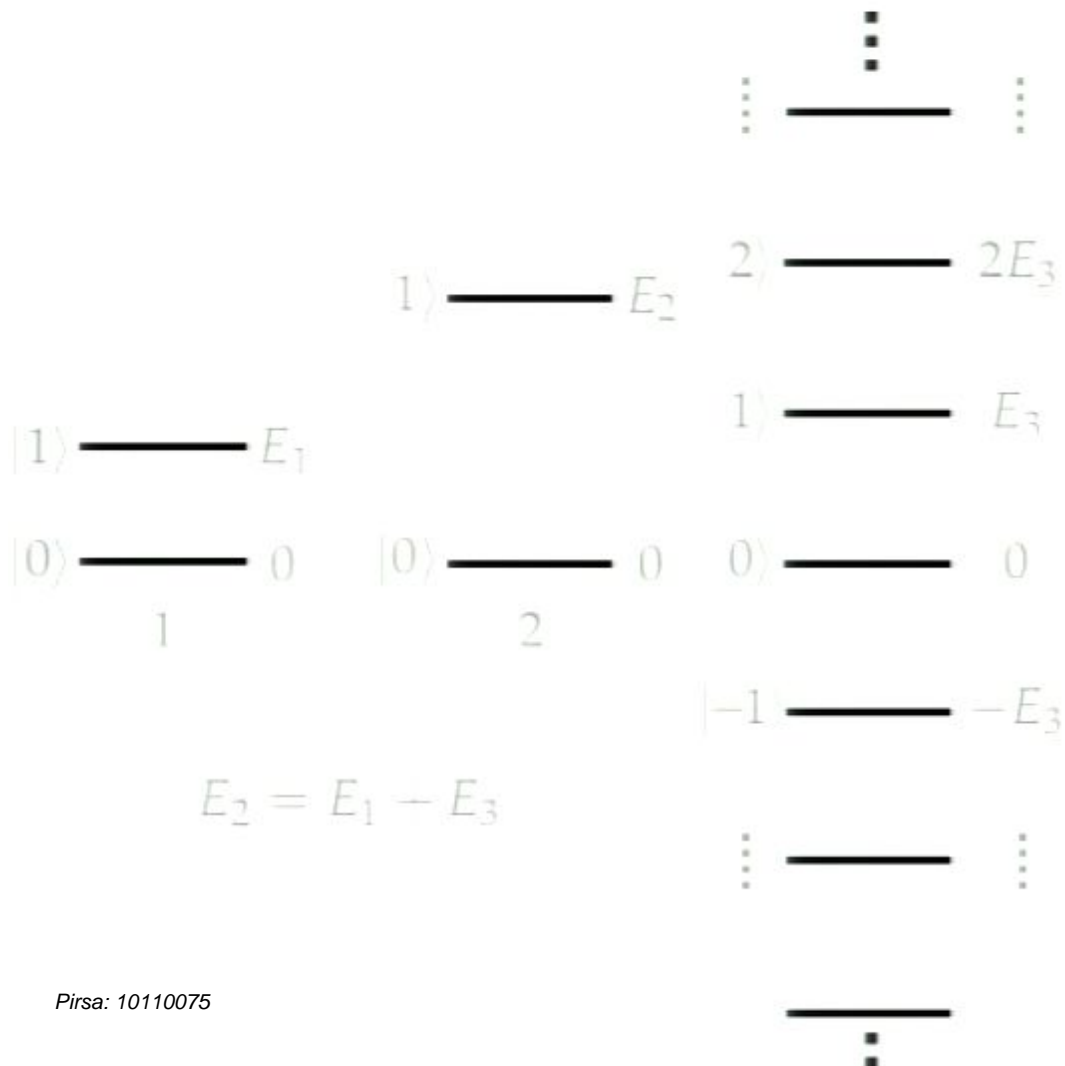


model engine

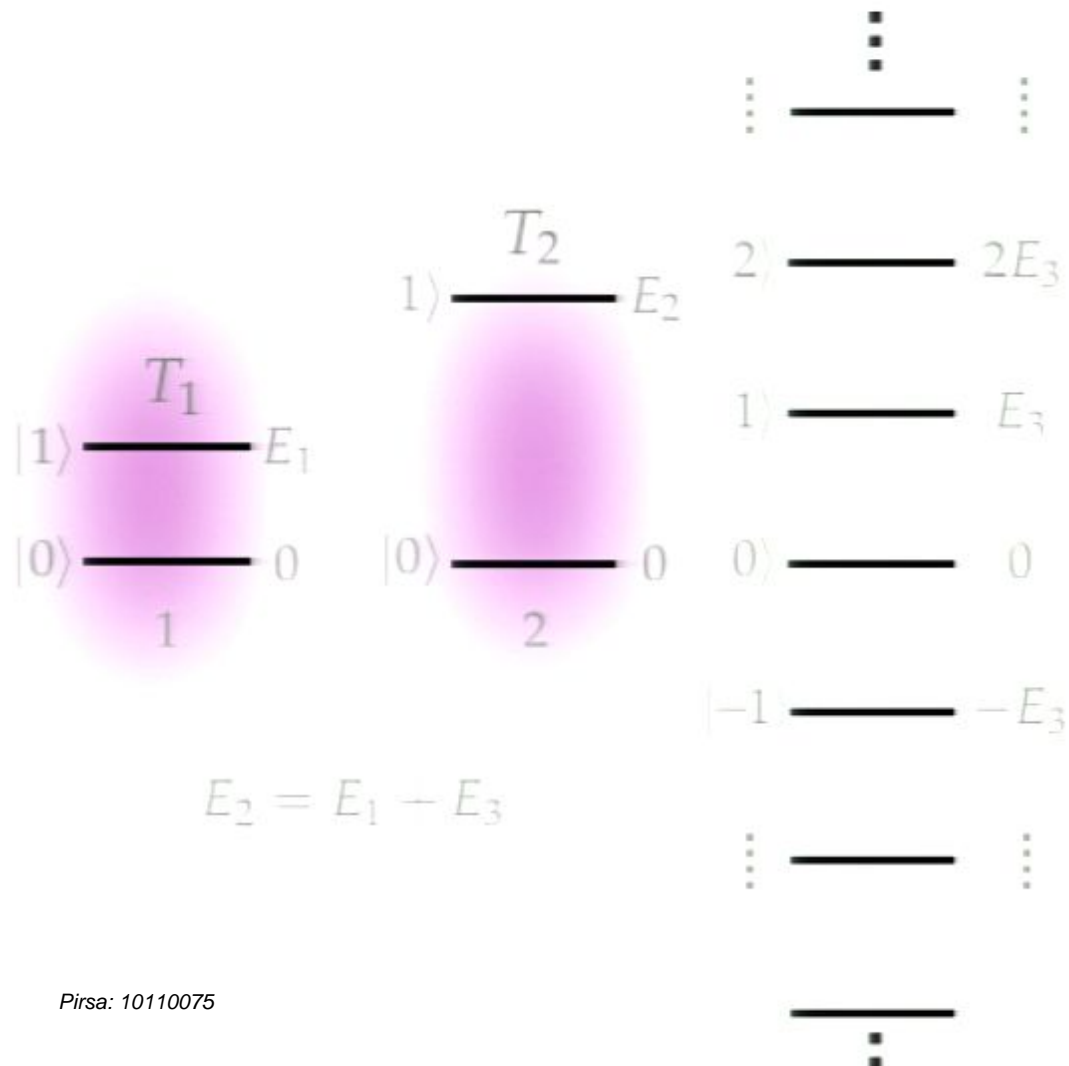
model engine



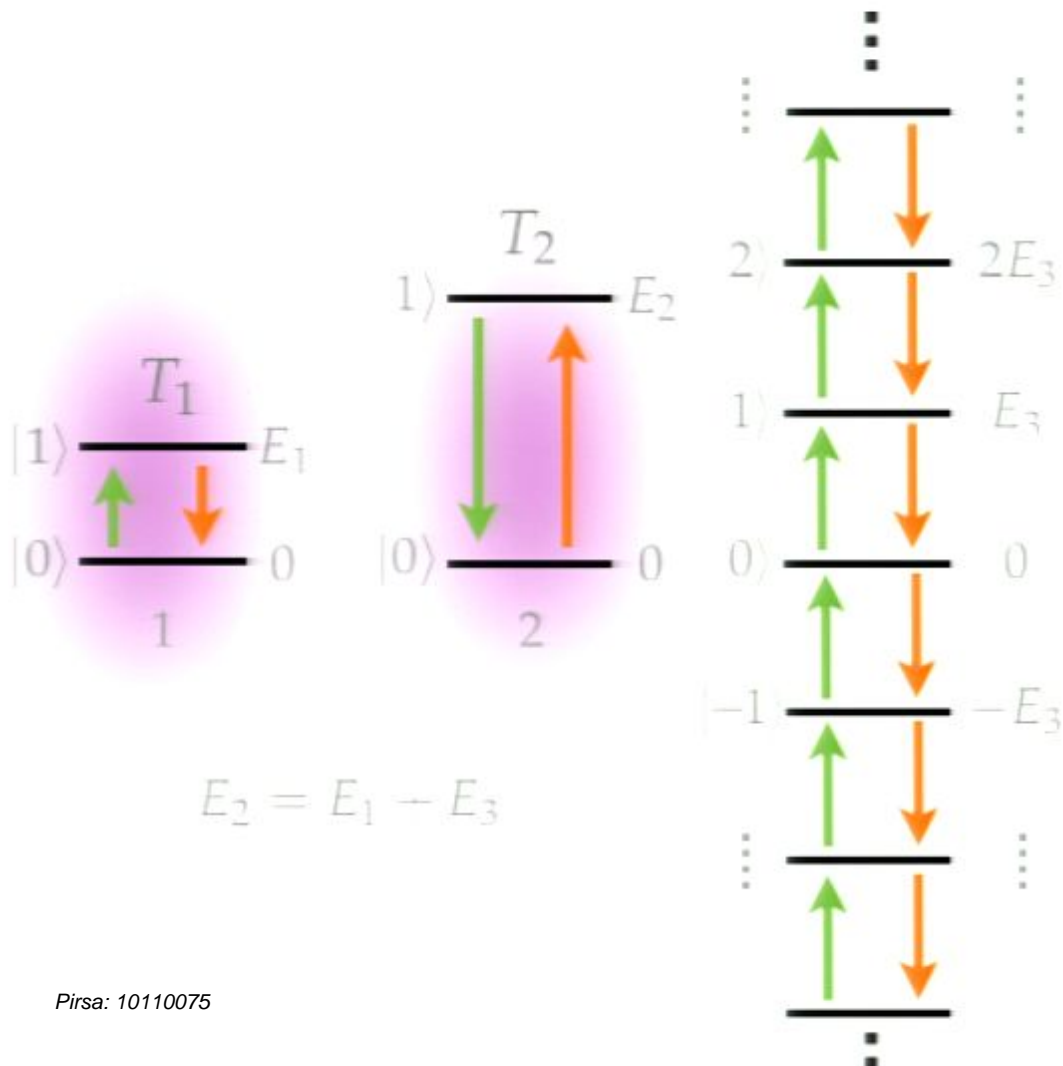
model engine



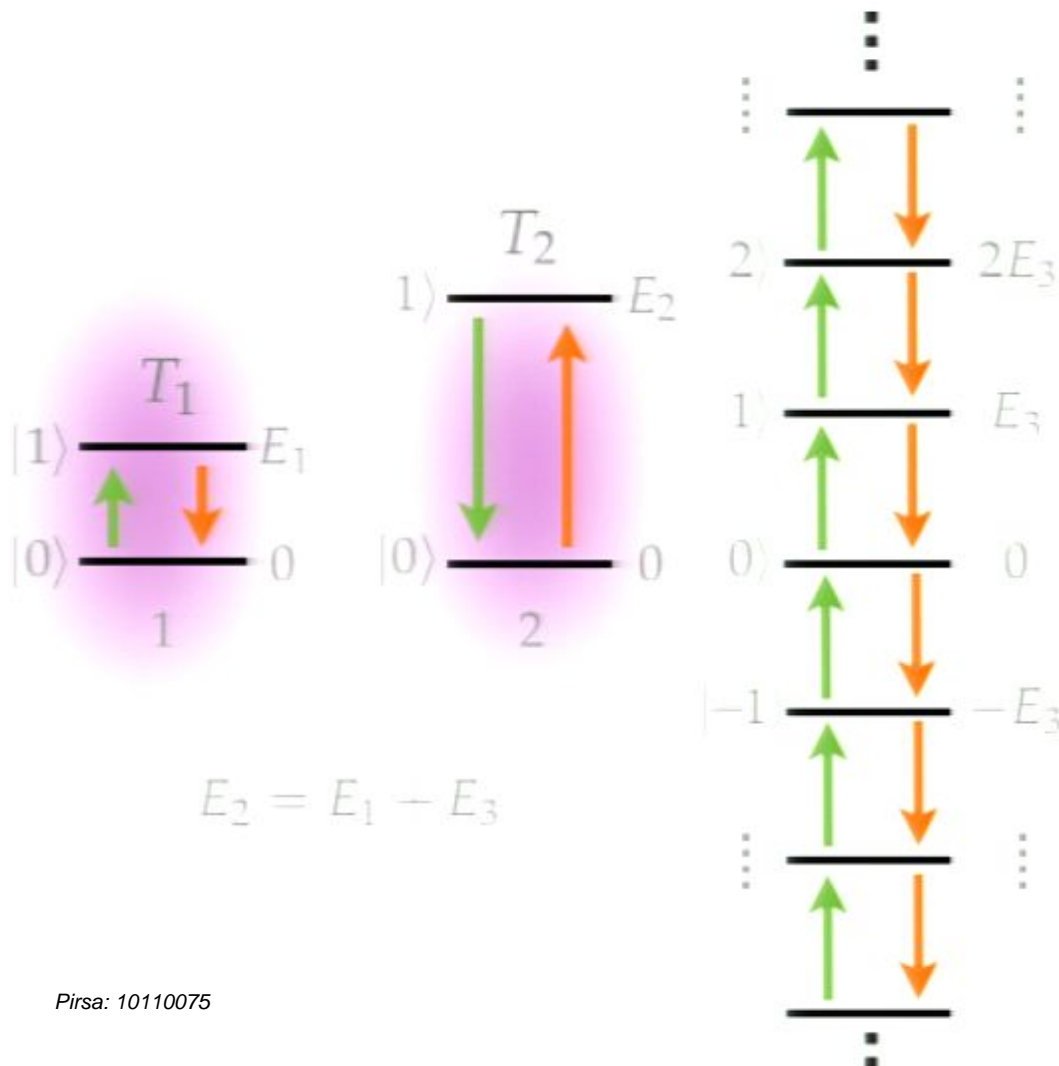
model engine



model engine



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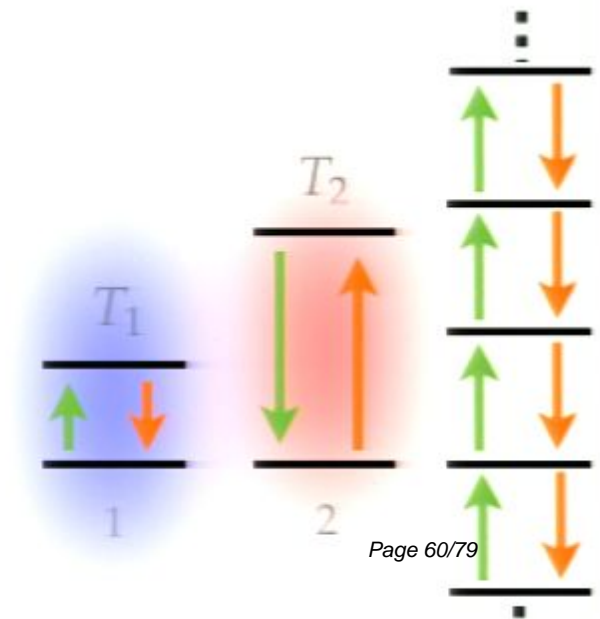
want $p(\text{raise}) > p(\text{lower})$

i.e. $p(01) > p(10)$

$$e^{-E_2/T_2} > e^{-E_1/T_1}$$

$$\boxed{\frac{E_1}{T_1} > \frac{E_2}{T_2}}$$

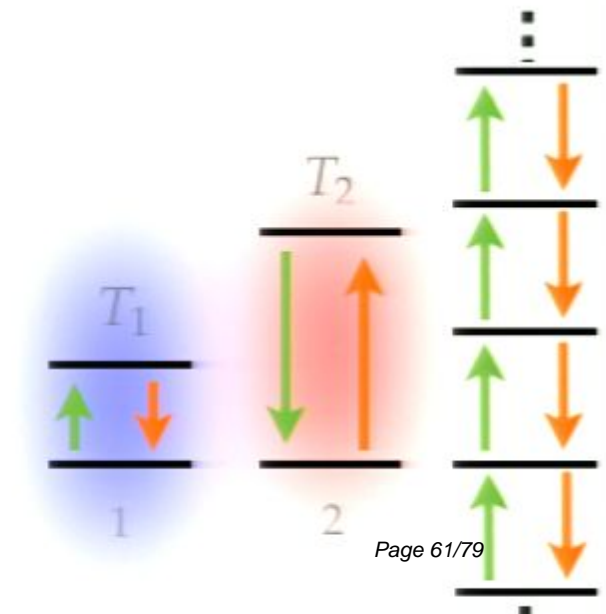
engine solution



engine solution

$$\frac{\partial \rho}{\partial t} = -i[H_0 + H_{int}, \rho] + \sum_i p_i (\tau_i \otimes \text{tr}_i \rho - \rho)$$

$$H_{int} = g \sum_{n=-\infty}^{\infty} (|01, n\rangle \langle 10, n+1| + |10, n+1\rangle \langle 01, n|)$$

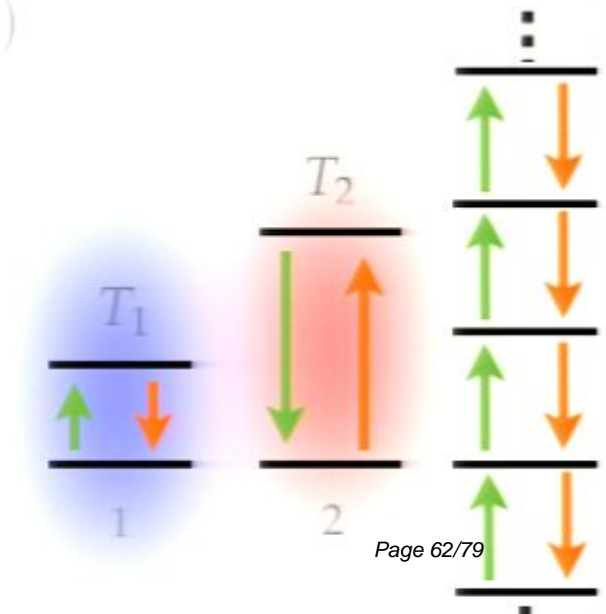


engine solution

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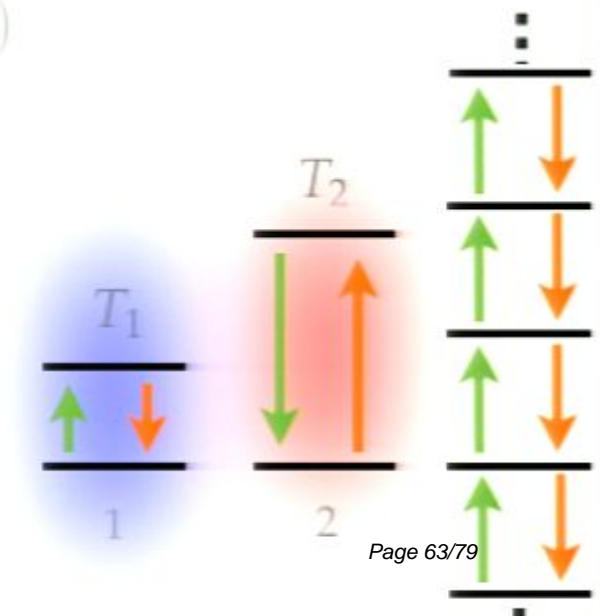
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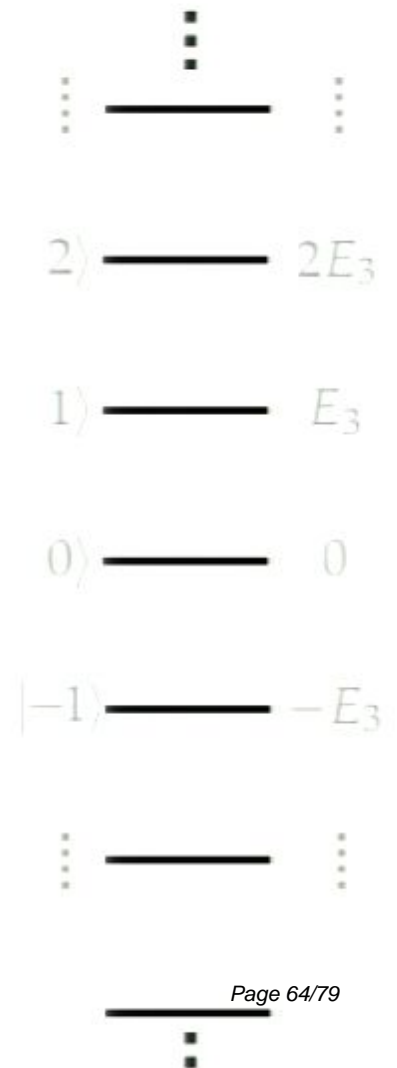
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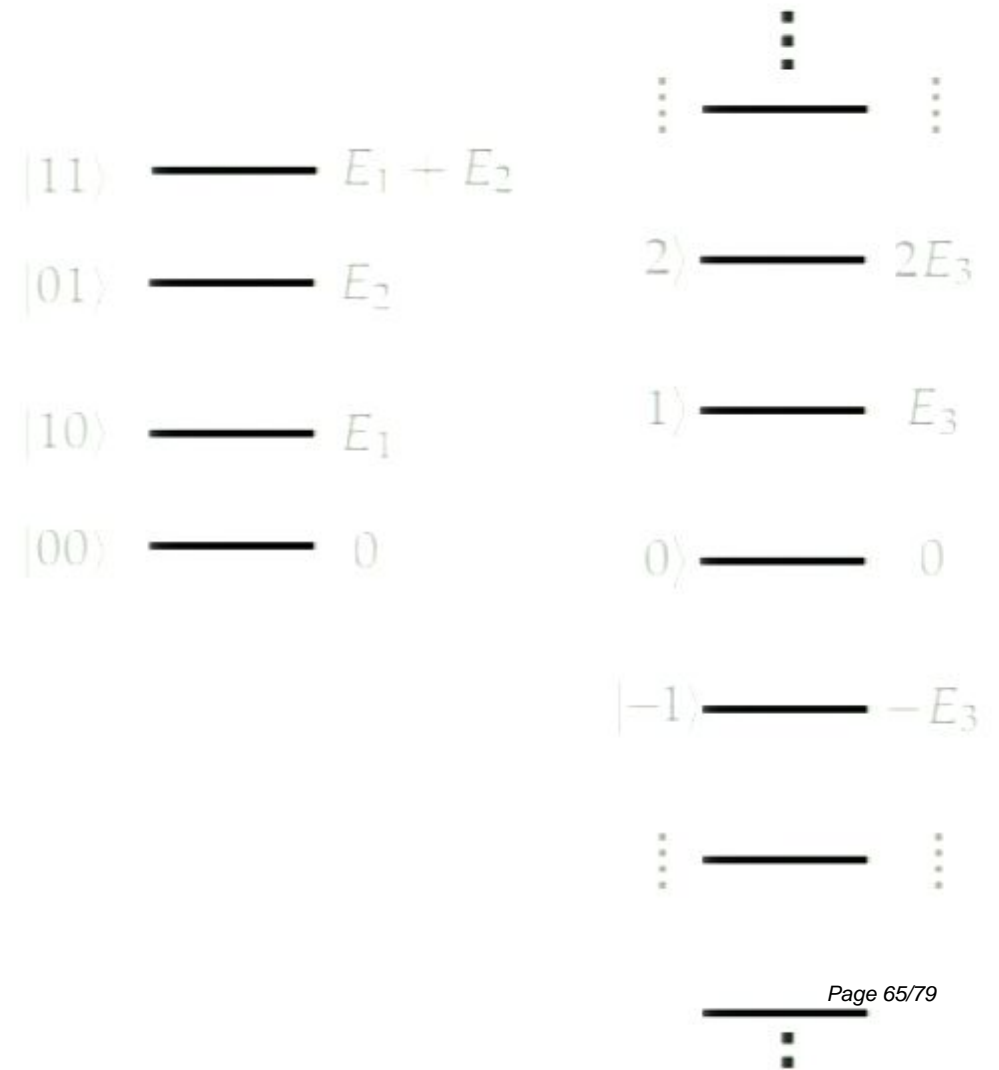
$$\frac{d}{dt} \langle E_w \rangle > 0 \quad \text{iff} \quad \frac{E_1}{T_1} > \frac{E_2}{T_2}$$



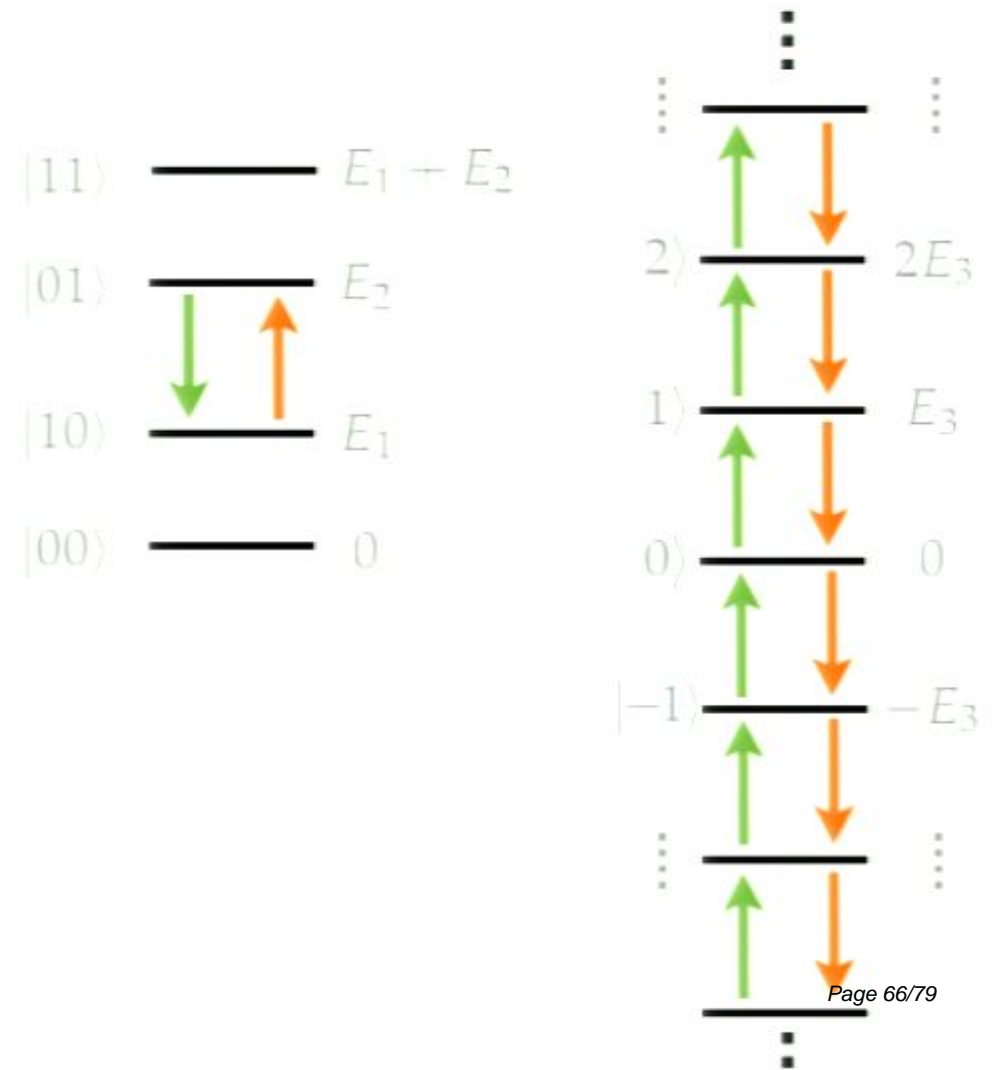
virtual transition



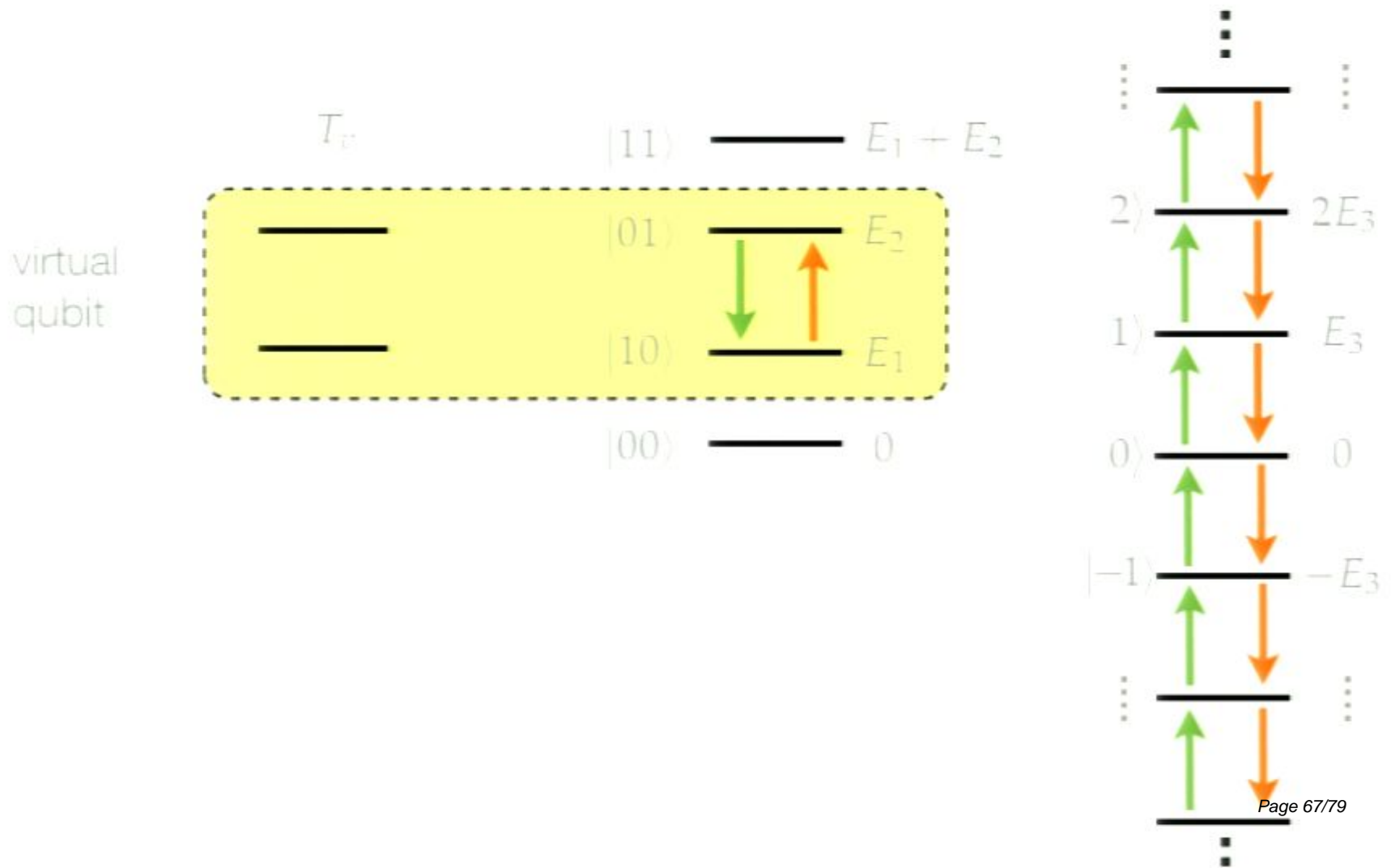
virtual transition



virtual transition

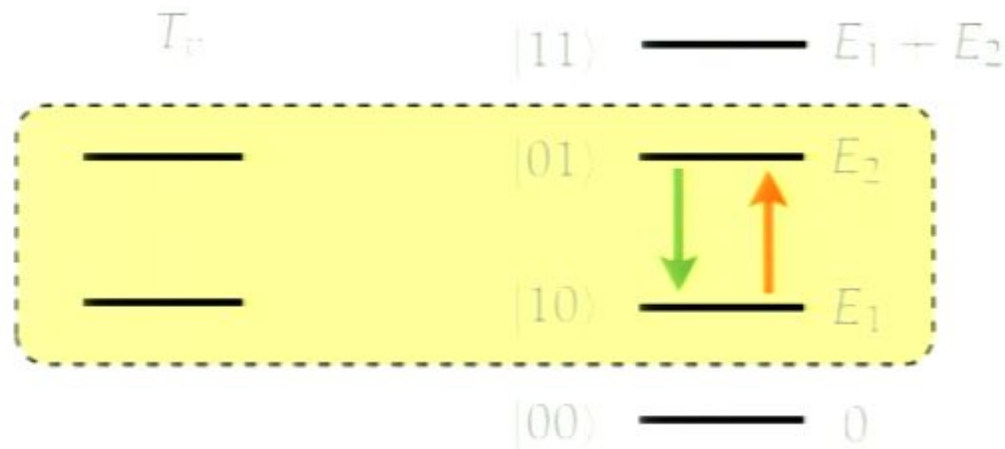


virtual transition



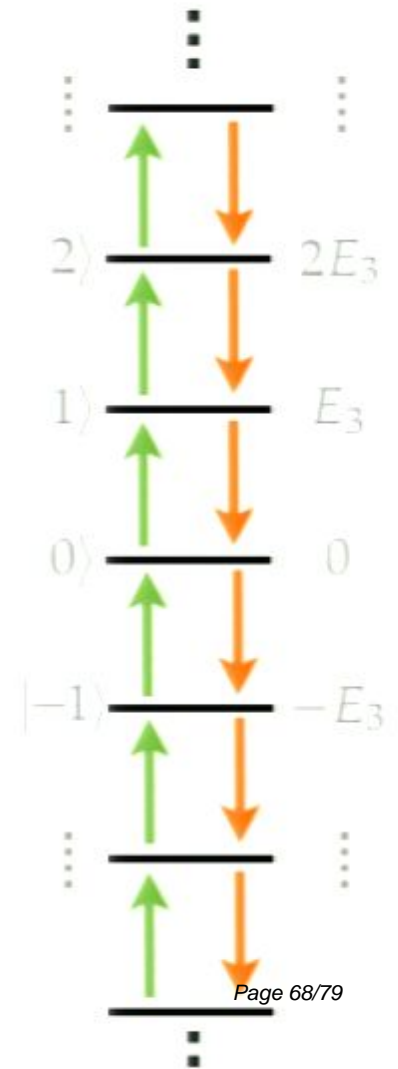
virtual transition

virtual qubit



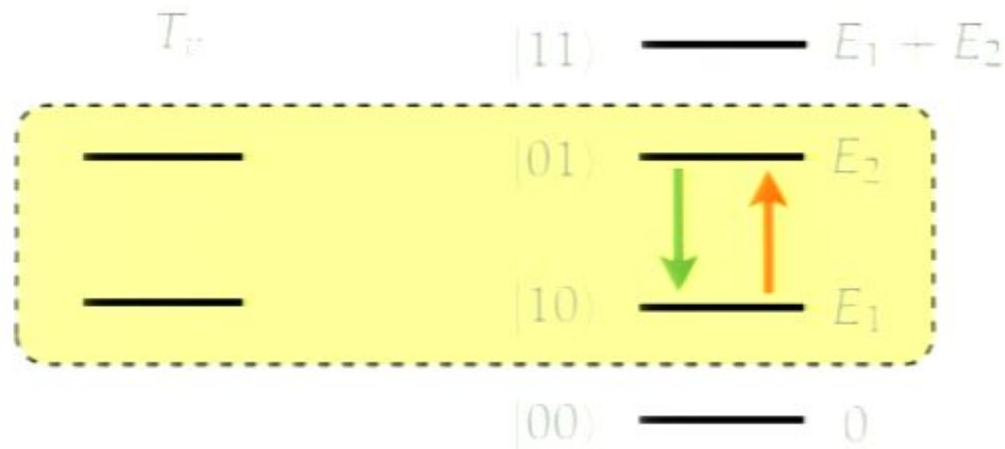
condition for engine to work

$$\frac{E_1}{T_1} > \frac{E_2}{T_2} \rightarrow T_v = \frac{E_2 - E_1}{\frac{E_2}{T_2} - \frac{E_1}{T_1}} < 0$$



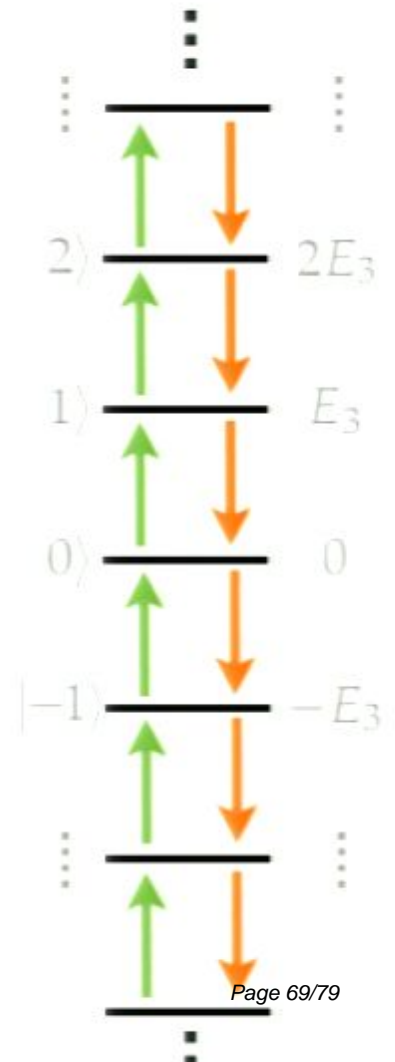
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unified viewpoint

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unified viewpoint

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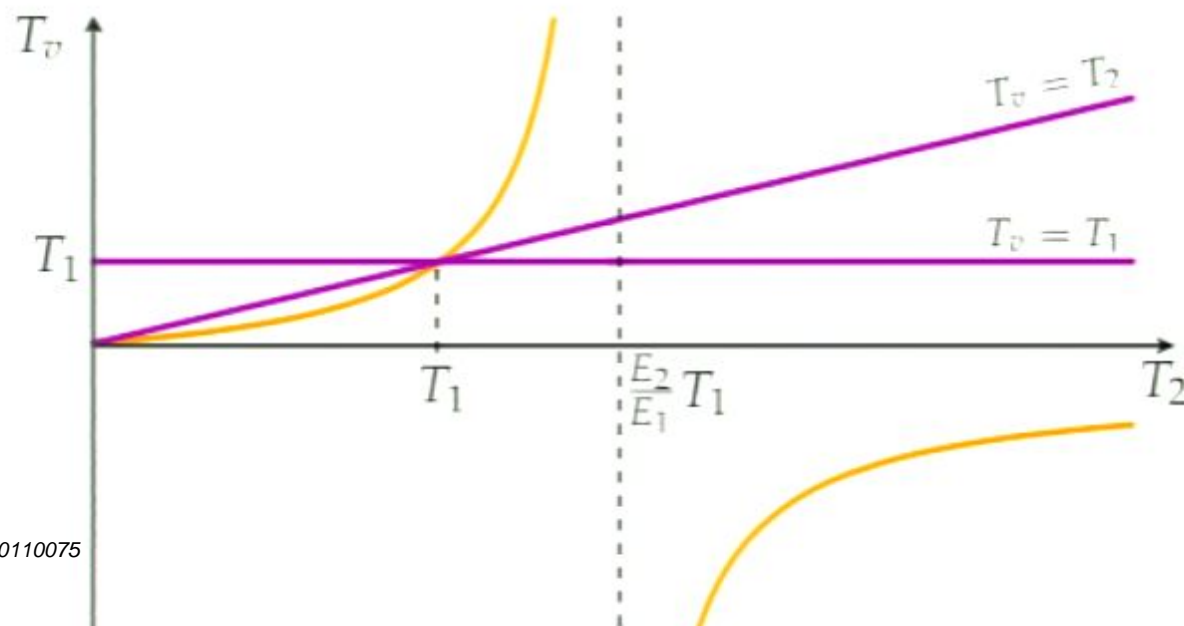
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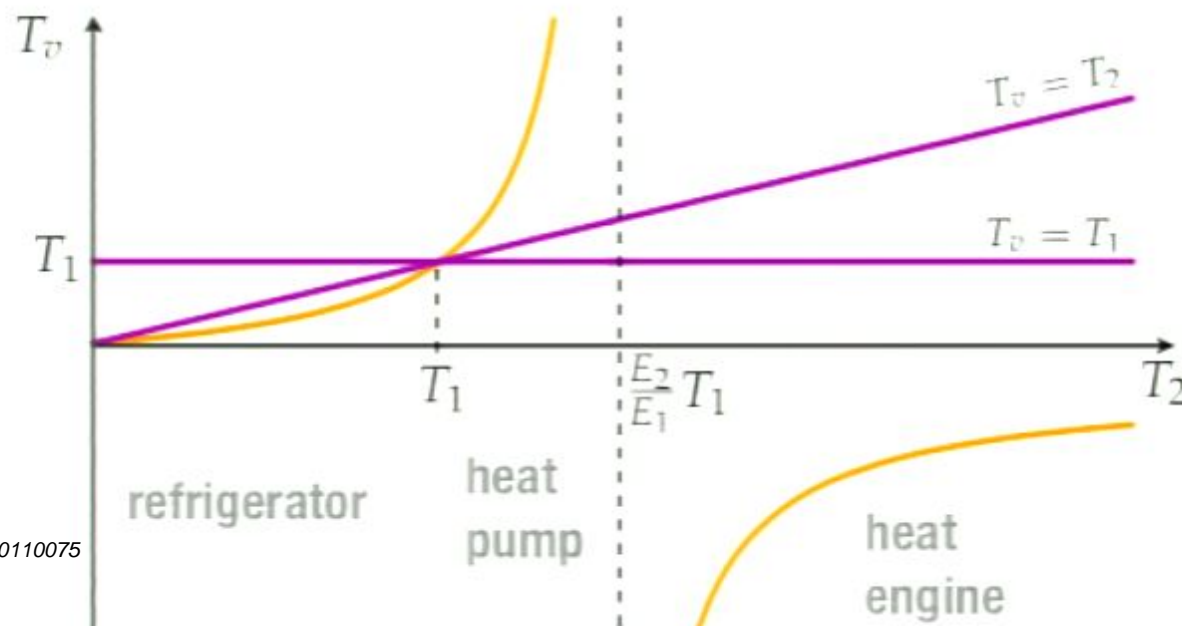
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open questions

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- quantify the 'quality' of work
- third law of thermodynamics

conclusions

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- work intimately related to population inversion
- unified view of thermal machines at quantum level