

Title: Heisenberg's Uncertainty Principle

Date: Aug 16, 2008 09:00 AM

URL: <http://pirsa.org/08080022>

Abstract: We will review the uncertainty principle of quantum mechanics, first formulated by Werner Heisenberg in 1927, and the role they played in the famous debate between Einstein and Bohr on the meaning of quantum theory. Along the way we will focus on questions like: what do we mean by "uncertainty", and how do we express that in the theory? What, in fact, is a physical property? Does a theory like quantum mechanics provide a description of physical reality? Interestingly, some of these questions do not have a unique answer.





particle / wave duality

particle / wave duality

momentum

$$p = \frac{h}{\lambda}$$

Energy

$$E = h\nu$$

particle / wave duality

momentum

$$p = \frac{h}{\lambda}$$

Energy

$$E = h\nu$$

(De Broglie)
1924.

particle / wave duality

momentum

$$p = \frac{h}{\lambda}$$

Energy

$$E = h\nu$$

(De Broglie)
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localized.

particle / wave duality

momentum

$$p = \frac{h}{\lambda}$$

Energy

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(De Broglie)
1924.

localized.

spread out

particle / wave duality

momentum

Energy

localized.

$$p = \frac{h}{\lambda}$$
$$E = h\nu$$

spread out

(De Broglie)
1924.

Heisenberg (1927)

particle

momentum

Energy

localized.

Heisenberg (1927)

p

$$E = h \nu$$

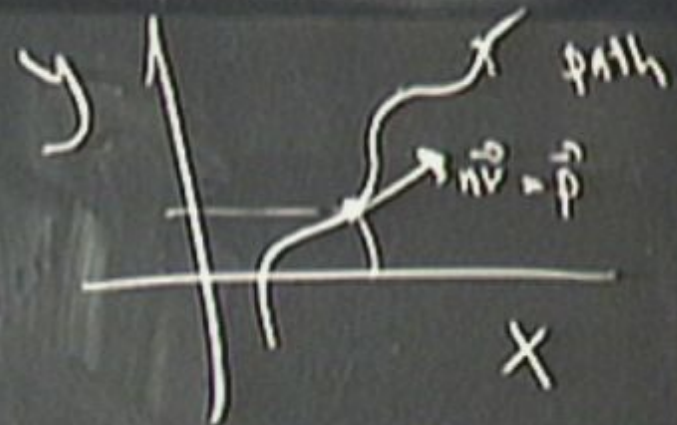
spread out

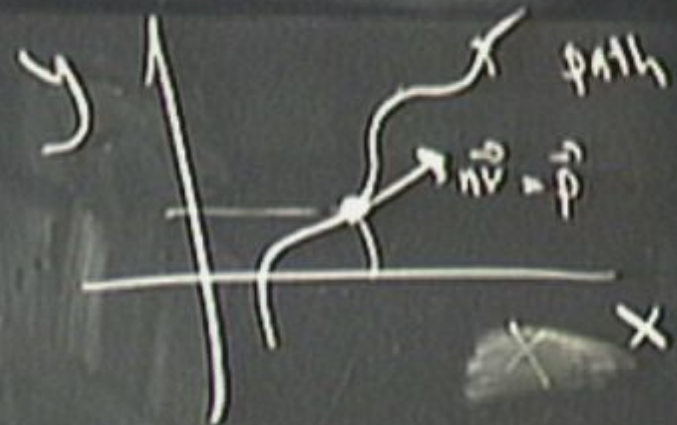
what's path of particle?

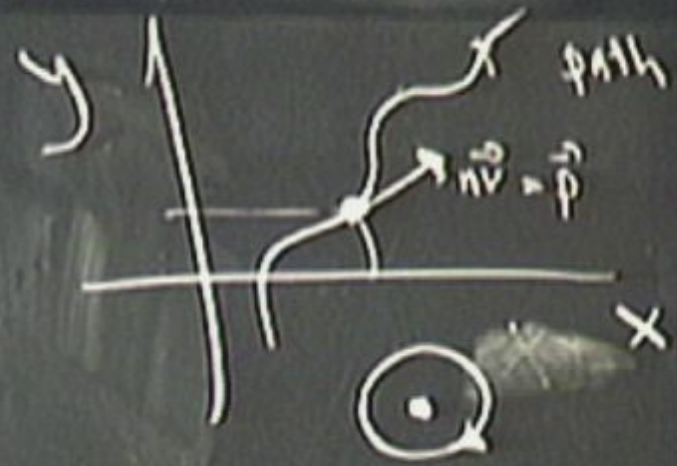
h

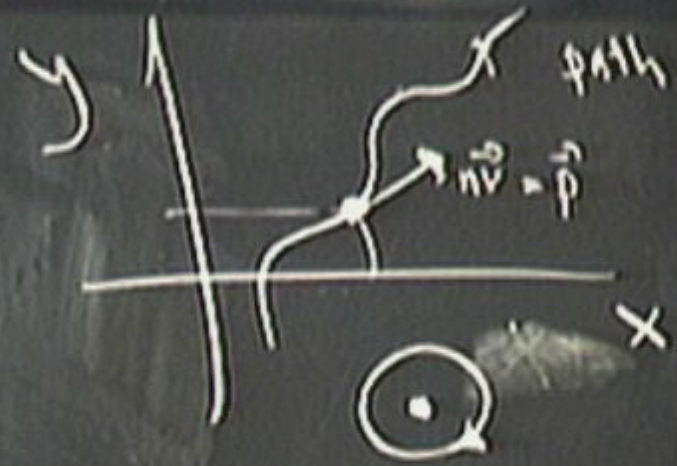
λ

(De Broglie)
1924

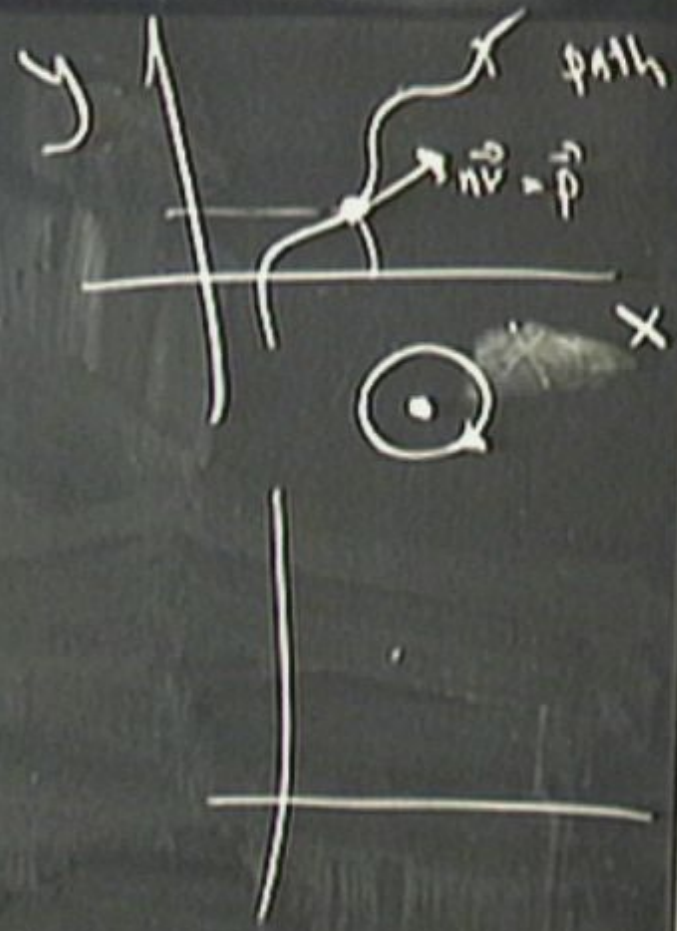


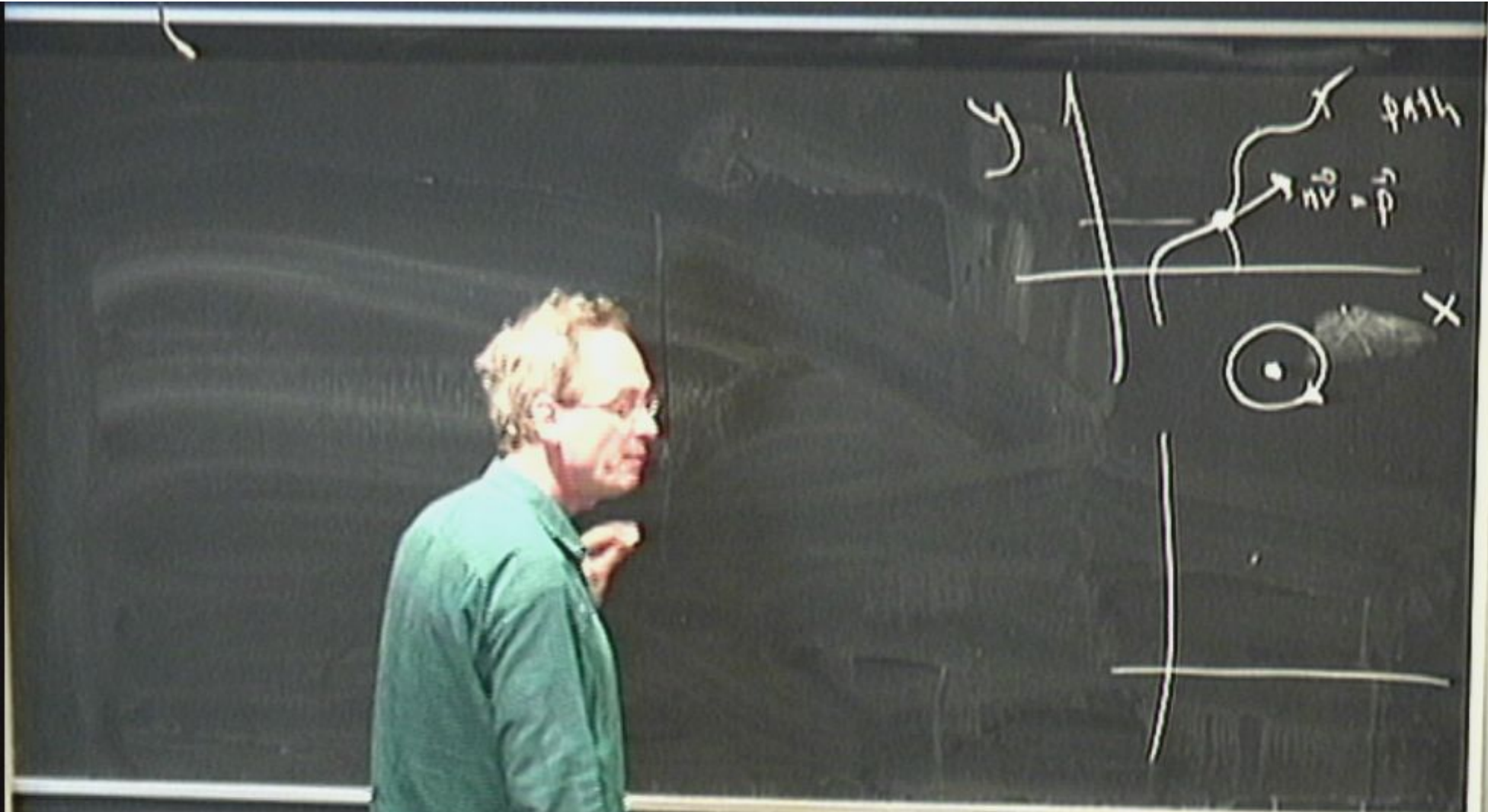


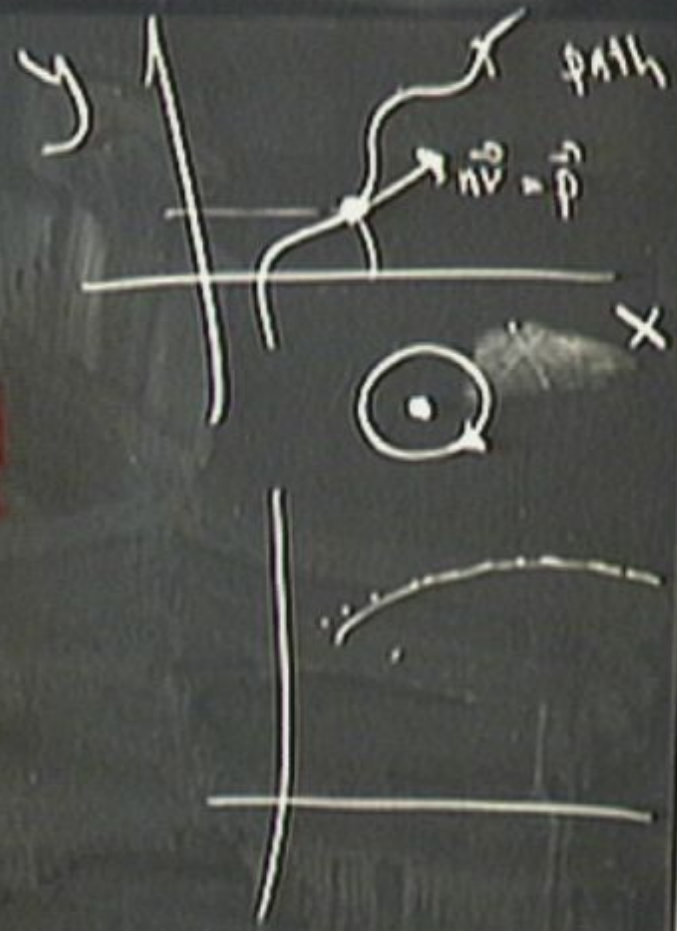


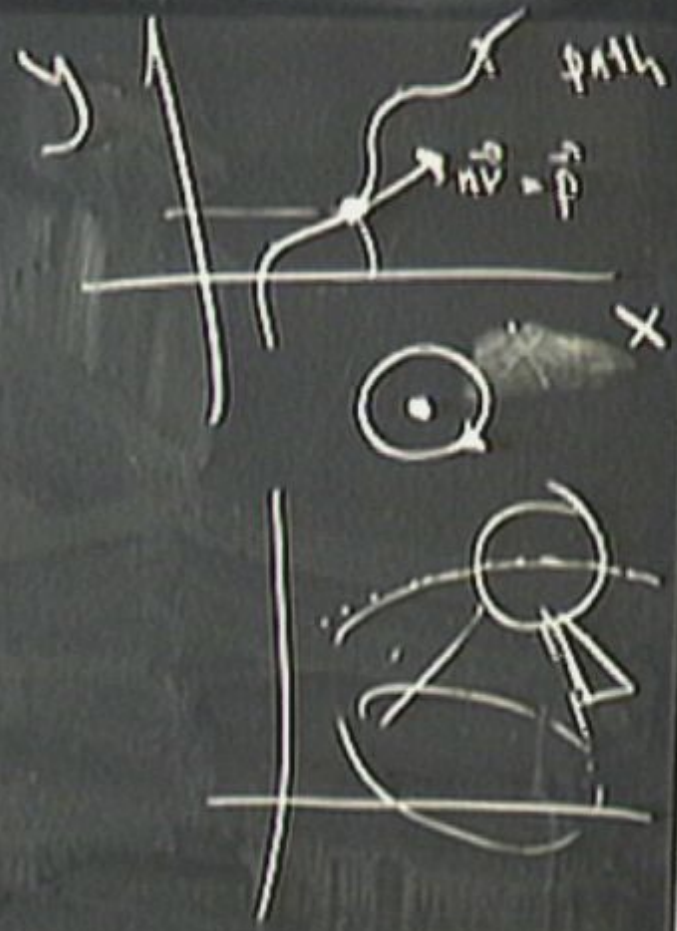


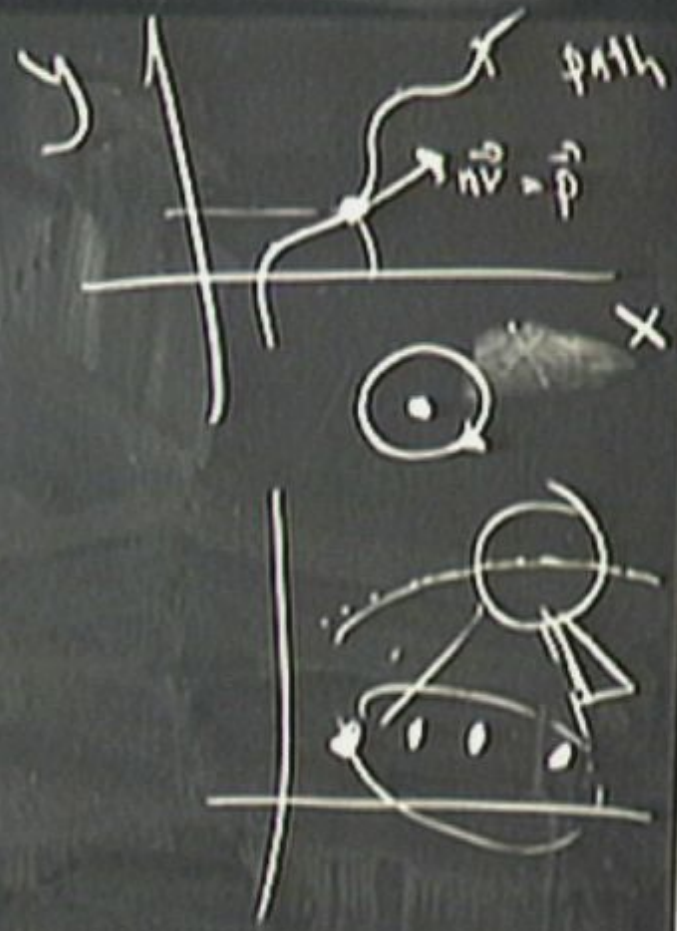
LA 100
UNIVERSITY OF CALIFORNIA
SANTA BARBARA

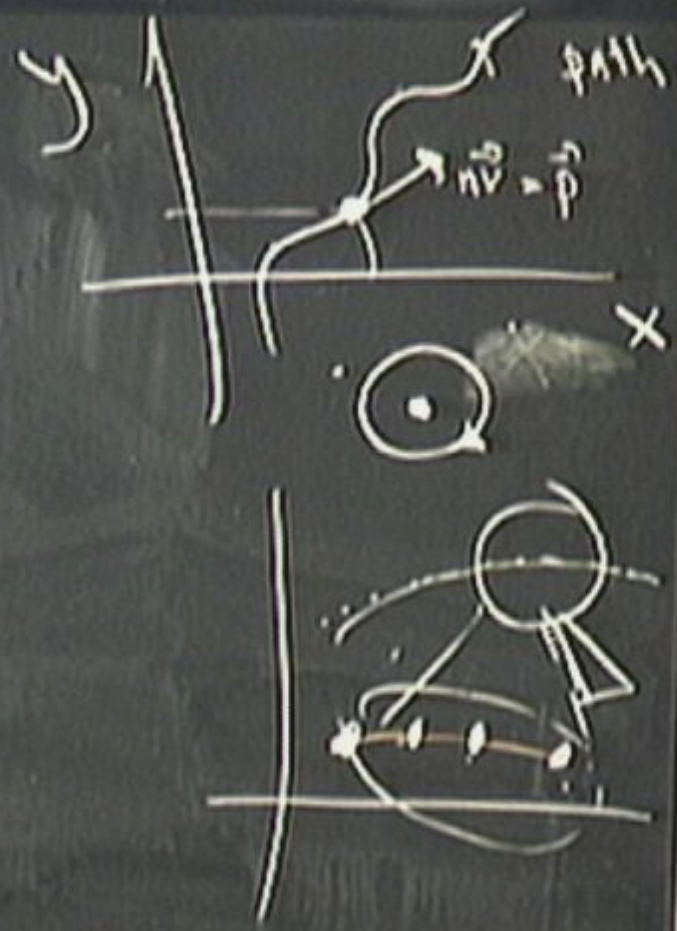


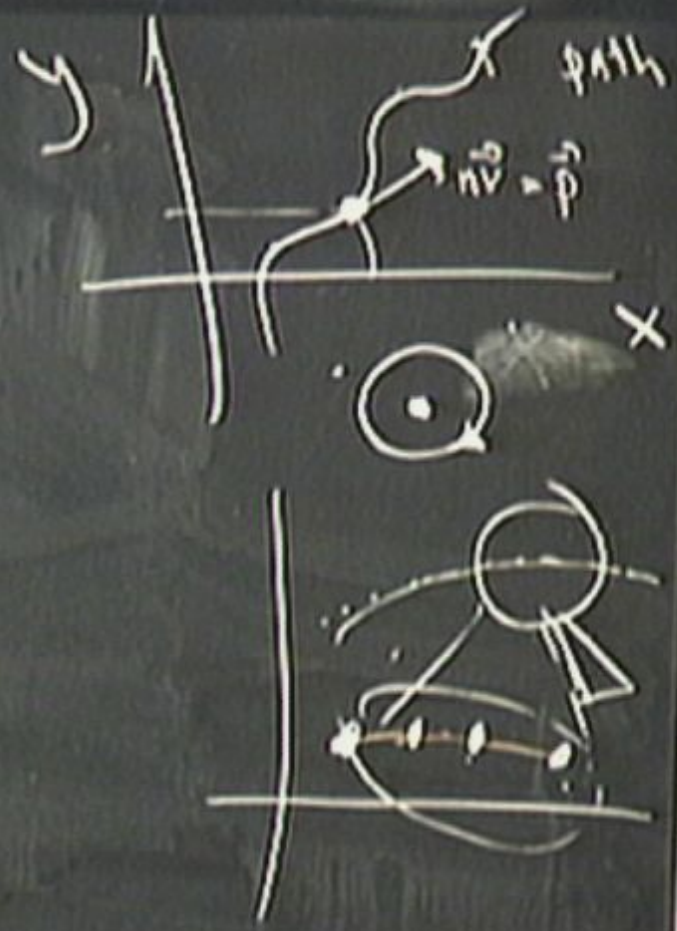
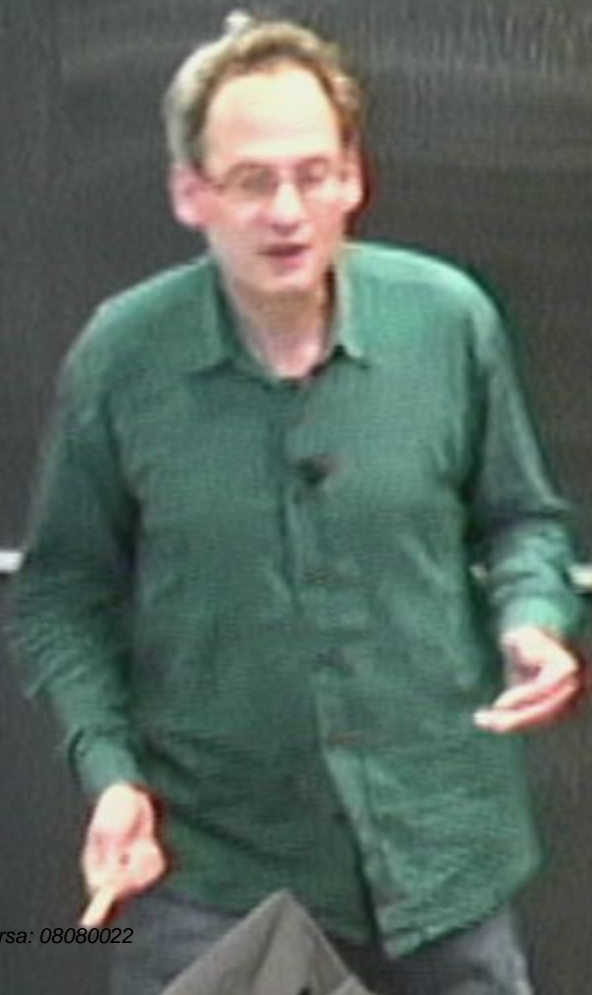


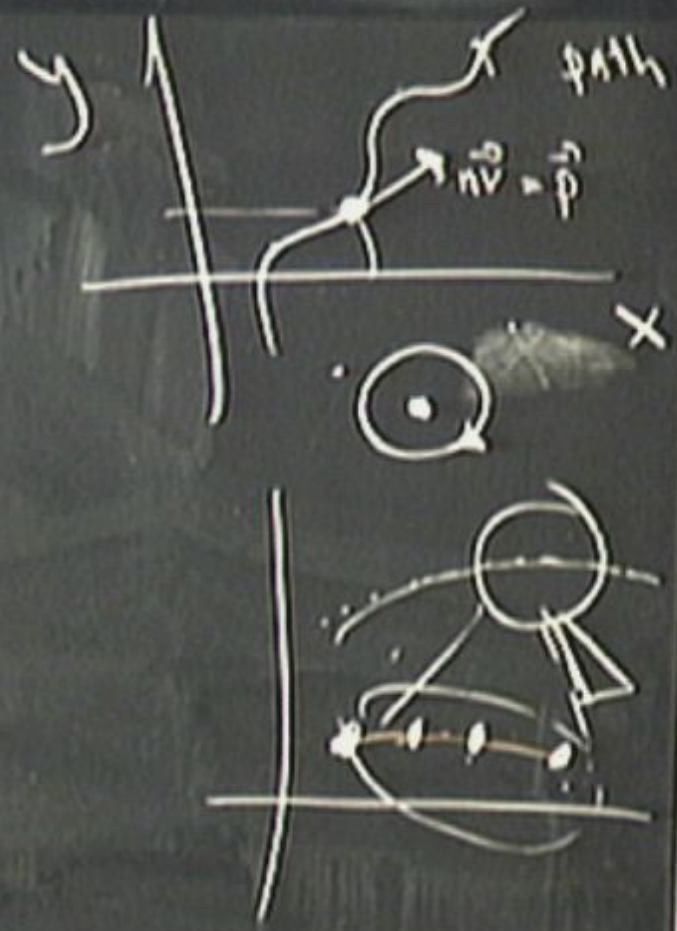




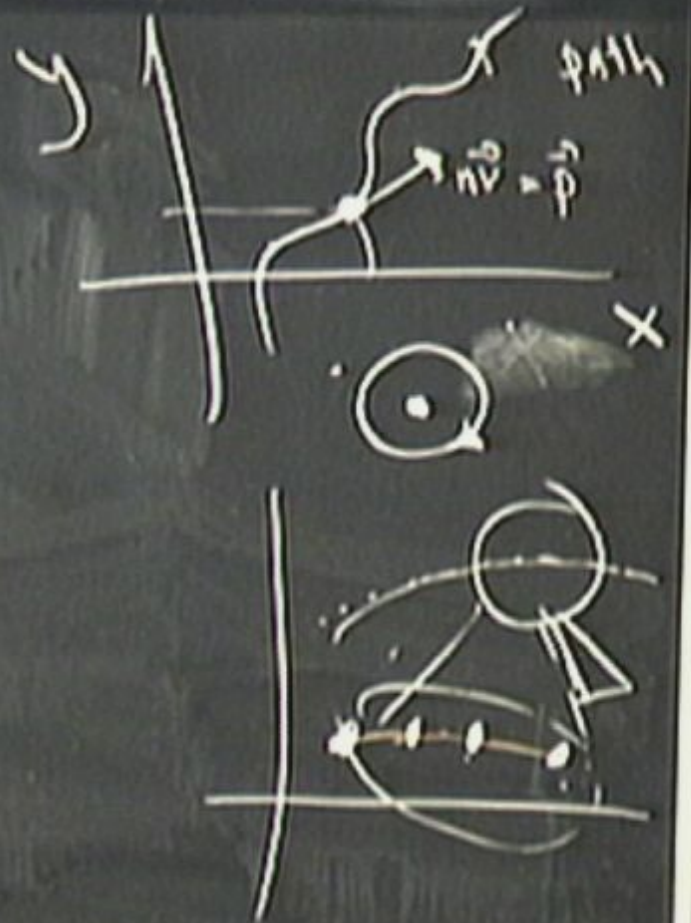






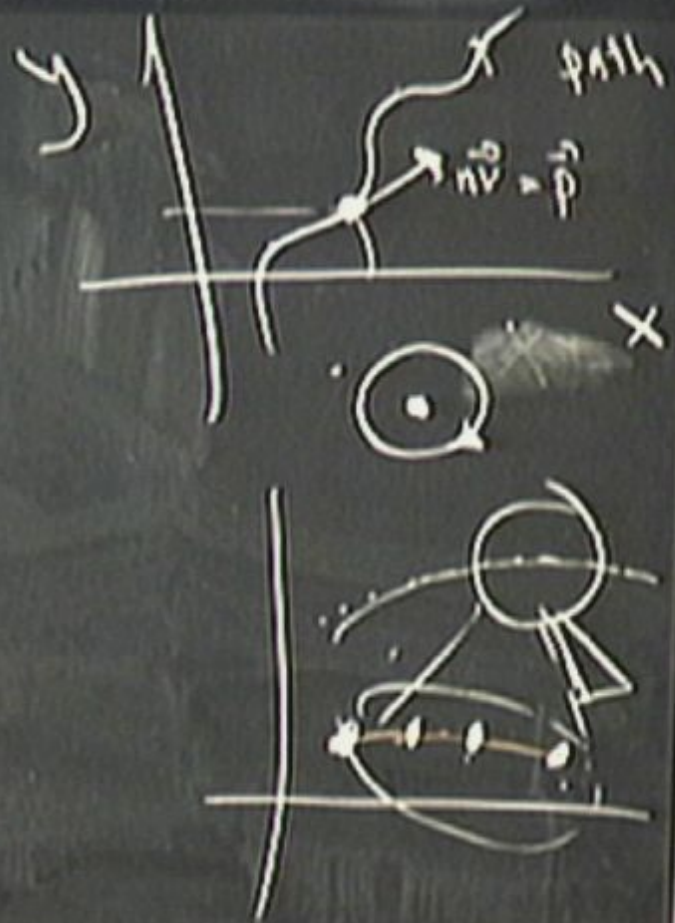


Principle



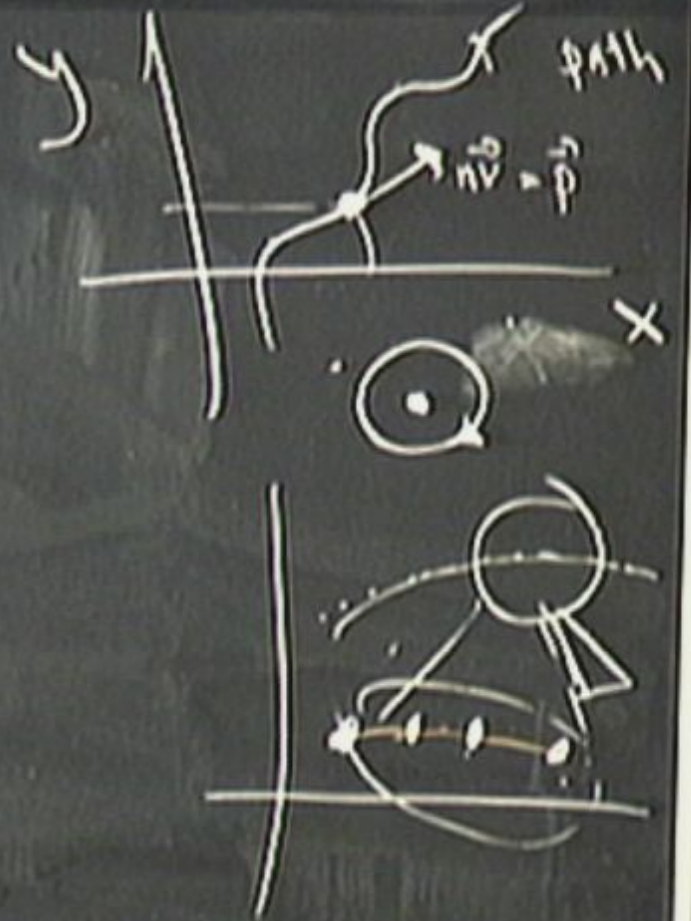
Principle

If you want to give meaning to phrase "position of electron" you must specify a way to observe it.



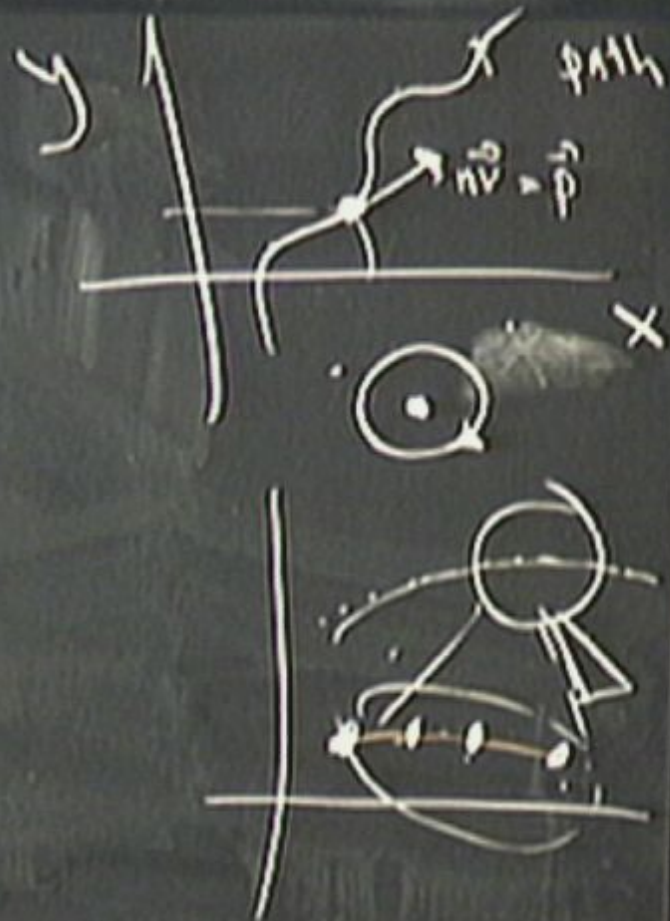
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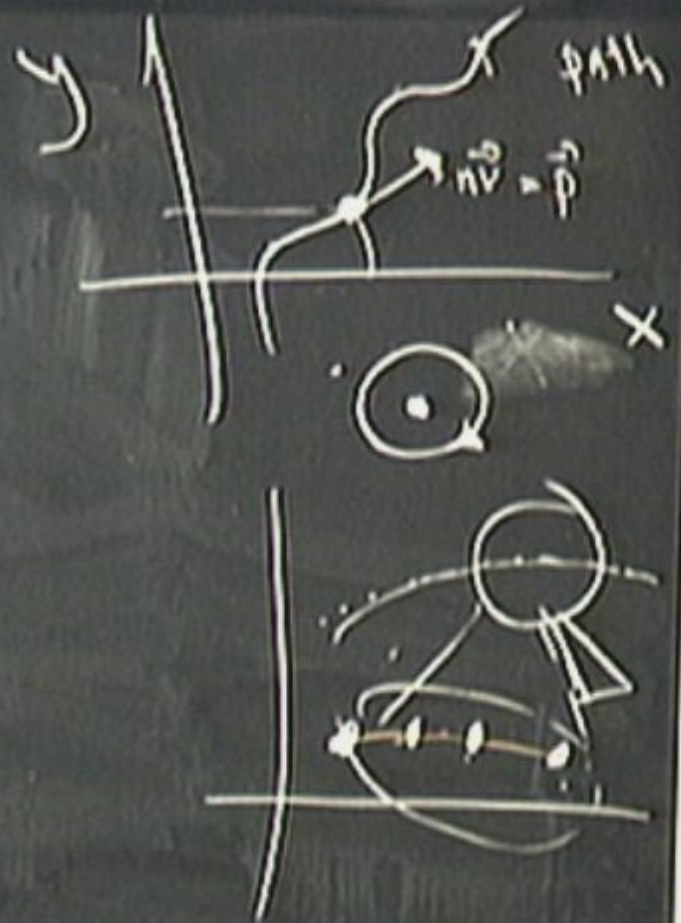
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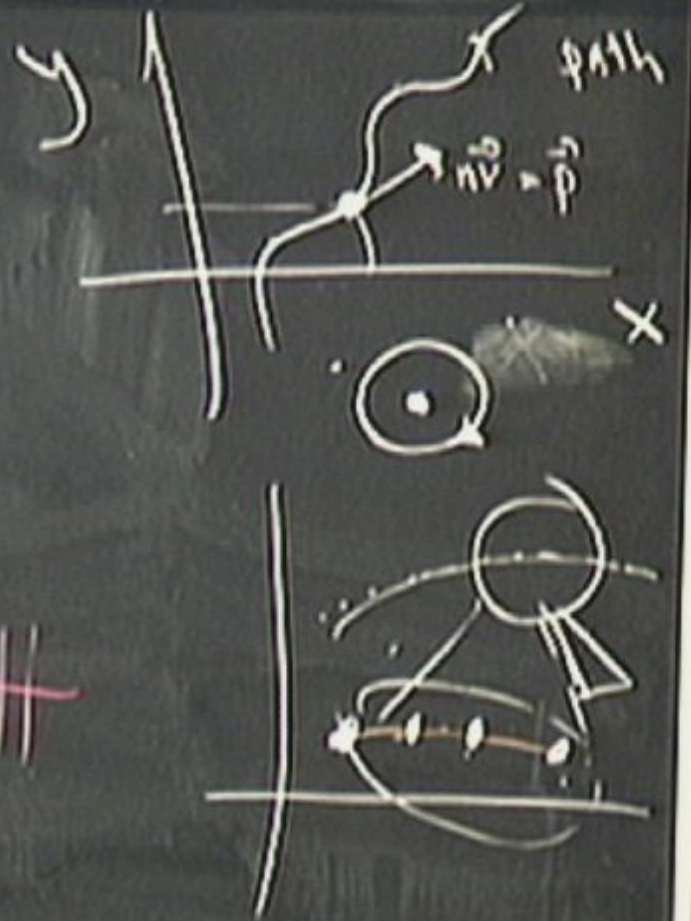
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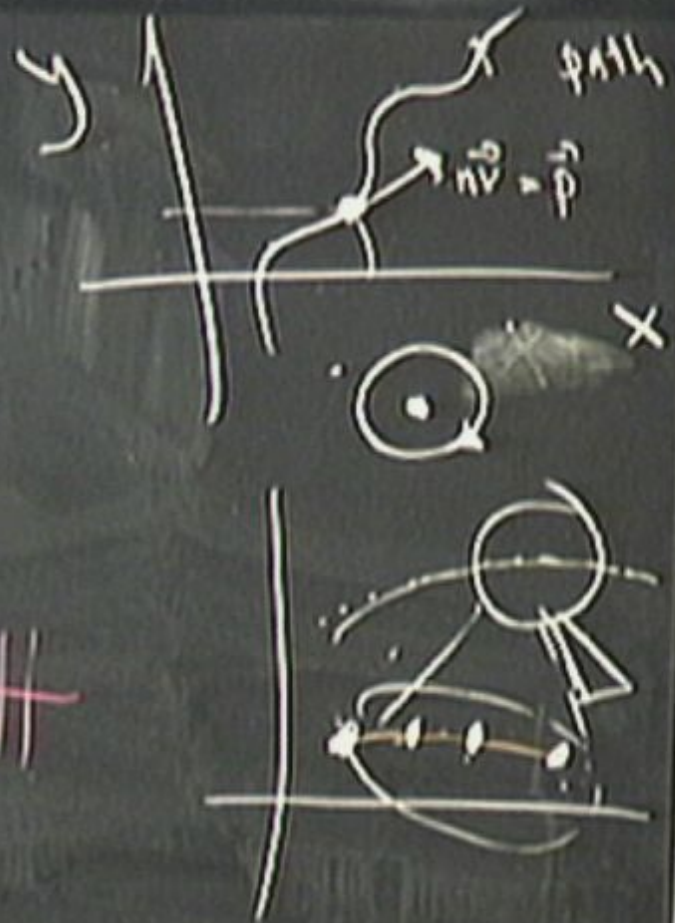
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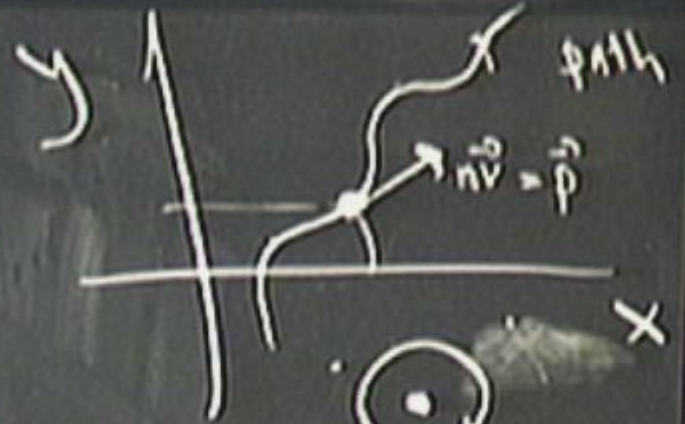
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Principle

If you want to give meaning to phrase
"position of electron" you must
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resolving power
 Δx

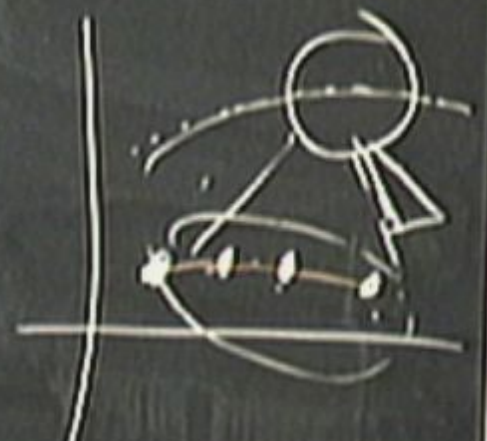
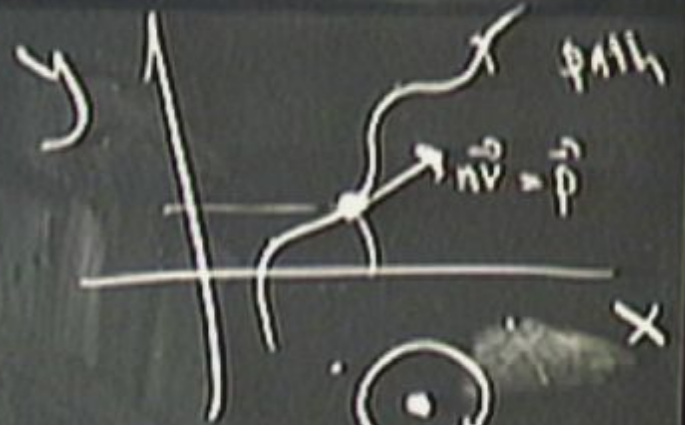


Principle

If you want to give meaning to phrase
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resolving power

$$\Delta x = \frac{\lambda}{\sin \theta}$$

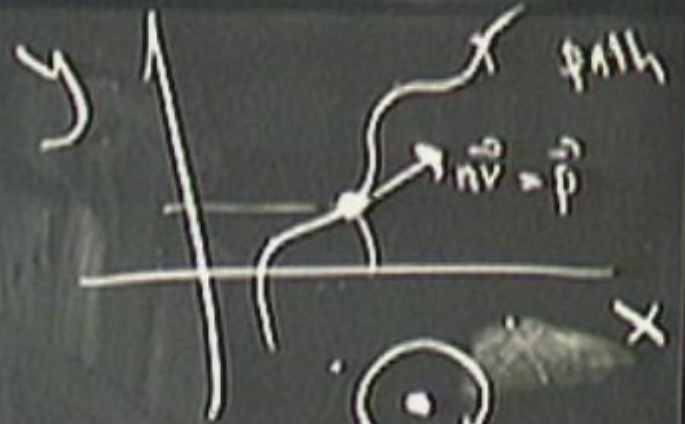


Principle

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Heisenberg principle

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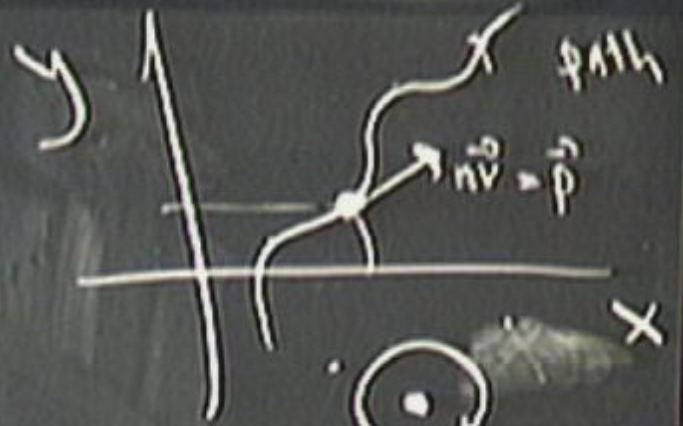


Principle

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$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

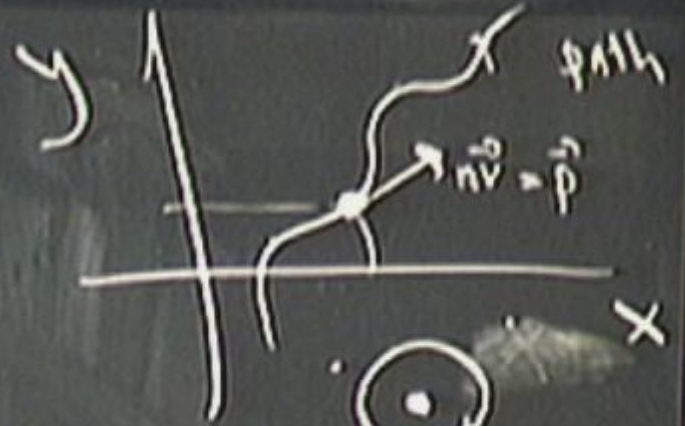


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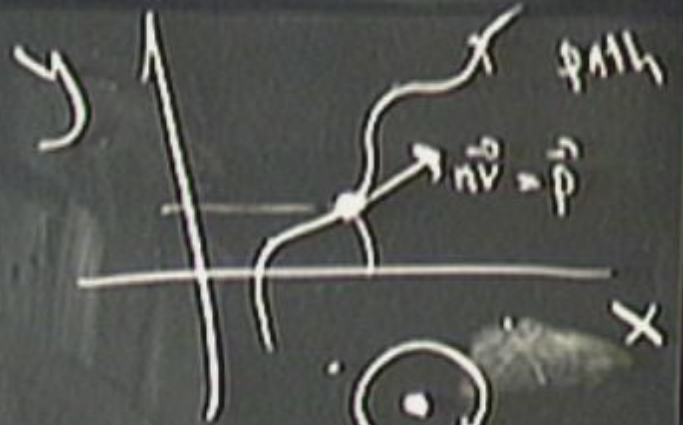


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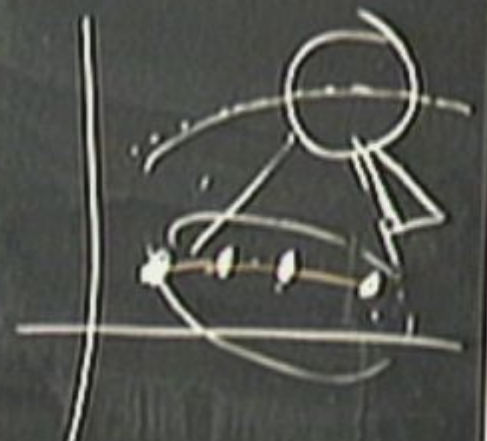
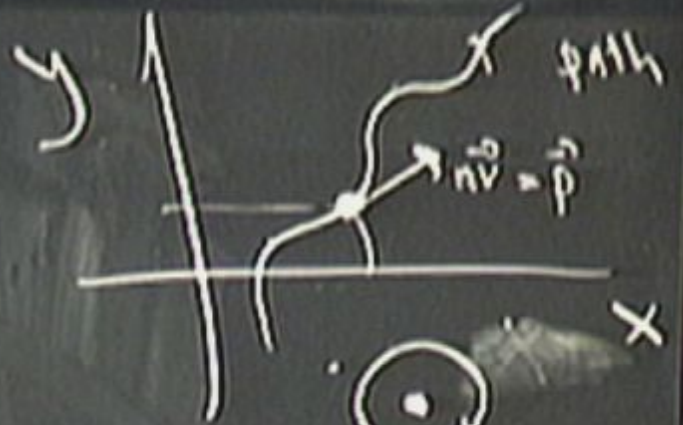


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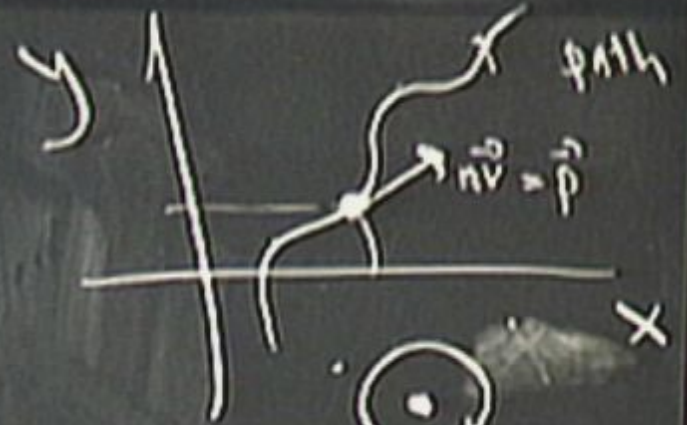
Principle

If you want to give meaning to phrase
"position of electron" you must
specify a way to ~~write~~ write it.

resting power

$$\delta x =$$

use y



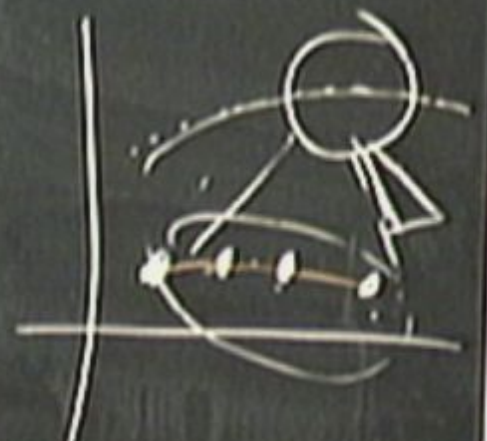
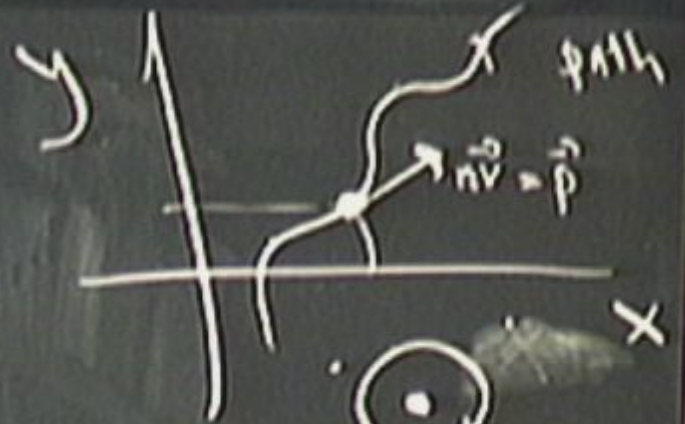
Principle

If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

resolving power

$$\Delta x = \frac{\lambda}{\sin \theta}$$

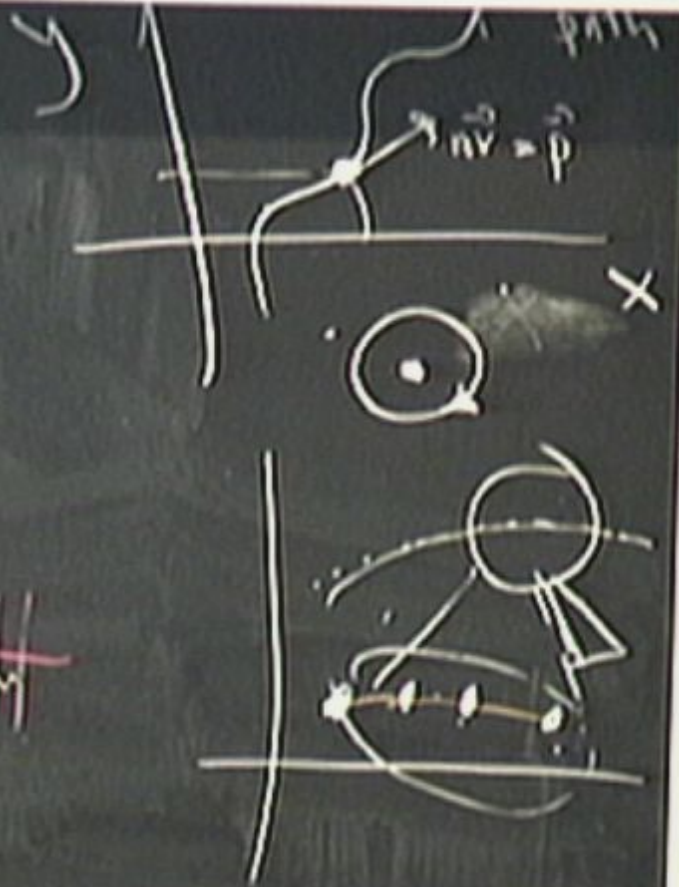
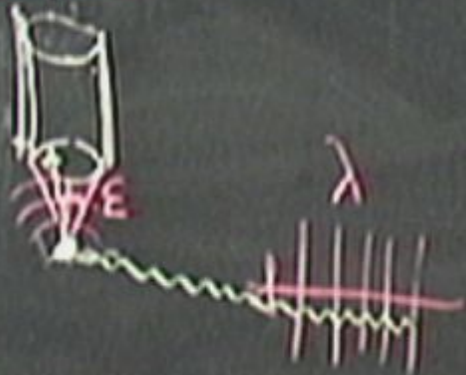
use γ -rays



Principle

If you want to give meaning to phrase
"position of electron" you must
find a way to observe it.

$\Delta x = \frac{\lambda}{\sin \theta}$
x-rays!



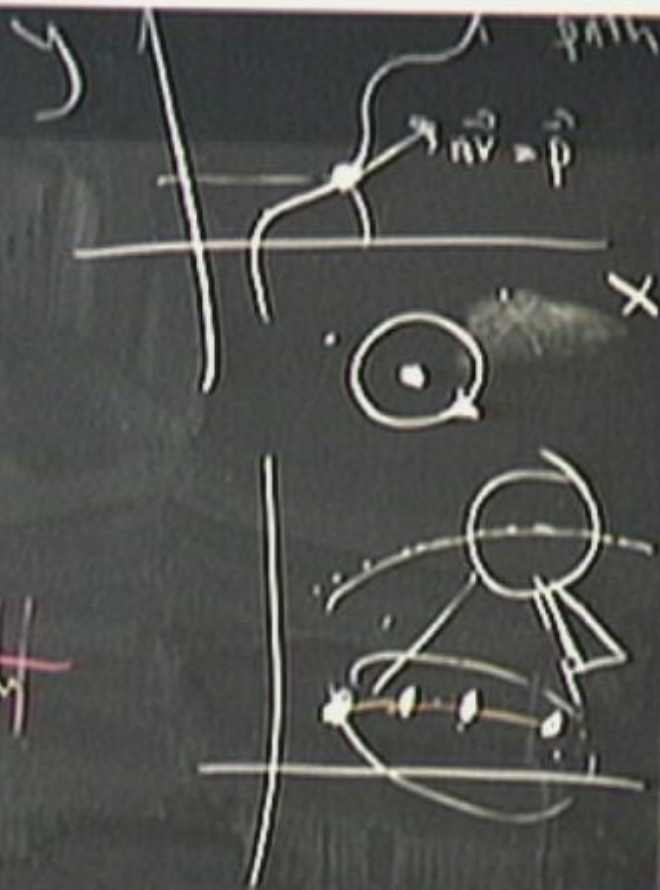
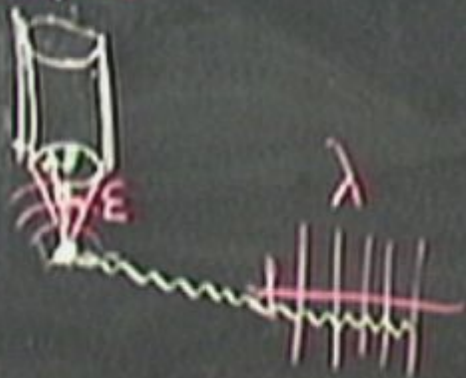
Principle

If you want to give meaning to phrase
"position of electron" you must
specify a way to observe it.

resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

use γ -rays!



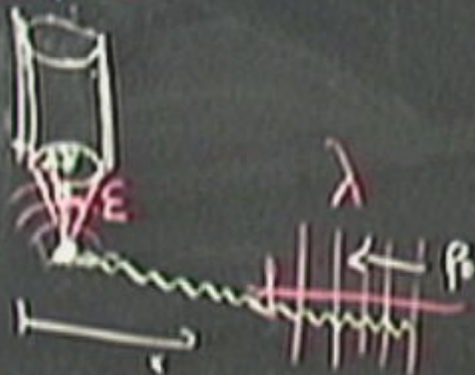
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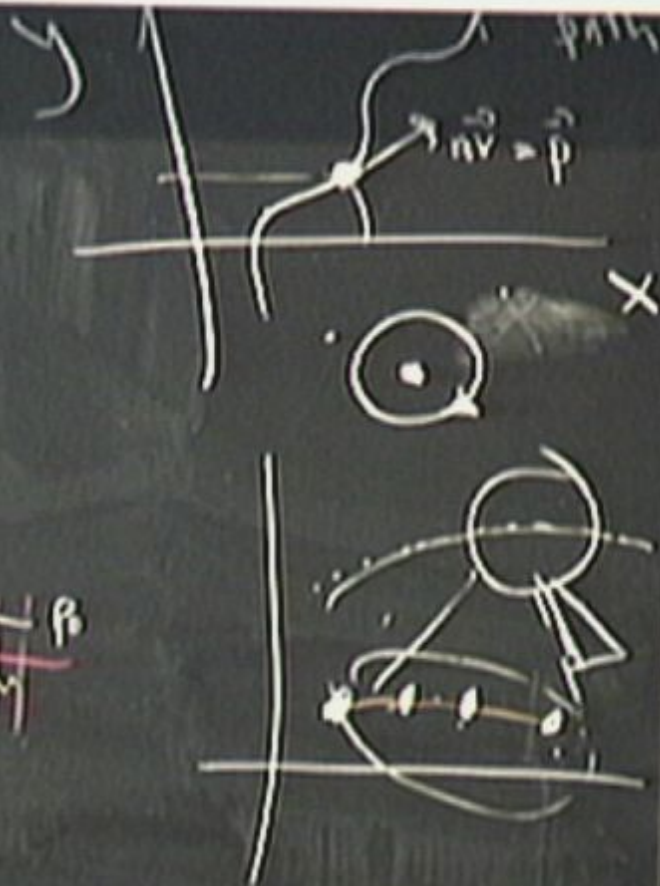
resolving power

$$\Delta x = \frac{\lambda}{\sin \theta}$$

use γ -rays!



$$p_x = p_0 \sin \theta$$



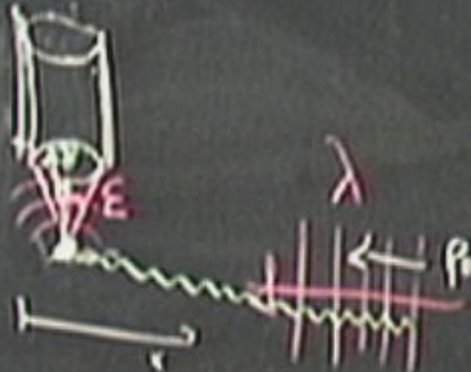
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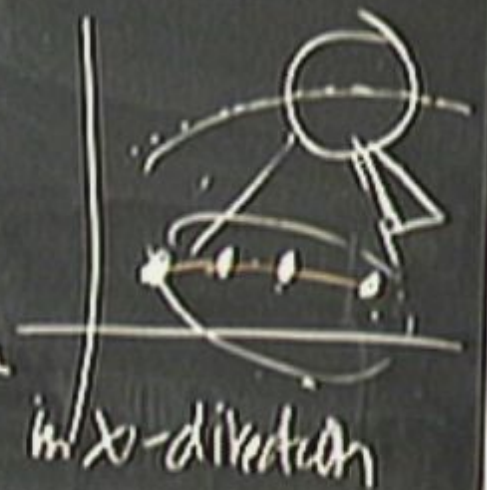
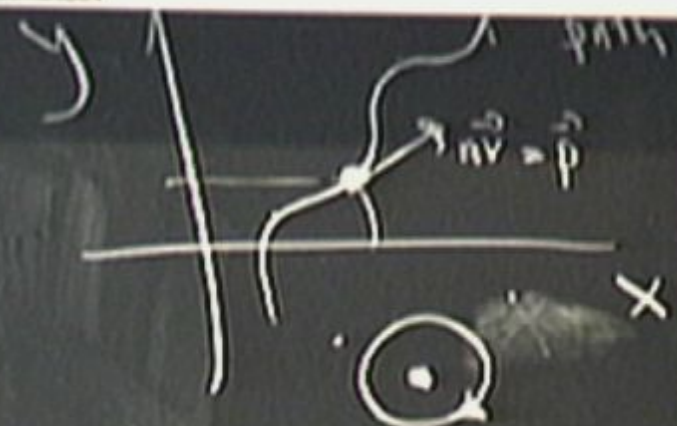
resolving power

$$\Delta x = \frac{\lambda}{\sin \theta}$$

use γ -rays!



$p_x = p_0 \sin \theta$ momentum of electron after collision with electron

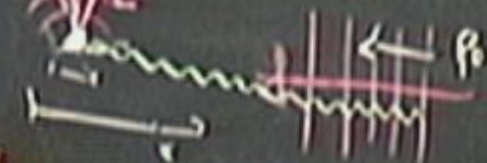


Principle

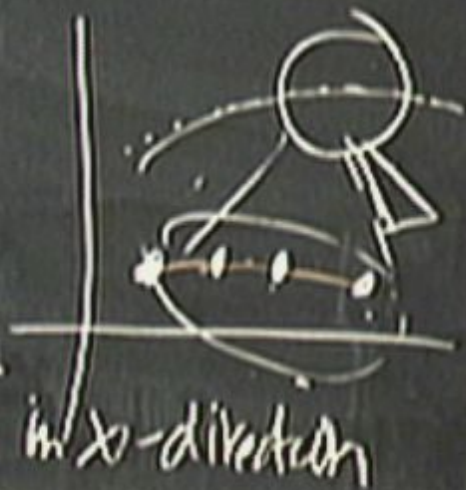
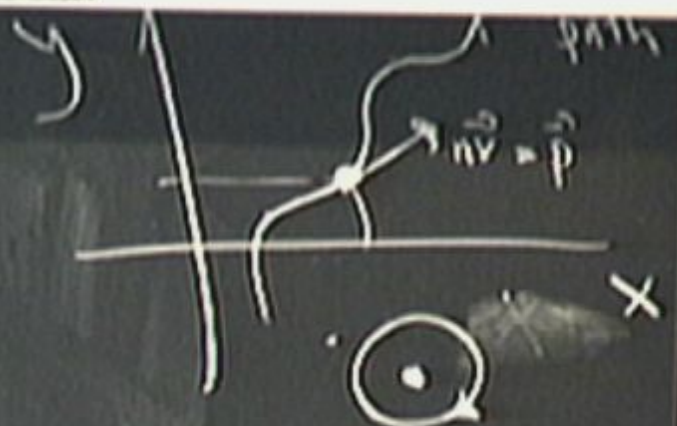
If you want to give meaning to phrase "position of electron" you must specify how to observe it.

resolving

Δx
 Δp
 ΔE



$p_0 \sin \theta$ Number of photons after reflection with electron



Principle

If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

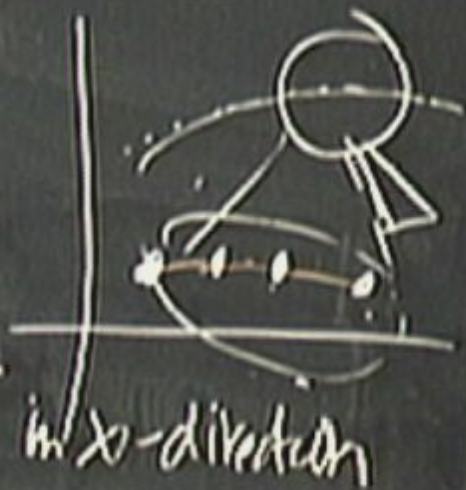
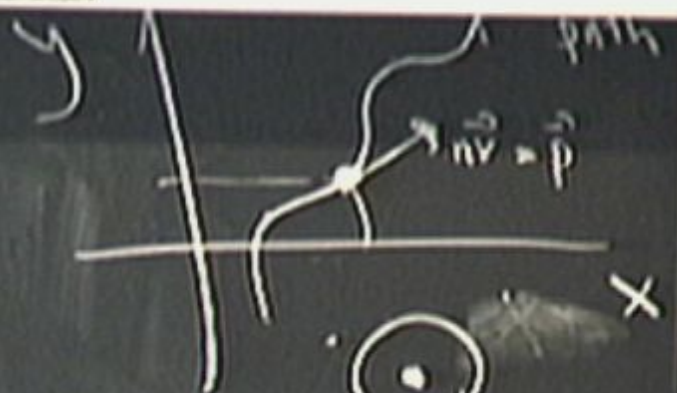
resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

Use γ -rays $|\theta| < \epsilon$

$$p_x = p_0 \sin \theta$$

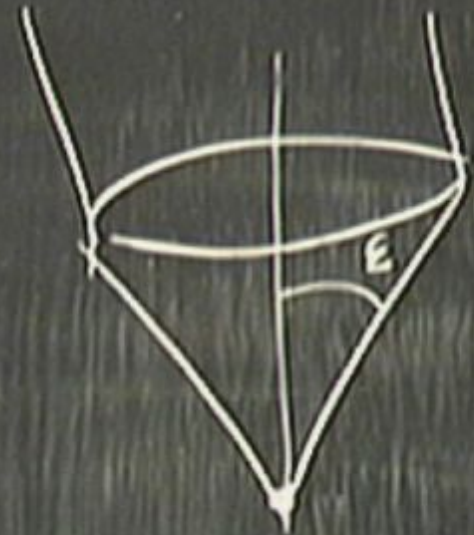
Number of photons after collision with electron



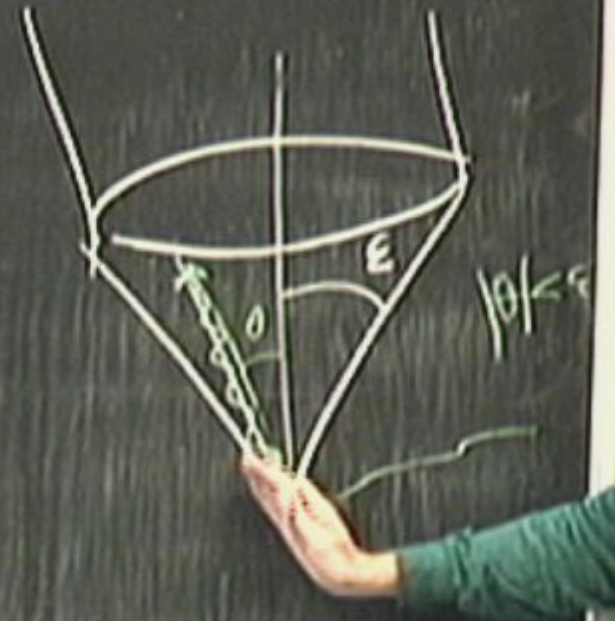
Uncertainty in momentum of photon $\delta p_x(\text{ph})$

uncertainty in momentum of photon $\delta p_x(\text{ph}) \sim p_0 \sin \epsilon$

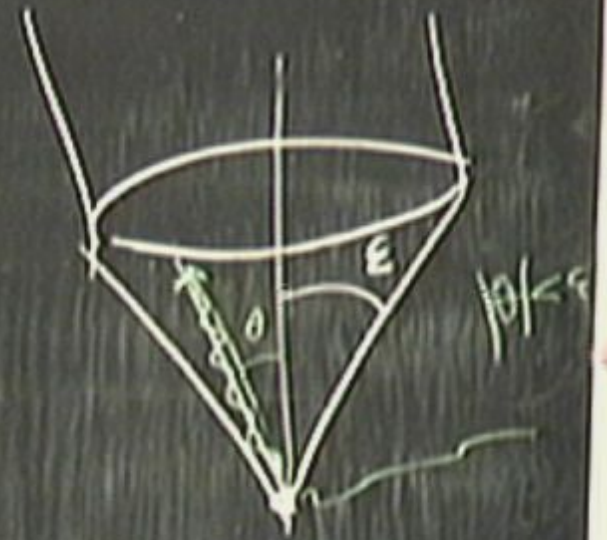
Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$



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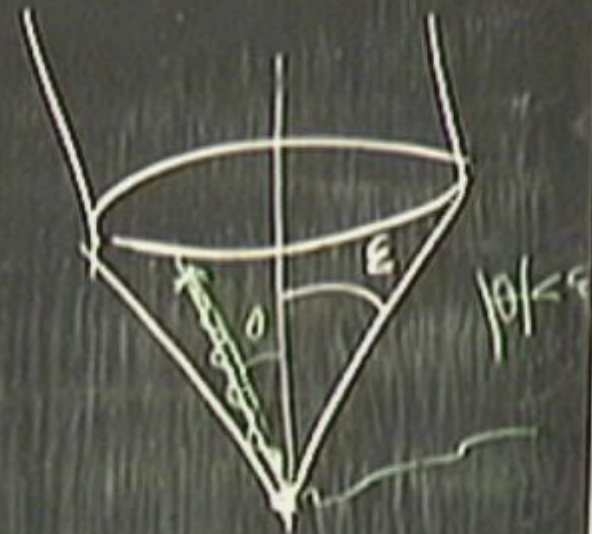
Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$



Uncertainty in momentum of photon $\delta p_x(\text{ph}) \sim p_0 \sin \epsilon$

" " " of detector after collision is

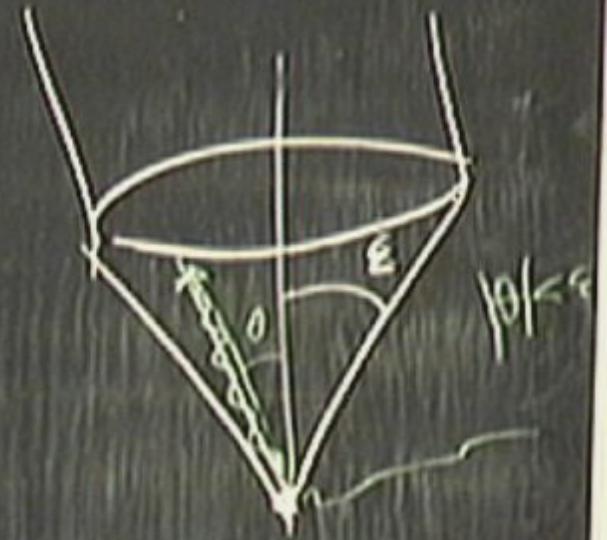
$$\delta p_x \approx p_0 \sin \epsilon$$



Uncertainty in momentum of photon $\delta p_x(\text{ph}) \sim p_0 \sin \epsilon$

" " " of electron after collision is

$$\delta p_x \sim p_0 \sin \epsilon \\ = \frac{h}{\lambda} \sin \epsilon$$



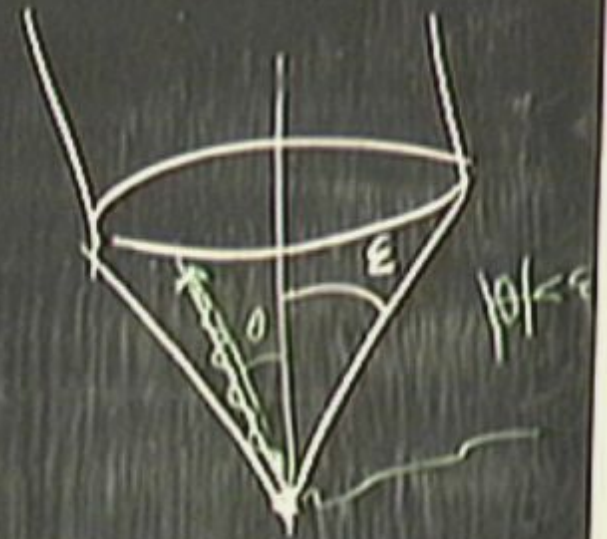
Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$

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$$\delta p_x \sim p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$



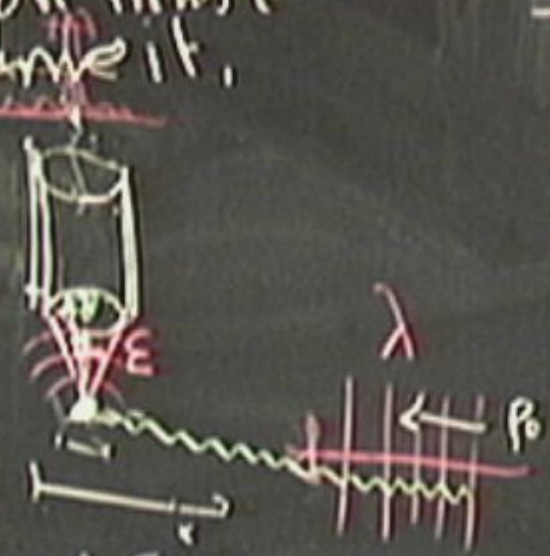
Principle

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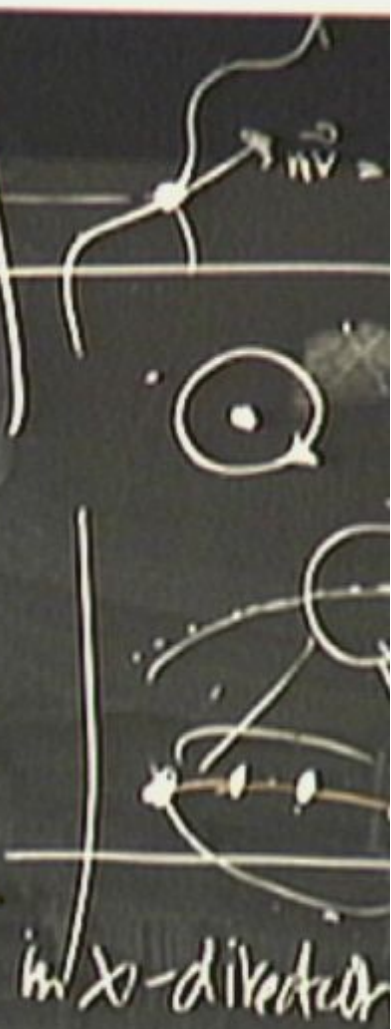
resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

use γ -rays $|\theta| < \epsilon$



$p_x = p_0 \sin \theta$ momentum of photon after collision with electron



Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$

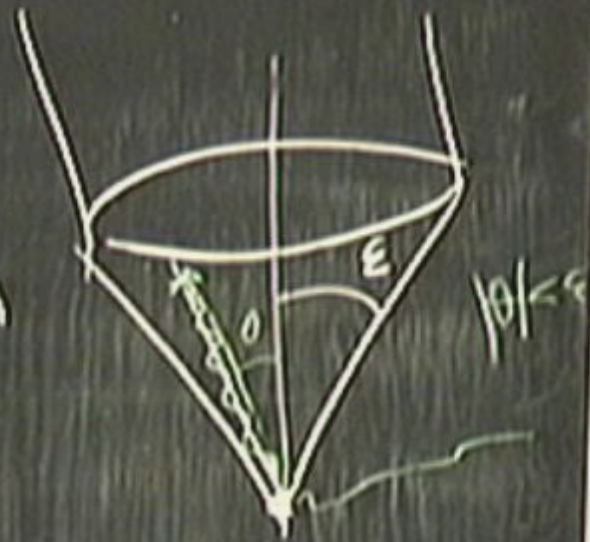
" " " of detector after collision is

$$\delta p_x \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$

$$\delta x \delta p_x \sim h$$



Principle

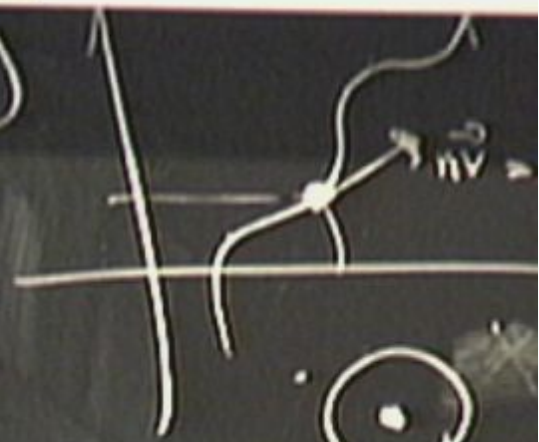
If you want to give meaning to phrase "position of electron" you must specify a way to

resolving power

$$\Delta x = \lambda$$

use γ -ray

$= p_0 - m \cdot v$ momentum of photon after collision with electron



Principle

If you want to give meaning to phase
"position of electron" you must
specify a way ~~to~~ define it.

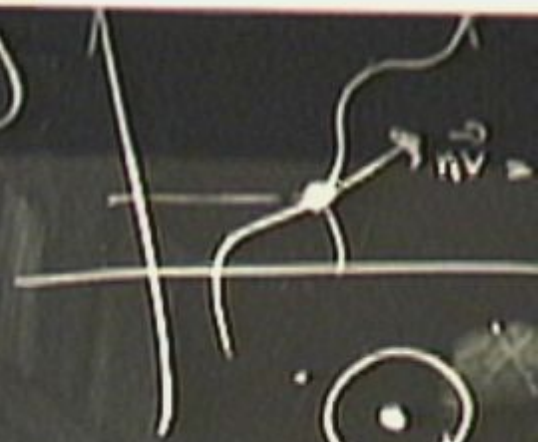
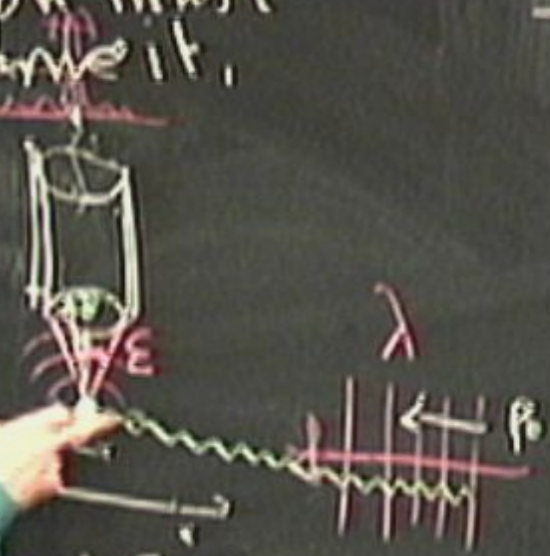
resolving power

$$\Delta x =$$

$$\lambda \sin \theta$$

$= p_0 \sin \theta$ number of photons
after collision with electron

in x -direction



Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$

" " " of detector after collision is

$$\delta p_x \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$



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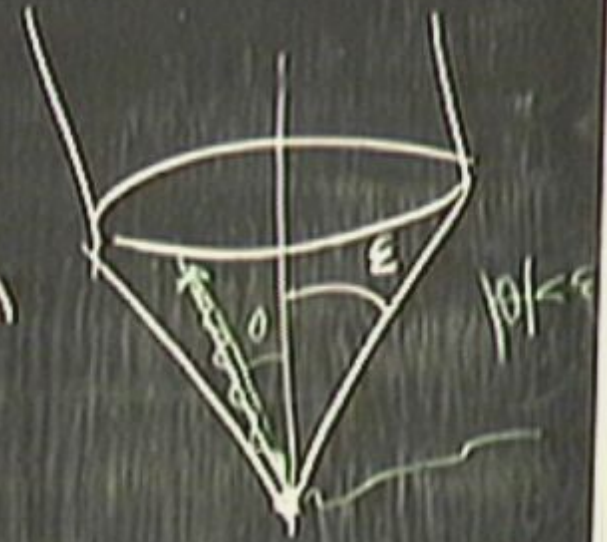
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$$\delta x =$$

$$\delta x \delta p_x \sim h$$



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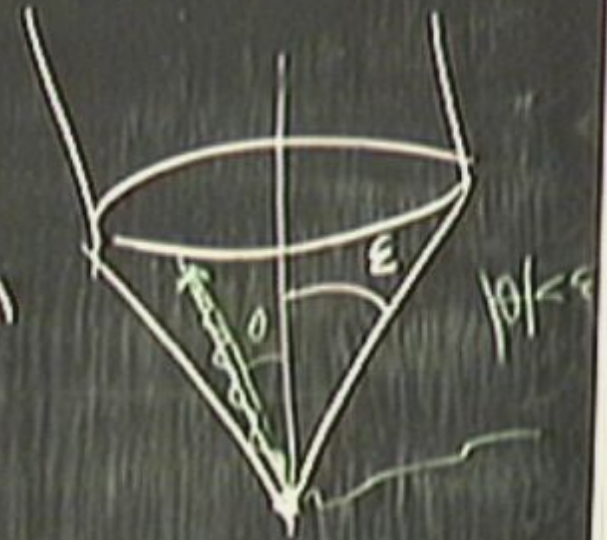
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Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$

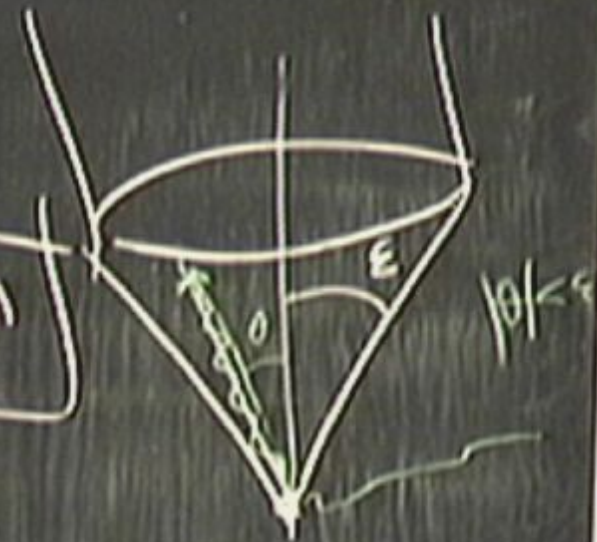
" " " of detector after collision is

$$\delta p_x^{det} \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$

$$\delta x \delta p_x \sim h$$



Uncertainty in momentum of photon $\delta p_x^{(ph)} \sim p_0 \sin \epsilon$

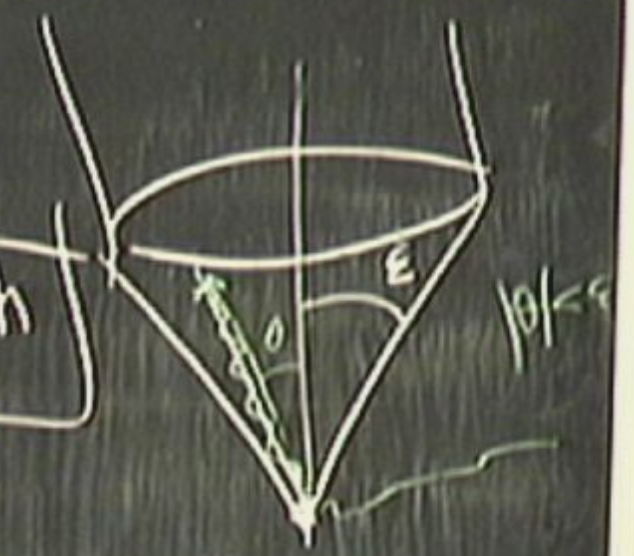
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$$\delta p_x^{(el)} \sim p_0 \sin \epsilon$$

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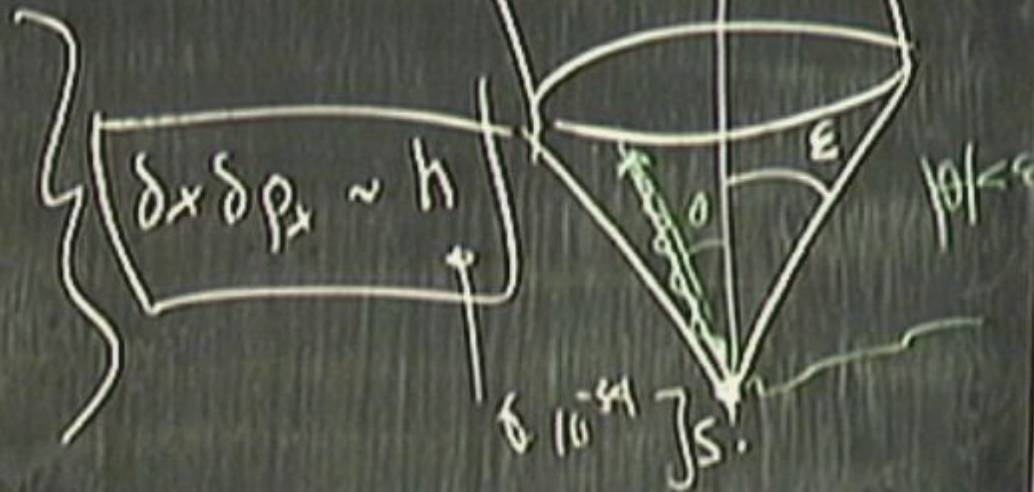
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$$\delta x = \frac{\lambda}{\sin \epsilon}$$



Principle

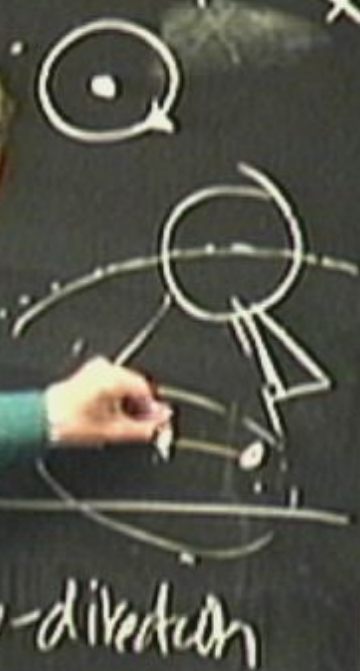
If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

Use γ -rays $|\theta| < \epsilon$

$$P_x = P_0 \sin \theta \text{ after}$$



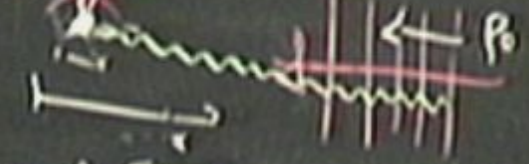
Principle

If you want to give meaning to phrase "position of electron" you must specify way to observe it.

Resolution

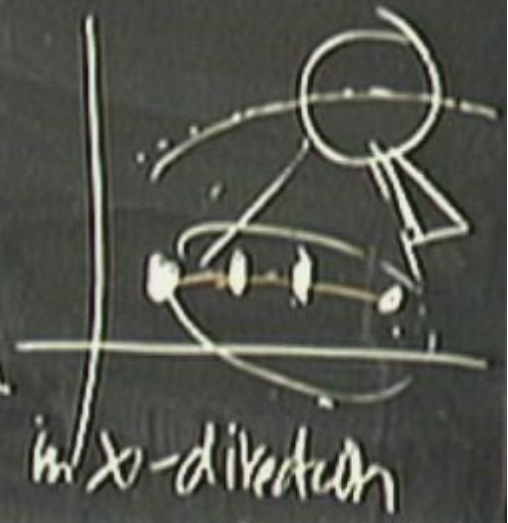
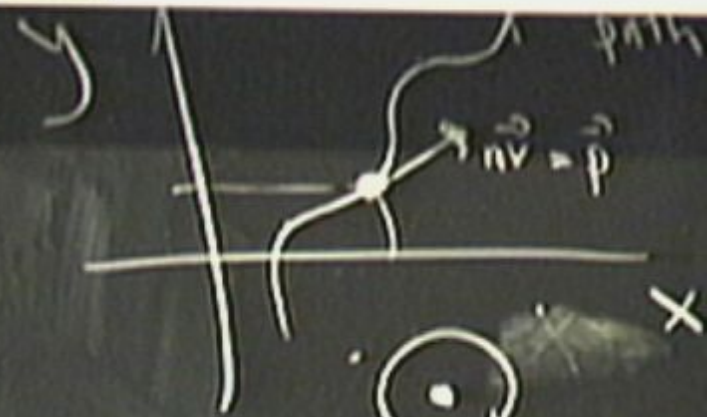
$$\frac{\lambda}{\sin \epsilon}$$

λ
 $\sin \epsilon$
 $\sim \text{mms}$
 $|\theta| < \epsilon$



$p_x = p_0 \sin \theta$

Number of photons after collision with electron



Principle

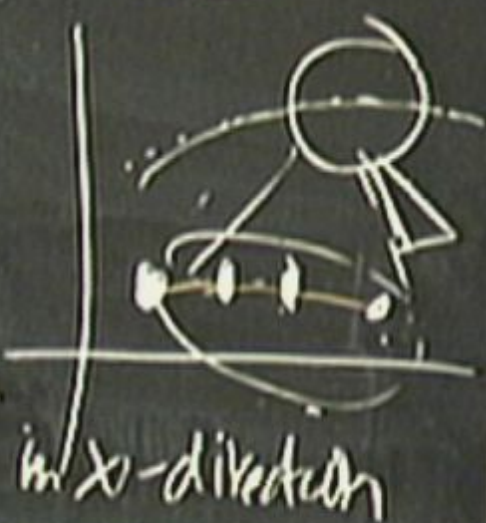
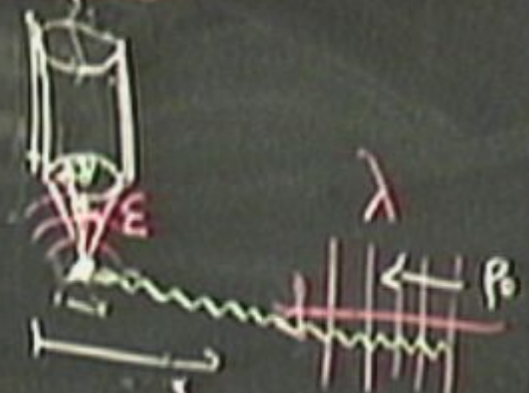
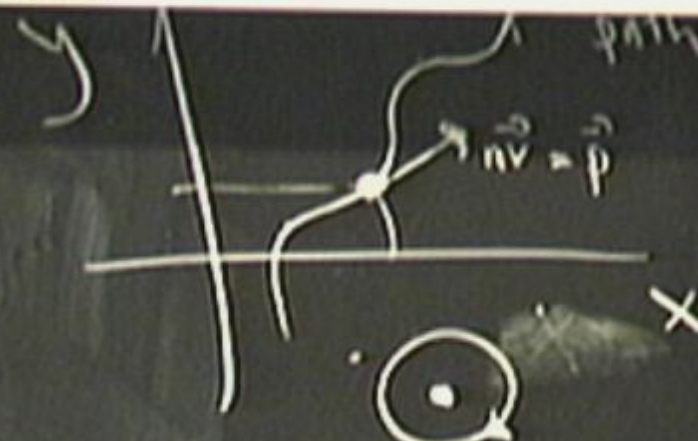
If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

Use γ -rays $|\theta| < \epsilon$

$P_x = P_0 \sin \theta$ amount of photon after collision with electron



Principle

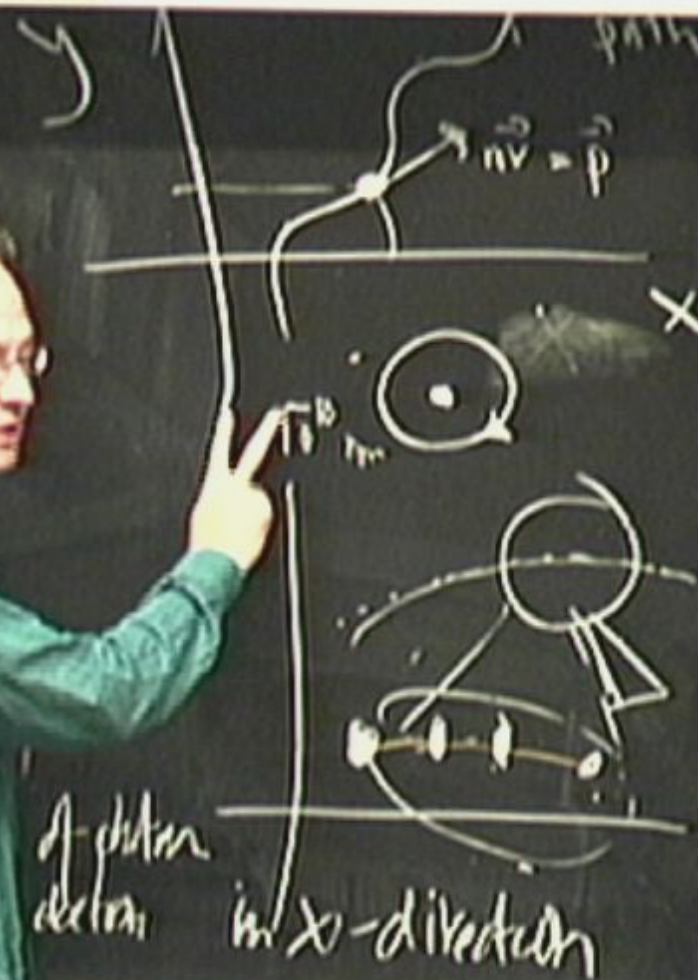
If you want to give meaning to phrase
 "position of electron" you must
 specify a way to observe it.

resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

use γ -rays $|\theta| < \epsilon$

$$p_x = p_0 \sin \theta$$



Principle

If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

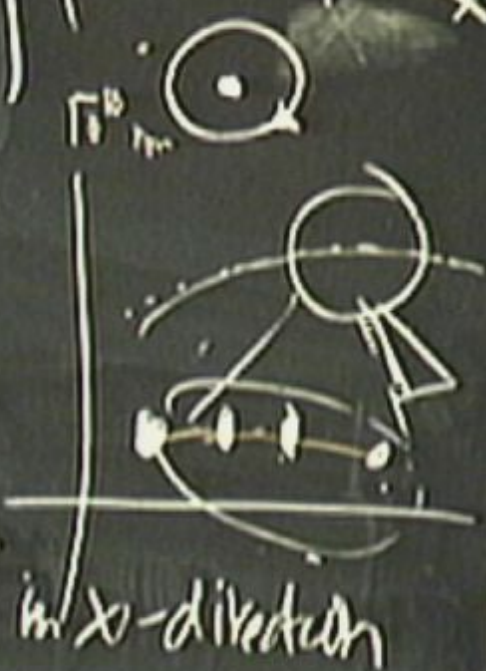
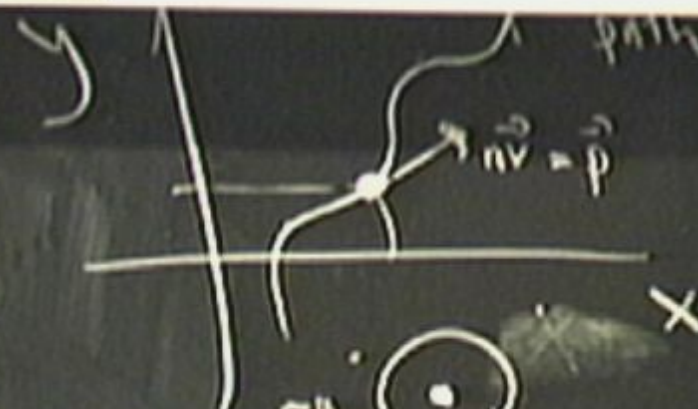
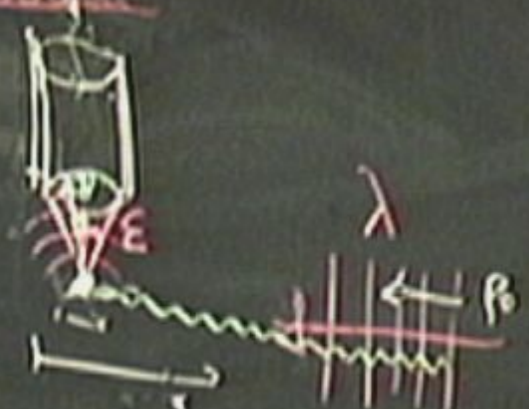
uncertainty principle

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

Use γ -rays
 $|\theta| < \epsilon$

$$p_x = p_0 \sin \theta$$

Number of photons after collision with electron



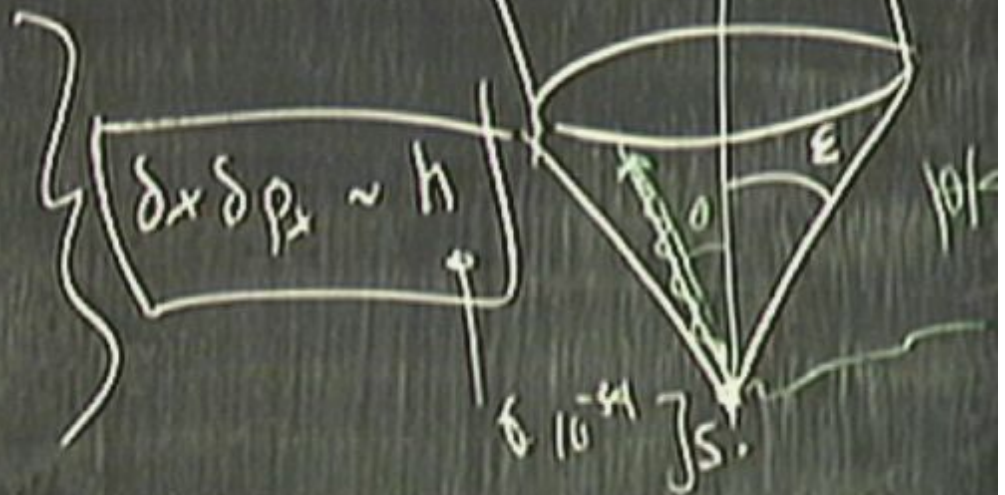
Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$

" " " of detector after collision is

$$\delta p_x^{det} \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$



\vec{p}

Principle

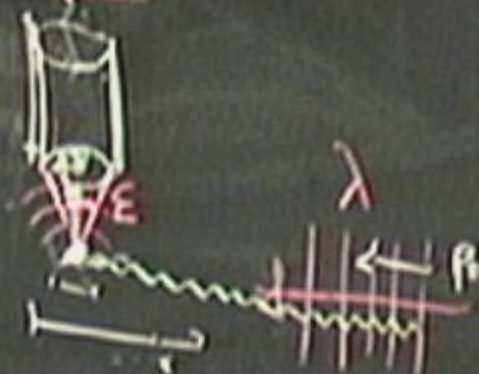
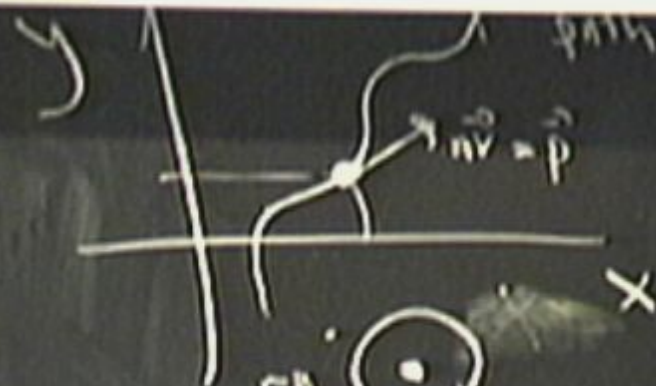
If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

resolving power

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

use γ -rays
 $|\theta| < \epsilon$

$p_x = p_0 \sin \theta$ - number of photons after collision with electron



Uncertainty in momentum of photon $\delta p_x(p_h) \sim p_0 \sin \epsilon$

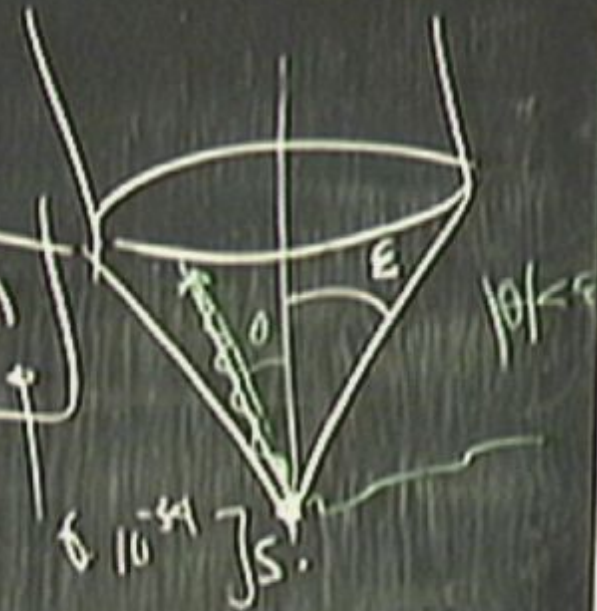
" " " " of detector after collision is

$$\delta p_x^{det} \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$

$$\delta x \delta p_x \sim h$$



$\vec{N} \vec{p}$

Heisenberg's principle is a philosophical one and hence debatable.

Uncertainty in momentum of photon $\delta p_x(\text{ph}) \sim p_0 \sin \epsilon$

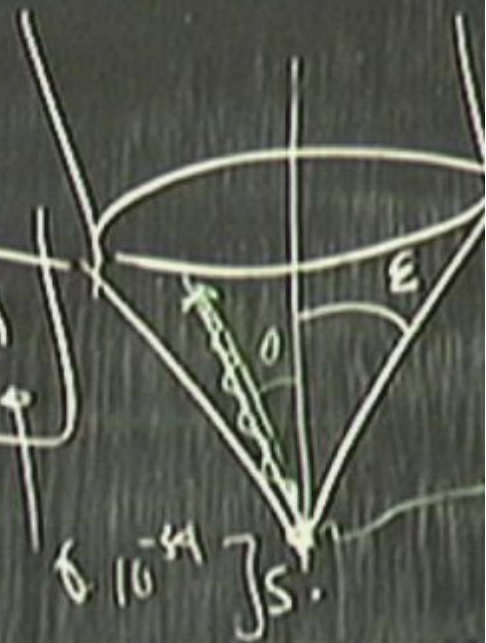
" " " of electron after collision is

$$\delta p_x^{\text{el}} \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

$$\Delta x \Delta p_x \sim h$$



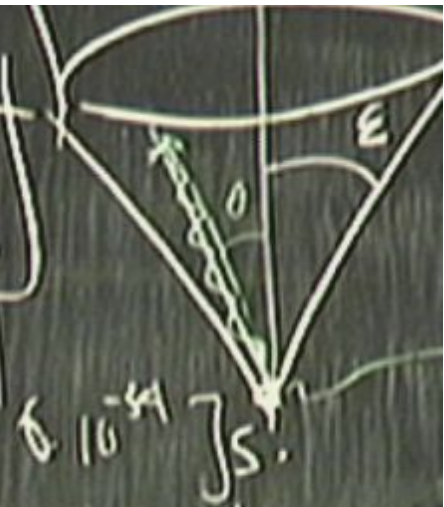
Δp Heisenberg's principle is a philosophical one and hence debatable.

$$\delta p_x \approx p_0 \sin \epsilon$$

$$= \frac{h}{\lambda} \sin \epsilon$$

$$\delta x = \frac{\lambda}{\sin \epsilon}$$

$$\delta x \delta p_x \sim h$$



$\nabla \cdot \vec{p}$ · Heisenberg's principle is a philosophical one
 and hence debatable.
 Heisenberg equates "uncertainty", "lack of meaning"

$$\Delta x = \frac{h}{5mc}$$

NB. Heisenberg's principle is a philosophical one and hence debatable.
Heisenberg equates "uncertainty", "lack of meaning", "experimental uncertainty"

$6 \cdot 10^{-34}$ Js.

$$\Delta x = \frac{h}{5m\omega}$$

N.B. Heisenberg's principle is a philosophical one
and hence debatable.

$6 \cdot 10^{-34}$ Js.

→ Heisenberg equates "uncertainty", "lack of meaning",
"experimental inaccuracy", "indefiniteness",
"indeterminacy".

$$\Delta x = \frac{h}{5m\epsilon}$$

$6 \cdot 10^{-34} \text{ Js}$

N.B. Heisenberg's principle is a philosophical one and hence debatable.

→ Heisenberg equates "uncertainty", "lack of meaning", "experimental inaccuracy", "indefiniteness", "indeterminacy" etc.

Principle

If you want to give meaning to phrase "position of electron" you must specify a way to observe it.

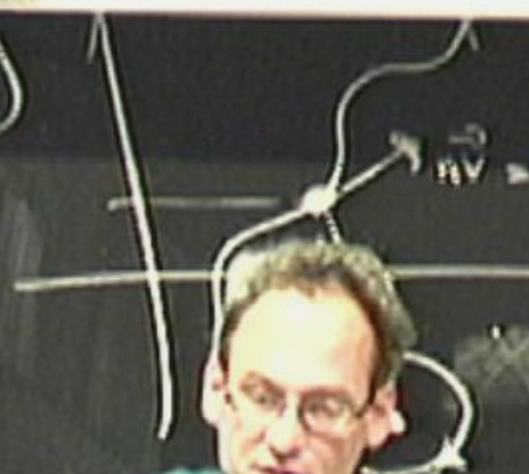
resolving power

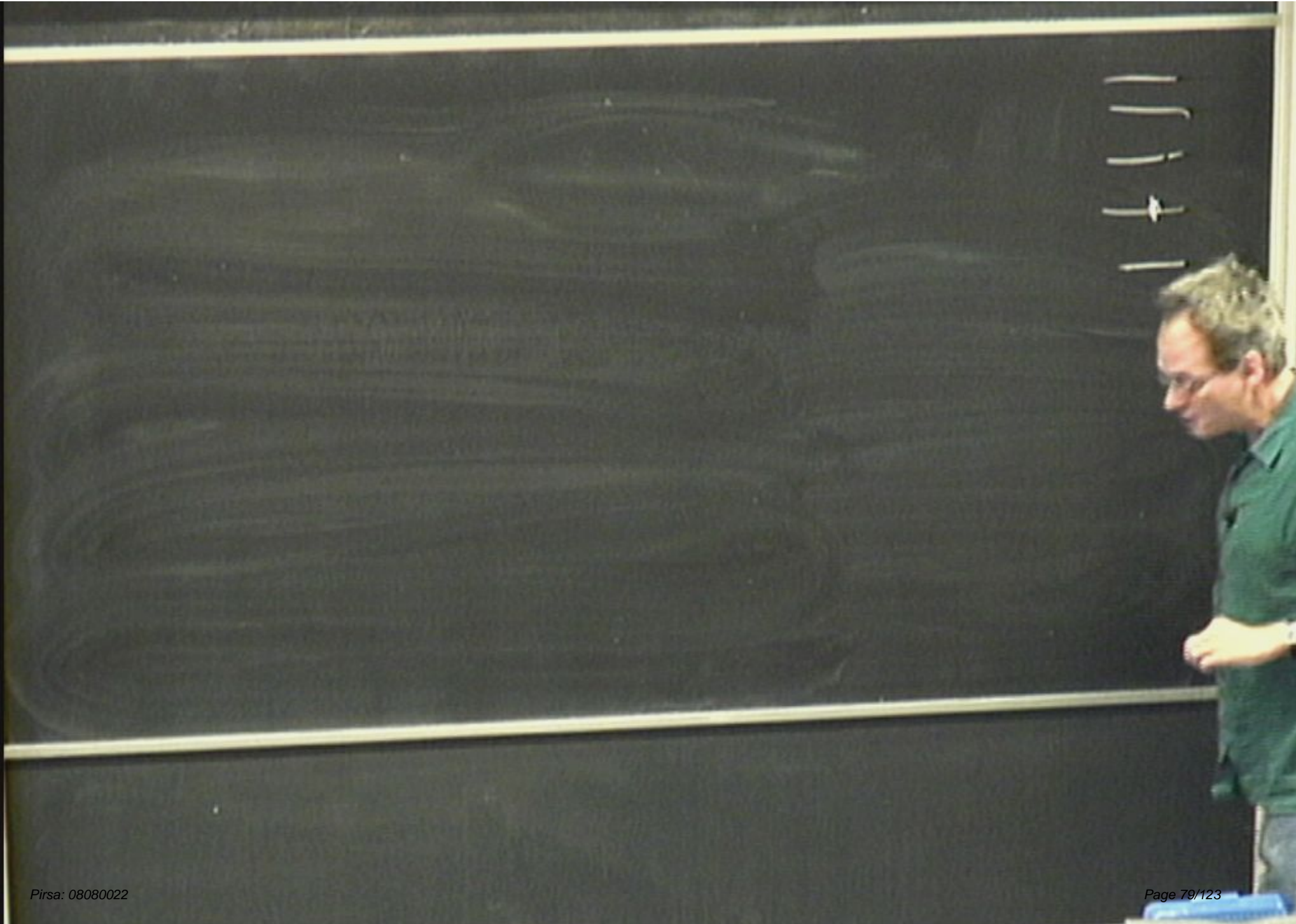
$$\Delta x = \frac{\lambda}{\sin \epsilon}$$

use γ -rays $|\theta| < \epsilon$

$$P_x = P_0 \sin^2 \theta$$

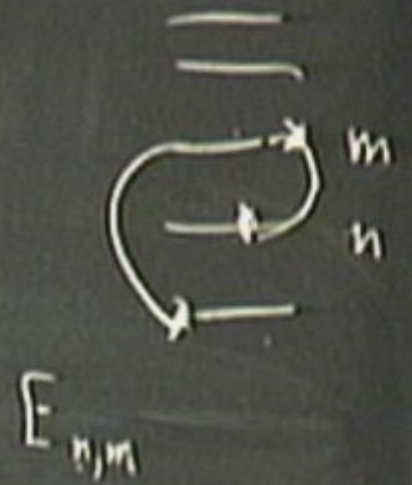
fraction of photons after collision with electron

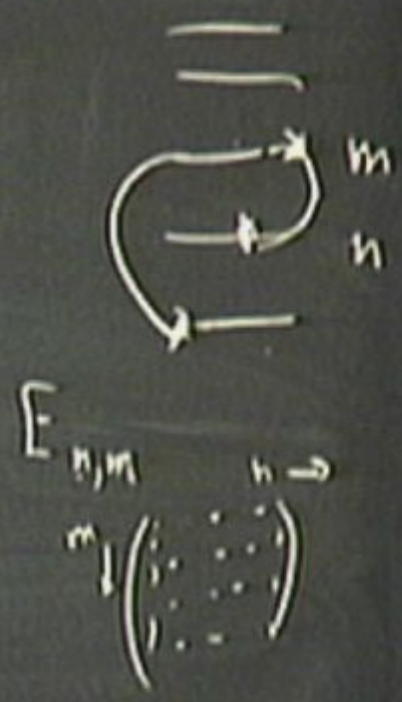


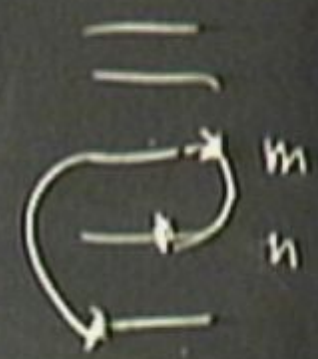


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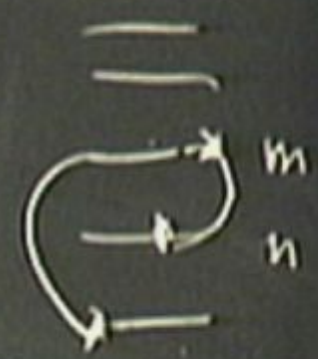




$$E_{n,m} \quad n \rightarrow$$

$$m \downarrow \begin{pmatrix} \cdot & \cdot & \cdot \\ \cdot & \cdot & \cdot \\ \cdot & \cdot & \cdot \\ \cdot & \cdot & \cdot \end{pmatrix}$$

matrices



$$E_{n,m} \quad n \rightarrow$$

$$m \downarrow \begin{pmatrix} \cdot & \cdot & \cdot \\ \cdot & \cdot & \cdot \\ \cdot & \cdot & \cdot \end{pmatrix}$$

matrices

Heisenberg 1925. matrix mechanics
 Schrödinger 1926 wave mechanics.

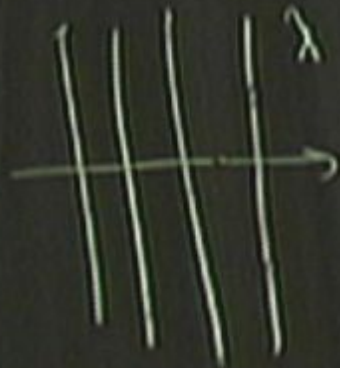
Single slit diffraction.



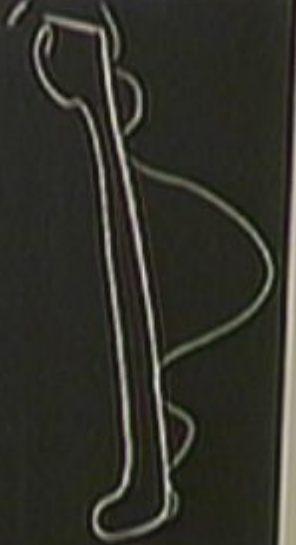
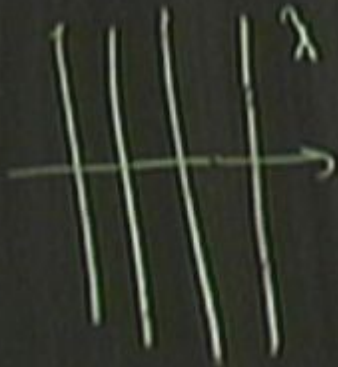
Single slit diffraction.



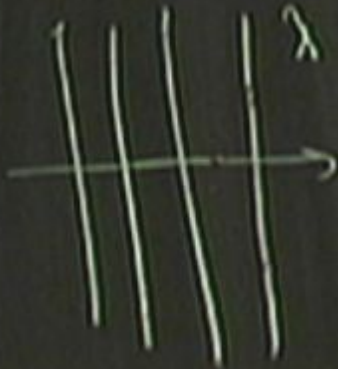
Single slit diffraction.



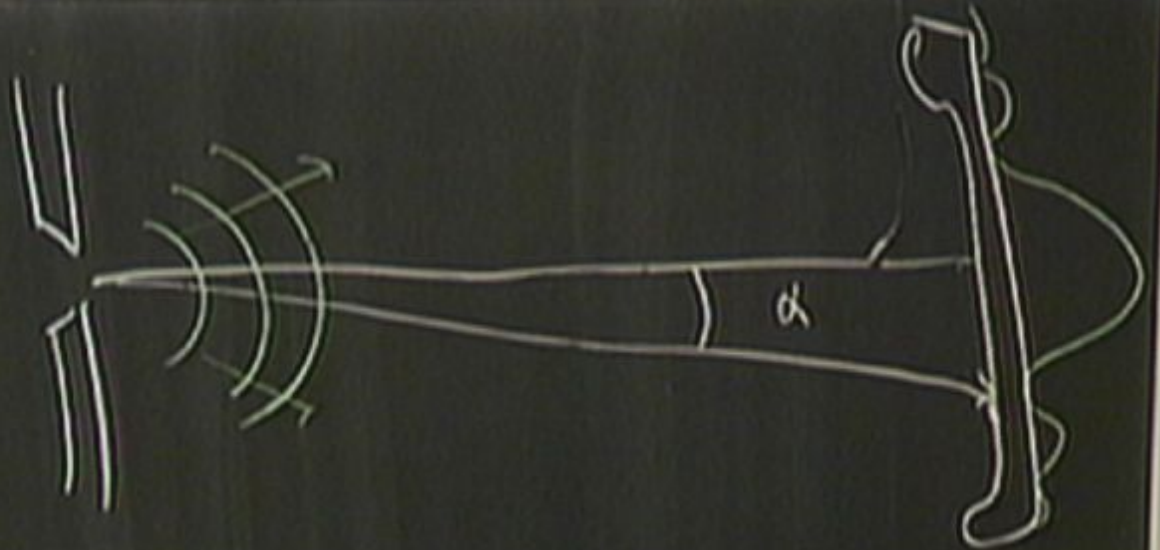
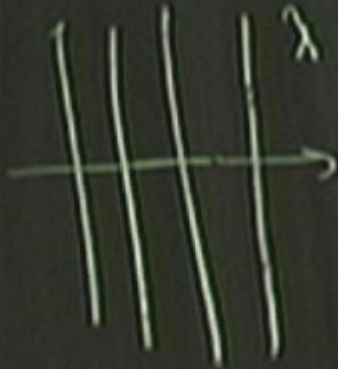
Single slit diffraction.



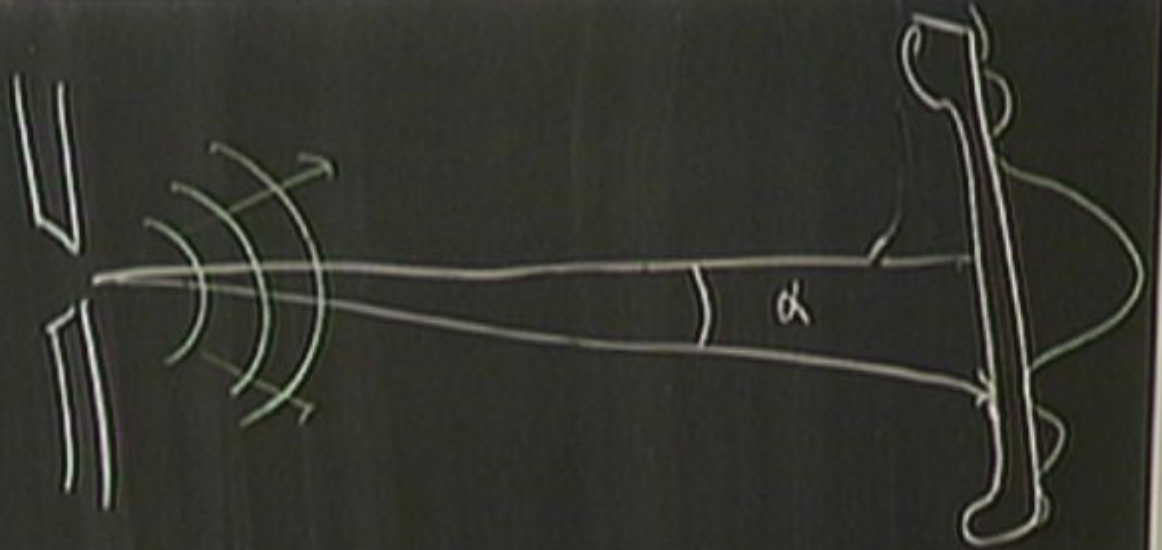
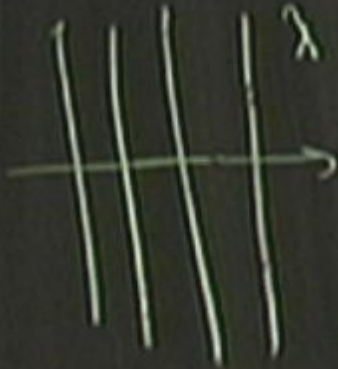
Single slit diffraction.



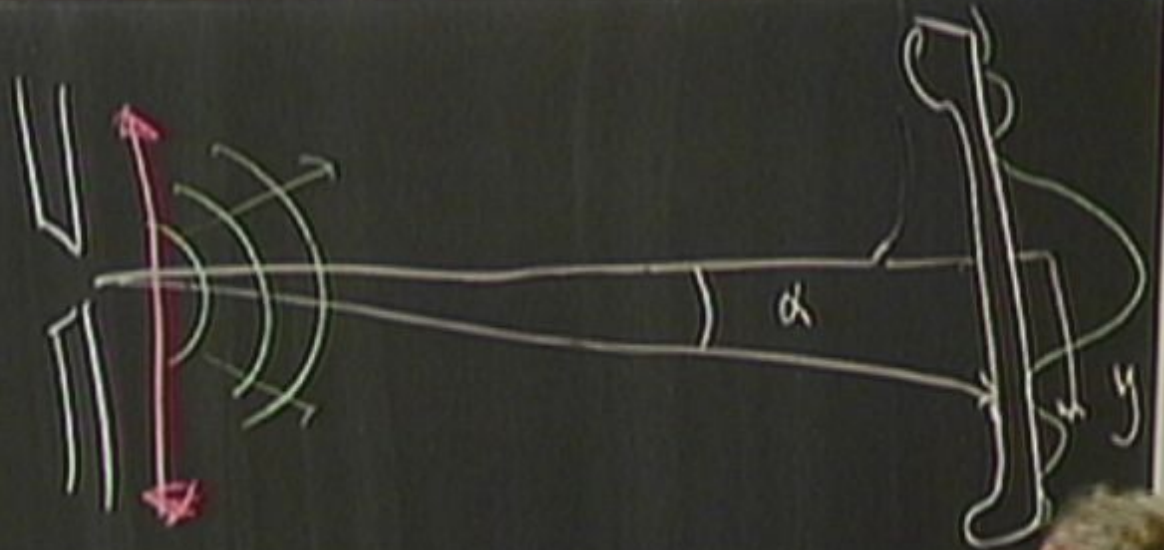
Single slit diffraction.



Single slit diffraction.

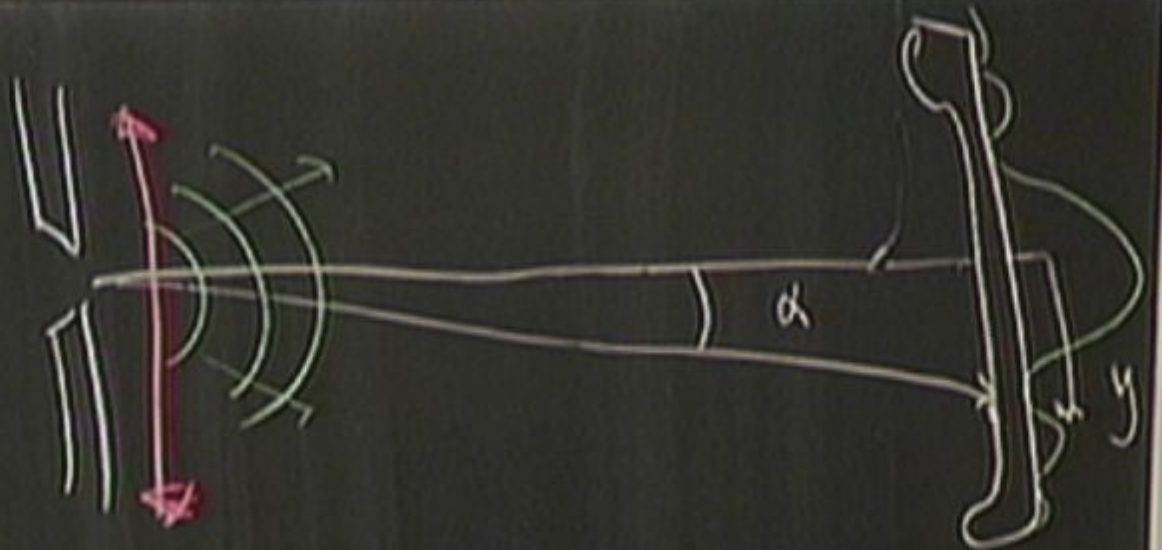
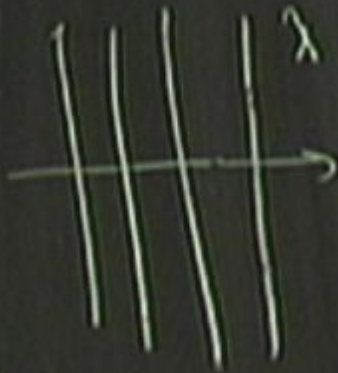


Single slit diffraction.



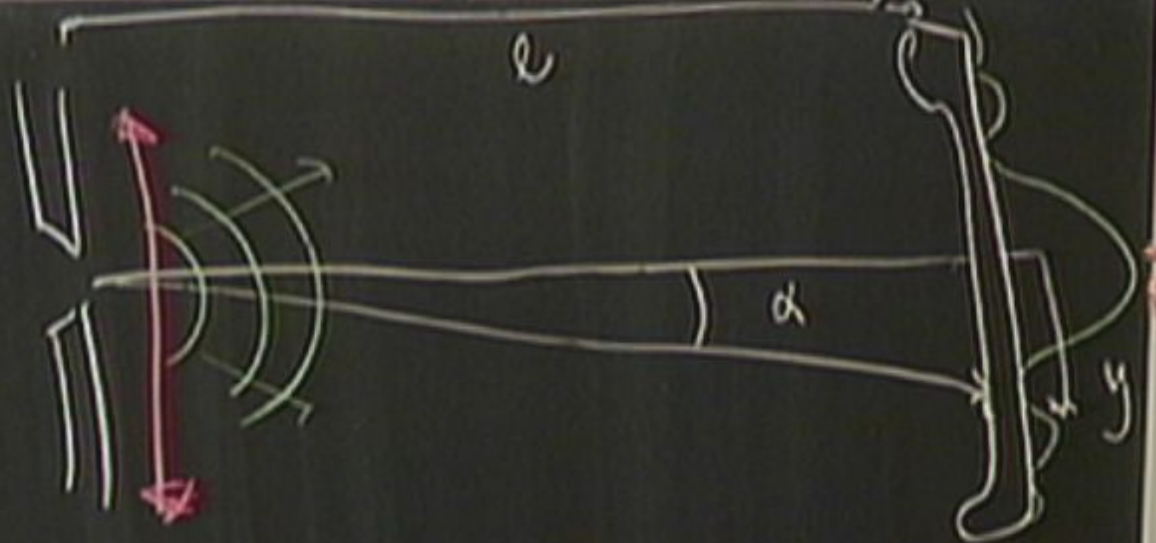
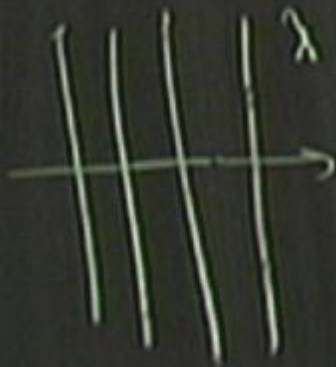
θ

Single slit diffraction.



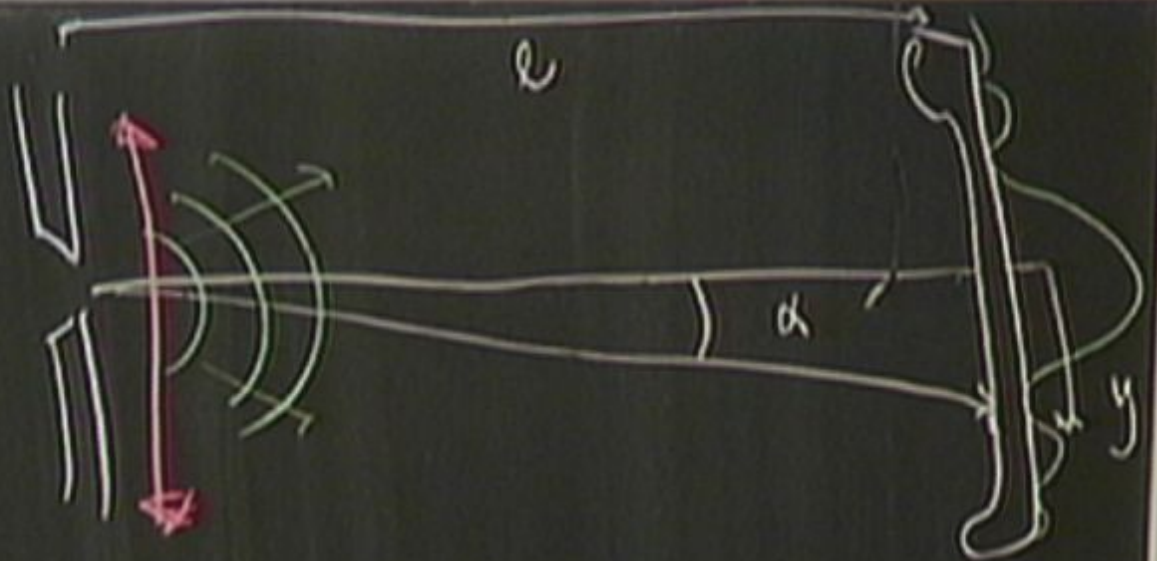
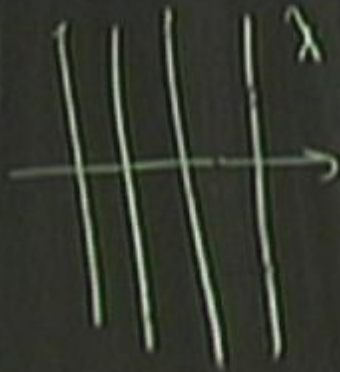
$$P = P_0 \text{sinc} \alpha$$
$$\approx P_0 \frac{y}{L}$$

Single slit diffraction.

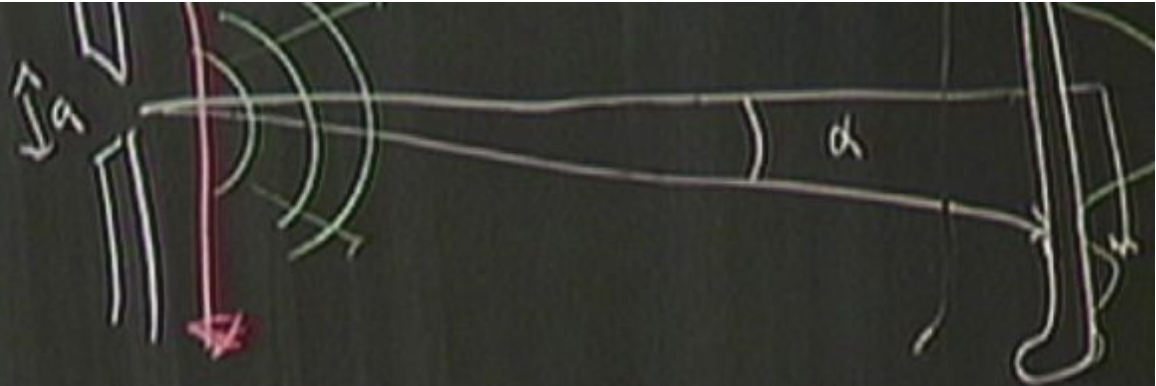
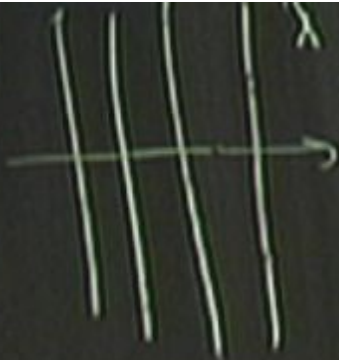


$$P = P_0 \sin \alpha$$
$$P \approx P_0 \frac{y}{l}$$

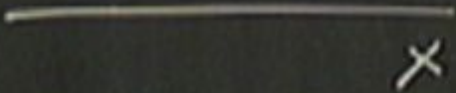
Single slit diffraction.

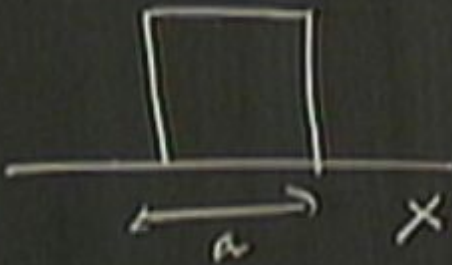
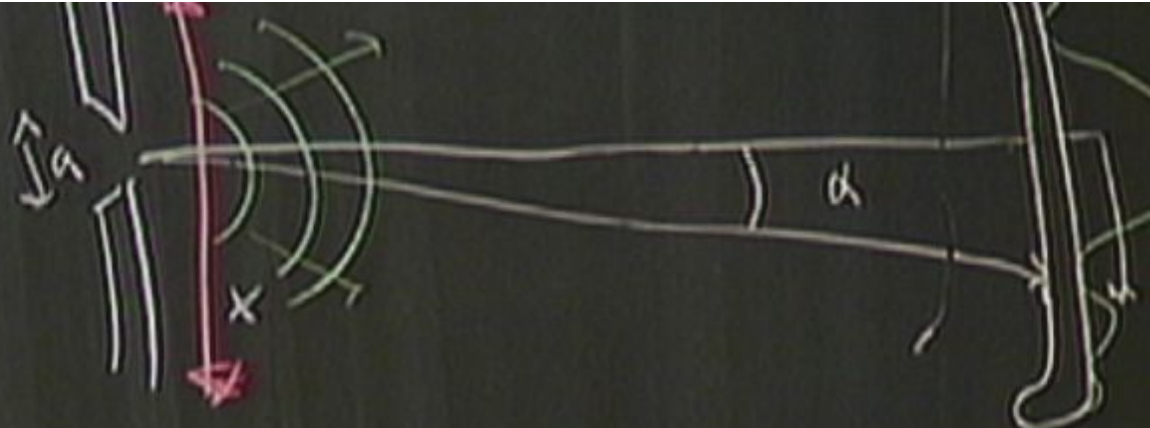
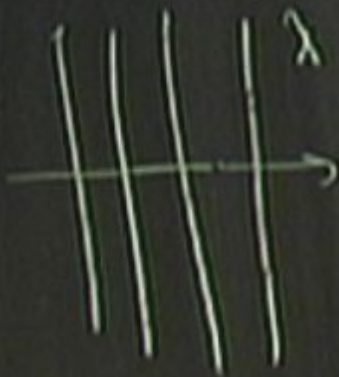


$$P = P_0 \text{sinc } \alpha$$
$$= P_0 \frac{y}{l}$$



$$P = P_0 \sin \alpha$$
$$P_0 \frac{y}{L}$$

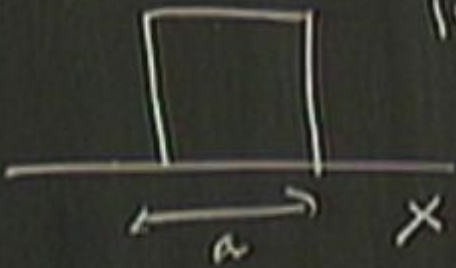
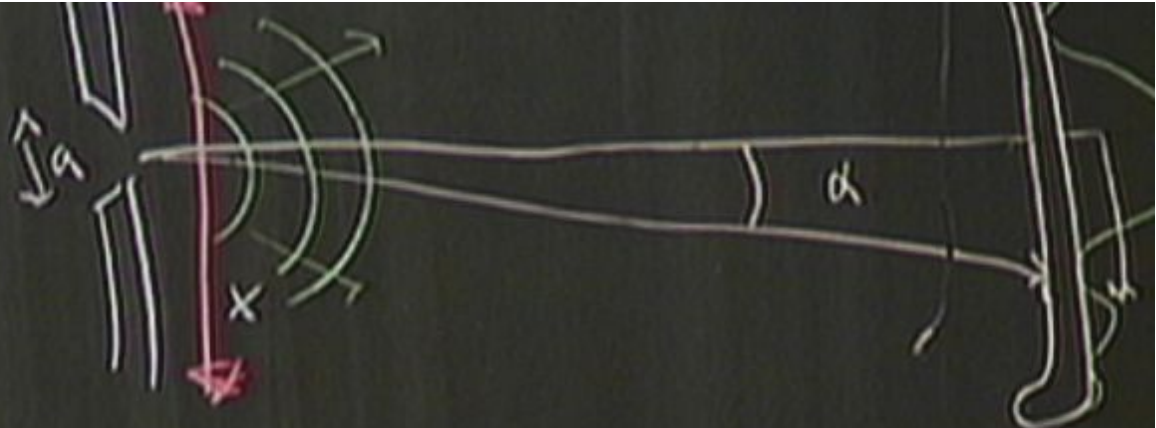
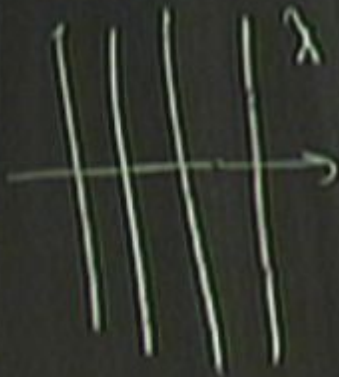




$$P = P_0 \sin \alpha$$

$$P \approx P_0 \frac{y}{r}$$

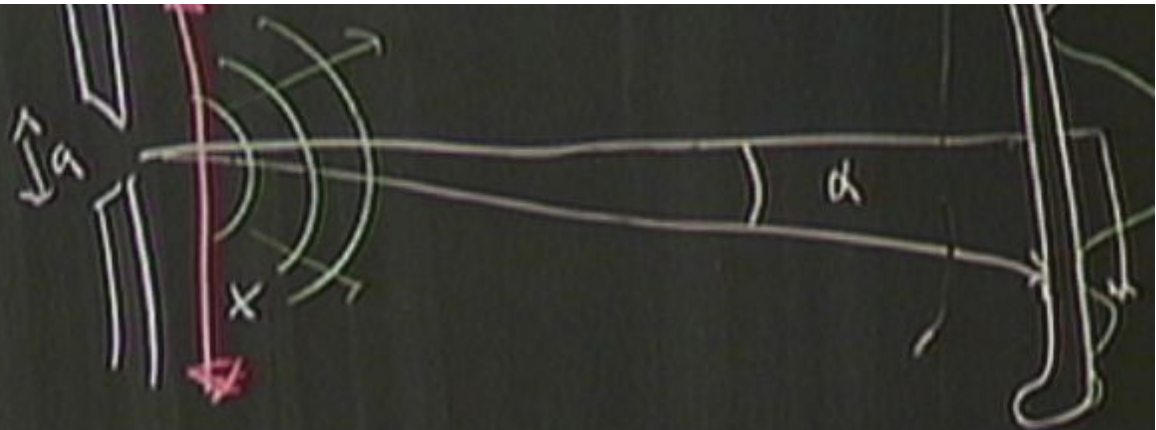
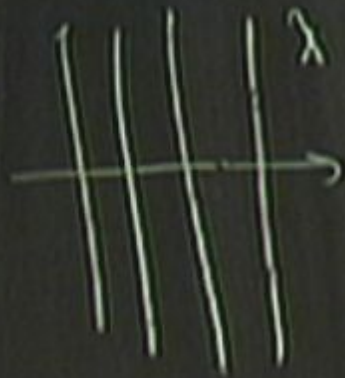




$$14(\omega)^2$$

$$P = P_0 \sin \alpha$$

$$P \approx P_0 \frac{a}{\lambda}$$



$\psi(x)$

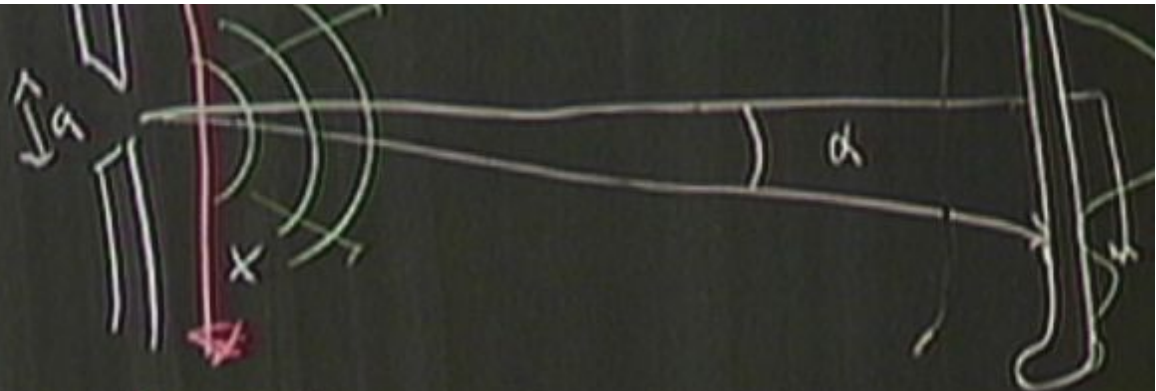
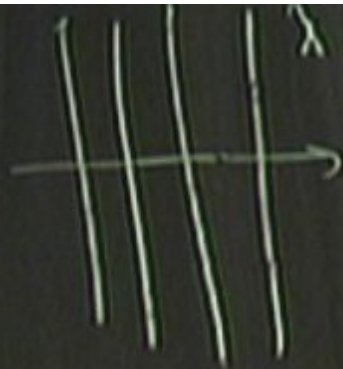


$$|\psi(x)|^2$$

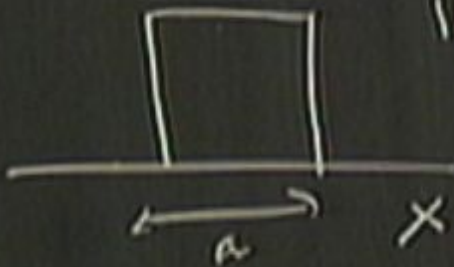
$$P = P_0 \sin \alpha$$

$$P_0 \frac{y}{r}$$



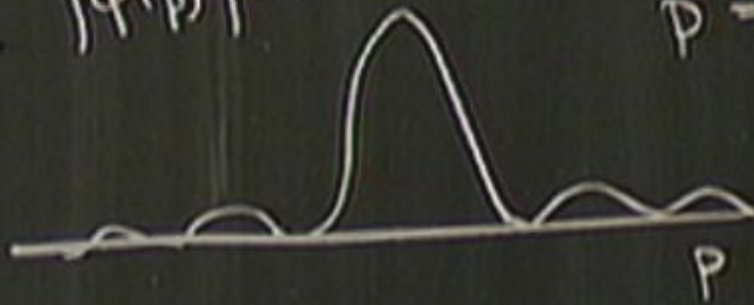


$\psi(x)$

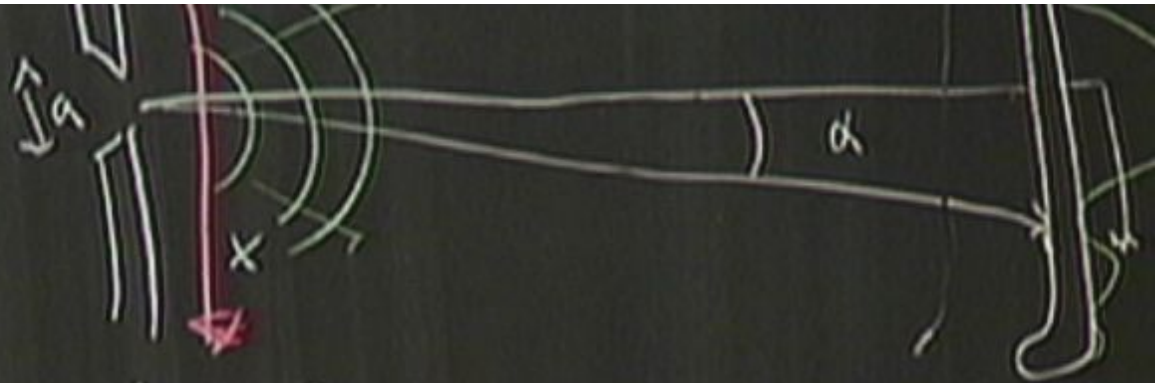
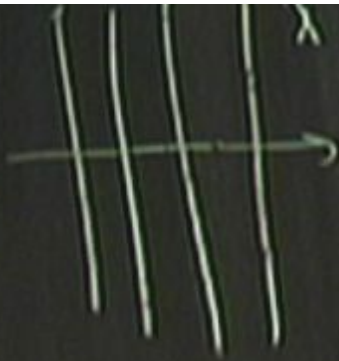


$|\psi(x)|^2$

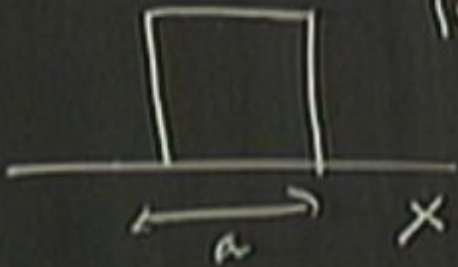
$|\varphi(p)|^2$



$p = p_0 \sin \alpha$
 $E p_0 \frac{y}{L}$



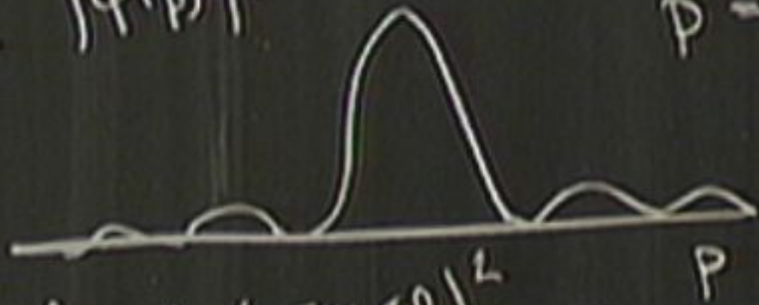
$\psi(x)$



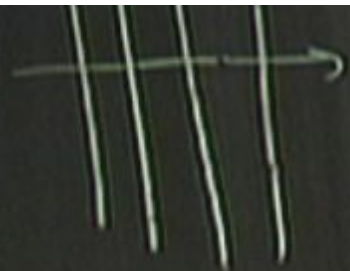
$|\psi(x)|^2$

$|\varphi(p)|^2$

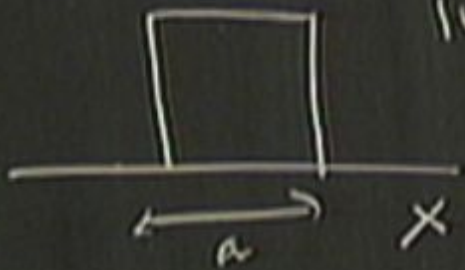
$|\varphi(p)|^2 = \frac{a}{\pi} \left(\frac{\sin \frac{ap}{p}}{ap} \right)^2$



$P = P_0 \sin \alpha$
 $E P_0 \frac{y}{L}$

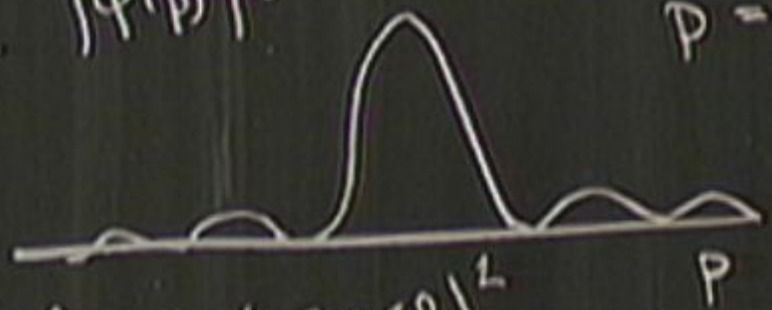


$\psi(x)$



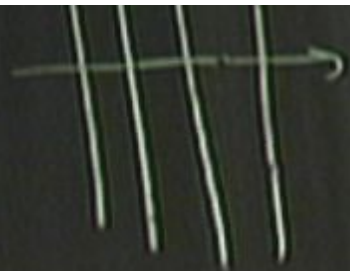
$|\psi(x)|^2$

$|\varphi(p)|^2$

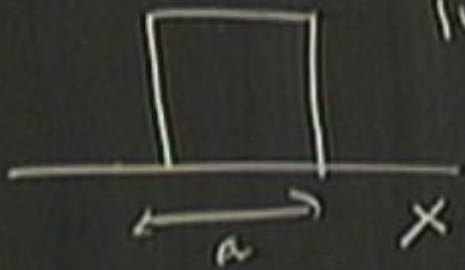


$p = p_0 \sin \alpha$
 $\approx p_0 \frac{y}{L}$

$|\varphi(p)|^2 \approx \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$

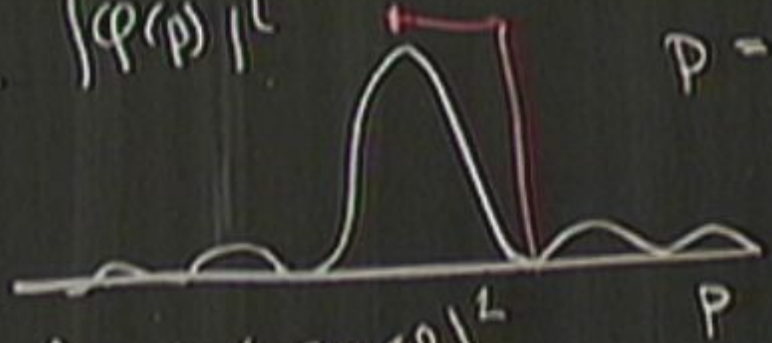


$\psi(x)$



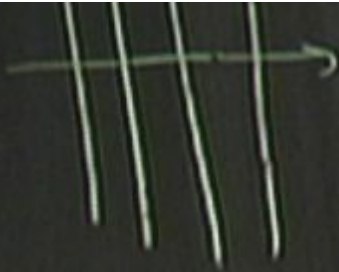
$|\psi(x)|^2$

$|\varphi(p)|^2$

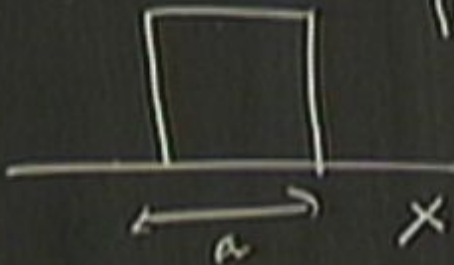


$p = p_0 \sin \alpha$
 $\approx p_0 \frac{y}{L}$

$|\varphi(p)|^2 \approx \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$



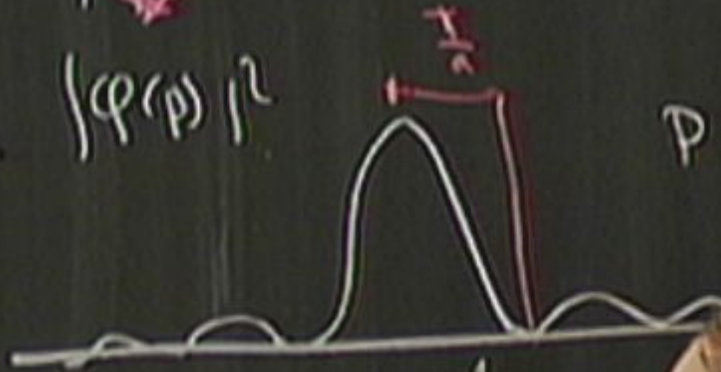
$\psi(x)$



$k=1$

$|\psi(x)|^2$

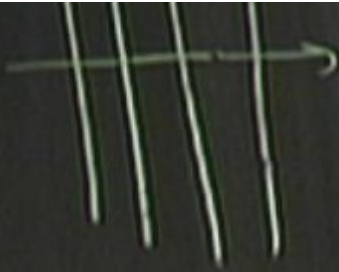
$|\varphi(p)|^2$



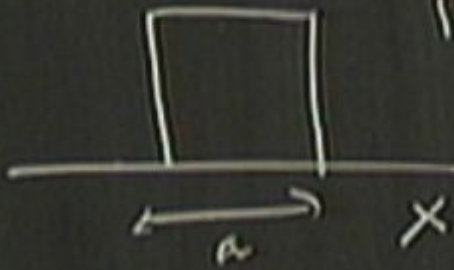
$p = p_0 \sin \alpha$
 $\approx p_0 \frac{y}{L}$

$|\varphi(p)|^2 \approx \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$





$\psi(x)$

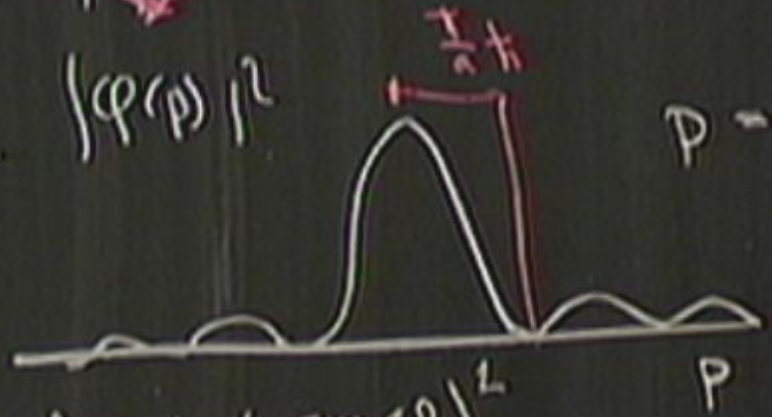


$k=1$

$|\psi(x)|^2$

$|\varphi(p)|^2$

$|\varphi(p)|^2 \sim \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$



$p = p_0 \sin \alpha$
 $\approx p_0 \frac{\lambda}{L}$

Define

$$\Delta x = \sqrt{\langle \hat{x}^2 \rangle - \langle \hat{x} \rangle^2}$$

Define

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

$$\text{where } \langle x^2 \rangle = \int x^2 |\Psi(x)|^2 dx$$

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etc.

Then $\Delta x \Delta p \geq \frac{\hbar}{2}$ (Heisenberg uncertainty principle)

Define

$$\Delta x = \sqrt{\langle \hat{x}^2 \rangle - \langle x \rangle^2}$$

$$\Delta p = \sqrt{\langle \hat{p}^2 \rangle - \langle p \rangle^2}$$

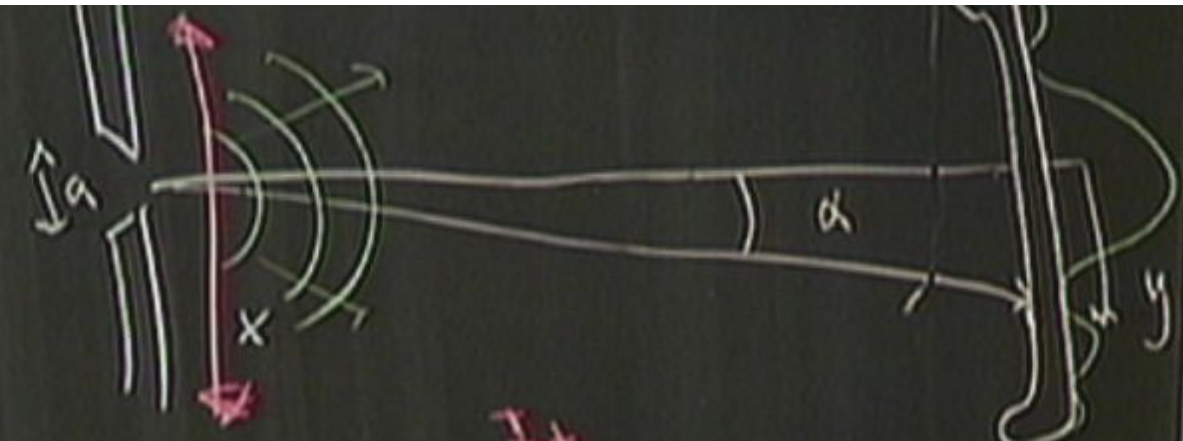
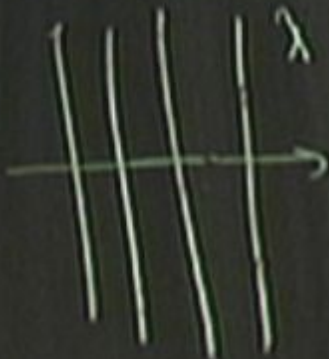
$$\text{Then } \Delta x \Delta p \geq \frac{\hbar}{2}$$

(Kannard
1927)

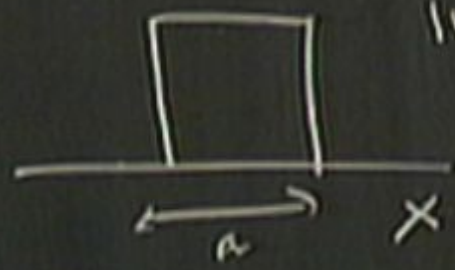
$$\text{where } \langle \hat{x}^2 \rangle = \int x^2 |\Psi(x)|^2 dx$$

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etc.



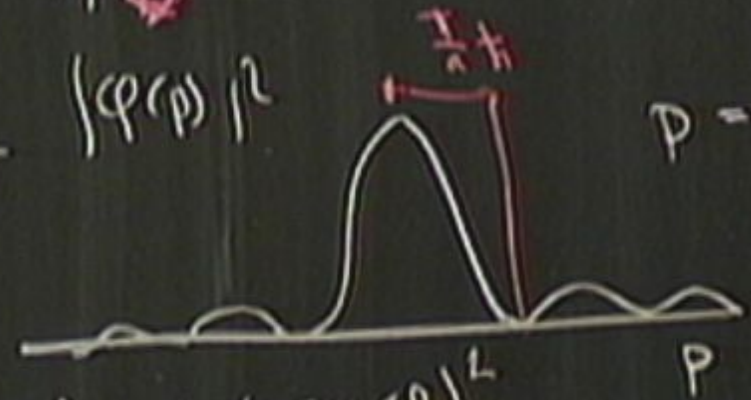
$\psi(x)$



$k=1$

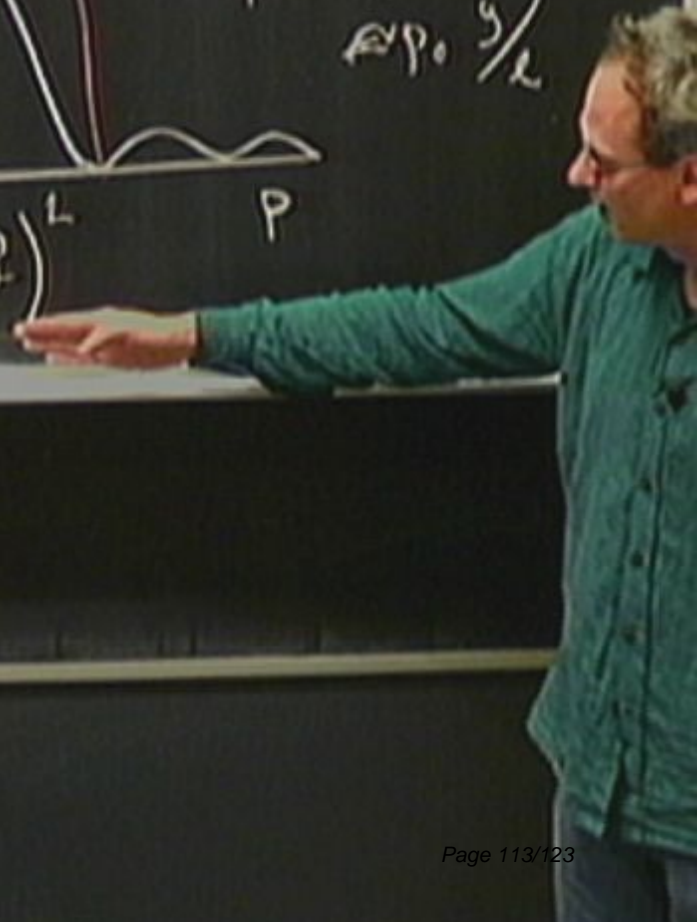
$|\psi(x)|^2$

$|\varphi(p)|^2$



$p = p_0 \sin \alpha$
 $\approx p_0 \frac{y}{L}$

$|\varphi(p)|^2 \approx \frac{a}{\pi} \left(\frac{\sin \frac{\pi a p}{a p}}{a p} \right)^2$



Define

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

$$\text{Then } \Delta x \Delta p \geq \frac{\hbar}{2}$$

where $\langle x^2 \rangle = \int x^2 |\psi(x)|^2 dx$

(Kannard
1927)

$$\langle x \rangle = \int x |\psi(x)|^2 dx$$

$$\langle p^2 \rangle = \int \psi^* \hat{p}^2 \psi dx$$

$\psi(x)$

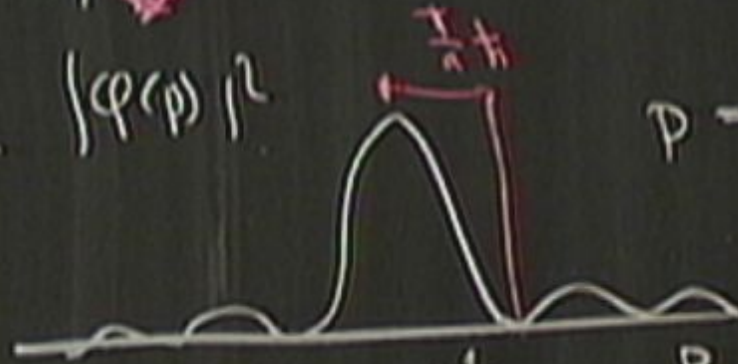


$k=1$

$\Delta x \sim a$

$|\psi(x)|^2$

$|\varphi(p)|^2$

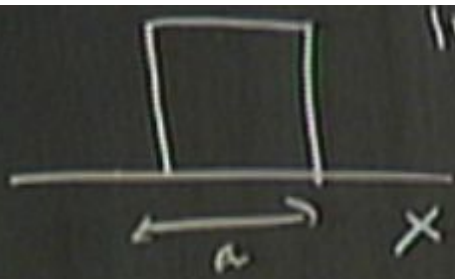


$p = p_0 \pm \Delta p$
 $\Delta p_0 \sim \frac{h}{a}$

$|\varphi(p)|^2 \sim \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$

$\Delta p = \infty$

$\langle p^2 \rangle = \int_{-\infty}^{\infty} p^2 |\varphi(p)|^2 dp$



$k=1$

$$\Delta x \sim a$$



$\Delta p \sim \frac{h}{a}$

$$|\psi(p)|^2 \sim \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$$

$$\Delta p = \infty$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

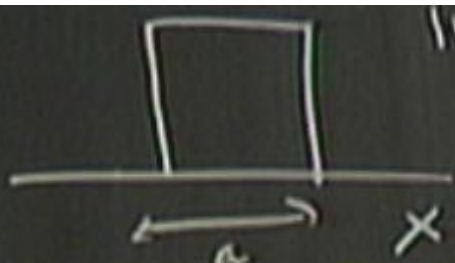
$$\Delta x \Delta p \geq \frac{h}{2}$$

(Kannard 1927)

$$\langle x \rangle = \int x |\psi(x)|^2 dx$$

$$\langle p^2 \rangle = \int_{-\infty}^{\infty} p^2 |\psi(p)|^2 dp$$

$k=1$



$$\Delta x \sim a$$

$$|\varphi(p)|^2 \sim \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$$

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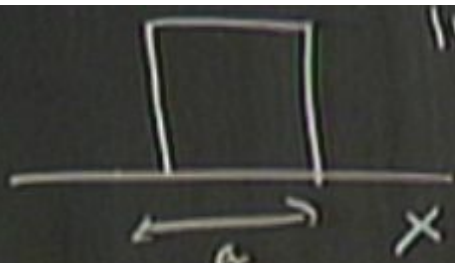
Then $\Delta x \Delta p \geq \frac{\hbar}{2}$

(Kannard 1927)

$$\langle x \rangle = \int x |\psi(x)|^2 dx$$

$$\langle p^2 \rangle = \int_{-\infty}^{\infty} p^2 |\varphi(p)|^2 dp$$

$k=1$



$$\Delta x \sim a.$$



$$|\varphi(p)|^2 \sim \frac{a}{\pi} \left(\frac{\sin ap}{ap} \right)^2$$

$$\langle p \rangle = 0$$

$$\Delta p = \infty$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

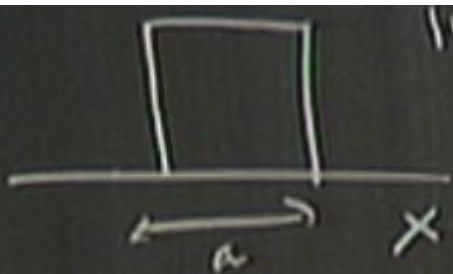
Then $\Delta x \Delta p \geq \frac{\hbar}{2}$

(Kannard 1927)

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$k=1$



$$\Delta x \sim a$$



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(Kannard 1927)

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Defines

$$\Delta x = \sqrt{\langle \hat{x}^2 \rangle - \langle x \rangle^2}$$

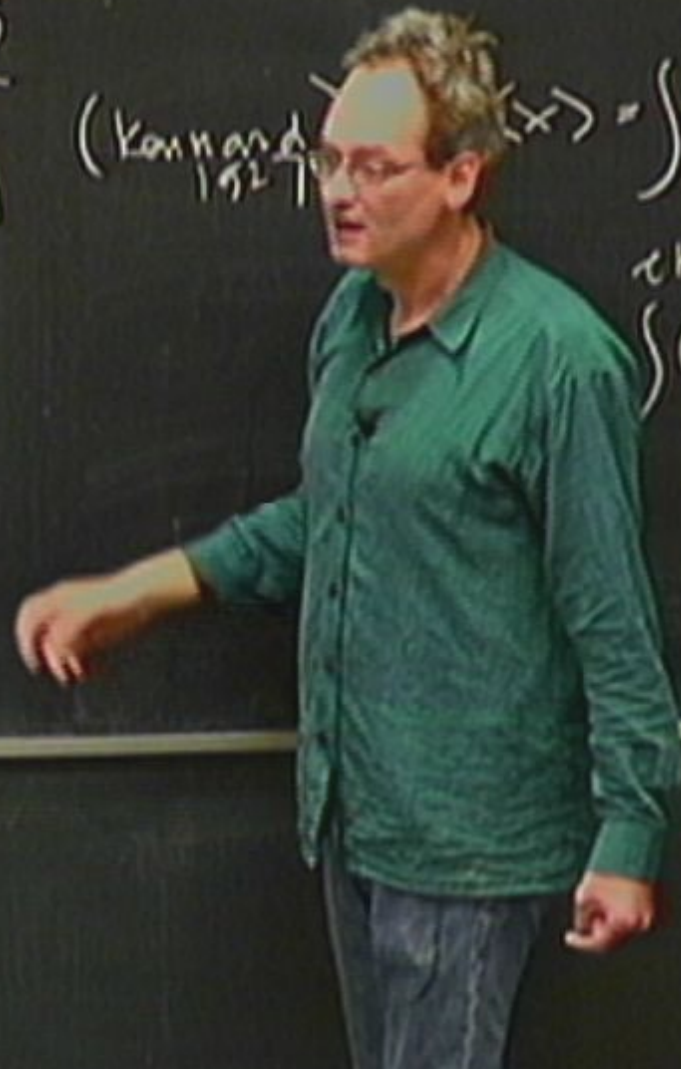
$$\Delta p = \sqrt{\langle \hat{p}^2 \rangle - \langle p \rangle^2}$$

$$\boxed{\text{Then } \Delta x \Delta p \geq \frac{\hbar}{2}}$$

where $\langle x^2 \rangle = \int x^2 |\psi(x)|^2 dx$

(Kohn and Sham) $\langle x \rangle = \int x |\psi(x)|^2 dx$

etc. $\int |\psi(p)|^2 dp$



Define

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2}$$

$$\text{Then } \Delta x \Delta p \geq \frac{\hbar}{2}$$

(Korn and
1927)

where $\langle x^2 \rangle = \int x^2 |\psi(x)|^2 dx$

$$\langle x \rangle = \int x |\psi(x)|^2 dx$$

$$\langle p^2 \rangle = \int \psi^* \hat{p}^2 \psi dx$$

Defines

$$\Delta x = \sqrt{\langle \hat{x}^2 \rangle - \langle \hat{x} \rangle^2}$$

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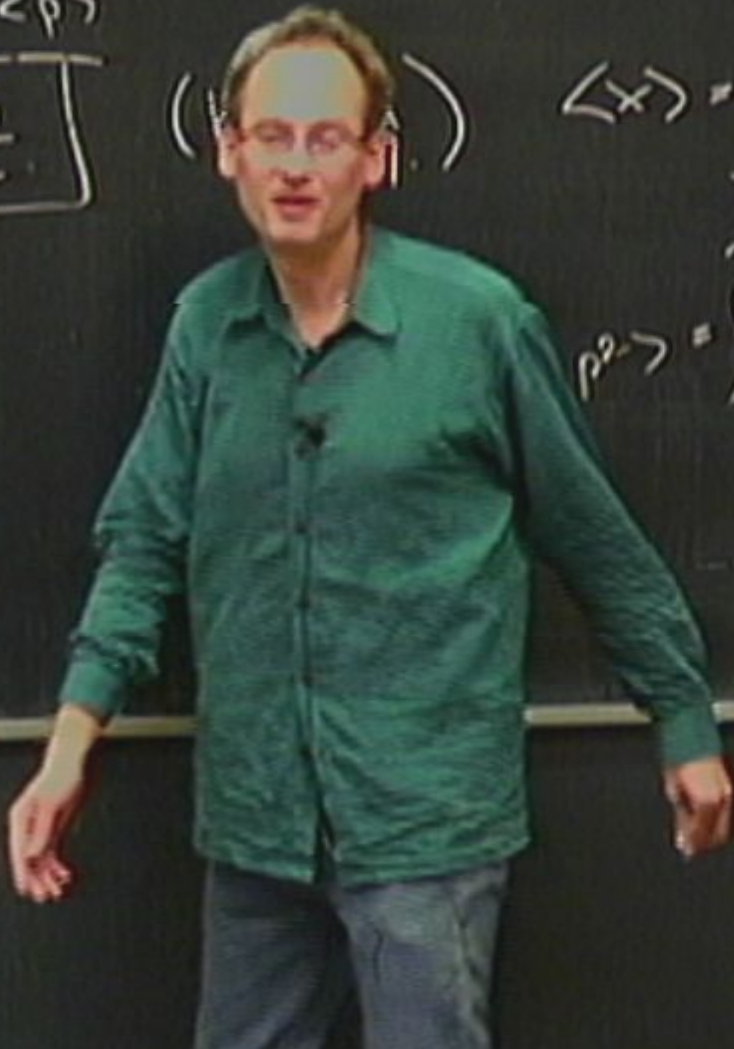
$$\text{Then } \Delta x \Delta p \geq \frac{\hbar}{2}$$

$$\hbar = \frac{h}{2\pi}$$

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Defines

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2}$$

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$$\text{Then } \Delta x \Delta p \geq \frac{\hbar}{2}$$

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$$\langle x \rangle = \int x |\psi(x)|^2 dx$$

$$\langle p^2 \rangle = \int \psi^* (-\hbar^2 \nabla^2) \psi dx$$

(Kannard 1927)