Title: Quantum Mechanics 12 - How Atoms Emit and Absorb Photons

Date: Aug 16, 2008 02:30 PM

URL: http://pirsa.org/08080087

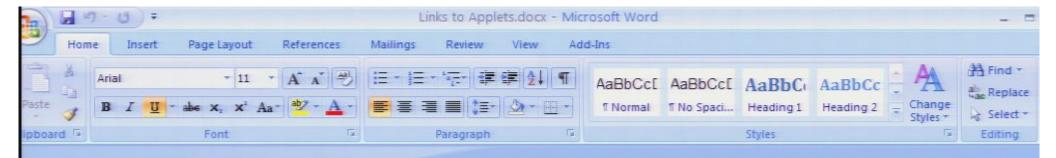
Abstract: Learning Outcomes: <br>

 $\hat{a}$  €¢ How the complex standing wave states of an electron in a one-dimensional box are  $\hat{a}$  ۾stationary states $\hat{a}$ ۥ in that the electron probability pattern is static (not changing with time). <br/> <br/> | changing with time |

 $\hat{a} \in \phi$  However, if the electron is put in a superposition of two such stationary states (with different energies), its probability pattern is not static, but rather oscillates back and forth; understanding how this oscillation is connected with photon emission and absorption. <br/> <br/> transfer of the electron is put in a superposition of two such stationary states (with different energies), its probability pattern is not static, but rather oscillates back and forth; understanding how this oscillation is connected with photon emission and absorption.

• Understanding how these ideas apply to the atom, and how a LASER works (Light Amplification by Stimulated Emission of Radiation).

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## Links to Applets

Falstad Applets

http://www.falstad.com/mathphysics.html

Standing Waves

http://www.ngsir.netfirms.com/englishhtm/StatWave.htm

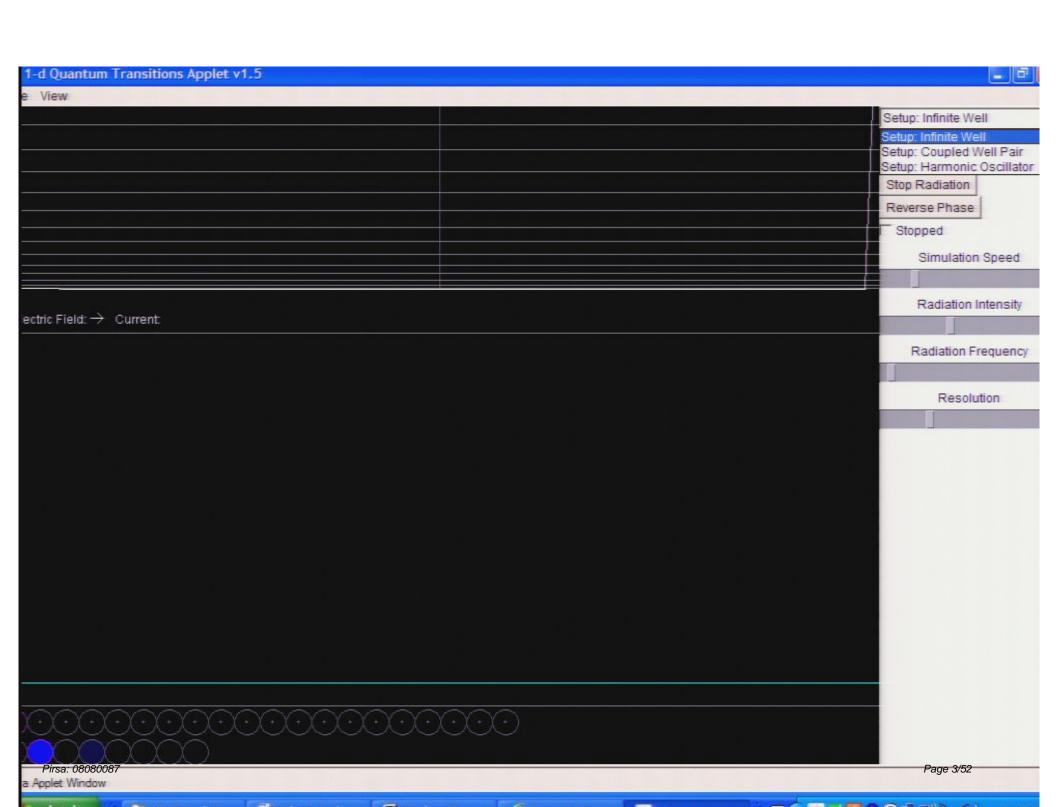
Superposition of Waves:

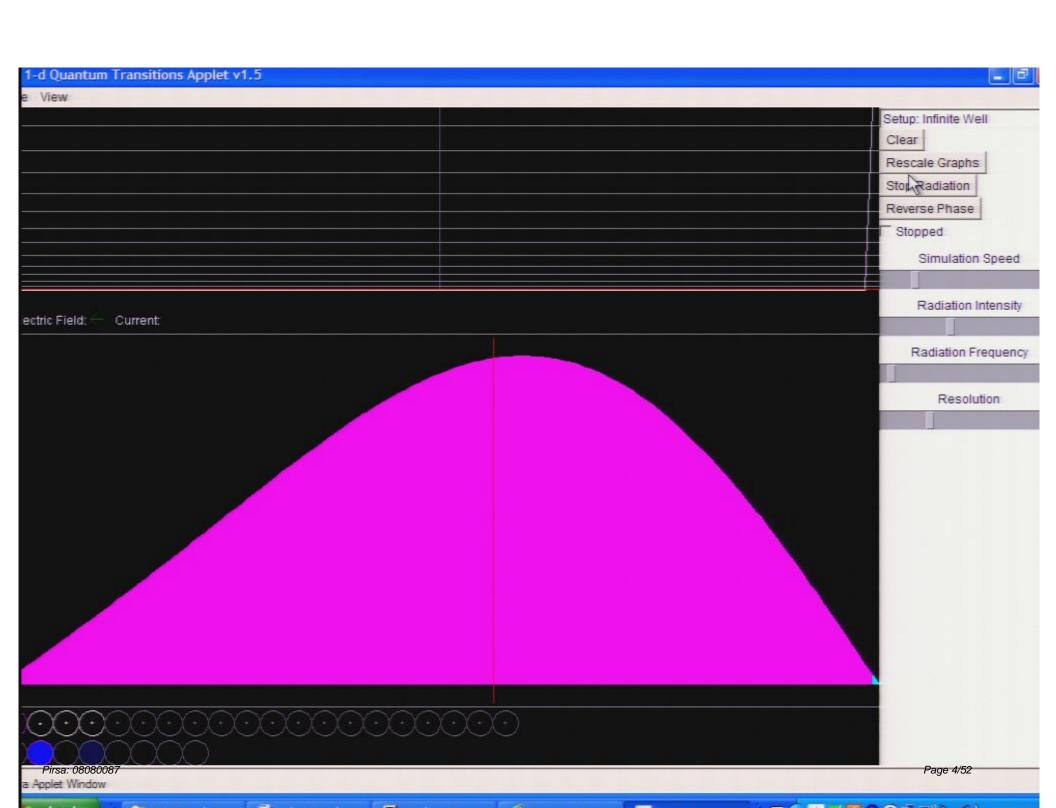
http://www.phy.ntnu.edu.tw/ntnujava/index.php?topic=19

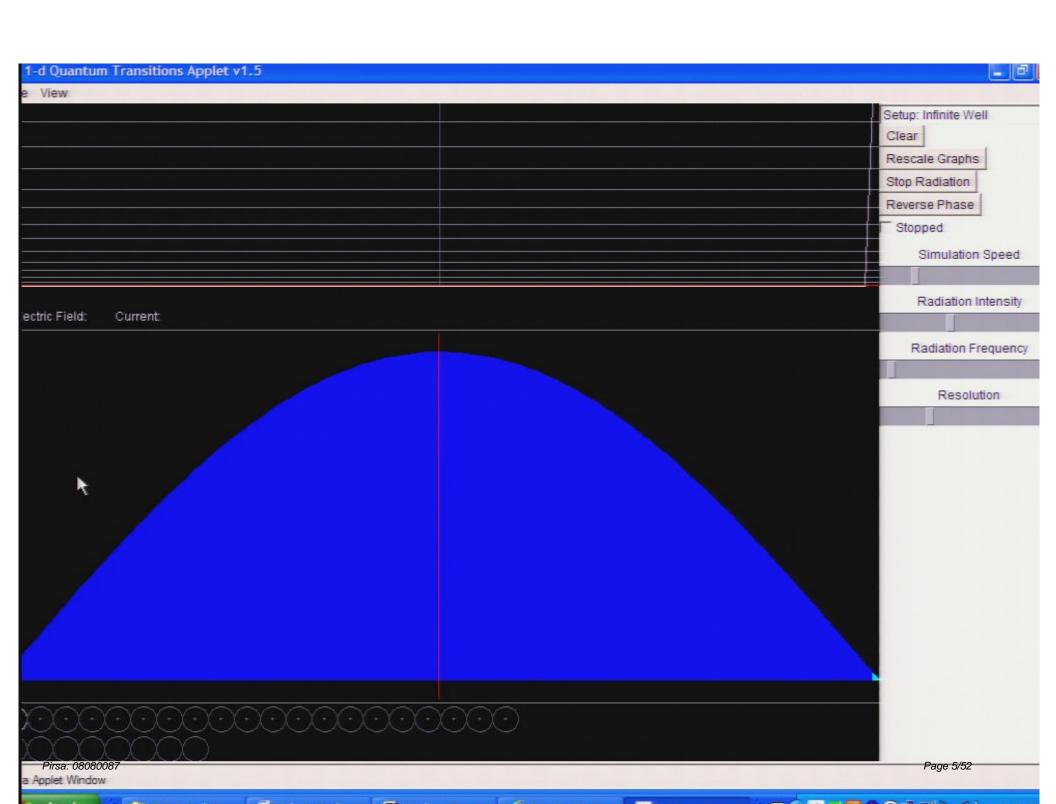
http://thespoon7.tripod.com/wave.htm

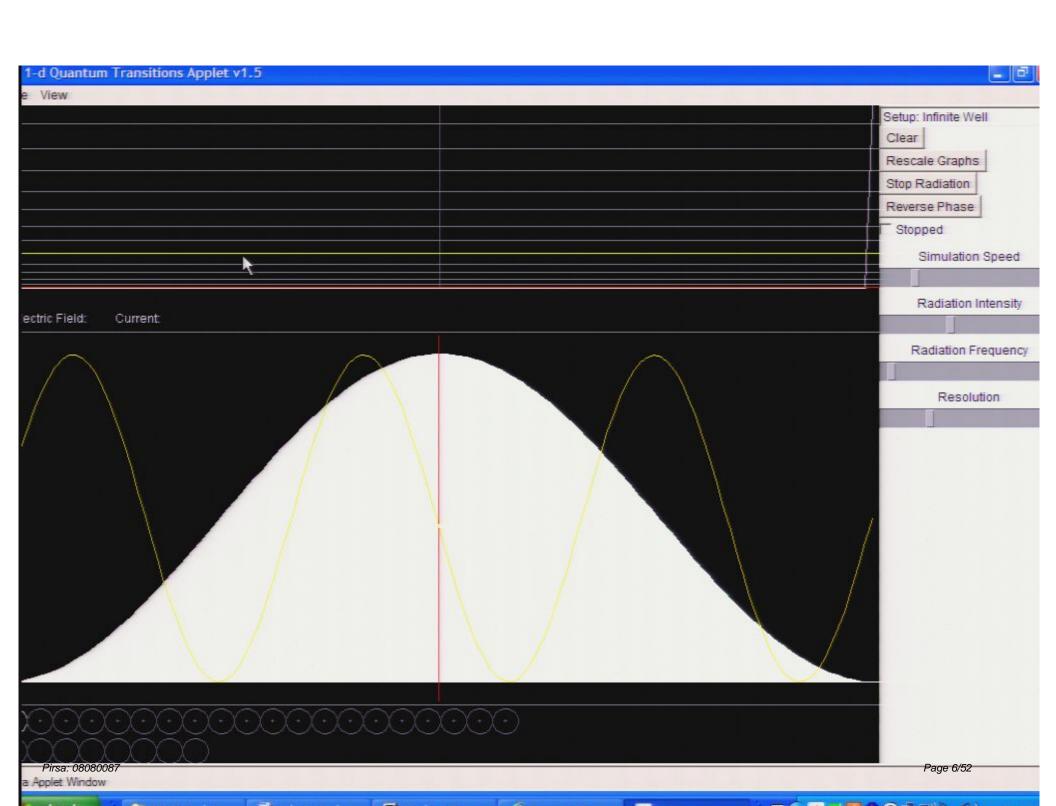
http://kingfish.coastal.edu/physics/physlets/Waves/superposition.html

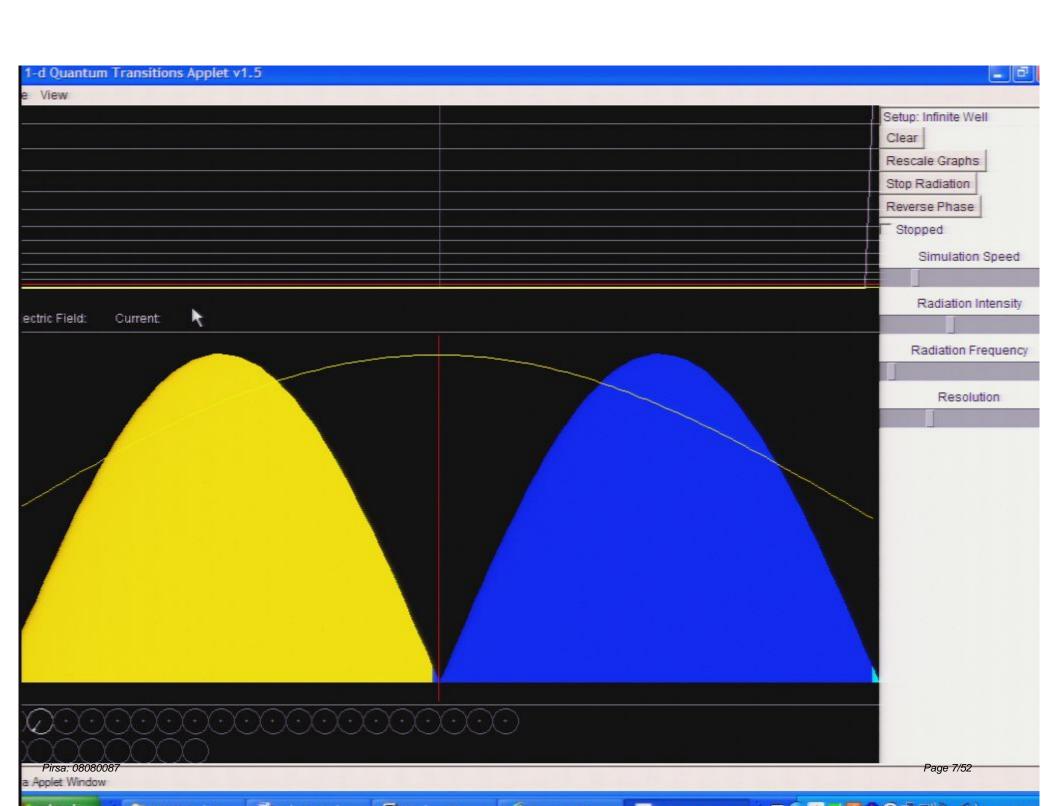
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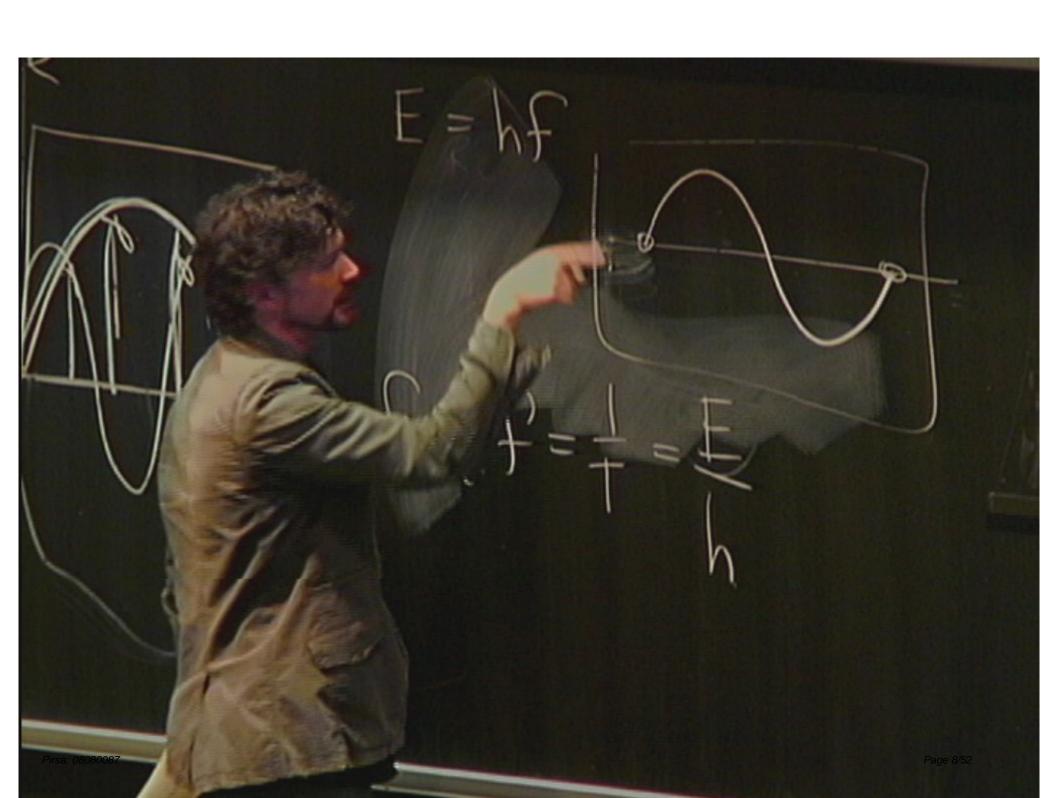


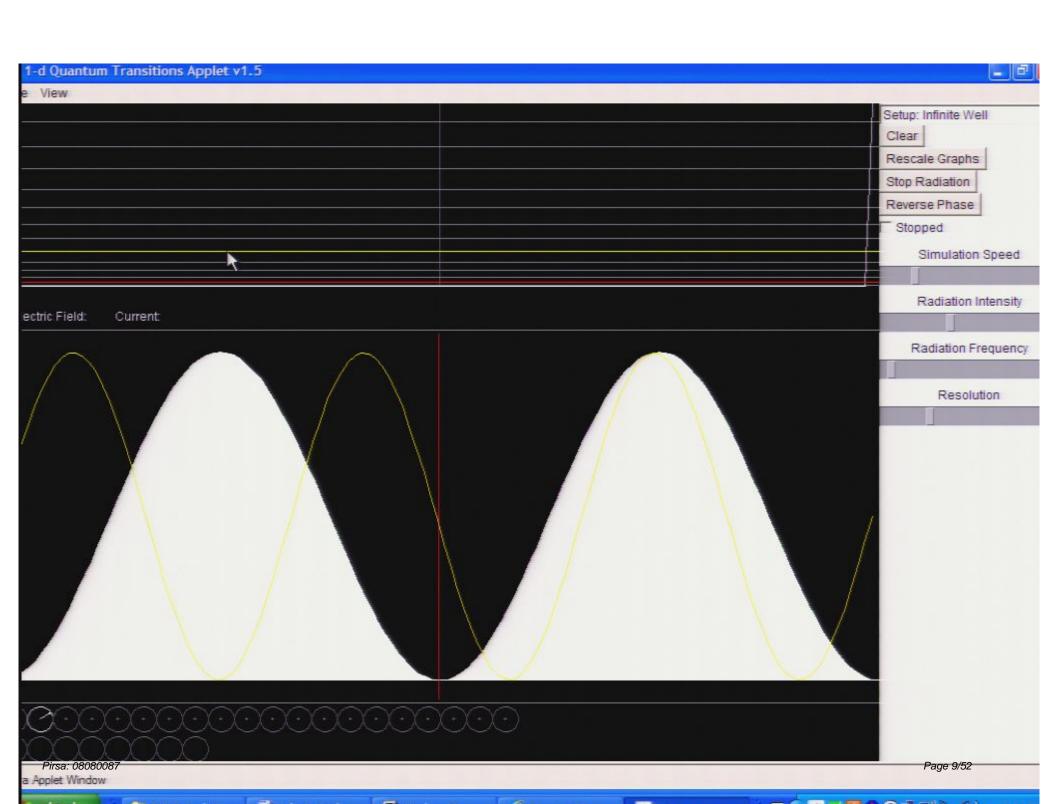










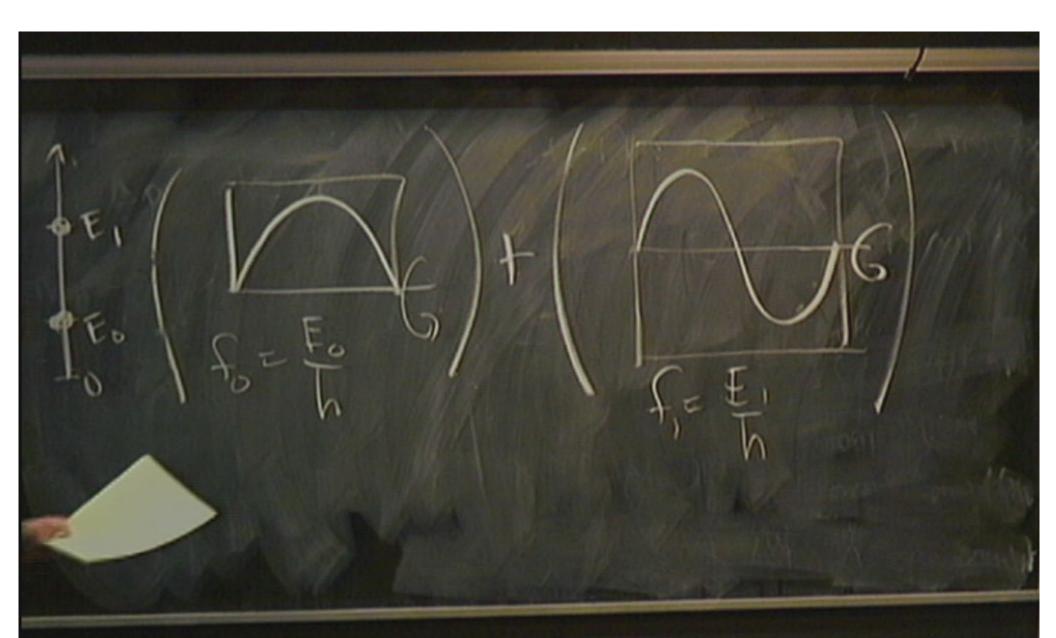


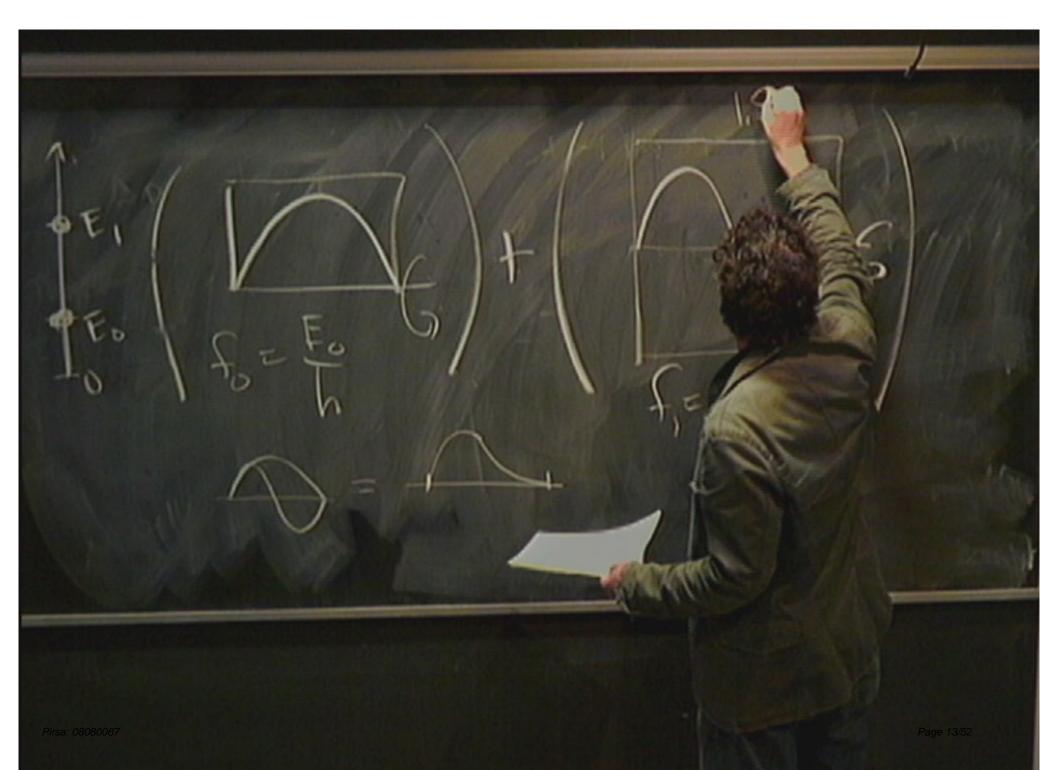
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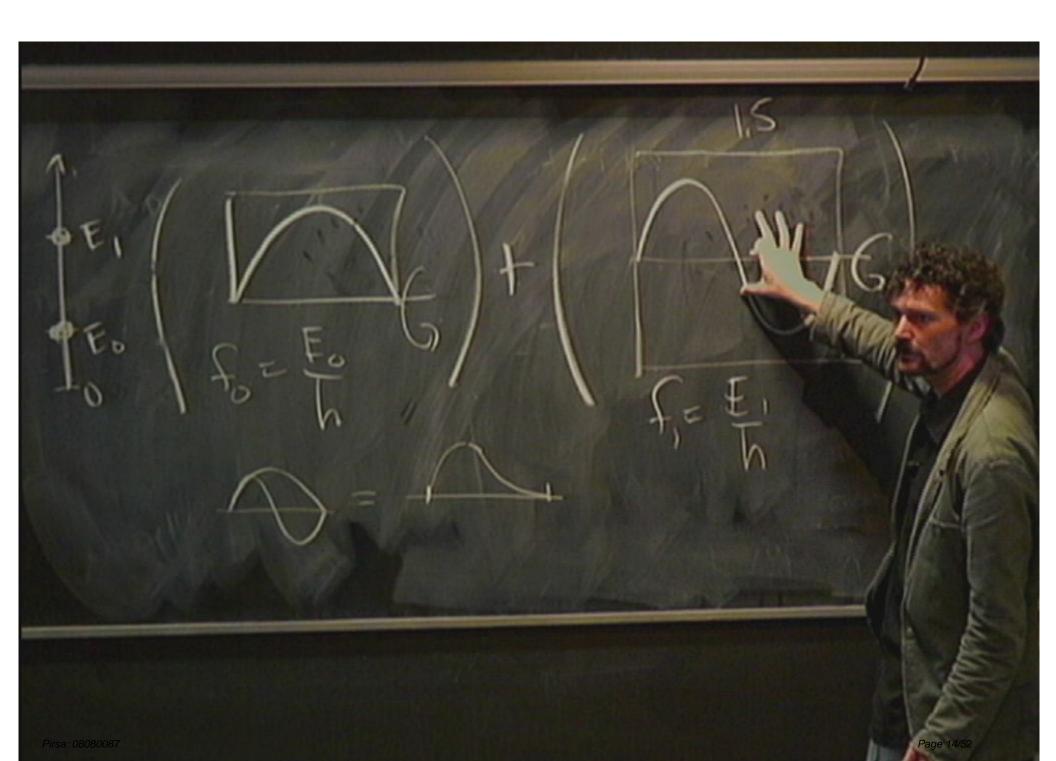
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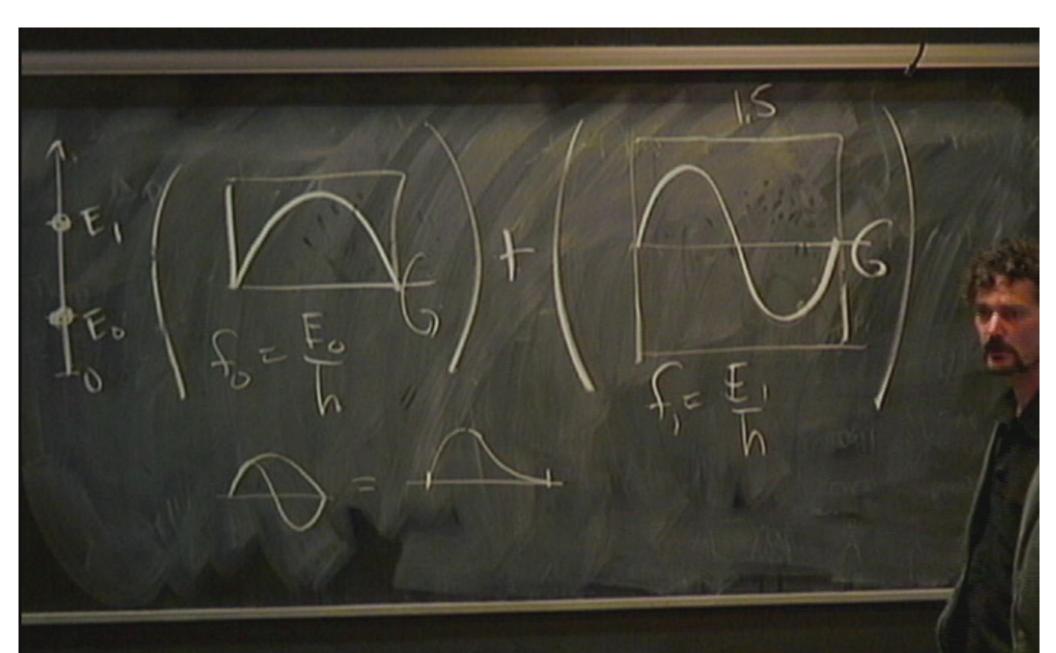
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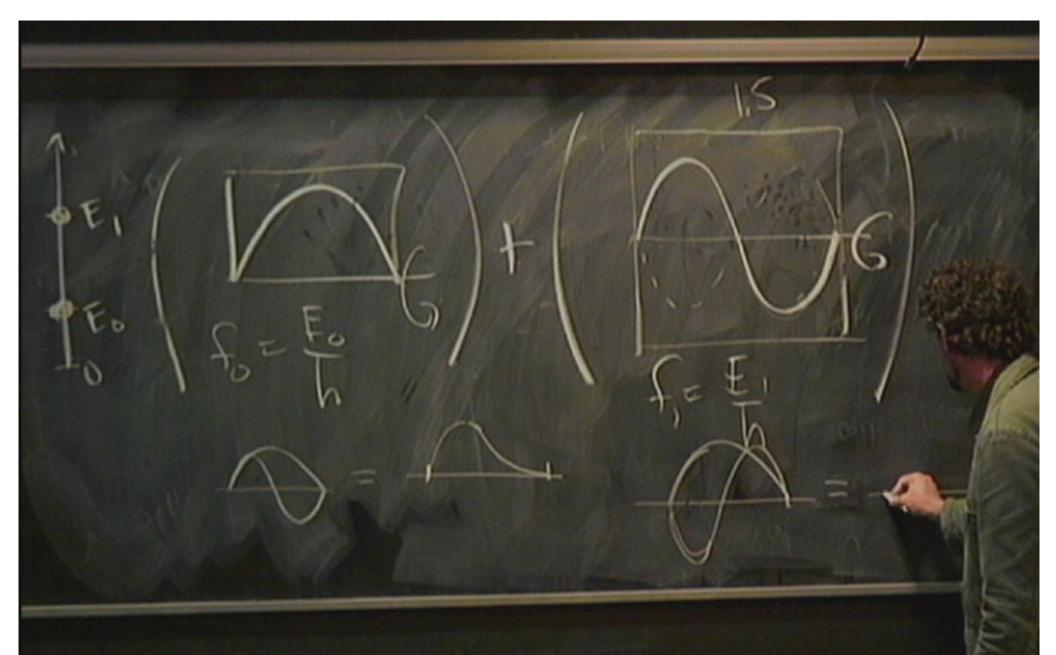
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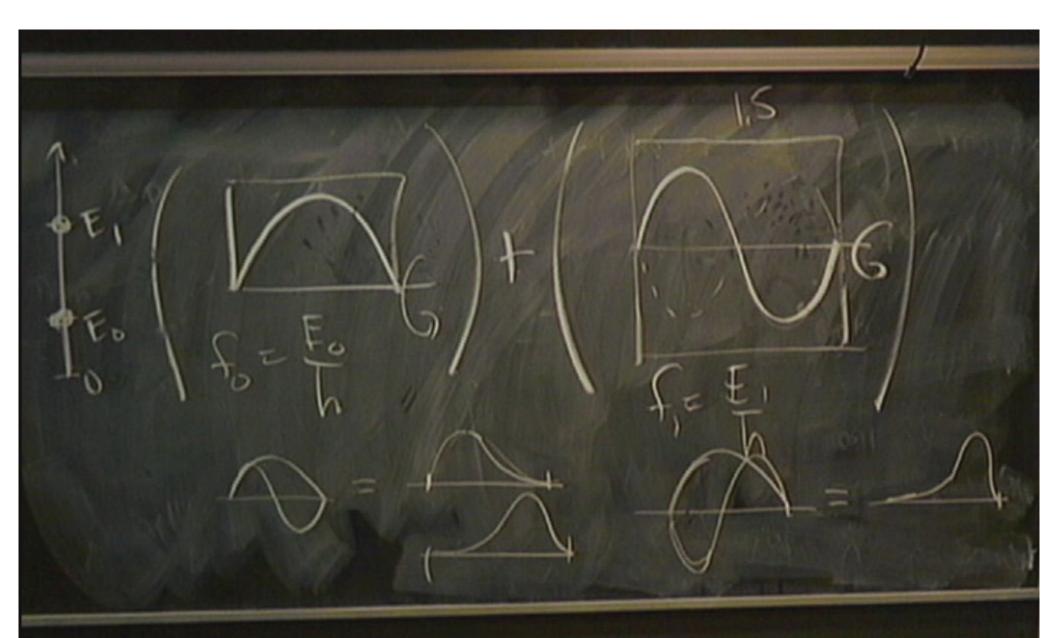


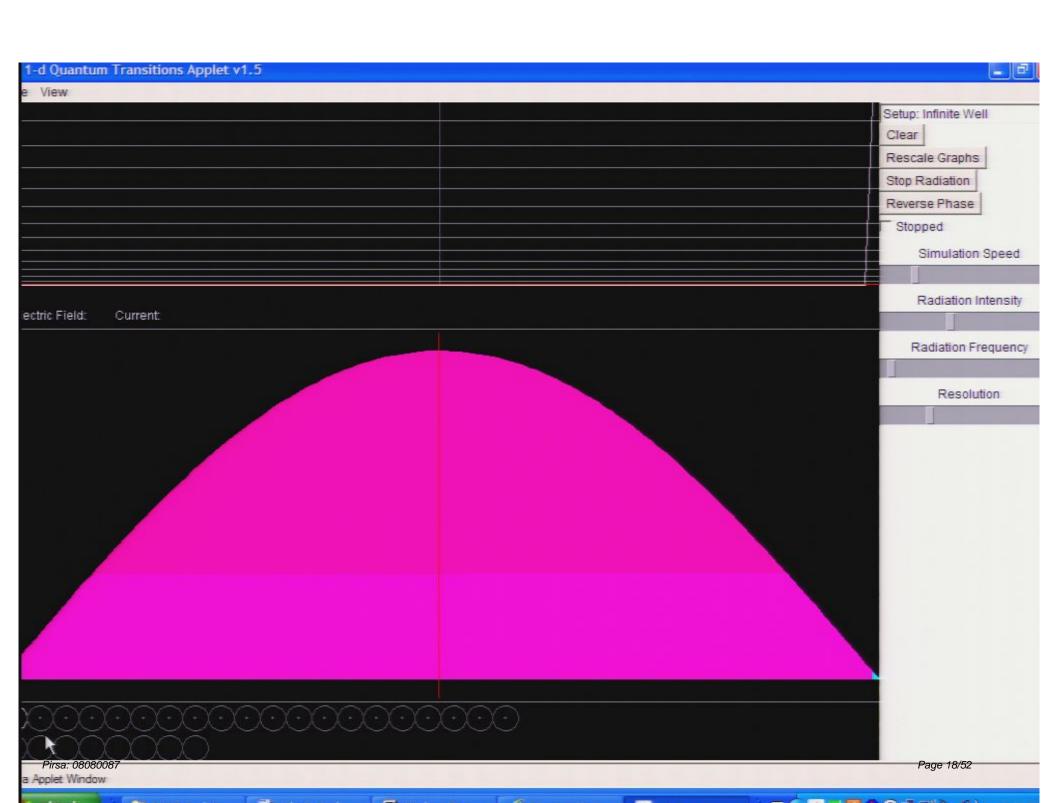


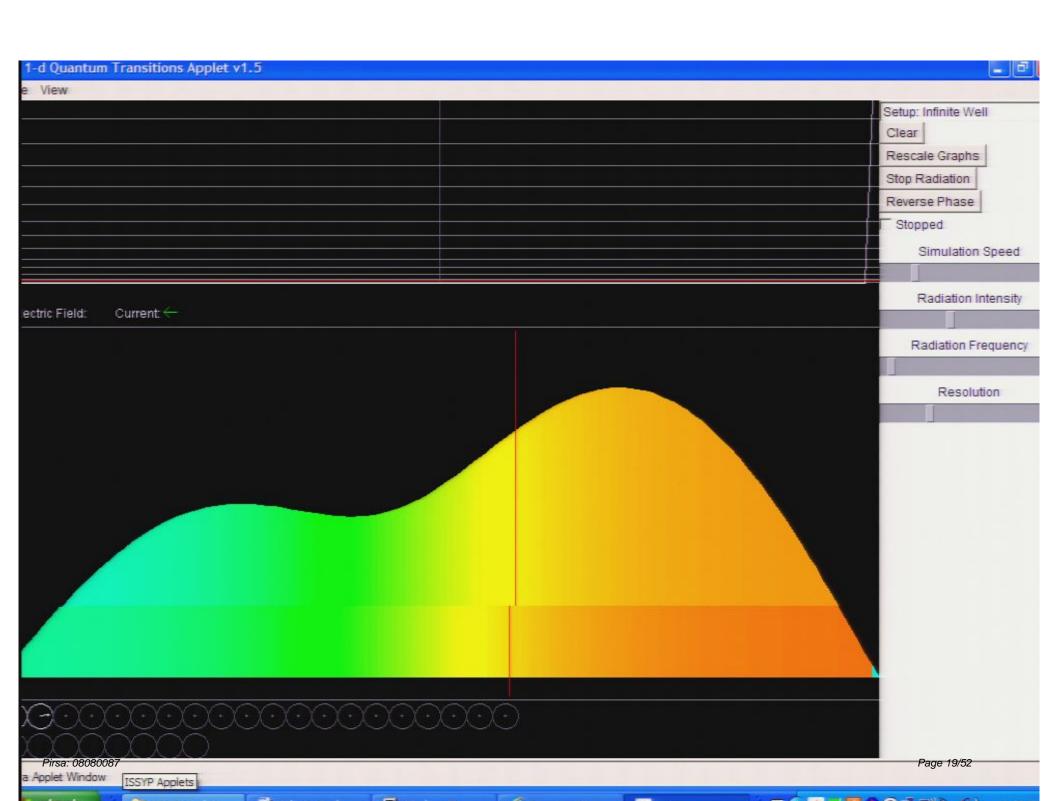


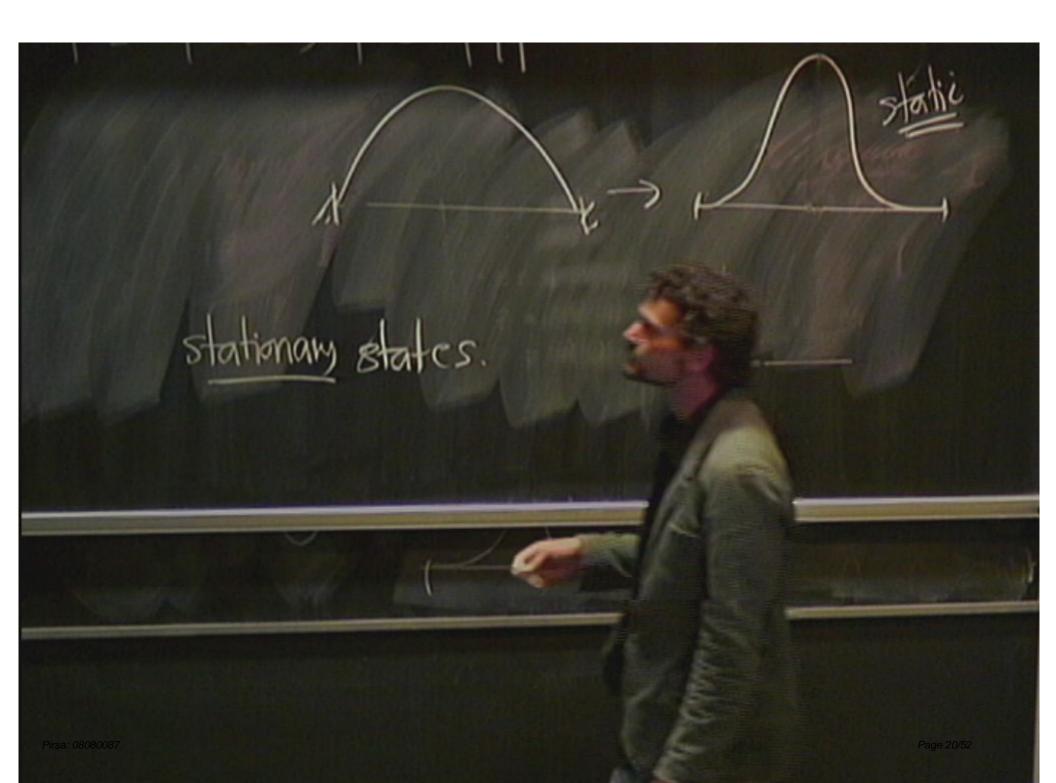


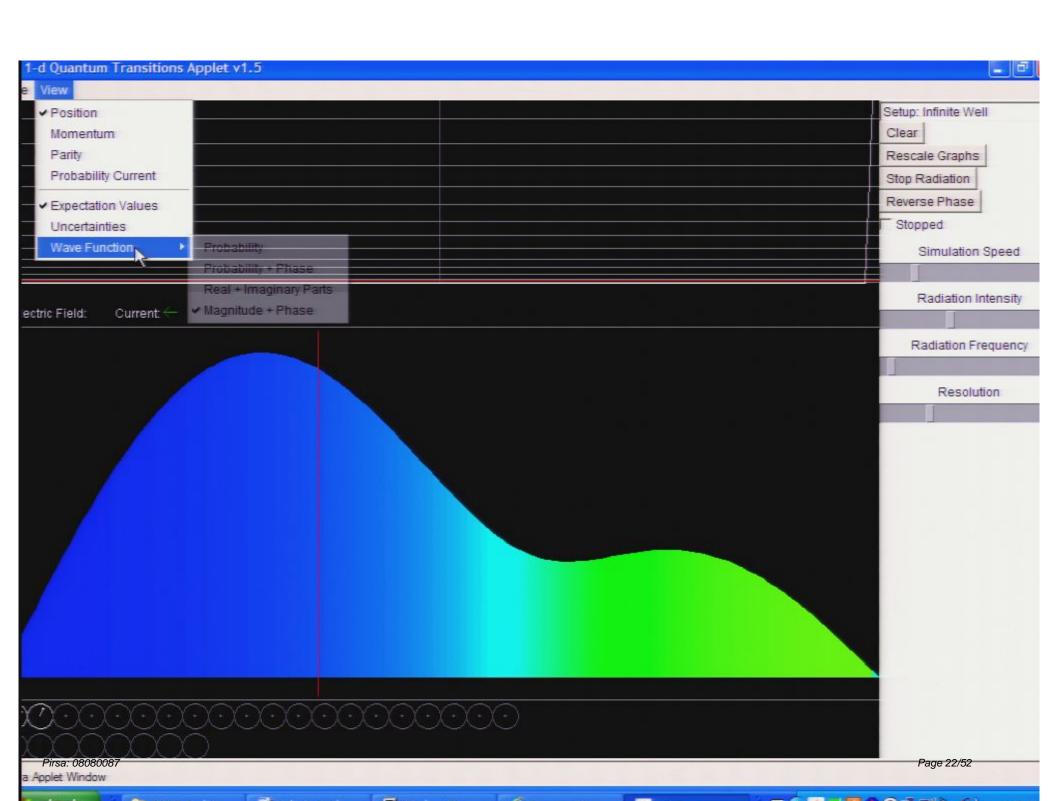


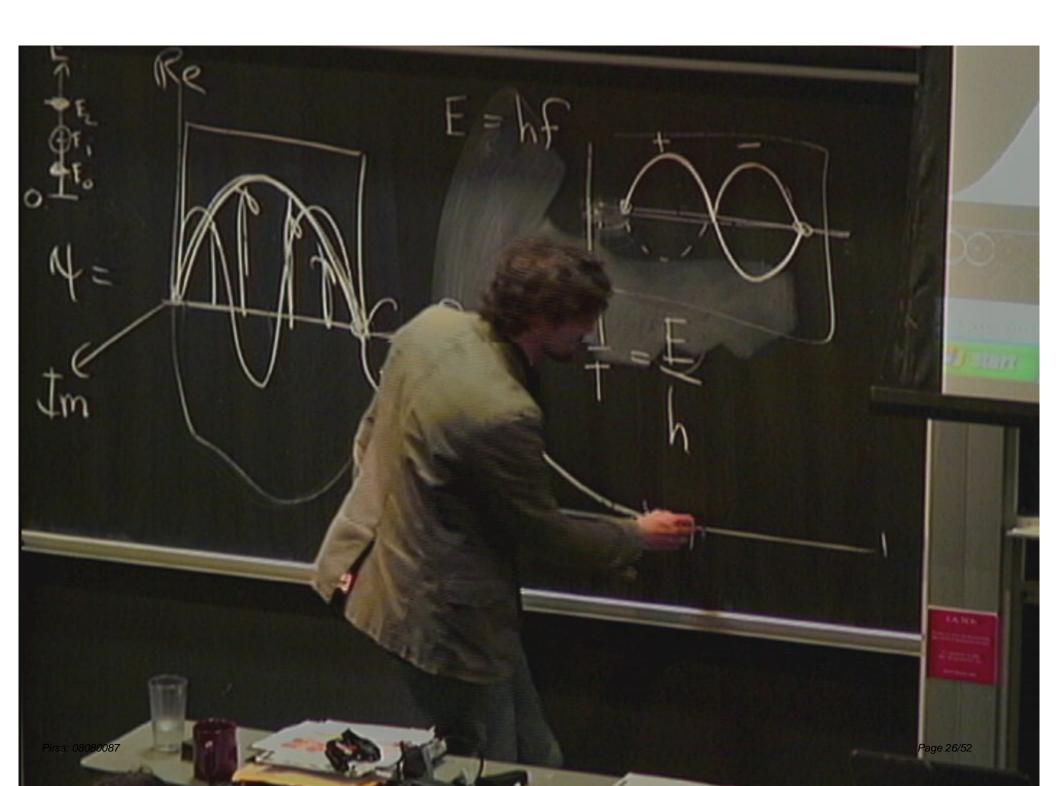


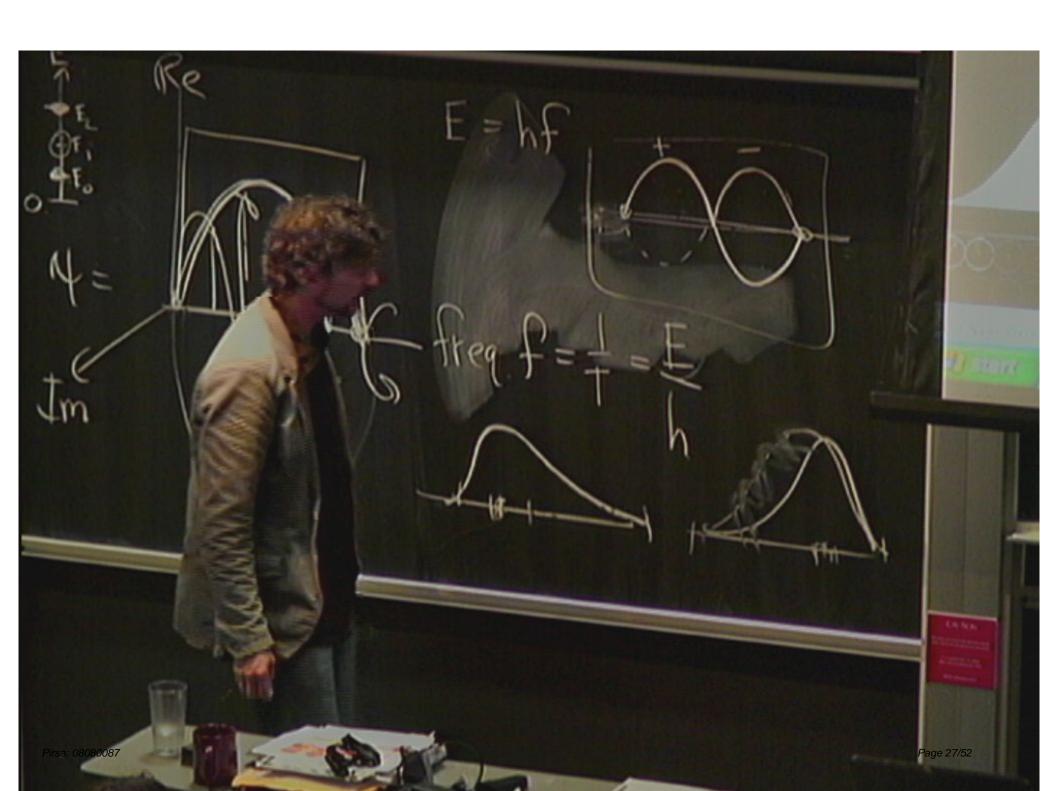


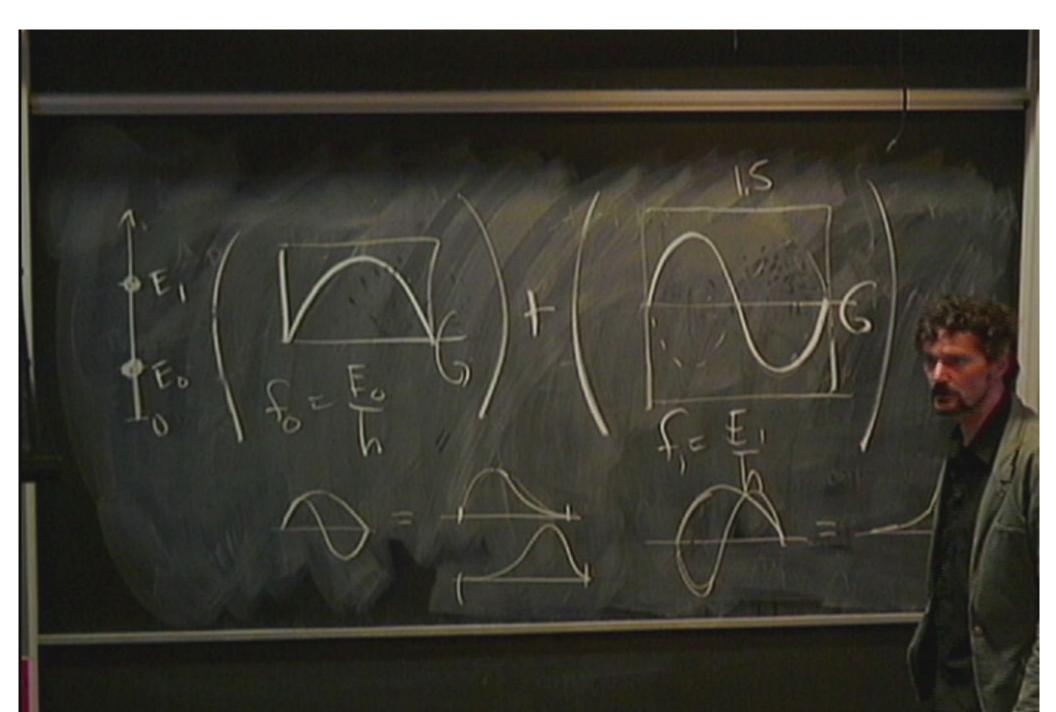






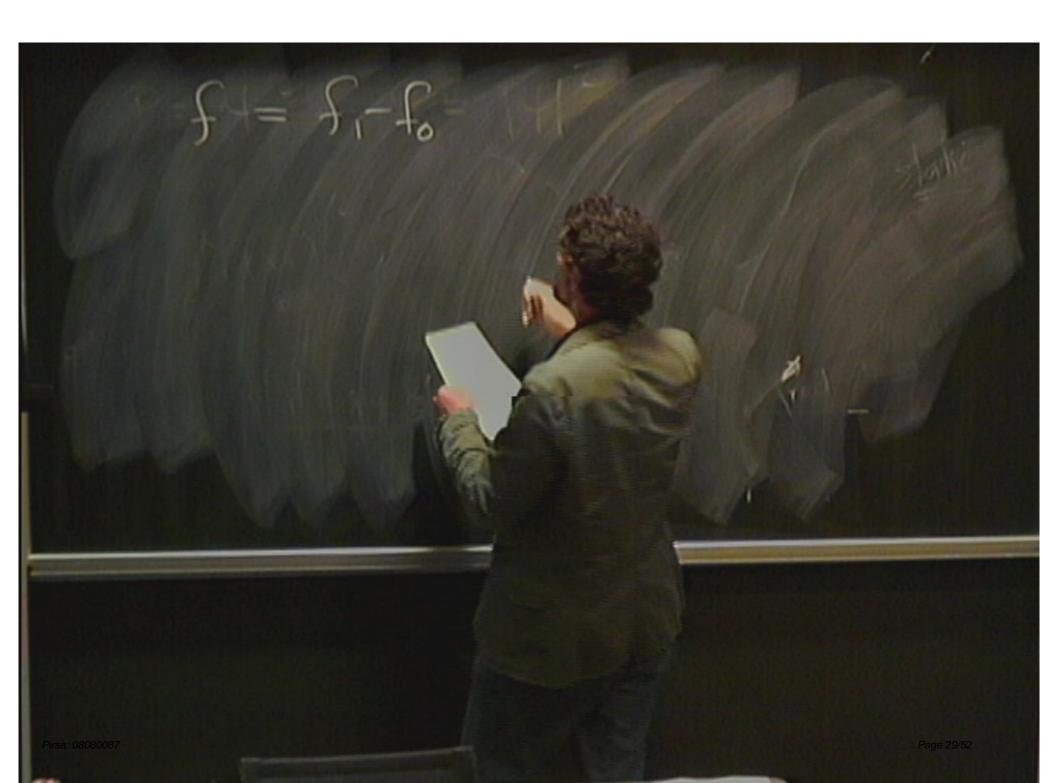






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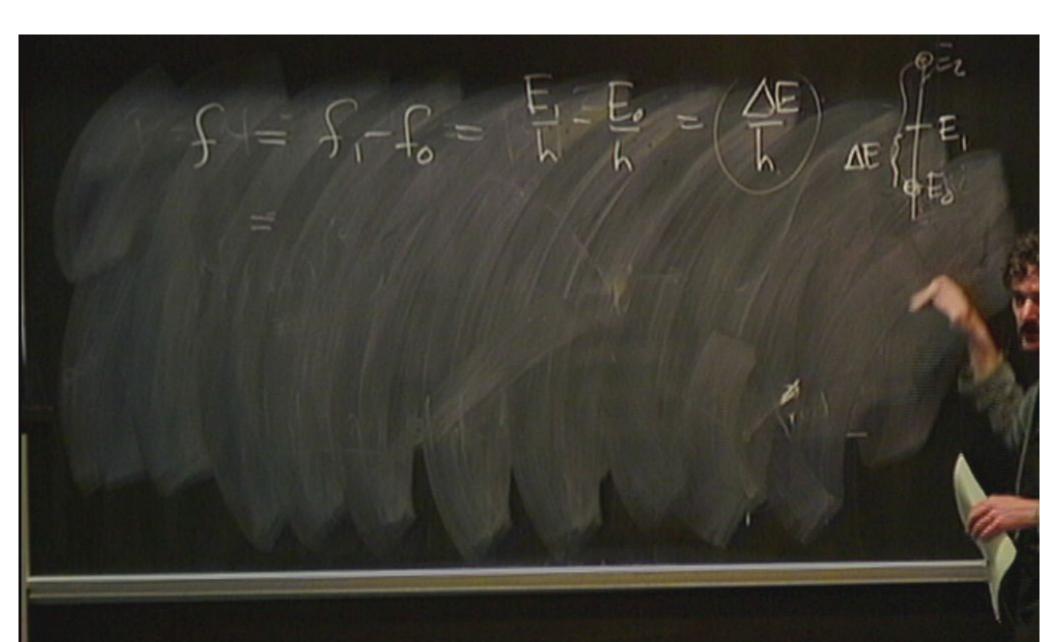
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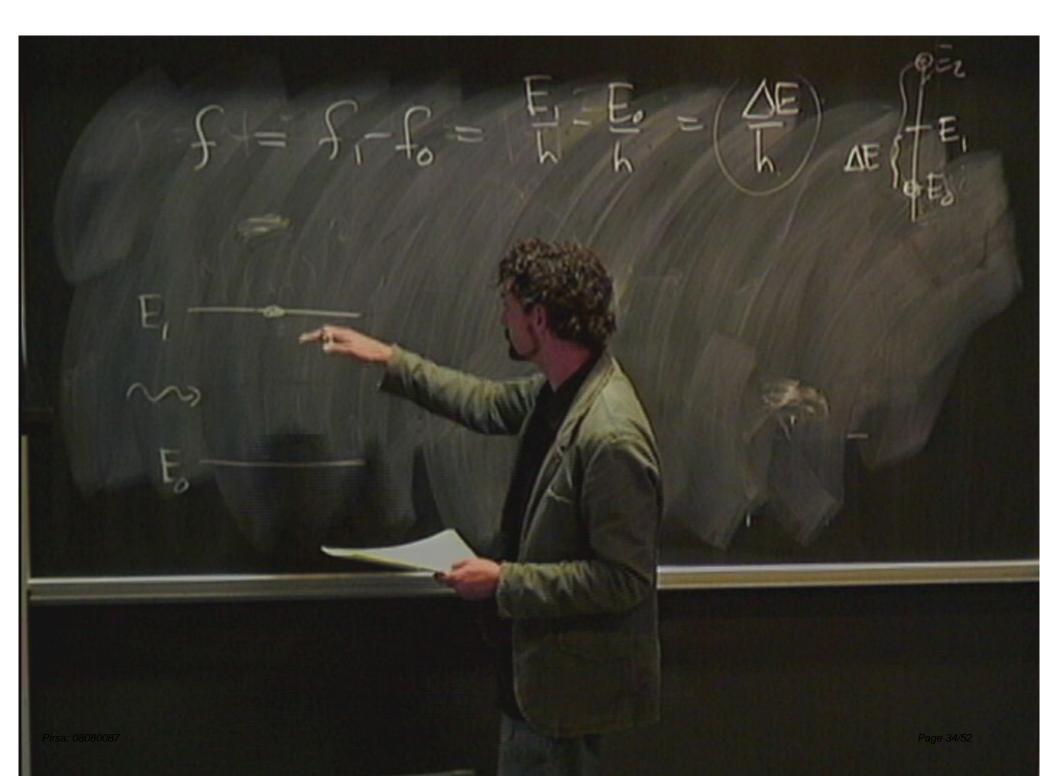
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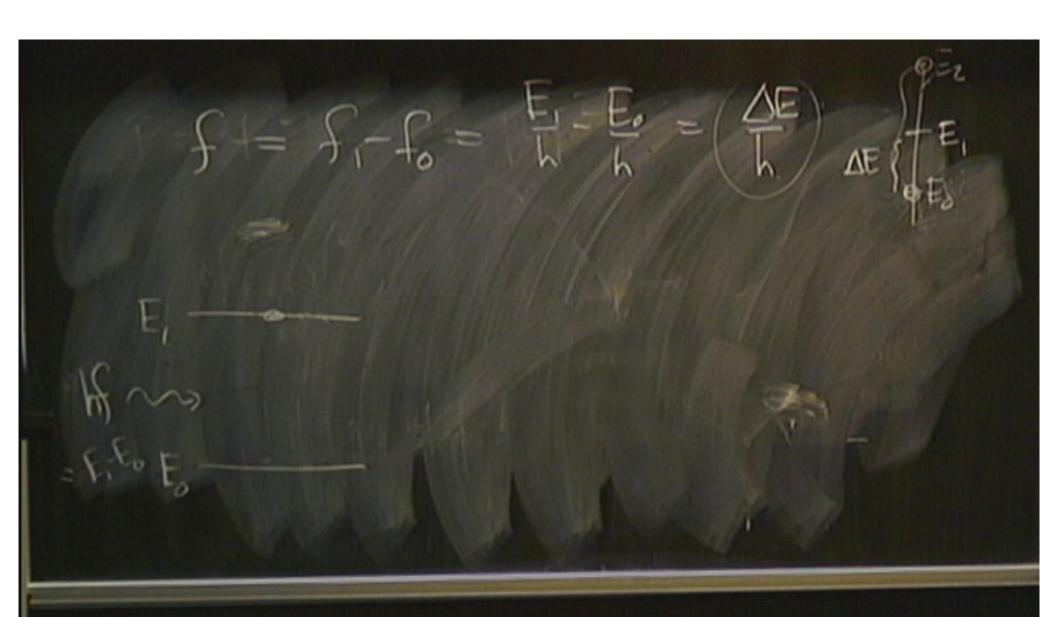


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= f,-fo= 5



